

# Low Cost, Long Cycle Life, Li-ion Batteries for Stationary Applications

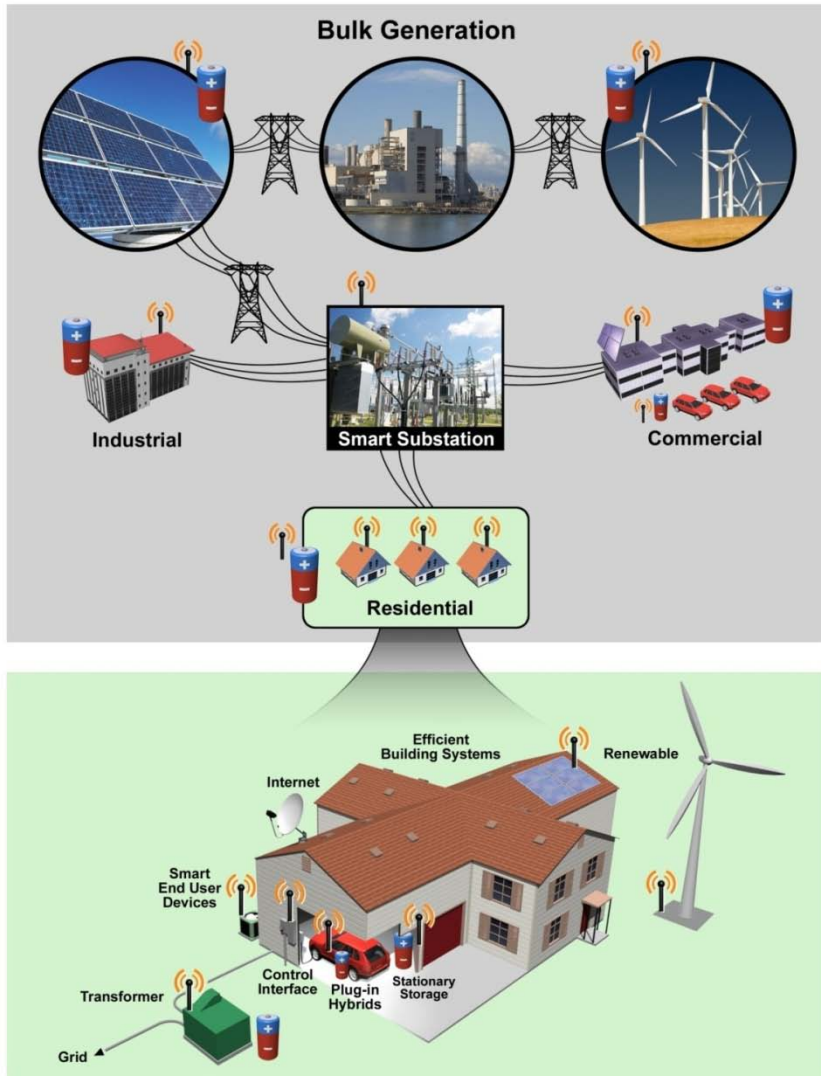
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# Introduction

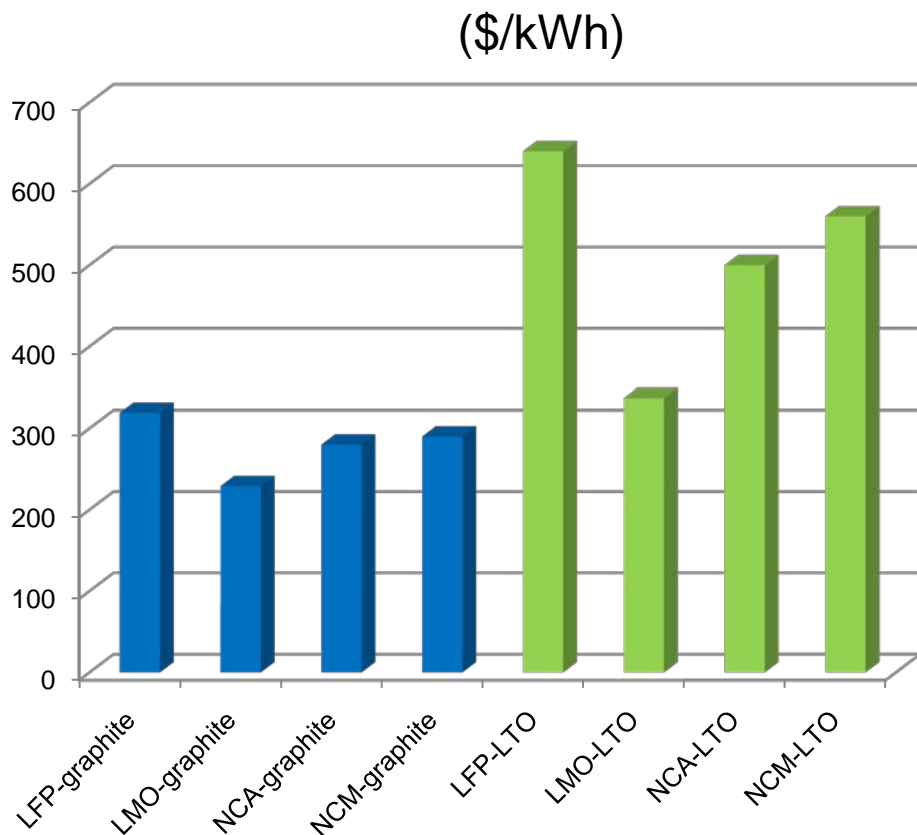


- ❑ Investigate the Li-ion battery for stationary energy storage to compliment the renewable energy resources such as solar/ wind power and load leveling for grid integration.
- ❑ Energy storage unit for possible community application ~kWh level.
- ❑ Li-ion battery energy storage with effective thermal management to improve its cycle and calendar life.
- ❑ Design Li-ion battery pack suitable for large scale energy storage system.

# Objectives

- ❑ Screen possible electrode materials and its combinations suitable for large scale Li-ion battery.
- ❑ Characterization of inherent heat generation of possible electrode candidates and its combinations.
- ❑ Cost analyses of the full cell battery per energy density.
- ❑ Improve the cycle life of Li-ion battery  $>1000$  cycles at 1C rate in coin cell configuration.

# Cost Analyses



**Figure 1.** Cost per kWh on various cathode / anode combination Li-ion battery (40kWh).<sup>1-3</sup>

Ref) <sup>1</sup> Entek projected cost

<sup>2</sup> TIAX estimate

<sup>3</sup> Estimate based on numbers provided by Andrew Jansen at ANL

Based on EV battery configuration.

Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub> anode is more expensive due to lower specific capacity 175mAh/g over graphite (372mAh/g).

LiFePO<sub>4</sub> cathode in full cell is slightly expensive due to lower potential 3.45V vs. Li<sup>+</sup>/Li.

Overall cost analyses depends on safety and cycle life of the full cell battery.

LFP : LiFePO<sub>4</sub>

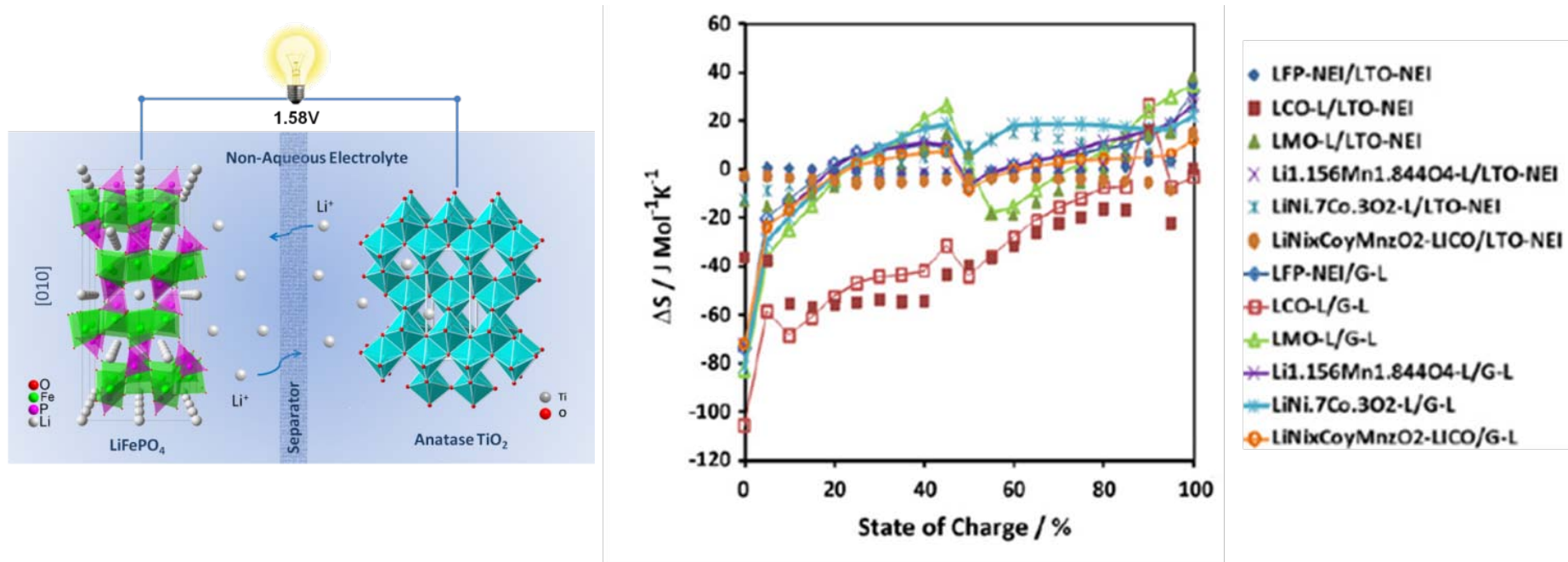
LMO: LiMn<sub>2</sub>O<sub>4</sub>

NCA: LiNi<sub>x</sub>Co<sub>y</sub>Al<sub>z</sub>O<sub>2</sub>

NCM: LiNi<sub>x</sub>Co<sub>y</sub>Mn<sub>z</sub>O<sub>2</sub>

LTO : Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub>

# Entropy vs. State of Charge (SOC)

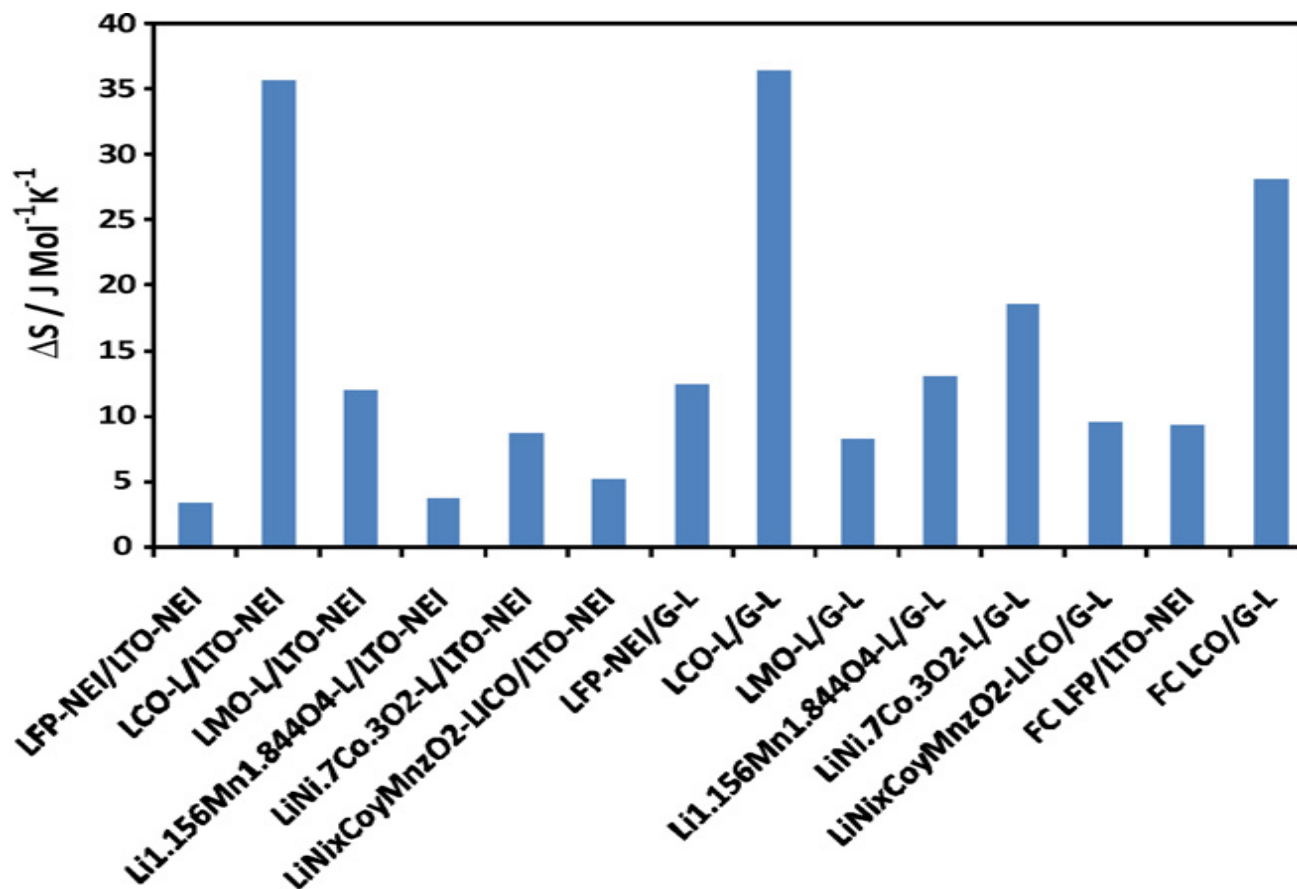


**Figure 2.** Entropy measurements on various electrode materials vs. state-of-charge (SOC).<sup>1</sup>

- ❑ Entropy changes at various state of charge (SOC) have been screened for selection of various electrode combinations with effective heat management and safety (thermal runaway).
- ❑ LiFePO<sub>4</sub> cathode and Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub> or TiO<sub>2</sub> anode combination show the low internal heat generation, due to minimum crystal structural changes “zero strain” during the Li insertion/extraction.
- ❑ LiFePO<sub>4</sub>, Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub> and TiO<sub>2</sub> redox potentials lie within stable window of conventional electrolyte where SEI layer formation is prevented.

Ref) <sup>1</sup> V.V. Viswanathan, D. Choi, D. Wang, W. Xu, S.Towne, R.E. Williford, J.-G. Zhang, J. Liu, Z. Yang, *Journal of Power Sources*, 195 (2010) 3720–3729

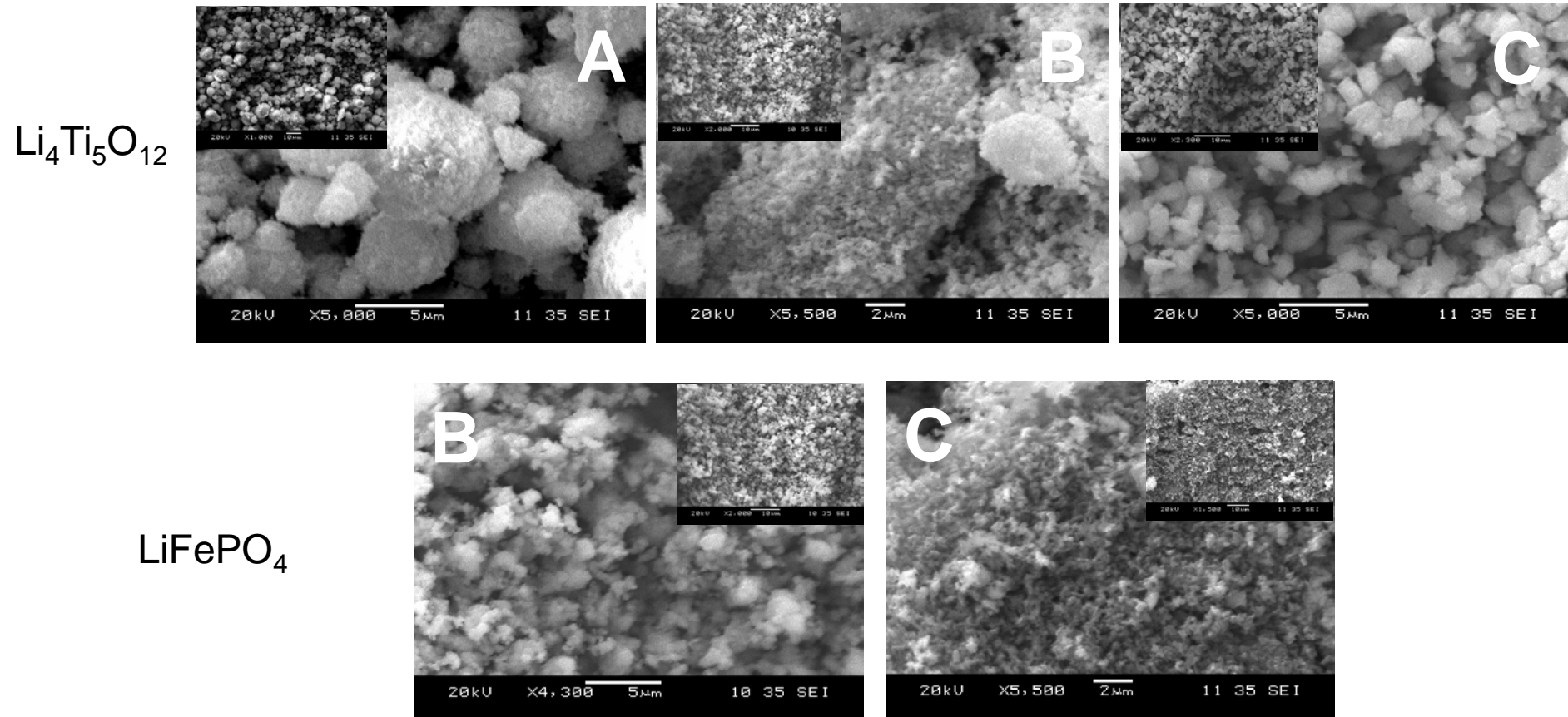
# Entropy Measurements



**Figure 3.** Average of absolute value of computed full cell entropy change over the 0~100% SOC range. FC LFP/LTO-NEI and FC LCO/G-L (last 2 columns) correspond to average of measured entropy change for full cells.



# Microstructures of $\text{LiFePO}_4$ & $\text{Li}_4\text{Ti}_5\text{O}_{12}$



**Figure 4.** SEM micrographs of  $\text{LiFePO}_4$  and  $\text{Li}_4\text{Ti}_5\text{O}_{12}$  electrode materials from different commercial sources (Company A, B, and C).

# Cathode & Anode Half-Cell Rate Analyses

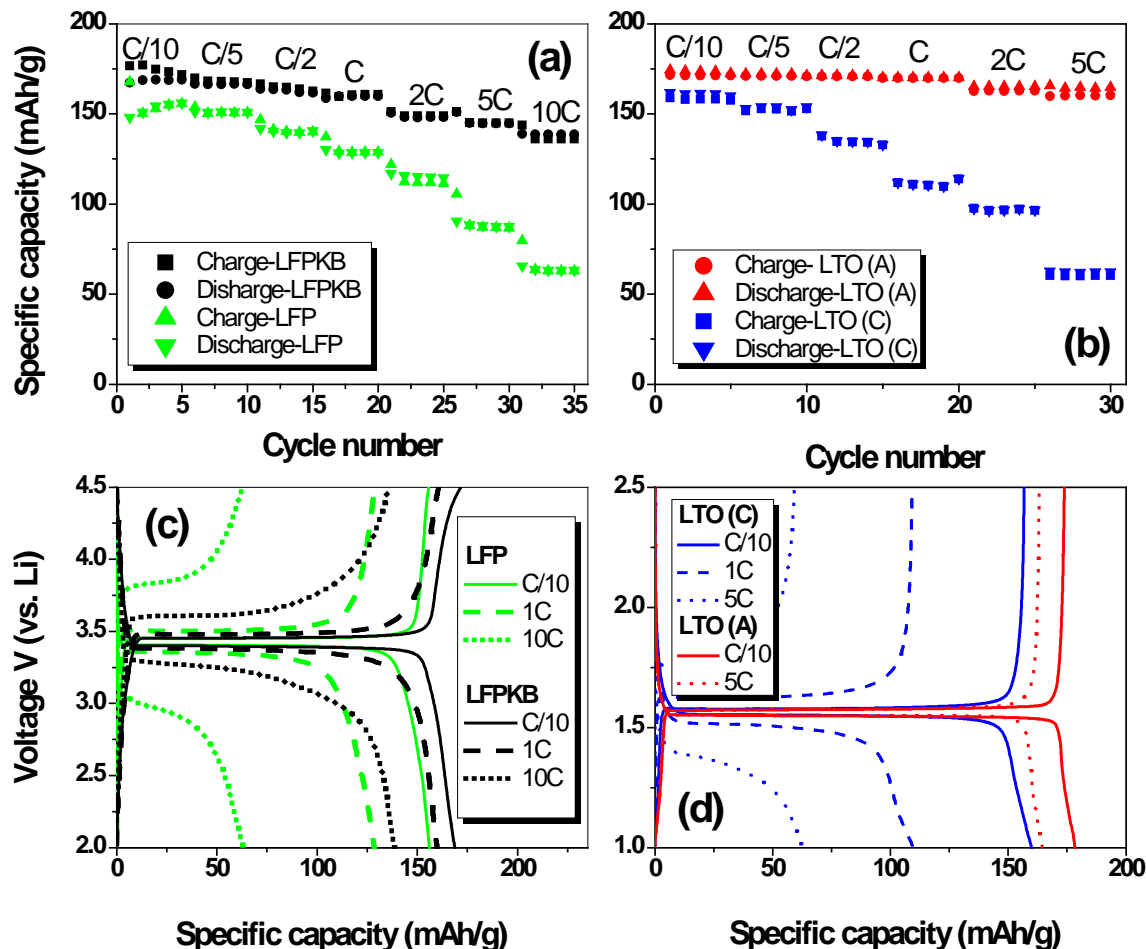


Figure 5. Electrochemical cycling at various C rates (a)  $\text{LiFePO}_4$  and Ketjen black modified  $\text{LiFePO}_4$ , (b)  $\text{Li}_4\text{Ti}_5\text{O}_{12}$  (company A and C) and voltage profiles of charge/discharge at various C rates (c)  $\text{LiFePO}_4$  and Ketjen black modified  $\text{LiFePO}_4$ , (d)  $\text{Li}_4\text{Ti}_5\text{O}_{12}$  (company A and C).

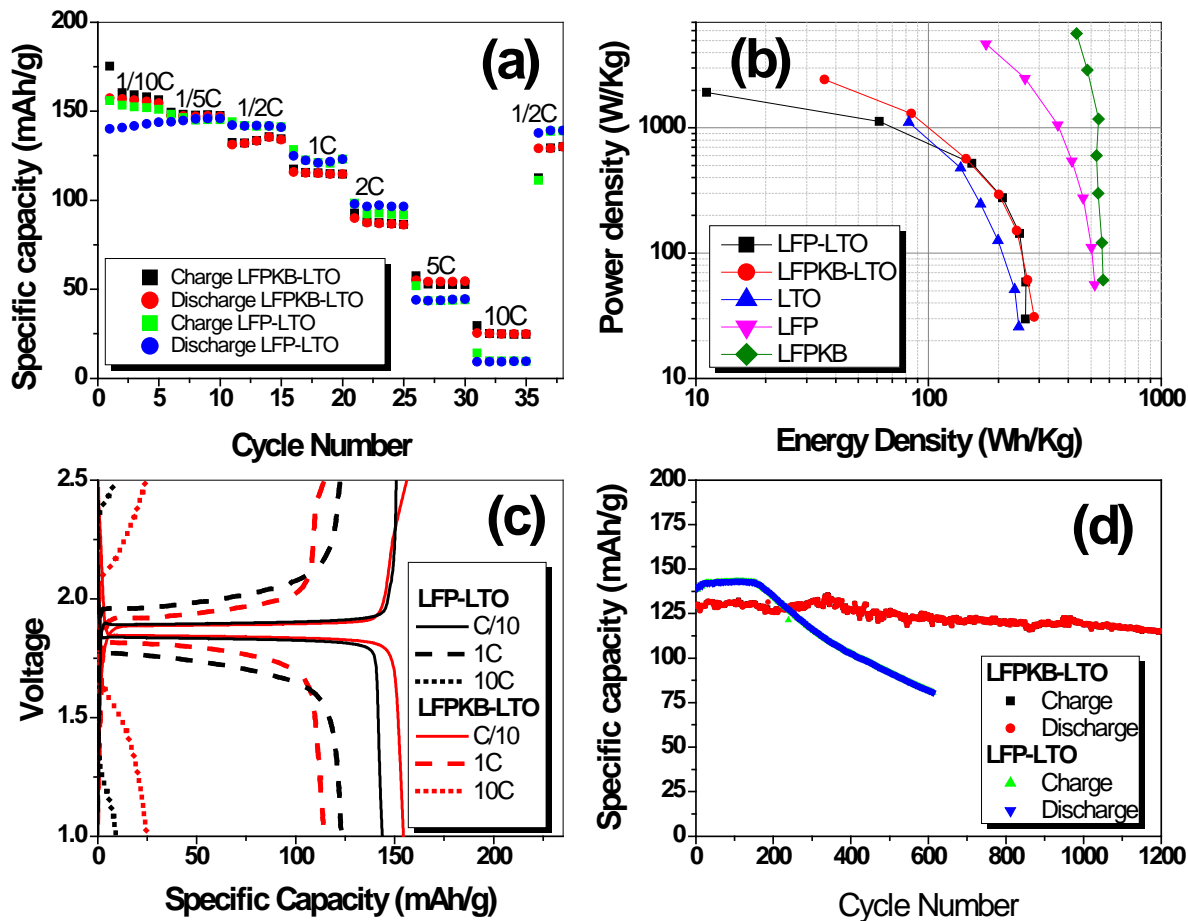
LiFePO<sub>4</sub> from company B and C was similar.  $\text{Li}_4\text{Ti}_5\text{O}_{12}$  from company C was not as good as those from company A and B.

Ketjen Black modified  $\text{LiFePO}_4$  by milling shows much improved rate performance.

Effective carbon mixing with electrode material and preparation step is critical for achieving the high rate and cycling stability



# LiFePO<sub>4</sub>-Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub> Full Cell Evaluation



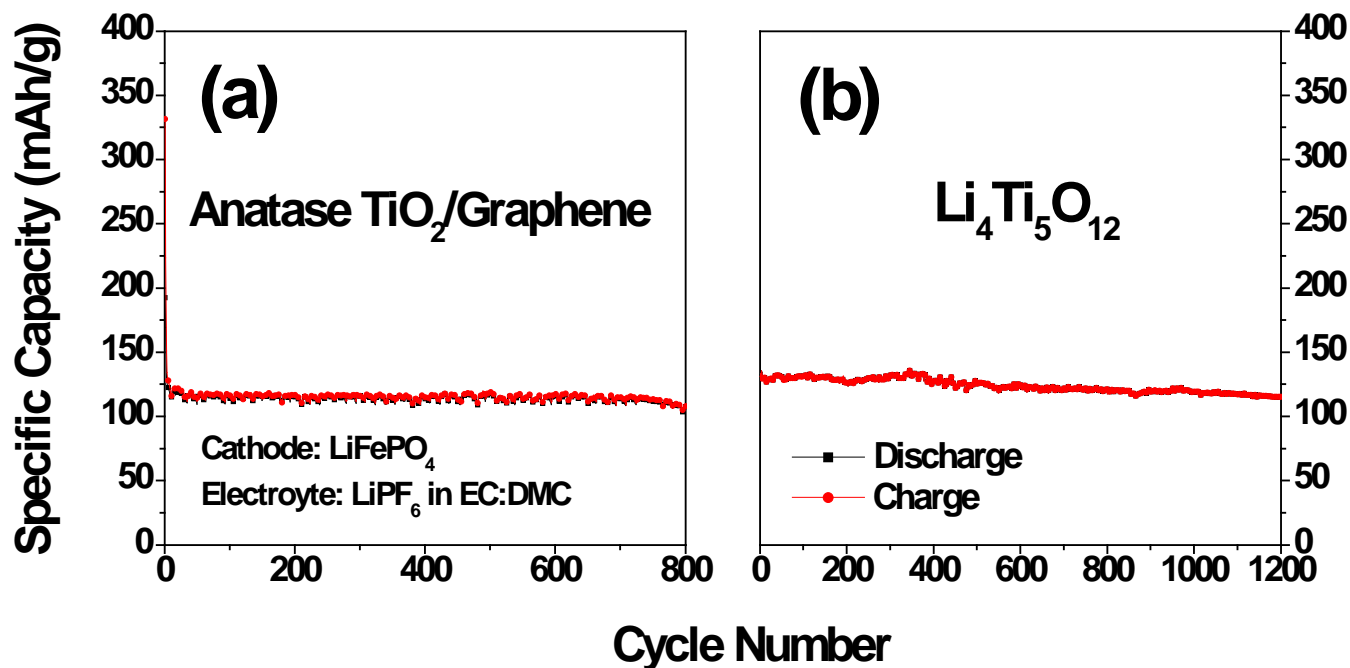
□ Ketjen Black modified LiFePO<sub>4</sub> by milling shows improvement in rate performance.

□ Effective carbon mixing with electrode material and electrode preparation is critical for achieving the high rate and cycling stability.

□ LiFePO<sub>4</sub> cathode plays important role in cycling stability.

Figure 6. Electrochemical cycling at various C rates of (a) LiFePO<sub>4</sub>-Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub> full cell with and without Ketjen black modified LiFePO<sub>4</sub>, (b) Ragone plot of full cell and half cell, (c) voltage profiles of charge/discharge at various C rates and (d) long term cycling of LiFePO<sub>4</sub>-Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub> full cell with and without Ketjen black modification.

# Full Cell Cycling Evaluation

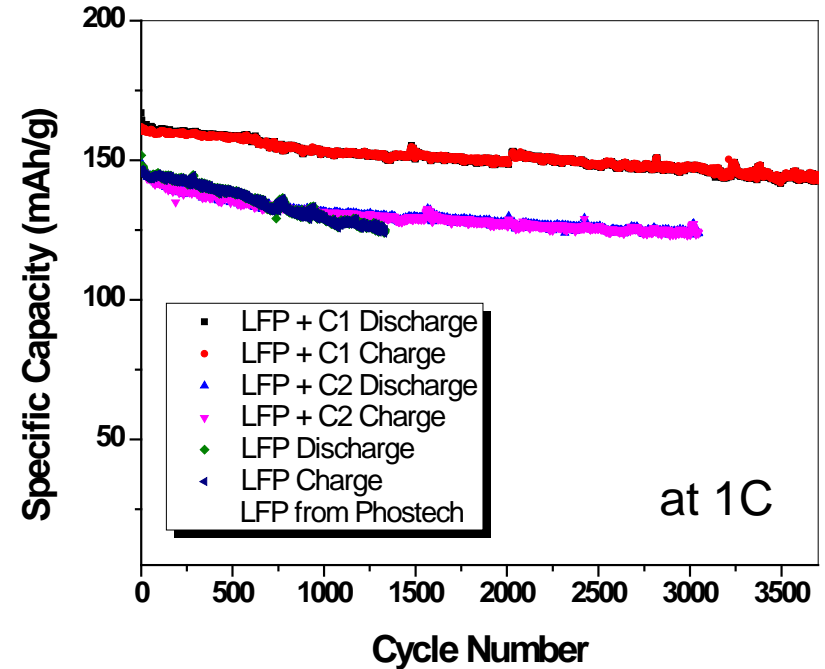
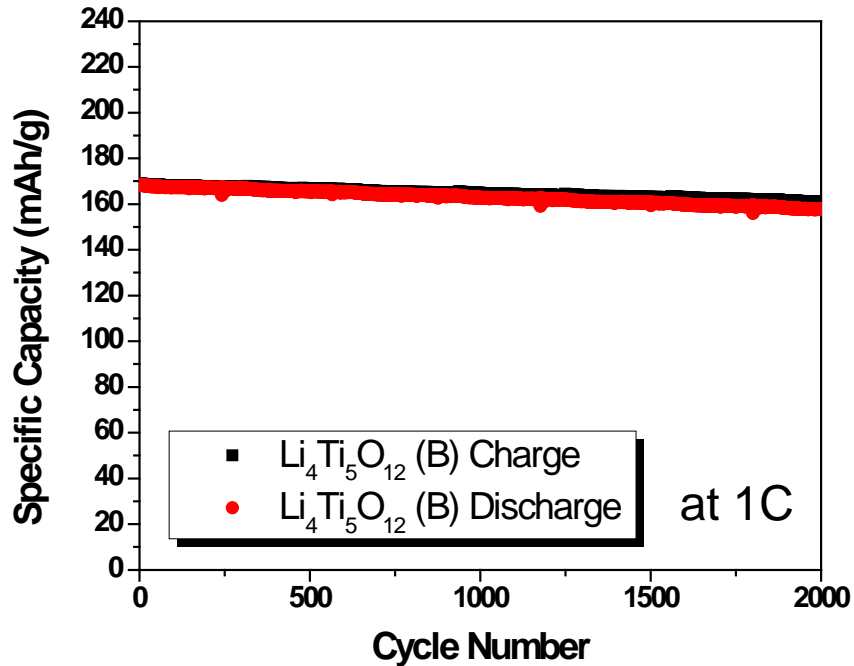


**Figure 7.** Cycling performance of Li-ion batteries made from LiFePO<sub>4</sub> cathode and (a) self-assembled anatase TiO<sub>2</sub>/graphene (5wt%) composite anode and (b) Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub> anode.<sup>1</sup>

- ❑ Over 1000cycles without negligible capacity fade have been achieved in coin cell.
- ❑ Full cell cycling stability depends on LiFePO<sub>4</sub> cathode stability.
- ❑ Effective carbon coating on LiFePO<sub>4</sub> cathode improves rate and cycle life.

Ref) <sup>1</sup> D. Choi, D. Wang, V. V. Viswanathan, I.-T. Bae, W. Wang, Z. Nie, J.-G. Zhang, G. L. Graff, J. Liu, Z. Yang, T. Duong, *Electrochem. Commun.* 12 (2010) 378–381.

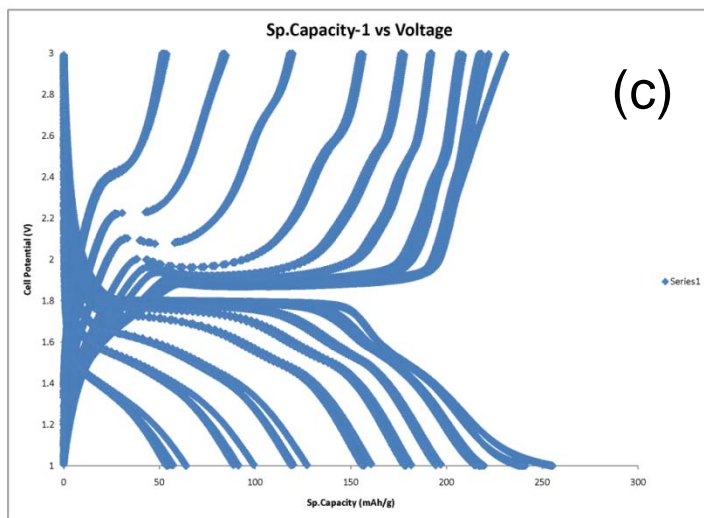
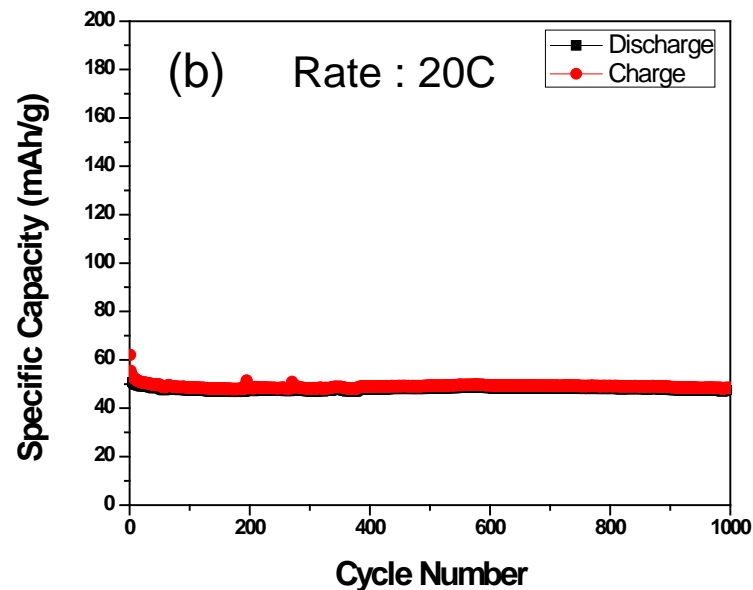
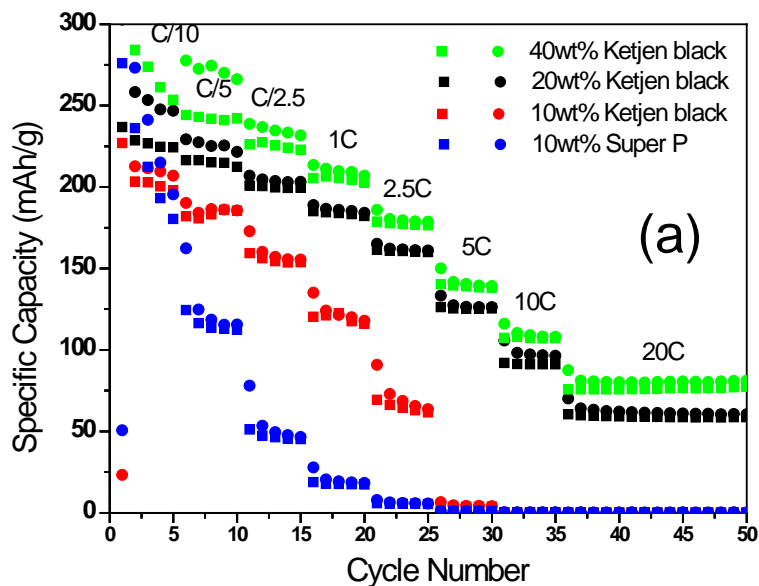
# $\text{LiFePO}_4$ and $\text{Li}_4\text{Ti}_5\text{O}_{12}$ Half Cell Cycling



**Figure 8.** Electrochemical cycling of  $\text{Li}_4\text{Ti}_5\text{O}_{12}$  and  $\text{LiFePO}_4$  cathode with various carbon (C1, C2 and super P) additives (10wt%).

- $\text{LiFePO}_4$  cathode plays important role in cycling stability.
- Optimized cycling of  $\text{LiFePO}_4$  cathode using different conductive carbon.
- On going test (>3500 cycles).

# Commercial anatase TiO<sub>2</sub> Anode



**Figure 9.** Electrochemical rate test on (a) Ketjen black modified anatase TiO<sub>2</sub> (b) extended cycling at 20C and (c) voltage profiles at various C-rates.

- ❑ Cheapest anatase TiO<sub>2</sub> commercially available have been tested for anode.
- ❑ Working to improve the rate performance through effective carbon mixing.
- ❑ Cycling stability is excellent.

# Summary

- ❑ Various electrode materials have been tested for entropy (reversible heat) measurements.  $\text{LiFePO}_4/\text{Li}_4\text{Ti}_5\text{O}_{12}$  cathode/anode combination is suitable Li-ion battery for stationary energy storage.
- ❑ Cost comparison on different electrode material combinations show ~\$600/kWh for  $\text{LiFePO}_4/\text{Li}_4\text{Ti}_5\text{O}_{12}$  but life cycle cost (\$/kWh/cycle) plus safety is expected to be competitive over other combinations.
- ❑ Rate and cycling tests were performed on half and full cell and electrodes are being further optimized. Cathode using  $\text{LiFePO}_4$  is critical for stable cycling.
- ❑ Cycling performance of >1000cycles at 1C rate was achieved at R.T.



# Future Tasks

- ❑ Enhance cycling stability >1000 cycles of  $\text{LiFePO}_4$  using cheaper route.
- ❑ Fabrication and testing of 18650 cell using  $\text{LiFePO}_4$  /  $\text{Li}_4\text{Ti}_5\text{O}_{12}$  combination electrodes.
- ❑ Calorimetric analyses of 18650 cell for total heat-generation.
- ❑ Cost analysis on 18650 cell with cost per cycle will be reported.