

# CAES Modeling



*Geomechanics Research Department*

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**Matt Kirk, Mark Grubelich, Steve Webb, Scott Broome**  
**SAND 2010-6940C**

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# Specific and existing problems, interests, needs



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**1-Potential Microbial and Chemical Impact of CAES in a Sandstone, M. Kirk**

**2-Assessment of Ignition/Explosion Potential in a Depleted Hydrocarbon Reservoir from Air Cycling Associated with CAES, M. Grubelich**

**3-Flow Analysis Parametric Study: S. Webb**

**4-Material Degradation (T-M-C-H effects) Due to Cyclic Loading, SJ Bauer and ST Broome**

**Started November 2009**



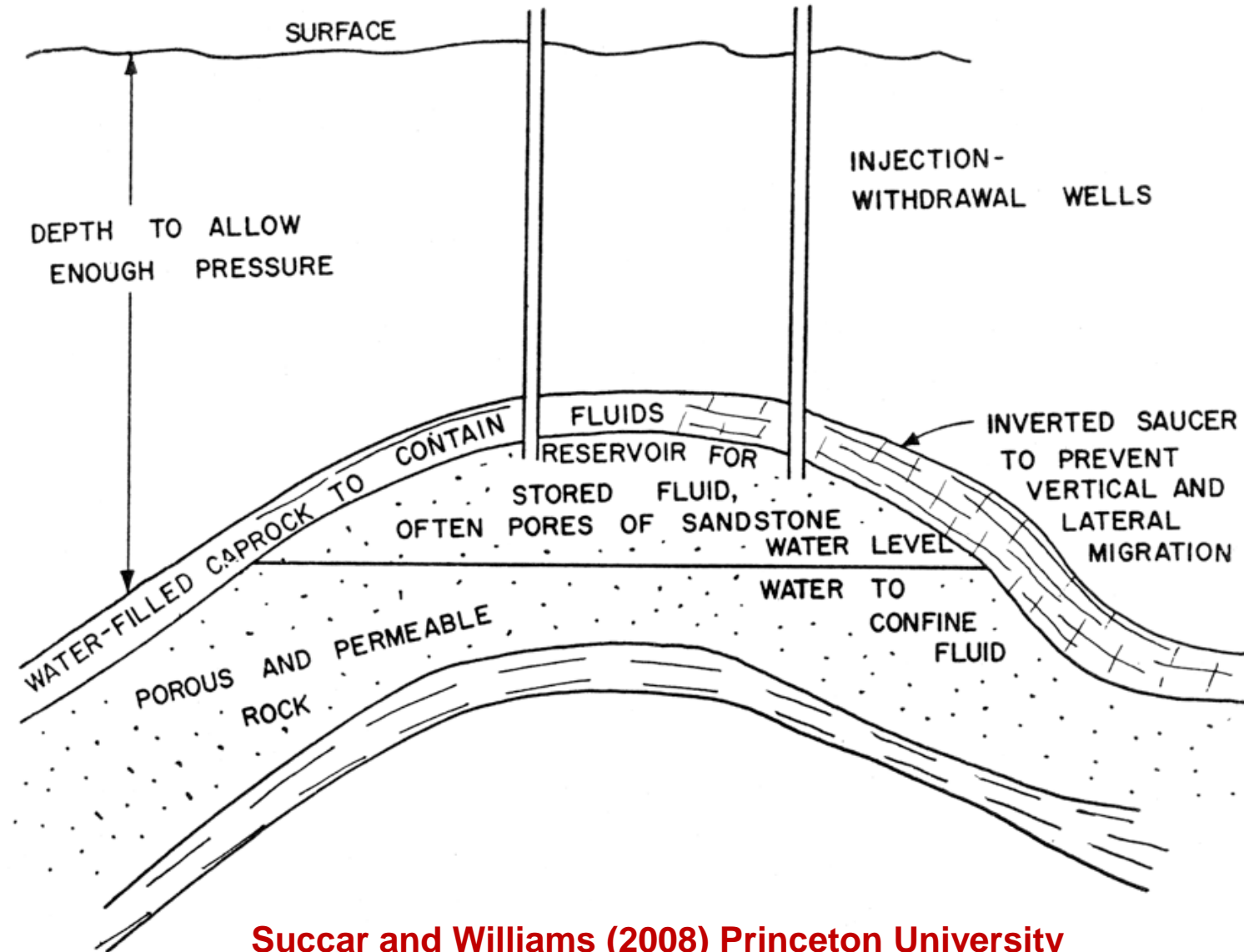
# **Potential Microbial and Chemical Impact of CAES in a Sandstone**

**Matthew Kirk**  
**Geochemistry Department**

# Compressed Air Energy Storage



Geomed



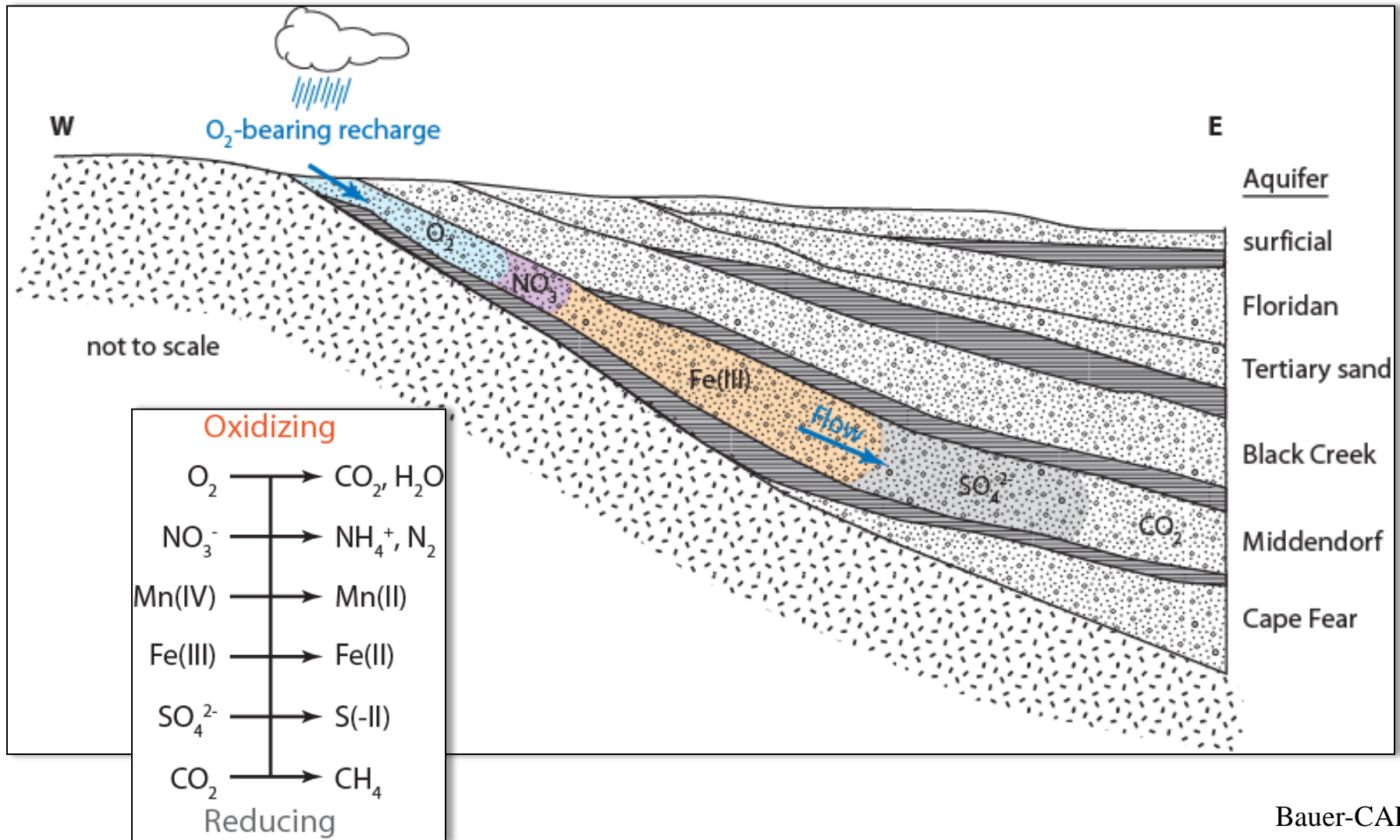
Succar and Williams (2008) Princeton University

Bauer-CAES

# Groundwater Microbiology



## Example: Middendorf coastal plain aquifer, South Carolina





# Conclusions: Potential Microbial and Chemical Impact of CAES in a Sandstone

- Sandstone evaluated in a reducing environment
- Microbial Fe(II) and Mn(II) oxidation will become favorable
- Pyrite oxidation could lead to considerable changes in pH, salinity, and mineralogy
- Microbiology and mineralogy changes would impact porosity



# Considerations for Explosion Potential for CAES in a Depleted Natural Gas Reservoir



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**Mark Grubelich**  
**Geothermal Energy**



# Results & Conclusions: Mitigation & Safety



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- **Purge reservoir before use**
- **Low pressure air cycling below UFL to remove gas (~90 psi)**
- **In-situ gas monitor**
- **Never draw down air below the LFL (370 psi)**
- **Insure no surface breach if ignition occurs (sufficient overburden)**
- **Monitor NG content entering surface equipment**
- **Further study required**
  - **Buoyancy issues, etc.**





# CAES Borehole Study: Steve Webb

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- **Objective**
  - **Look at Flow in Individual Boreholes**
  - **Simple 2-d Models**
  - **Estimate Number of Boreholes and CAES Footprint**
- **Assumptions**
  - **Representative Borehole/Formation Geometry**
  - **Include Two-Phase Behavior**
    - **Capillary Pressure and Relative Permeability**
    - **Bubble Formation**
    - **Air Injection and Withdrawal – 10 Weekly Cycles**

# Conclusions



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- **Permeability Variation Much More Important than Porosity Variation**
- **Procedure Can Quantify Differences Between Various Sets of Formation Parameters**
  - **Borehole Spacing, Number of Boreholes**
- **Borehole Arrays Will Be Investigated in the Future**

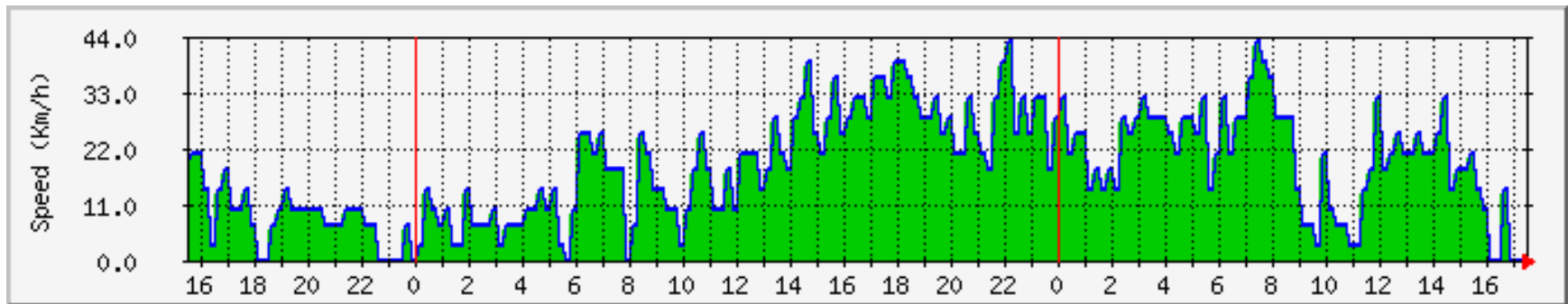


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# Material Degradation (T-M-C-H effects)

## Due to Cyclic Loading

### SJ Bauer and ST Broome



**Hourly fluctuations in wind speed could translate to frequent pressurization/depressurizations of salt caverns**



# Concluding Comments



- Preliminary cyclic tests completed on salt
- Change in volume strain observed
- Young's Modulus changes observed
- Acoustic emissions detected
- Cracks observed in thick sections
- Results consistent with previous work
- Implication that cyclic loading caused cracking at low differential stresses



# Summary/Conclusions



- 1- Sandstone in a reducing environment could effect biologic and mineralogic changes that could lead to changes in porosity and permeability**
- 2-Recommendations given for mitigation of potential use of a natural gas reservoir for CAES**
- 3- Permeability variation much more important than porosity variation; procedure can help determine borehole spacing, number of boreholes (CO\$T)**
- 4-Salt strength observed to degrade in cyclic loading**

# Work Products



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- 1- ***“Potential Effects of Compressed Air Energy Storage on Microbiology, Geochemistry, and Hydraulic Properties of Porous Aquifer Reservoirs”***, Kirk, Altman, and Bauer, SAND2010-4721  
***“Potential Subsurface Environmental Impact of Compressed Air Energy Storage in Porous Bedrock Aquifers”*** Env. Sci. & Tech. (in Prep, Kirk et al)
- 2- ***“Considerations for Explosion Potential for CAES in a Depleted Natural Gas Reservoir”*** , M. Grubelich
- 3- ***“Borehole and Formation Analyses in Reservoirs to Support CAES Development”*** , S. Webb
- 4- ***“Experimental Deformation of Salt in Cyclic Loading”***, S. Bauer and S. Broome, Solution Mining Research Institute April 2010 SAND2010-1805



# Statement about future tasks



- 1- Develop map for US regions with geology potentially suitable for CAES**
- 2- Borehole parametric study**
- 3-Continue evaluation of cyclic loading effects on salt and reservoir rocks**





**thanks**

**Questions?**



# **Potential Microbial and Chemical Impact of CAES in a Sandstone**

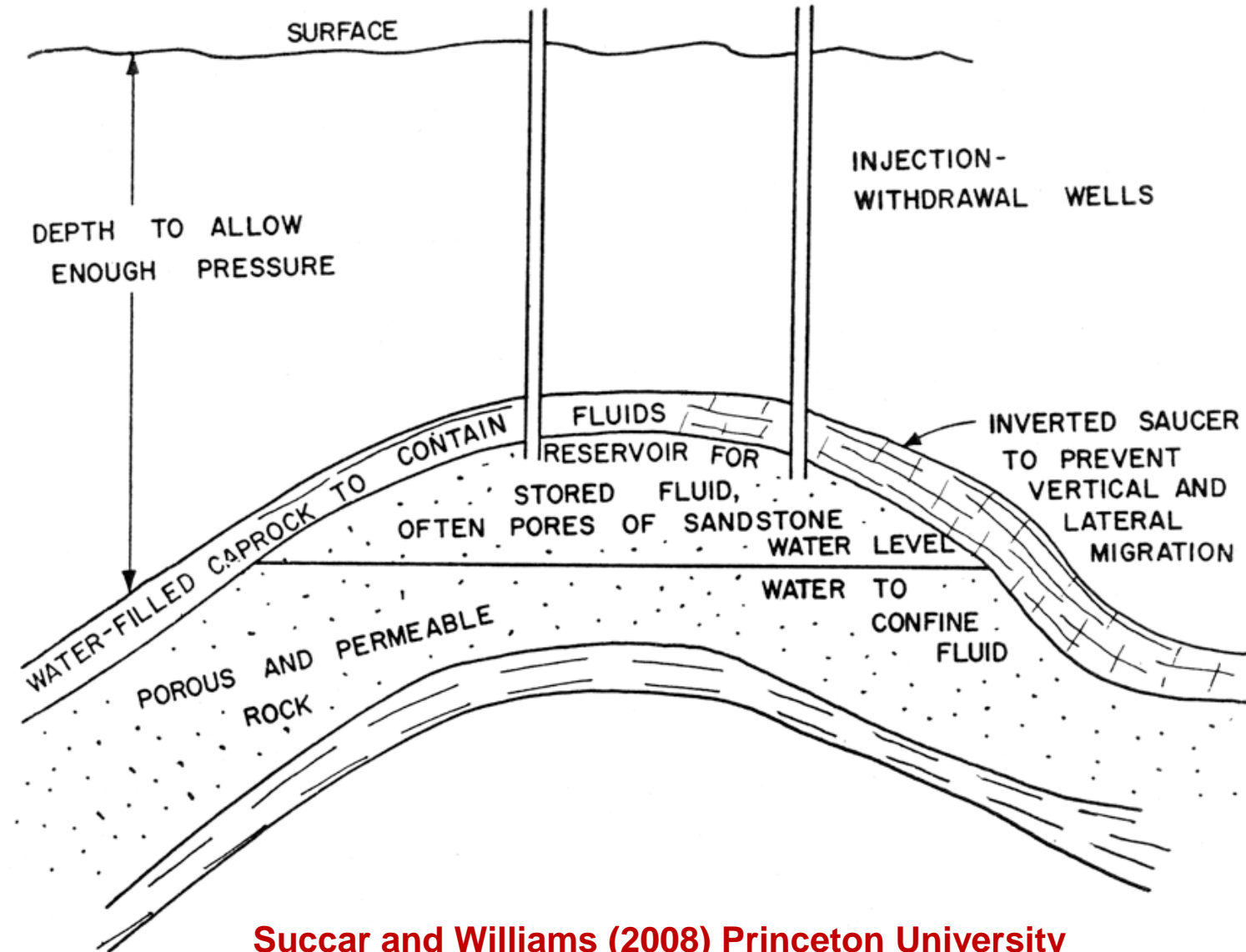
**Matthew Kirk**  
**Geochemistry Department**



# Compressed air energy storage



Geomed

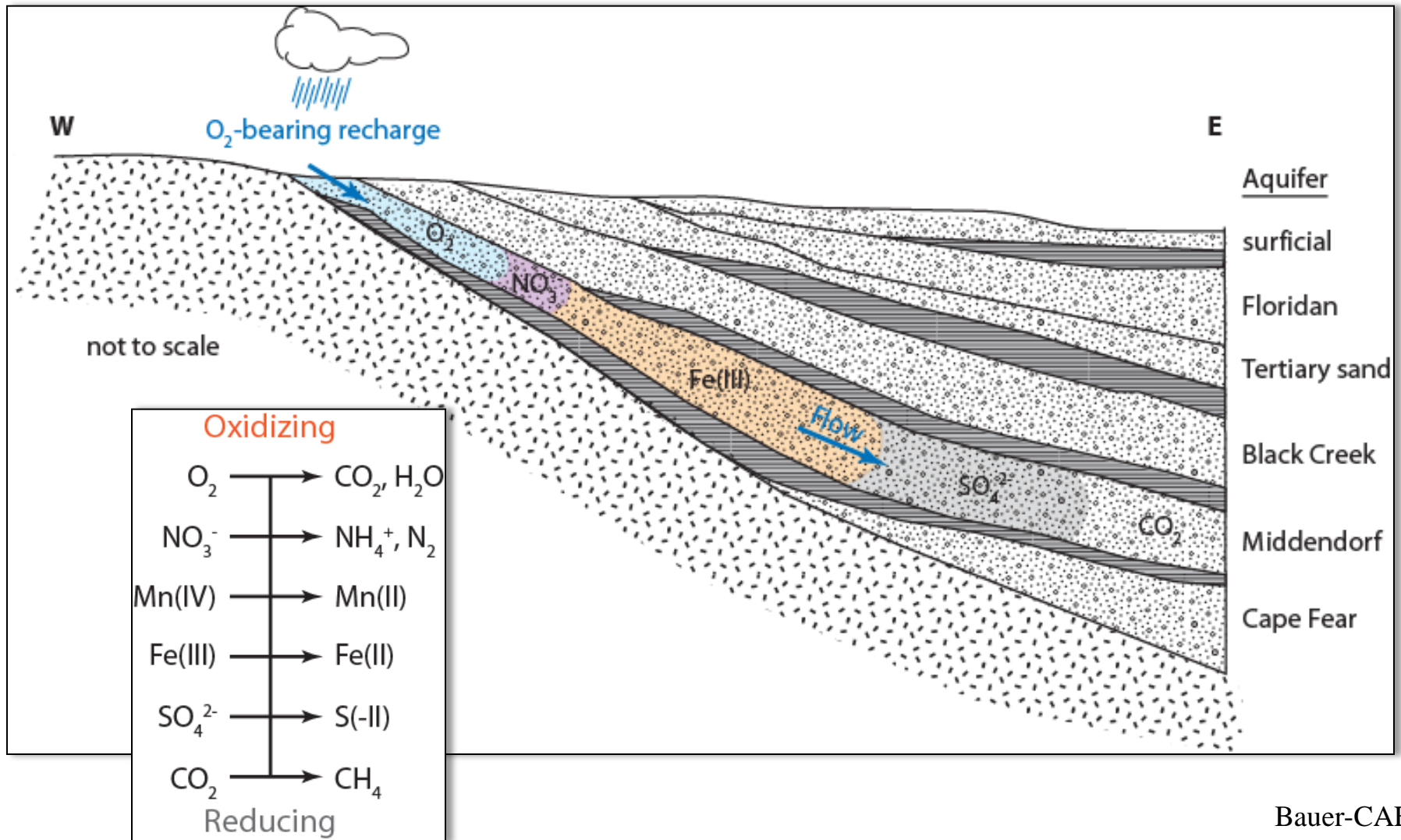


Succar and Williams (2008) Princeton University

# Groundwater microbiology



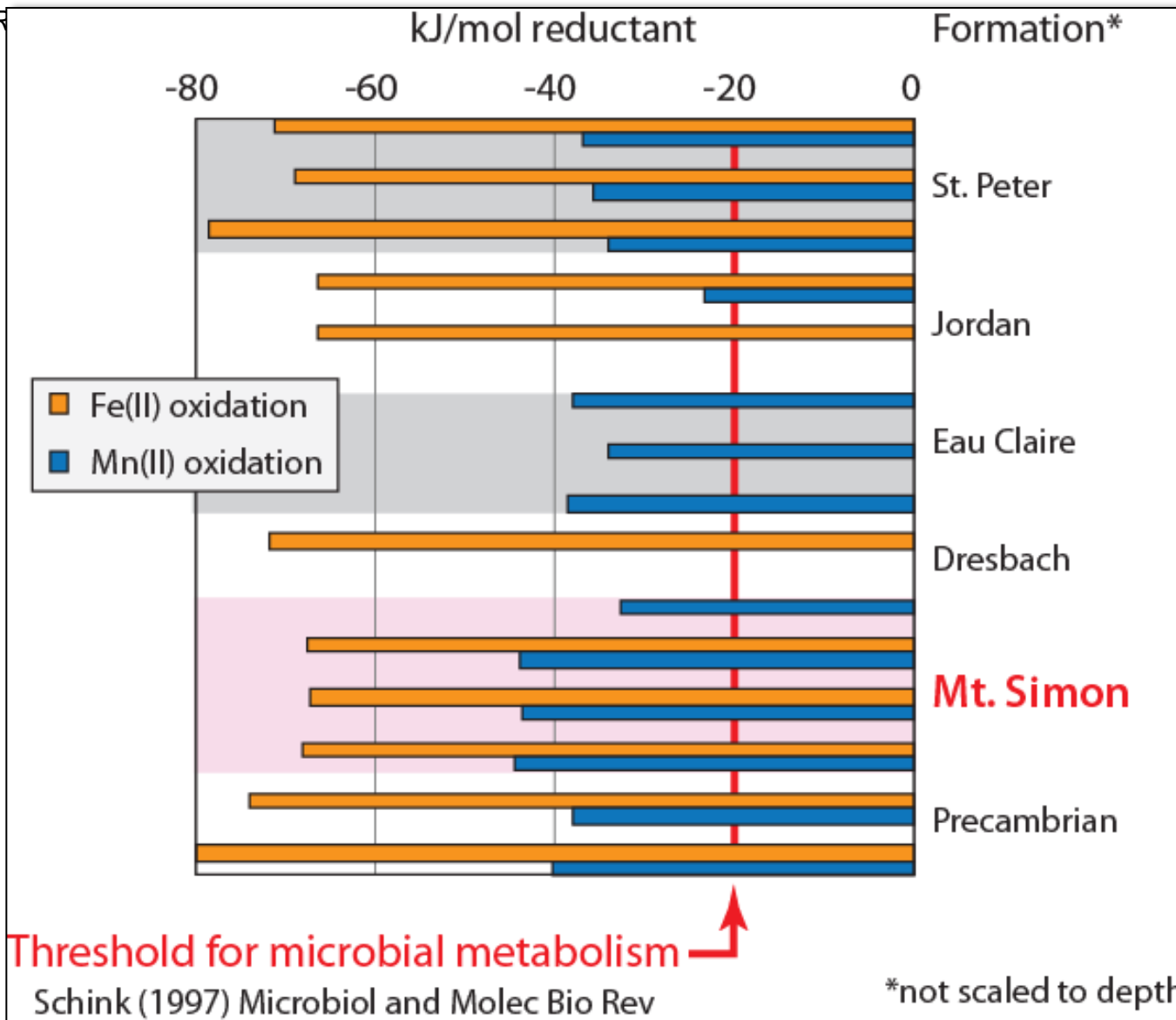
## Example: Middendorf coastal plain aquifer, South Carolina



# Metabolic energy available for Fe(II) and Mn(II) oxidation in the Mt. Simon



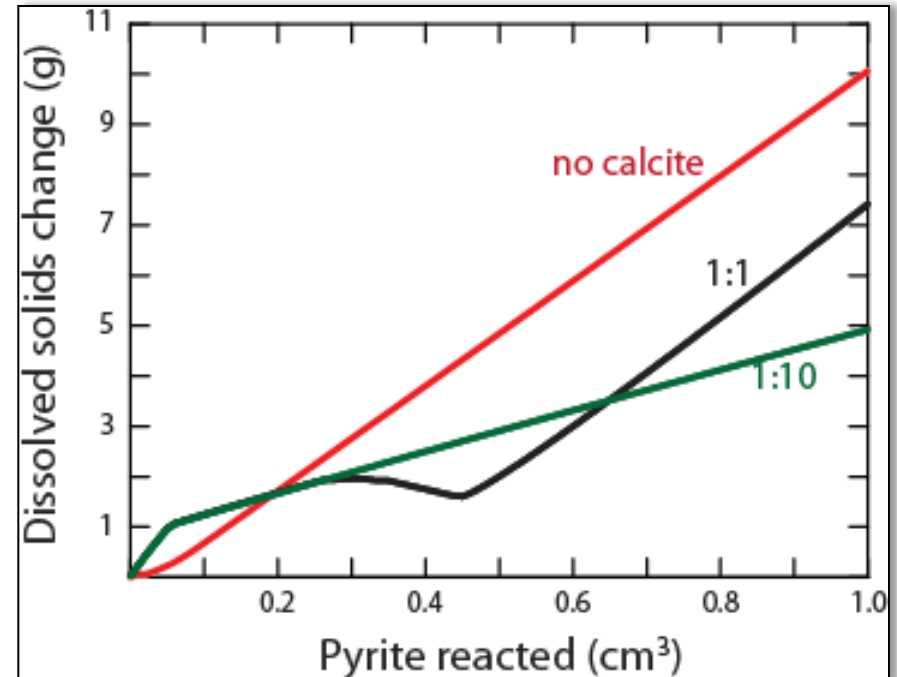
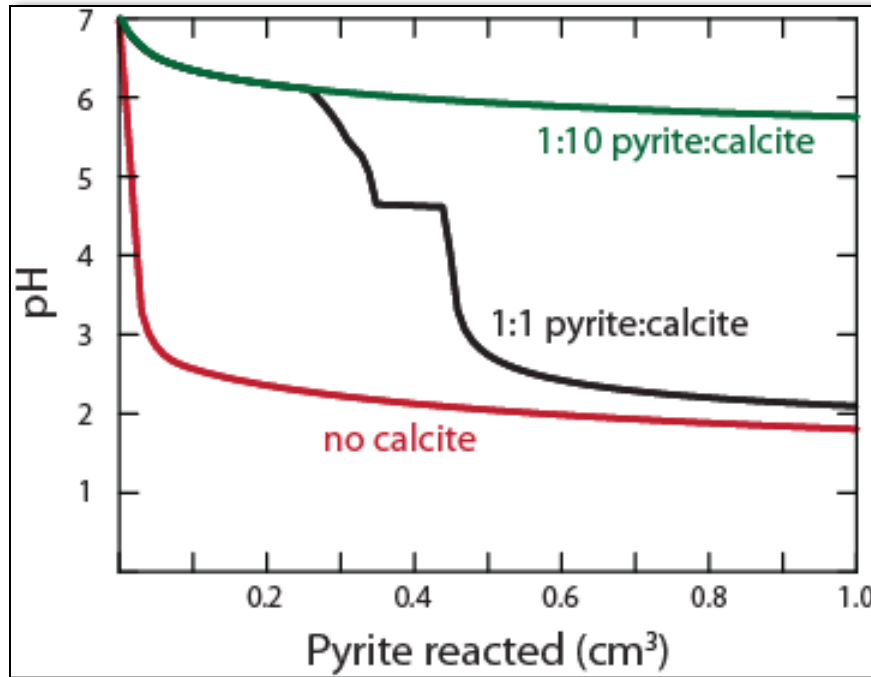
Geomechanics R



# Effect of pyrite oxidation on groundwater composition



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**Geochemist's Workbench reaction path model assuming 0.2 fO<sub>2</sub>**

- **no calcite:**  $\text{pyrite} + 3.75 \text{ O}_2 + 3.5 \text{ H}_2\text{O} \rightarrow \text{Fe(OH)}_3 + 2 \text{ SO}_4^{2-} + 4 \text{ H}^+$
- **with calcite:**  $\text{pyrite} + 2 \text{ calcite} + 3.75 \text{ O}_2 + 1.5 \text{ H}_2\text{O} \rightarrow \text{Fe(OH)}_3 + 2 \text{ SO}_4^{2-} + 2 \text{ Ca}^{2+} + 2 \text{ CO}_2$

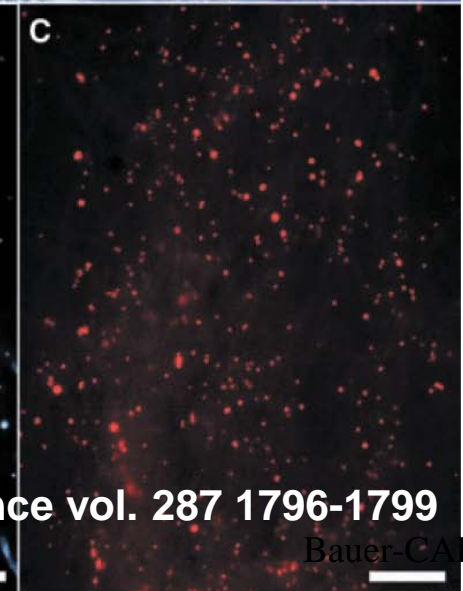
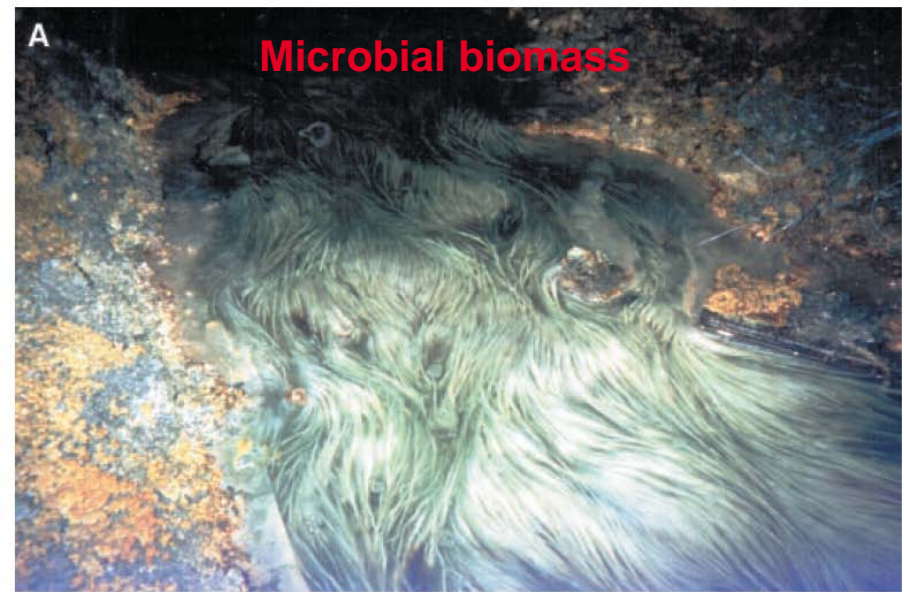
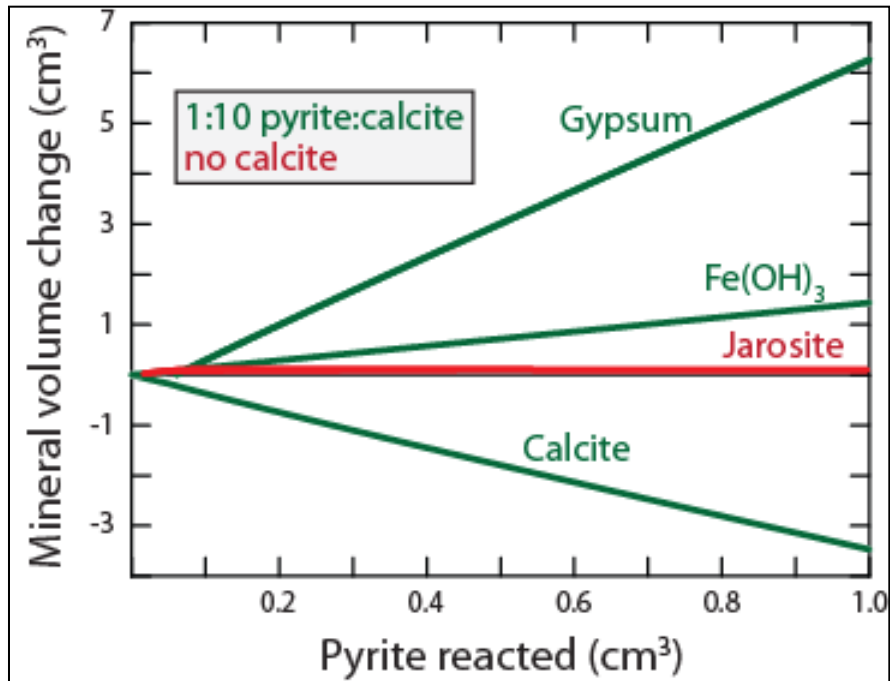


# Effect of Pyrite Oxidation on Porosity



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## Mineral volume



Edwards et al. Science vol. 287 1796-1799

Bauer-CAES



# Conclusions: Potential Microbial and Chemical Impact of CAES in a Sandstone

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# Considerations for Explosion Potential for CAES in a Depleted Natural Gas Reservoir



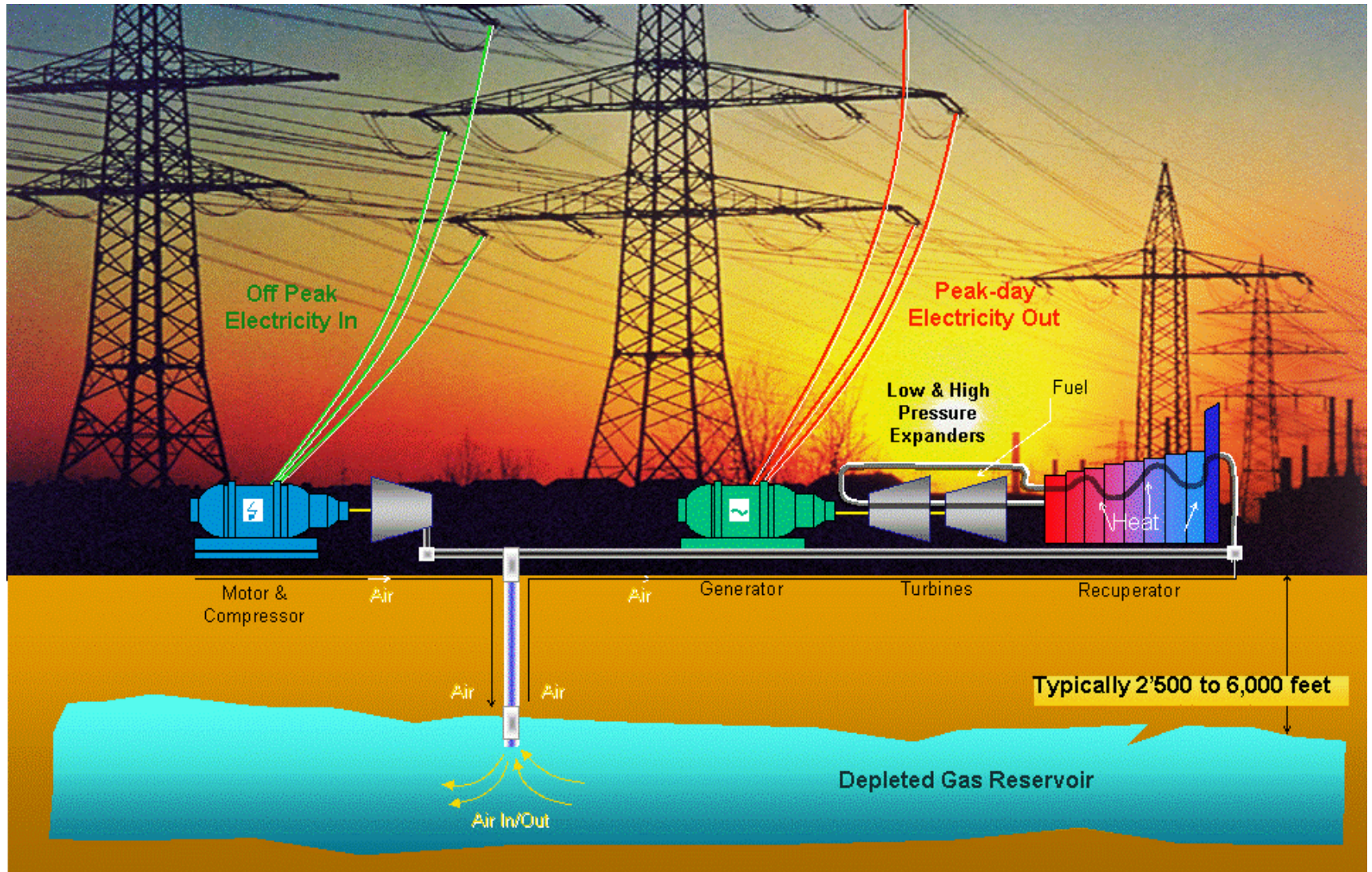
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**Mark Grubelich**  
**Geothermal Energy**



# Compressed Air Energy Storage

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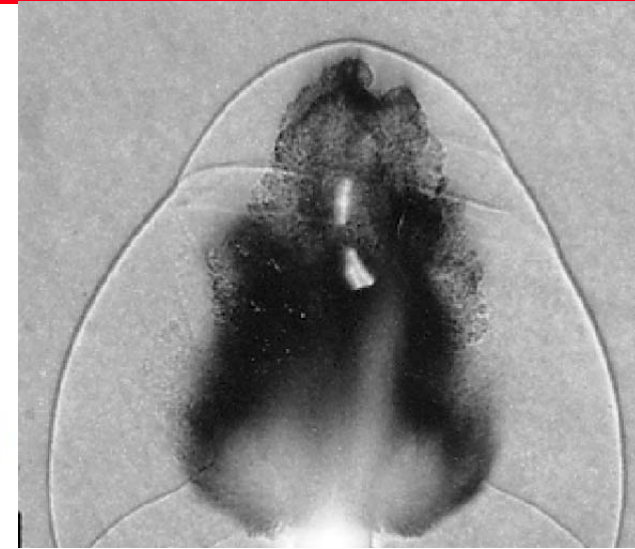
# Fuel, Oxygen & Ignition Source



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**Combustion or Deflagration**  
10's to 100's of ft/sec reaction rates.



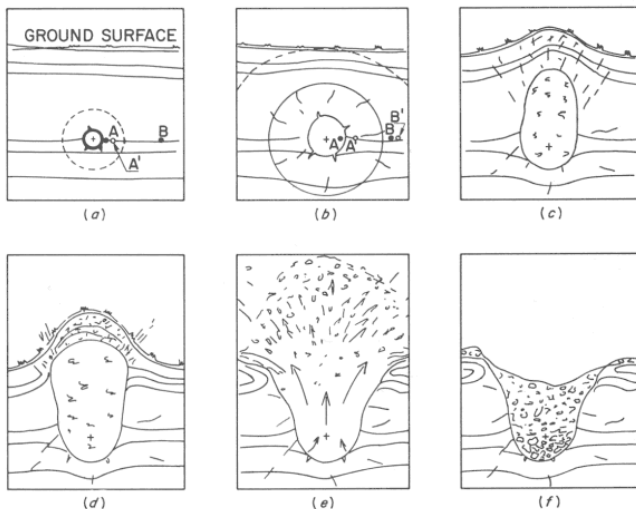
**Detonation, reaction proceeds at supersonic speeds (shock wave).**

# Why worry?



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- The pressure rise ratio for a confined deflagrating (unvented) fuel air mixture is **~9:1**
- The peak pressure ratio for a detonating fuel air mixture is **~ 18:1**
- Both events could be severe: (rough calculation in progress)



# Important Points

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## Depleted gas reservoir

- What does depleted mean?
- At atmospheric pressure?
  - What is the residual natural gas composition?
    - Why is this important?
      - Heavy hydrocarbons change the ignition window and decrease the ignition temperature

### Natural gas composition

Component	Typical Analysis (mole %)	Range (mole %)
Methane	95.2	87.0 - 96.0
Ethane	2.5	1.5 - 5.1
Propane	0.2	0.1 - 1.5
iso - Butane	0.03	0.01 - 0.3
normal - Butane	0.03	0.01 - 0.3
iso - Pentane	0.01	trace - 0.14
normal - Pentane	0.01	trace - 0.04
Hexanes plus	0.01	trace - 0.06
Nitrogen	1.3	0.7 - 5.6
Carbon Dioxide	0.7	0.1 - 1.0
Oxygen	0.02	0.01 - 0.1
Hydrogen	trace	trace - 0.02

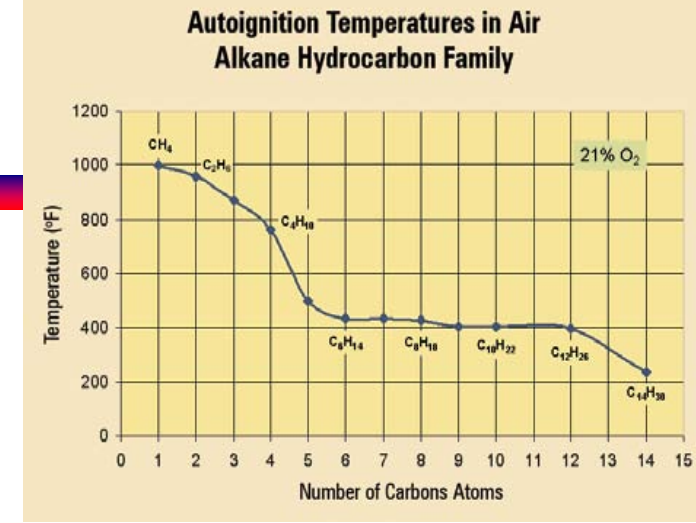


Table 6. — Limits of flammability of combustible vapors in air and oxygen at 25° C and 1 atm<sup>1</sup>

Combustible	Flammability limits, vol pct			
	Air		Oxygen	
	L <sub>25</sub>	U <sub>25</sub>	L <sub>25</sub>	U <sub>25</sub>
<b>HYDROCARBONS</b>				
Methane	5.0	15.0	5.0	61
Ethane	3.0	12.4	3.0	66
Propane	2.1	9.5	2.3	55
n-Butane	1.8	8.4	1.8	49
n-Hexane	1.2	7.4	1.2	<sup>2</sup> 52
n-Heptane	1.1	6.7	.9	<sup>2</sup> 47
Acetylene	2.5	100	≤2.5	100
Ethylene	2.7	36	2.9	80
Propylene	2.4	11	2.1	53
α-Butylene	1.6	10	1.8	58
Cyclopropane	2.4	10.4	2.5	60
Benzene	<sup>2</sup> 1.3	<sup>2</sup> 7.9	≤1.3	NA

# Ignition Window

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- **Lower Flammability Limit** (aka Lower Explosive Limit, LFL or LEL)
  - Below the LFL the mixture of fuel and air lacks sufficient fuel to react
  - Above the LFL deflagration or detonation possible
- **Upper Flammability Limit** (aka Upper Explosive Limit, UFL or UEL)
  - Above the UFL the mixture of fuel and air lacks sufficient air to react.
  - Below the UFL deflagration or detonation possible
- **~Ignition possible between 90 and 370 psi**
  - Assuming well mixed conditions and starting at 1 atmosphere NG
  - ~Below 90 psi too rich and above 370 psi too lean
  - Example: Flight 800 center tank explosion
    - Lean on the ground & rich at cruise altitude
    - Above the LFL and below the UFL during climb
    - Ignition source present
    - Boom!



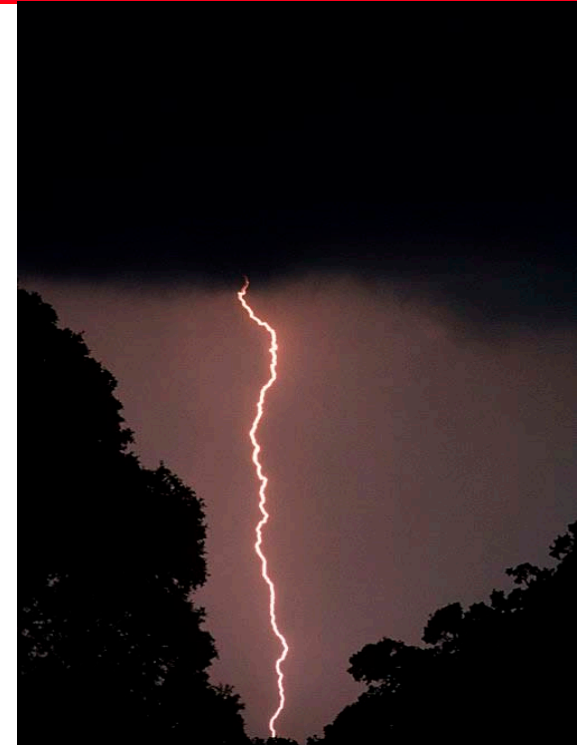
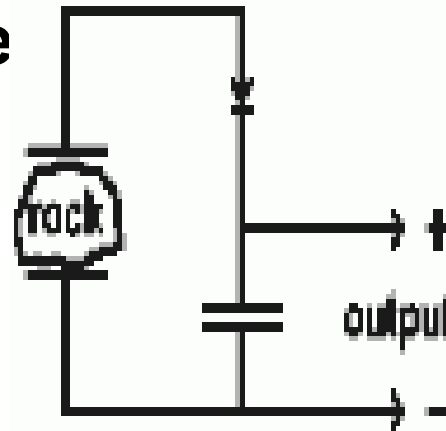




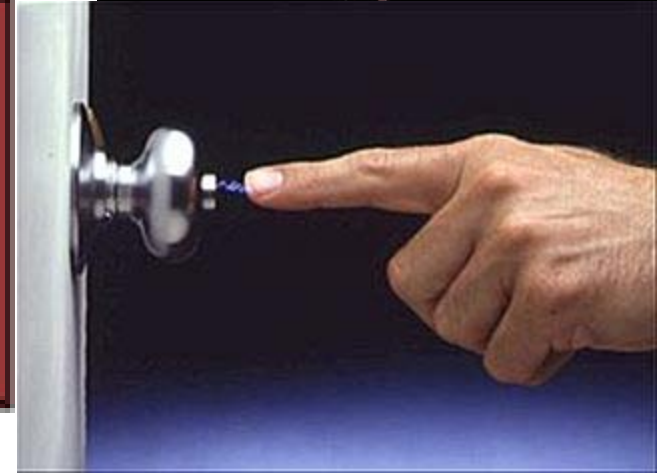
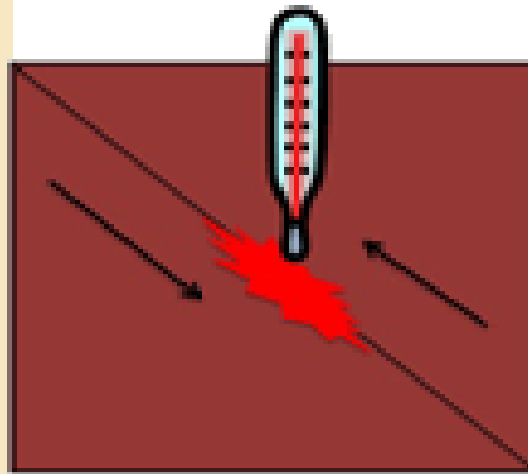
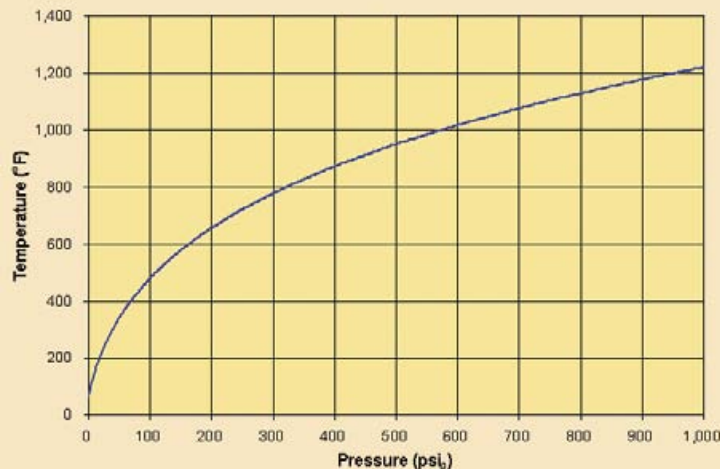
# Ignition Sources $0.3 \text{ mJ} = 0.0002 \text{ ft-lb} = \text{“not much”}$

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- **Adiabatic compression**
- **Piezo-electric discharge**
- **Static discharge**
- **Lightning strike**
- **Frictional heating**



**Adiabatic Compression of Air**



# Results & Conclusions: Mitigation & Safety



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- **Purge reservoir before use**
- **Low pressure air cycling below UFL to remove gas (~90 psi)**
- **In-situ gas monitor**
- **Never draw down air below the LFL (370 psi)**
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- **Monitor NG content entering surface equipment**
- **Further study required**
  - **Buoyancy issues, etc.**



# CAES Borehole Study



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**Stephen W. Webb**



Sandia is a multiprogram laboratory operated by Sandia Corporation, a Lockheed Martin Company,  
for the United States Department of Energy's National Nuclear Security Administration  
under contract DE-AC04-94AL85000.



# Conclusions



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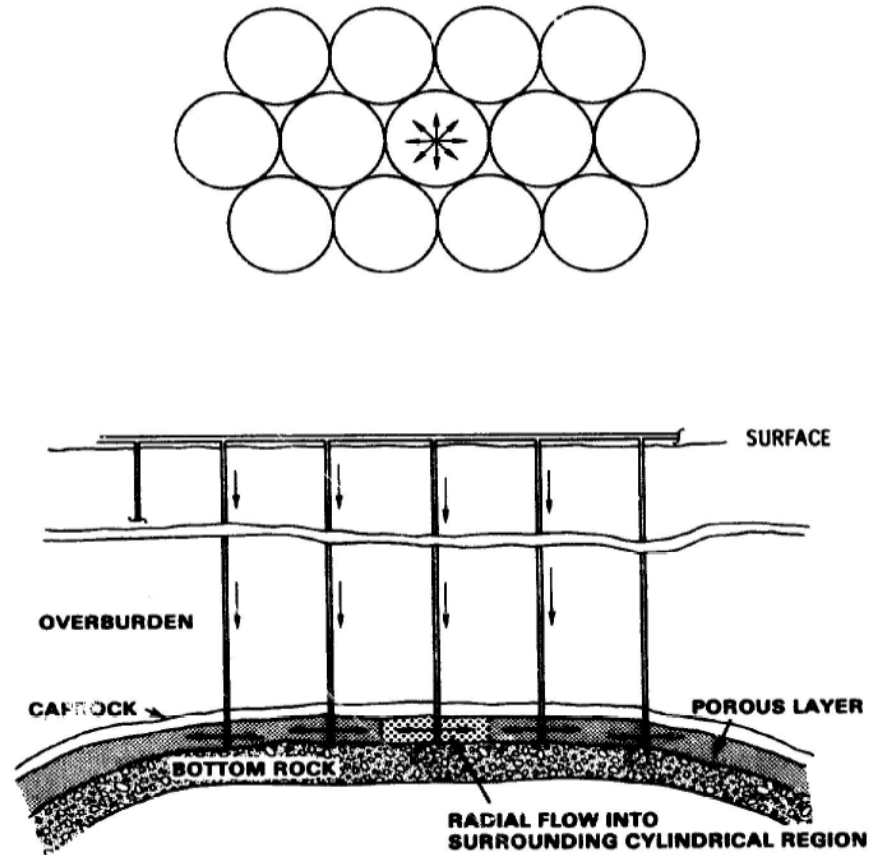
- **Permeability Variation Much More Important than Porosity Variation**
- **Procedure Can Quantify Differences Between Various Sets of Formation Parameters**
  - **Borehole Spacing, Number of Boreholes**
- **Borehole Arrays Will Be Investigated in the Future**



# Study geometry views



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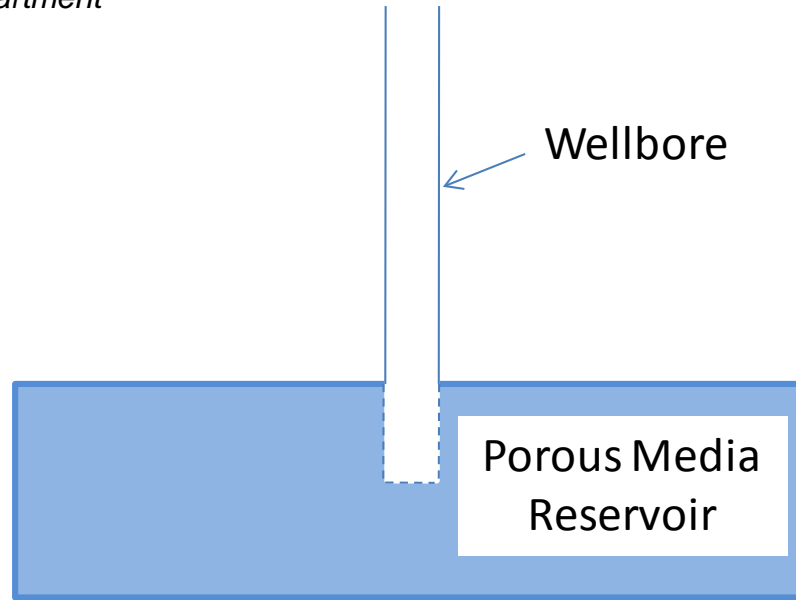


CAES Borehole Schematic (from Smith and Wiles, 1979)

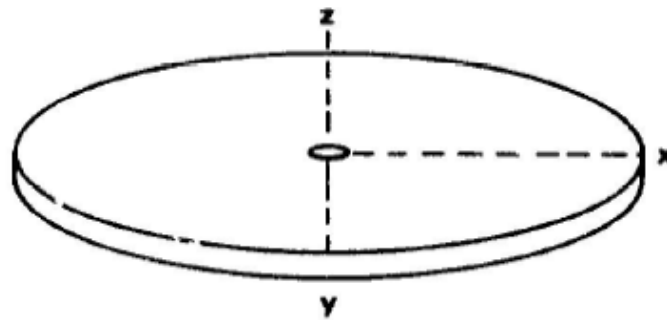
# CAES Borehole Study



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**Formation Radius Varies**



## Representative Borehole/Formation Geometry

# Study Parameters



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**Formation Height – 100 ft high**

**Depth – 2000 ft**

**Borehole Diameter – 7 inches**

**Partial Completion**

**Permeability – 100 mD to 2000 mD (500 mD Nominal)**

**Porosity – 0.1 to 0.3 (0.2 Nominal)**

**Formation Radius - Varies**

**Based on  $P_{\max}$  and  $P_{\min}$  Values**

**Mass Flows**

**See Cycle**

**Two-Phase Characteristic Curves**

**Leverett J-Function Scaling**

# Air Pressure Considerations



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**$P_{\min}$**

**Turbine Inlet Pressure = 45 bar (4.5 MPa)**

**Pressure Drop to Surface = ~5 bar (0.5 MPa)**

**Minimum Borehole Pressure = 5.0 Bar**

**$P_{\max}$**

**0.6 x Lithostatic = 8.4 MPa**

**Maximum Borehole Pressure = 8.4 MPa**



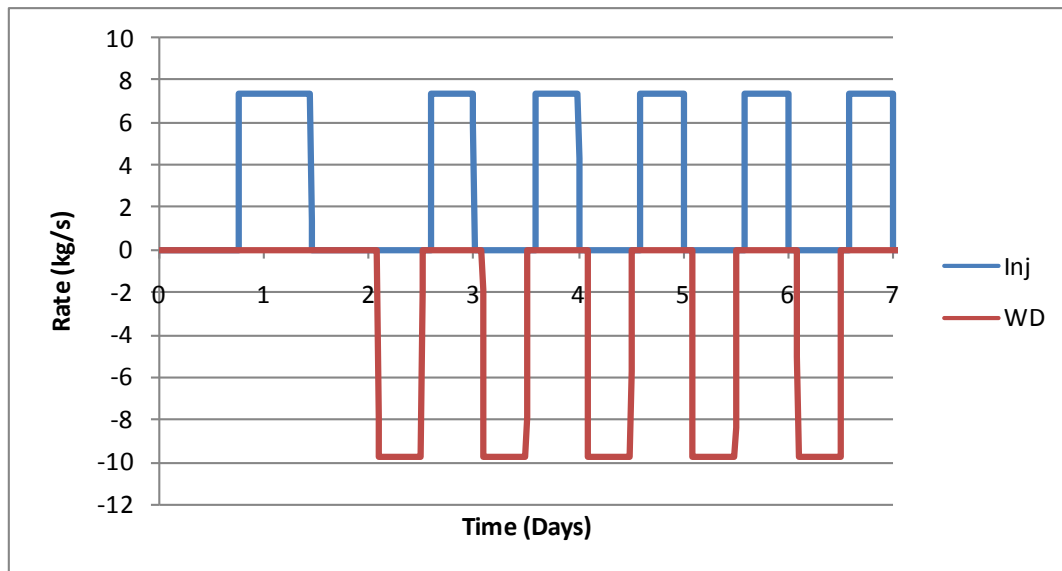
# Pressure Cycling Model



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## CAES Cycle

- Based on Smith and Liles (1979)
- 10% Mass Cycled Per Week
- 40% Air Added on the Weekend
- Mass Rates Based on Available Mass
  - » Function of Formation Radius, Porosity, Gas Saturation



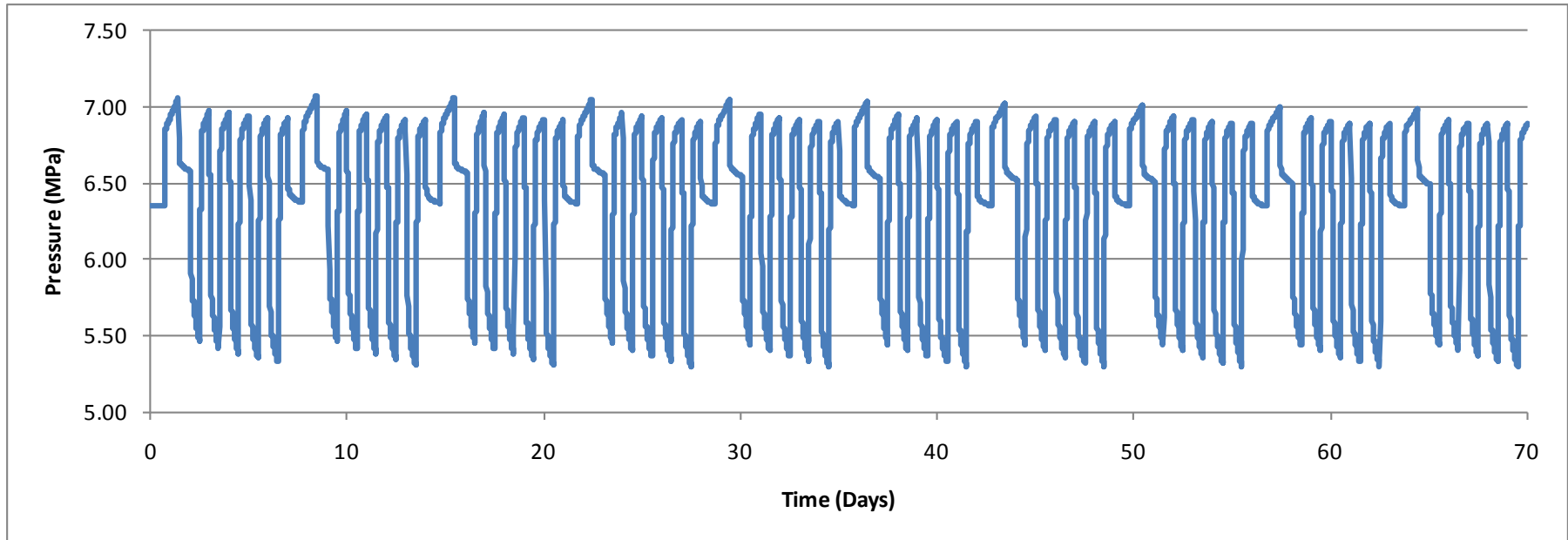
**Typical Cycle**

# Borehole pressure



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## Typical Cycle Results for Borehole Pressure – After Formation of Bubble



# Procedure for Given Permeability and Porosity



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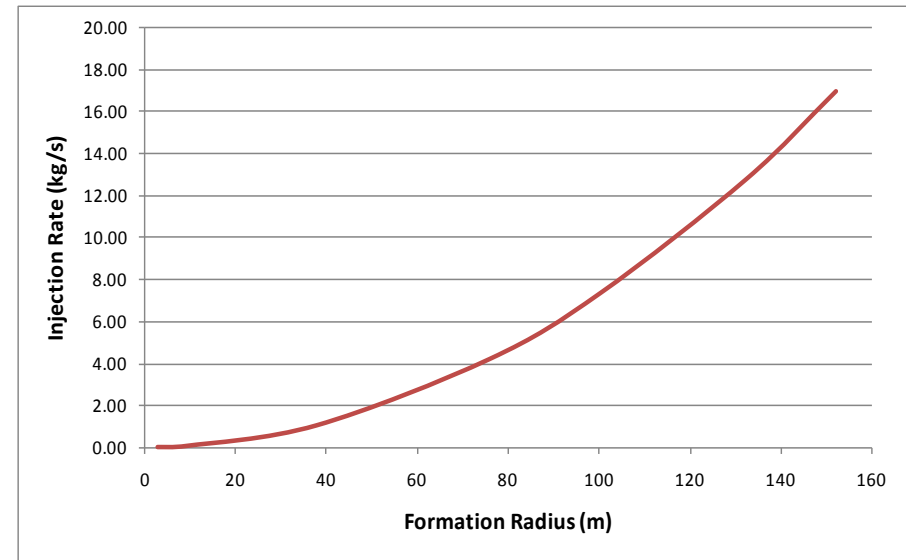
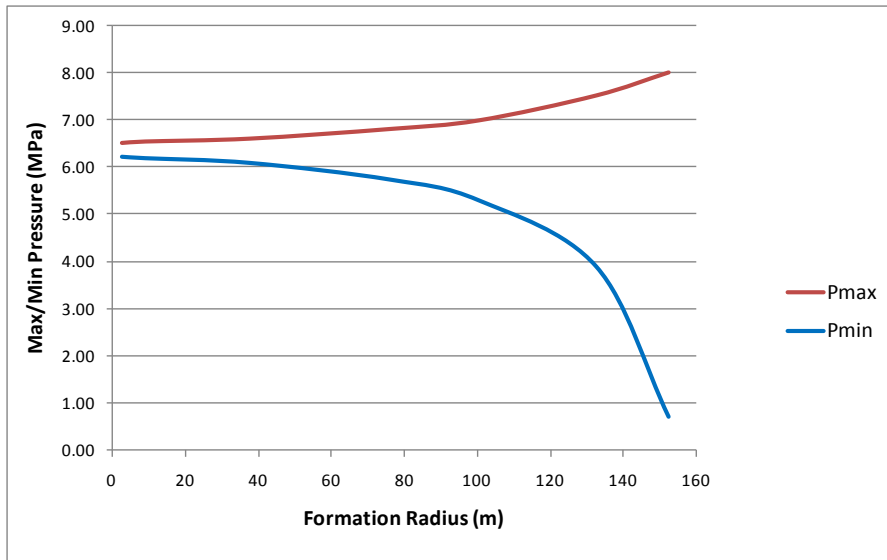
- **Formation Radius Increase**
  - **Mass Rates Increase – Larger Available Mass in Formation**
  - **$P_{\max}$  Increases**
  - **$P_{\min}$  Decreases**
- **Optimum Formation Radius and Mass Flow Rate When  $P_{\max}$  and/or  $P_{\min}$  Met**

# CAES Borehole Study



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## Typical Results ( $k = 500$ mD, $\phi = 0.2$ )



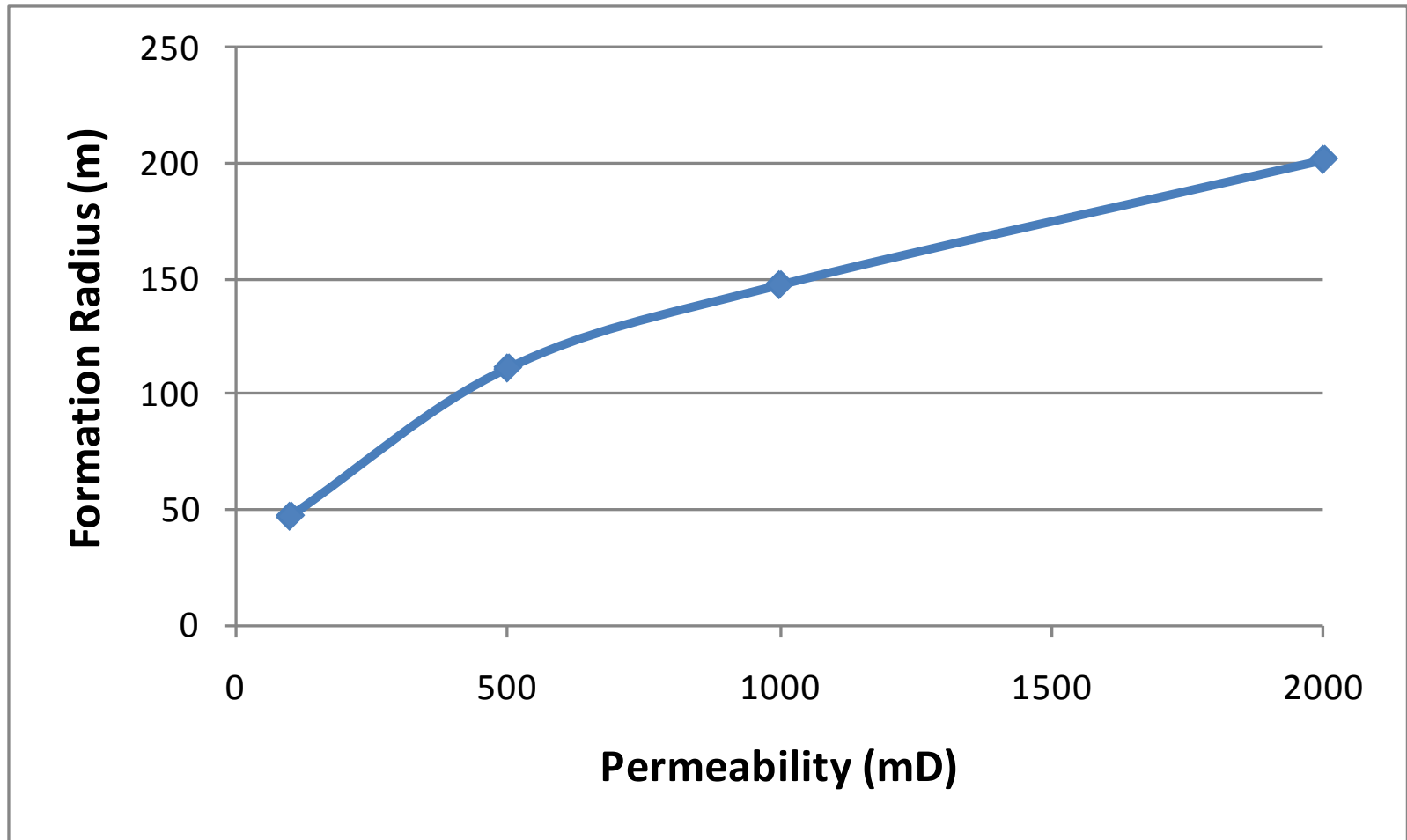
**Optimum Formation Radius = 111 m Based on  $P_{\min}$**



# Permeability Variation



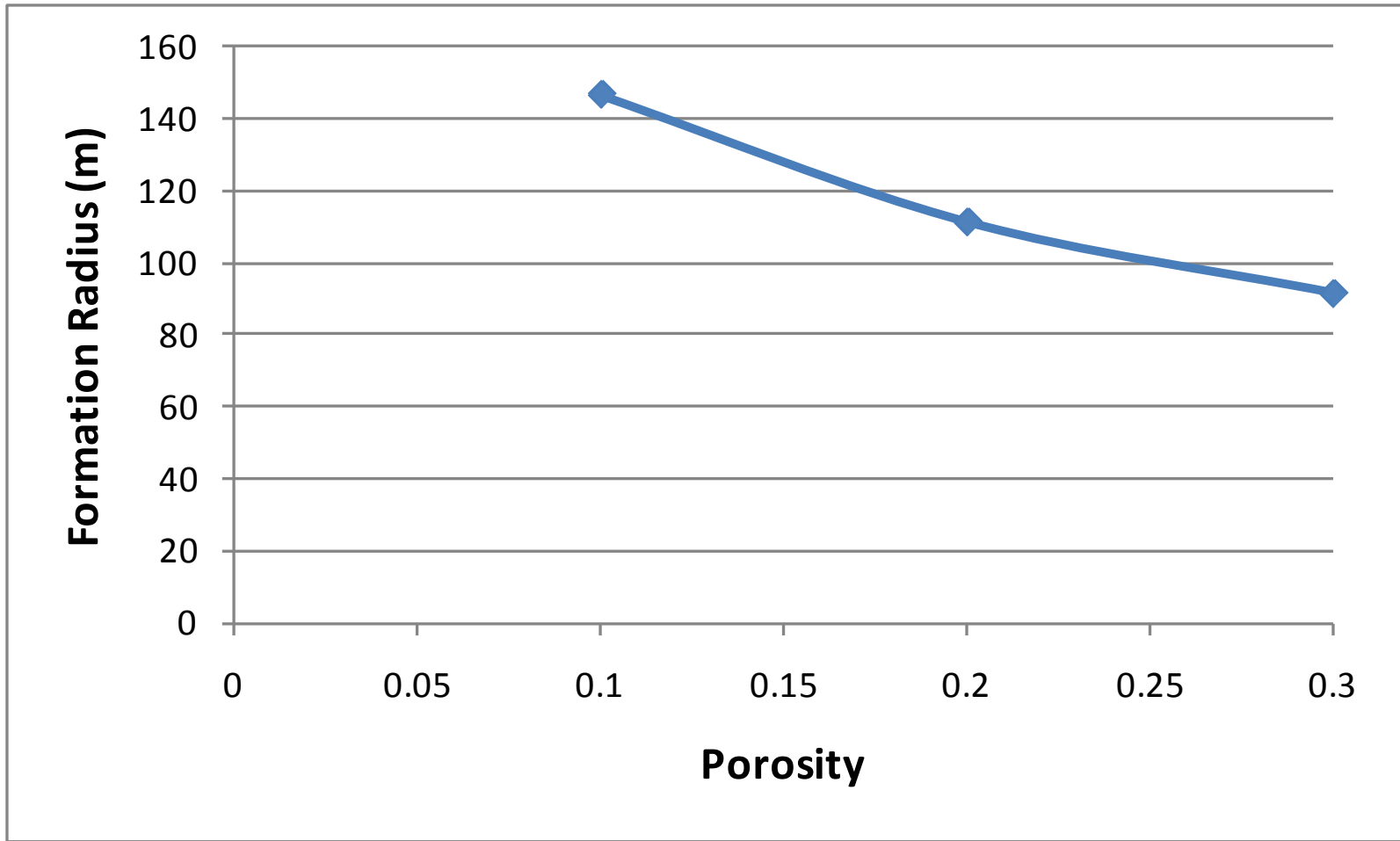
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# Porosity Variation



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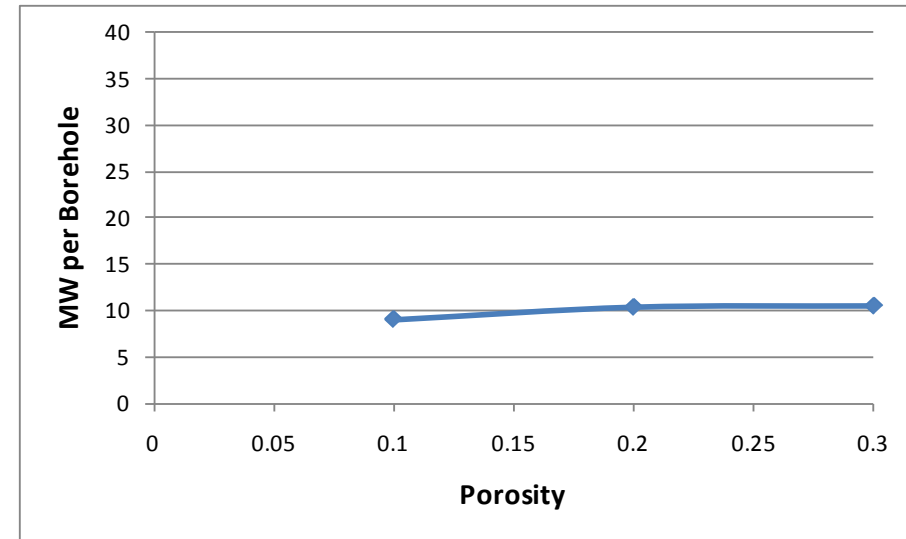
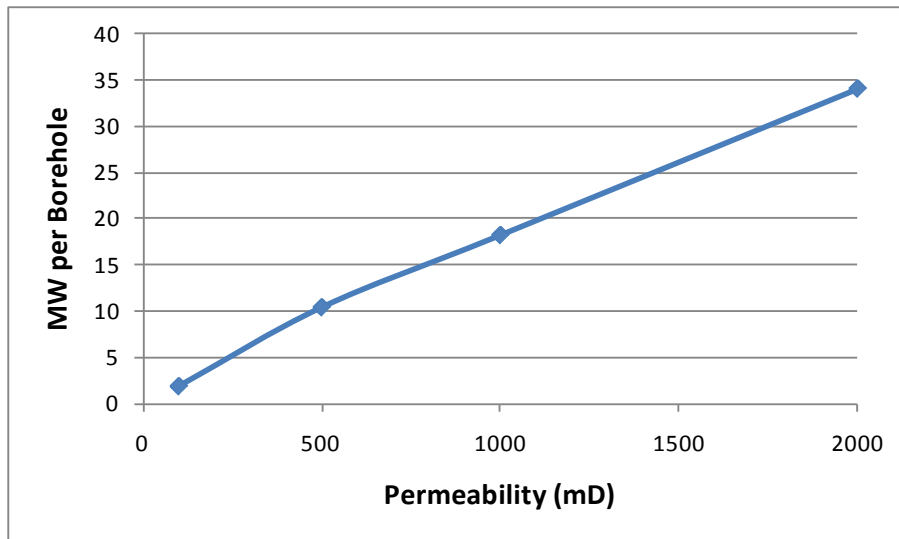
# Permeability/Porosity vs. Power



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## Using Typical Turbine Parameters

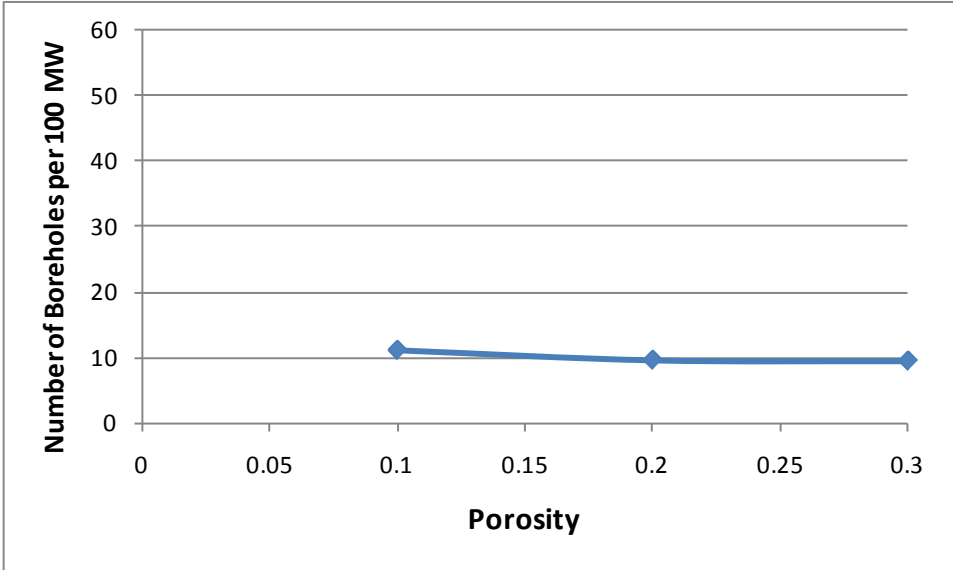
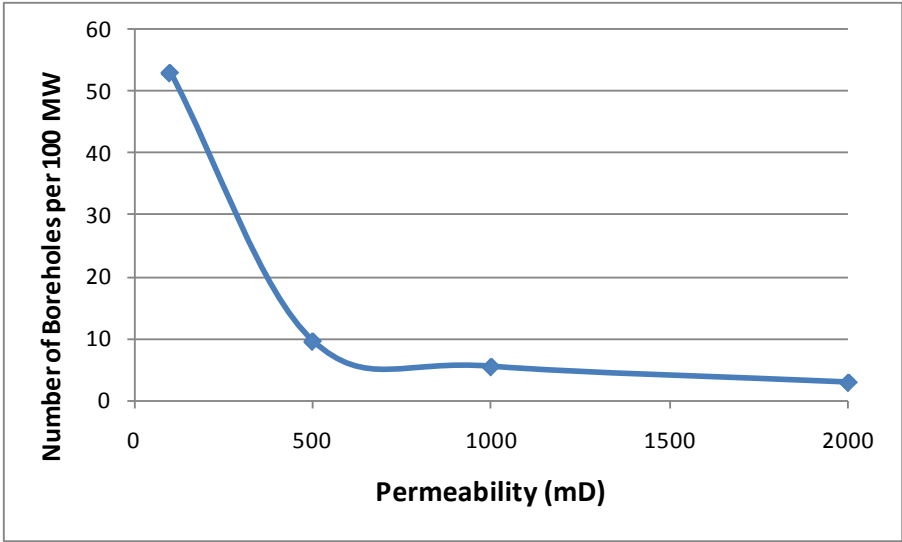
Based on Iowa CAES Power Density (~5 MW/m<sup>3</sup>) Scaled by Formation Pressure (Succar, 2008)



# Number of Boreholes vs Permeability & Porosity



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# Conclusions



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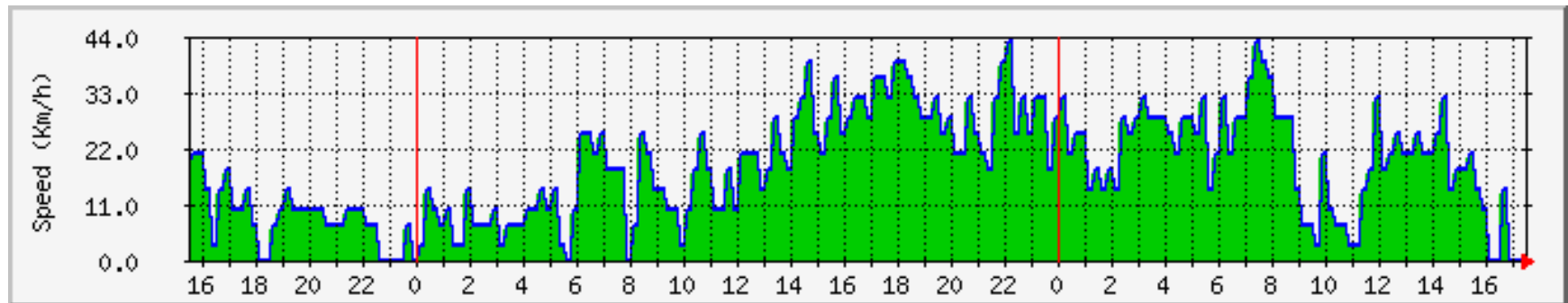
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# Material Degradation (T-M-C-H effects) Due to Cyclic Loading



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## *Experimental Deformation of Salt in Cyclic Loading* SJ Bauer and ST Broome

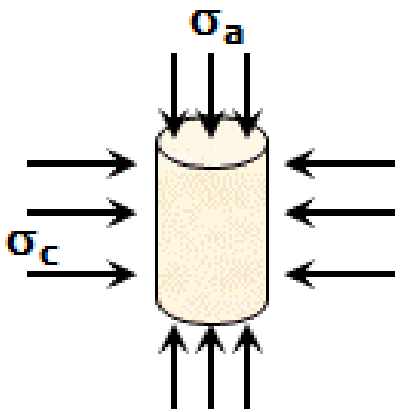


**Hourly fluctuations in wind speed could translate to frequent pressurization/depressurizations of salt caverns**

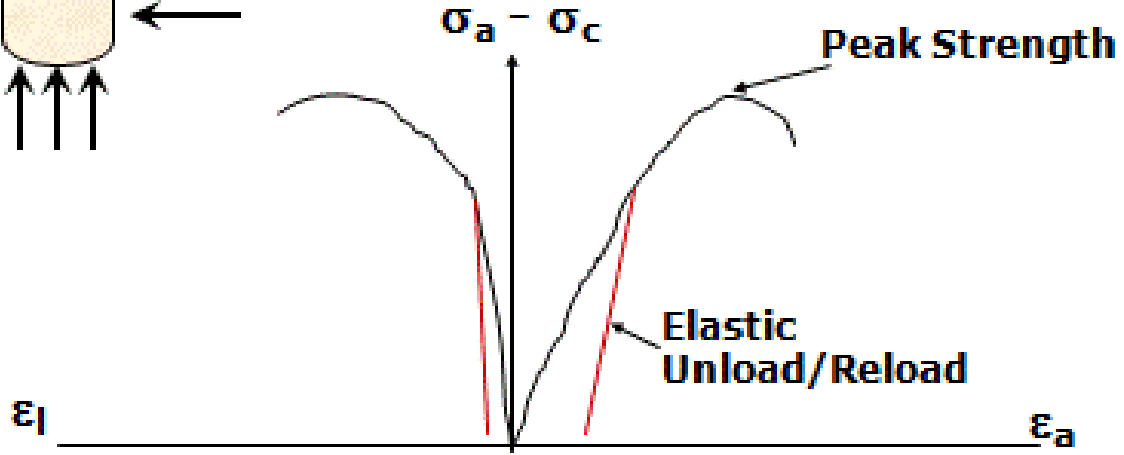


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## Quasi-Static Loading



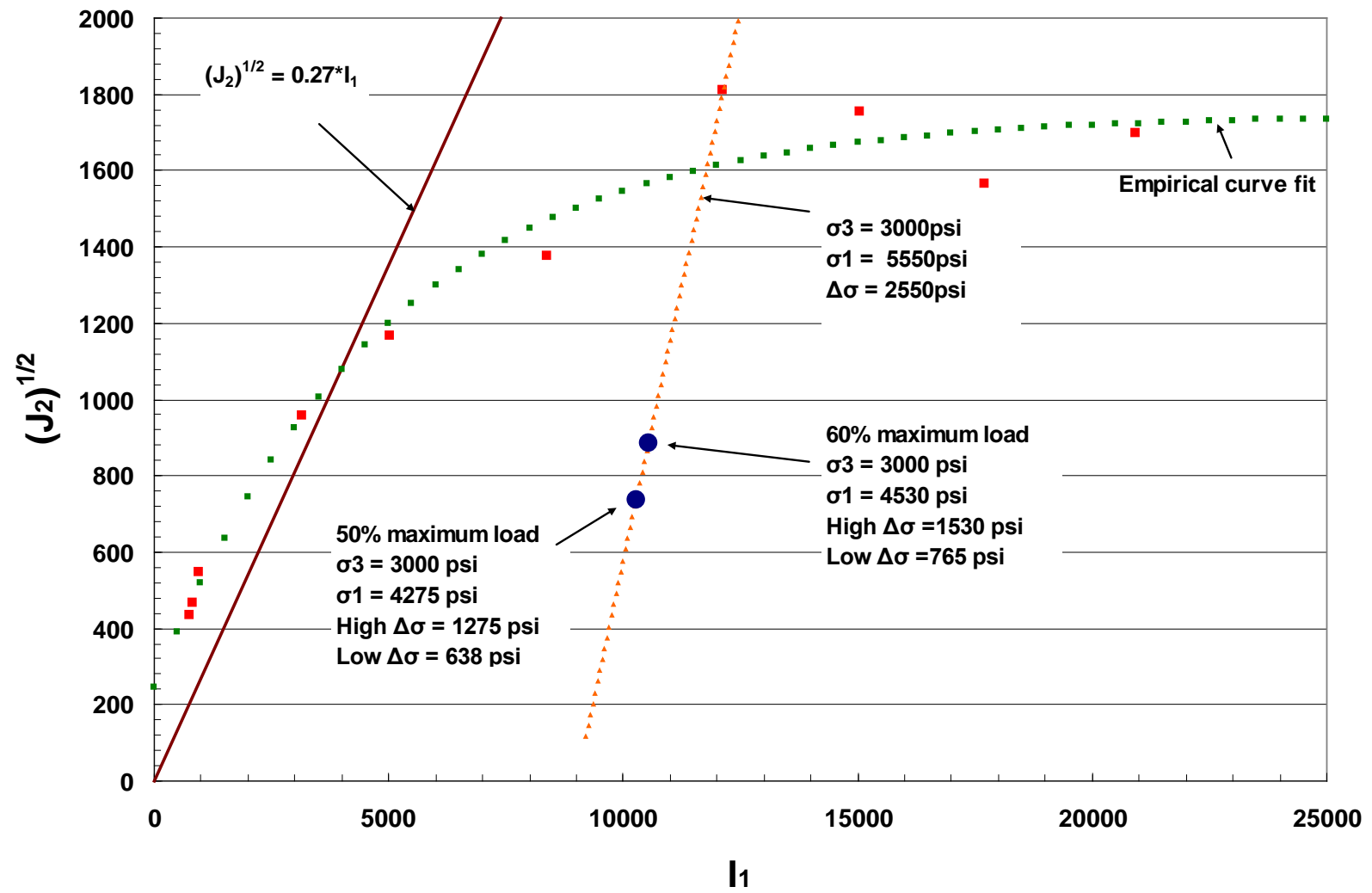
- Apply constant  $\sigma_c$
- Increase  $\sigma_a$  until specimen fails
- Measure strains



# Dilatant behavior of salt determined from quasi-static tests and stress states for this study



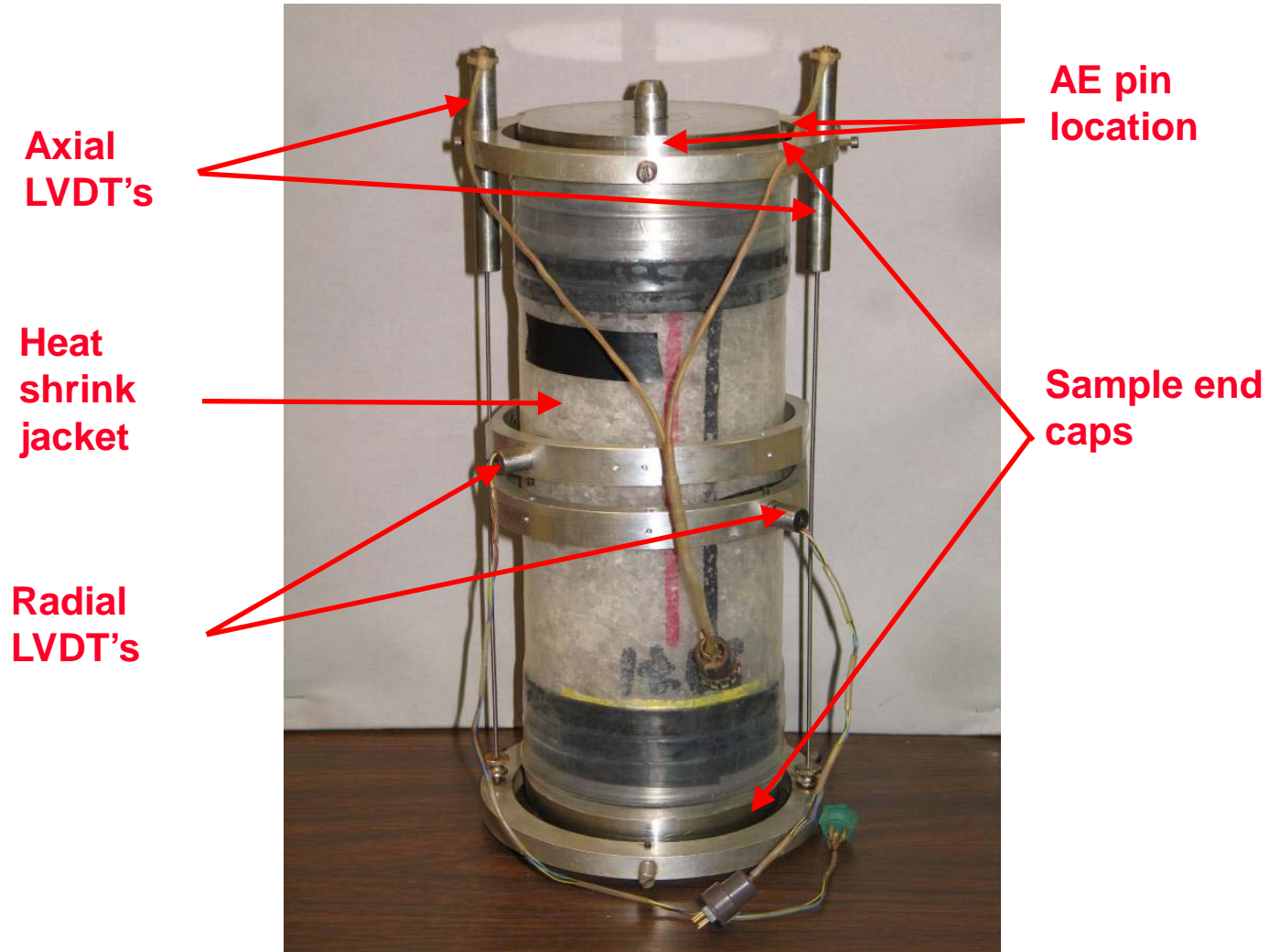
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# Test assembly



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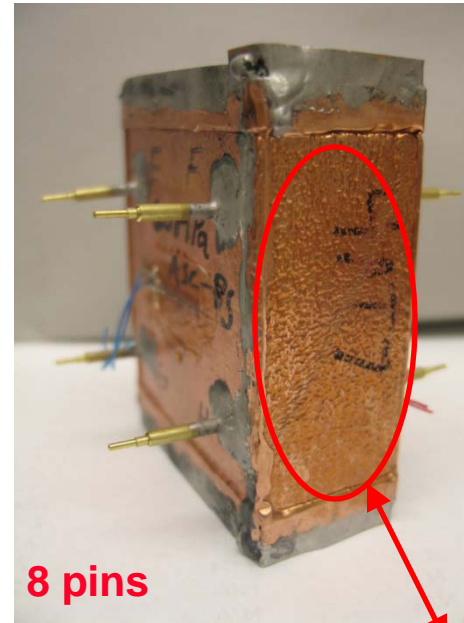




# Acoustic Emissions System

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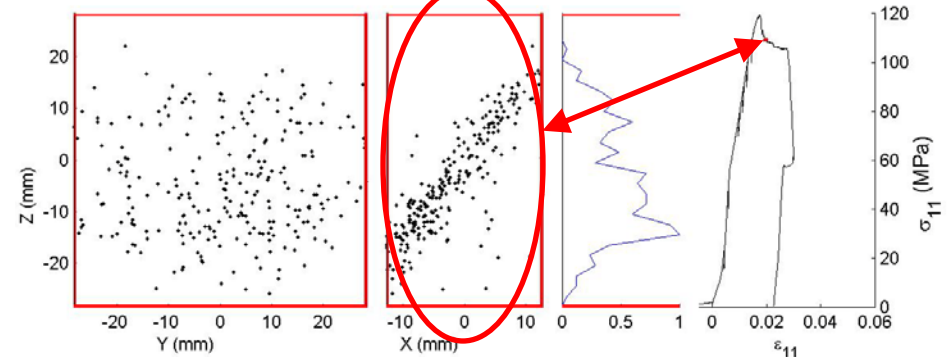
- Sample rates up to 25 MHz
- Typically acquire 3000 samples/event
- Tailor a discriminator to only sample events of a given criteria
- 60 dB amplifier
- Location of events is possible with many pins



8 pins



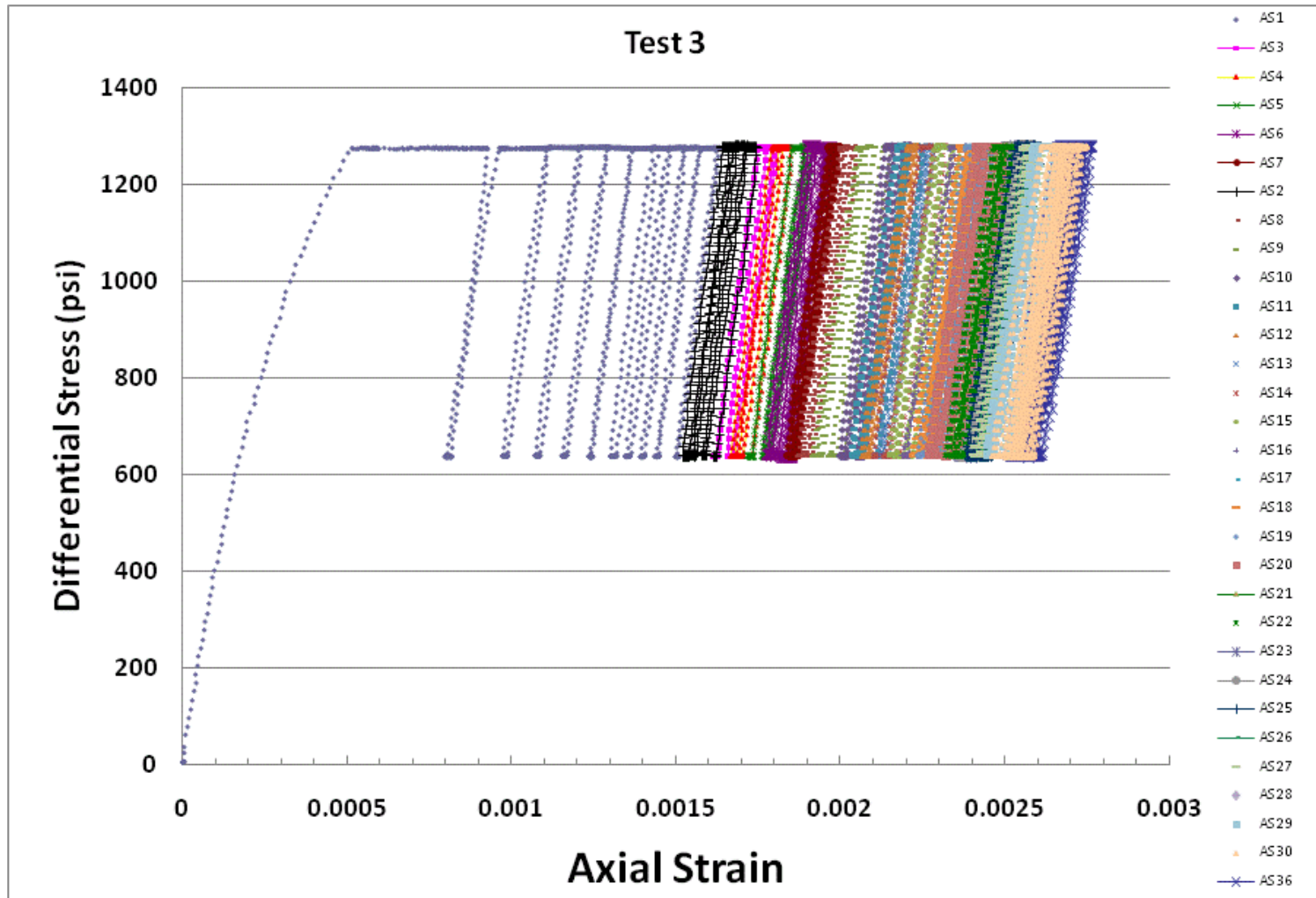
14 pins



# Differential stress versus axial strain, Test 3.



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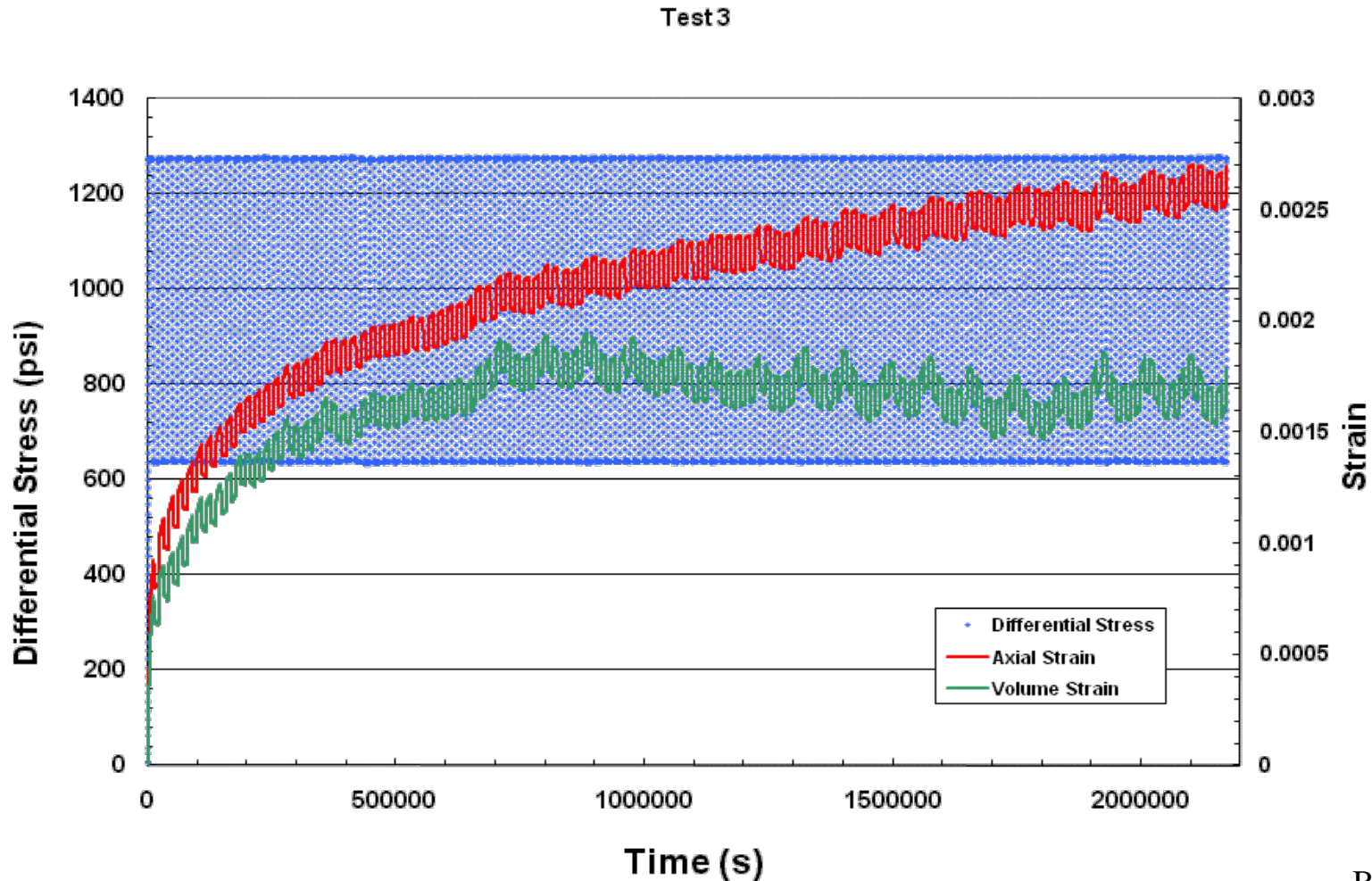


# Test data

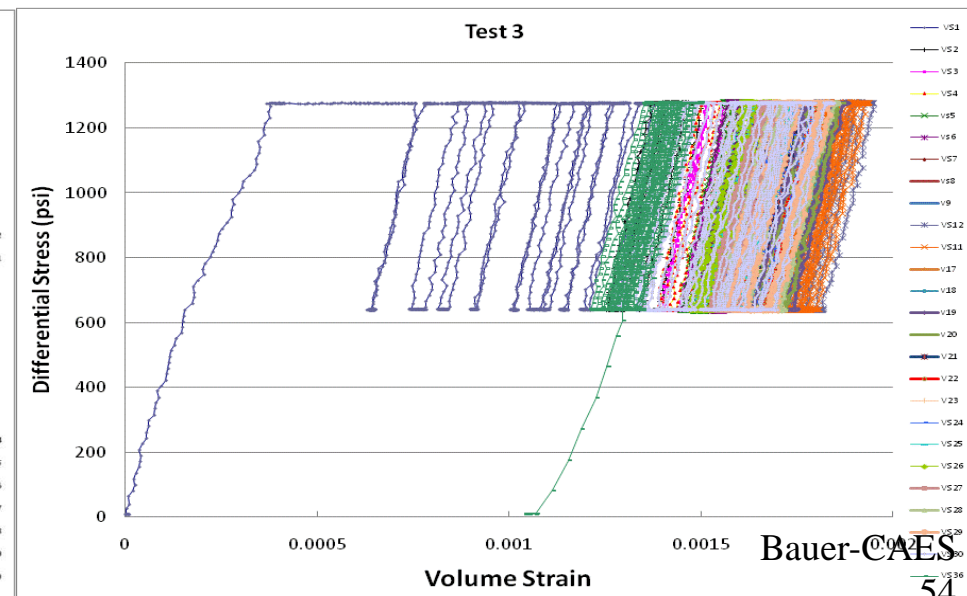
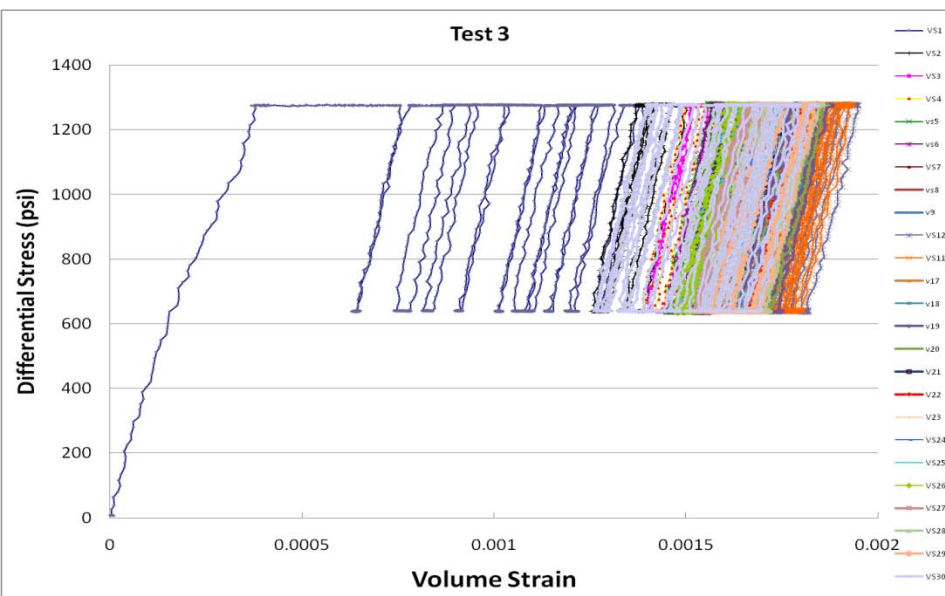
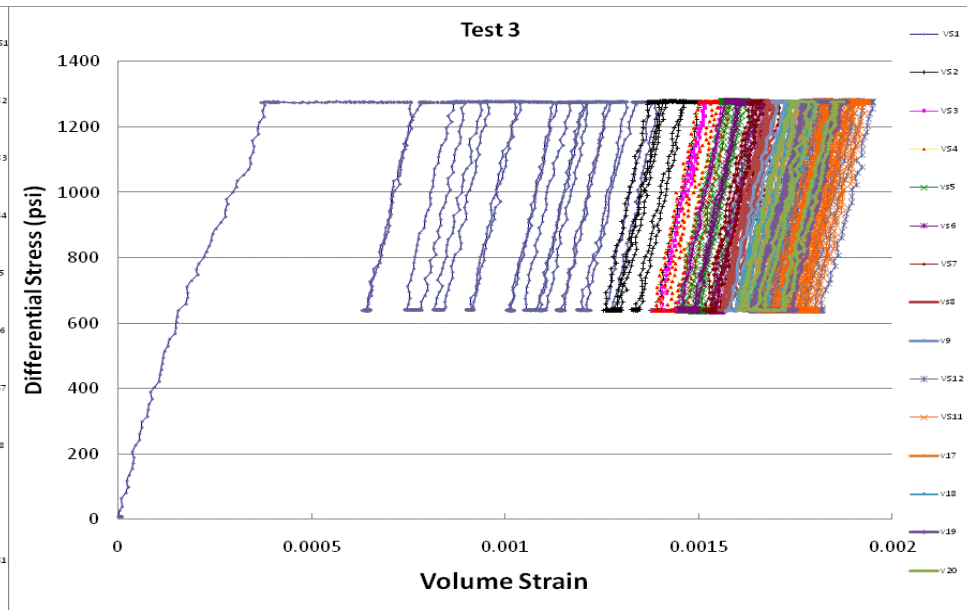
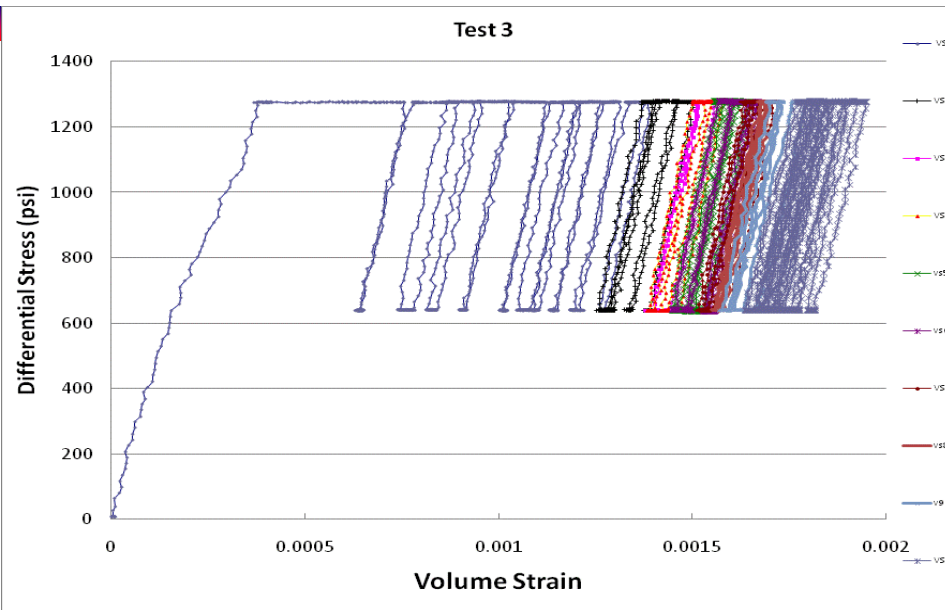


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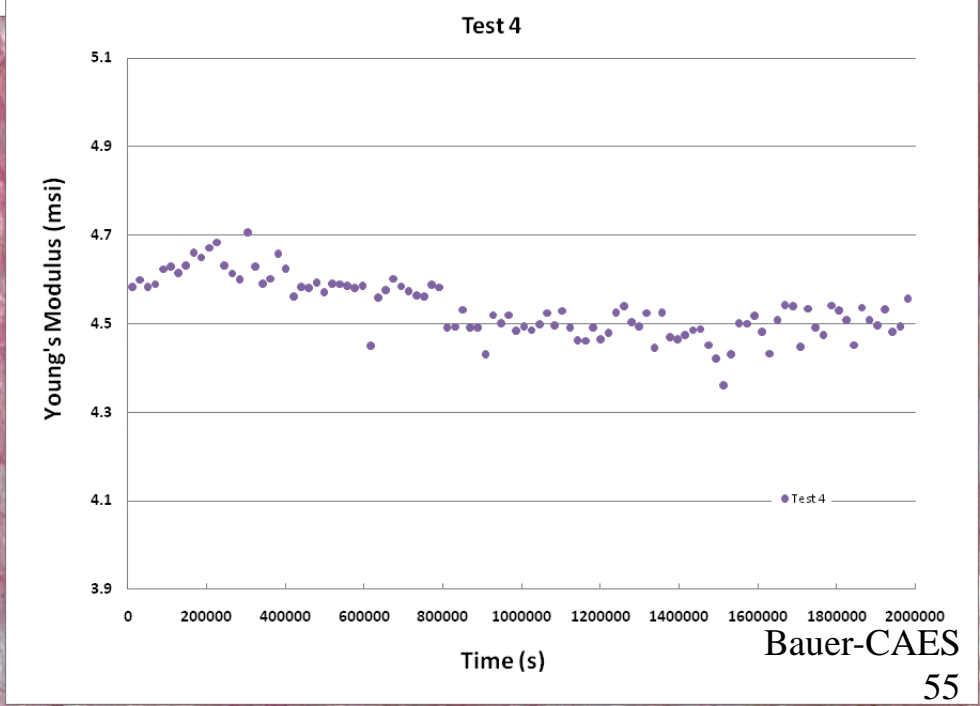
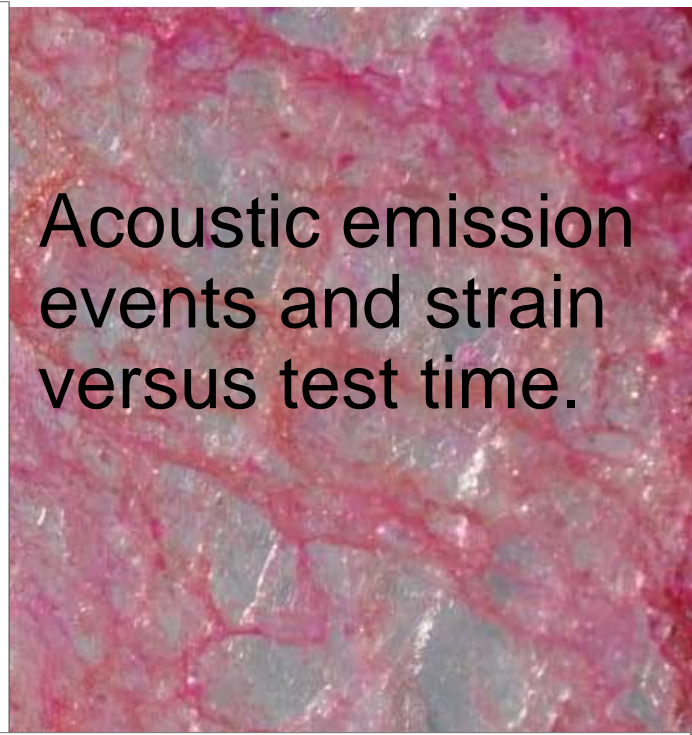
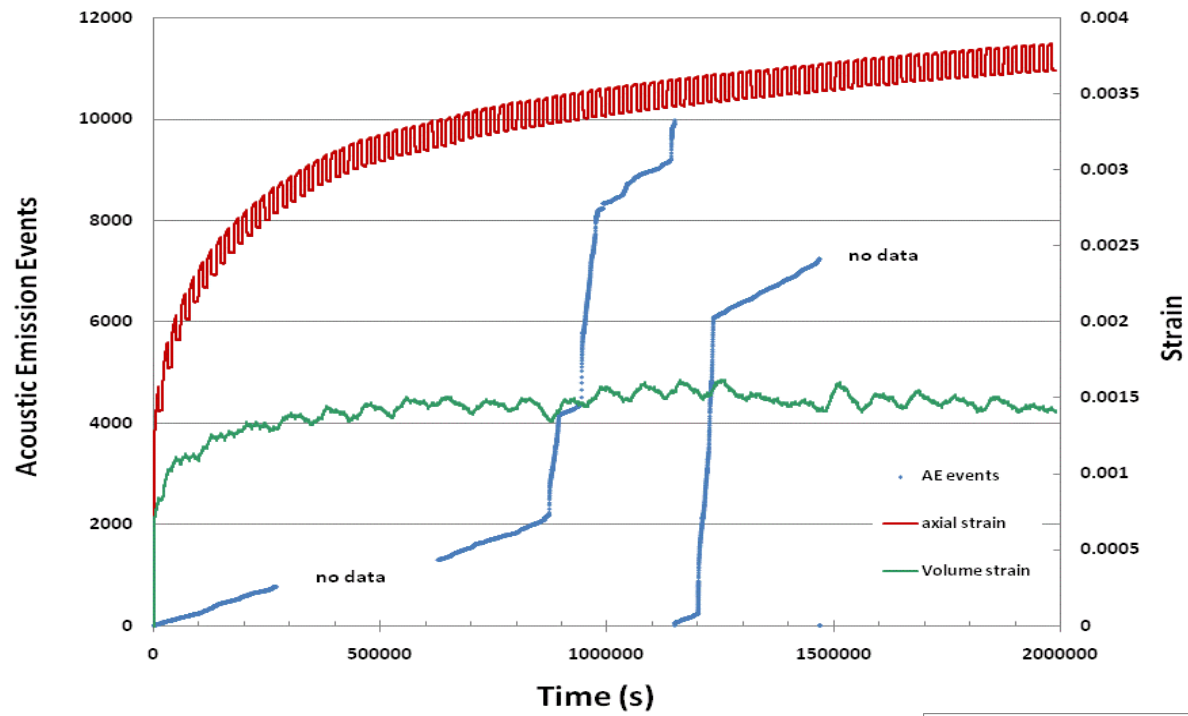
Differential stress, axial and volume strain versus time, Test 3.



# Differential stress versus volume strain, Test 3









# Concluding Comments



- Preliminary cyclic tests completed on salt
- Change in volume strain observed
- Young's Modulus changes observed
- Acoustic Emissions detected
- Cracks observed in thick sections
- Results consistent with previous work
- Implication that cyclic loading caused cracking at low differential stresses



**thanks**

**Questions?**