

Webinar

**Biogenic Carbon Tracking and
Measurement in Co-processing of Biogenic
Feeds in Petroleum Refineries**

September 27th, 2023



Presenters:

Dr. Zhenghua Li, Los Alamos National Laboratory

Dr. Sophie B. Lehmann, Pacific Northwest National Laboratory



This Webinar (9/27/2023)

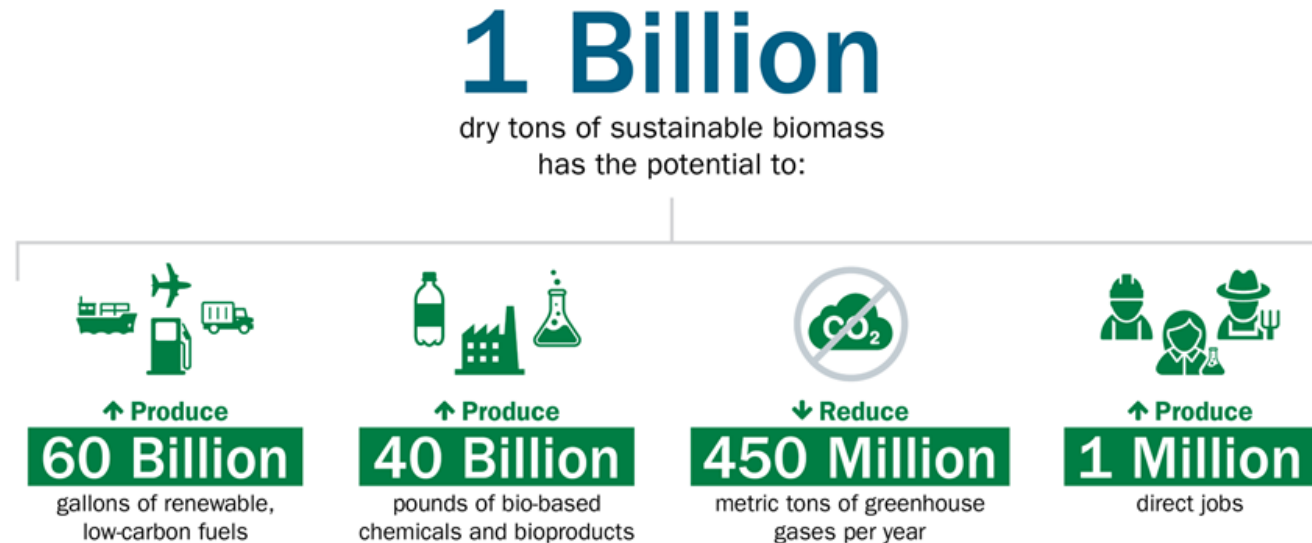
- Housekeeping
- • Overview of BETO
- Biogenic Carbon Tracking and Measurement by LSC ^{14}C and IRMS $\delta^{13}\text{C}$ 
- Laser Spectroscopy and Stable Carbon Isotopes for Tracking Biocarbon 
- Q&A

Last Week's Webinar (9/20/2023)

- Co-processing Fast Pyrolysis Bio-Oils and Hydrothermal Liquefaction Bio-crudes in Fluid Catalytic Cracking and Hydroprocessing in Refineries



- More GHG reductions, faster!
- Focusing on Sustainable Aviation Fuel (SAF) and other strategic transportation fuels
 - [BETO 2023 Multi-Year Program Plan](#)
 - [Sustainable Aviation Fuel Grand Challenge Roadmap](#)
 - [U.S. National Blueprint for Transportation Decarbonization](#)
- Unlocking the potential of the full range of renewable carbon resources



Bioenergy Technologies Office Program Areas

Renewable Carbon Resources



Conversion Technologies



Systems Development and Integration



Data, Modeling, and Analysis



- Support 3 billion gallons SAF by 2030
- Support 35 billion gallons SAF by 2050
- Cost-effective SAF and other strategic fuels with at least 70% greenhouse gas reductions
- Decarbonization of chemicals

Today's Presenters





Dr. Zhenghua Li
Senior Scientist
Los Alamos National Laboratory



Dr. Sophie B. Lehmann
Scientist
Pacific Northwest National Laboratory

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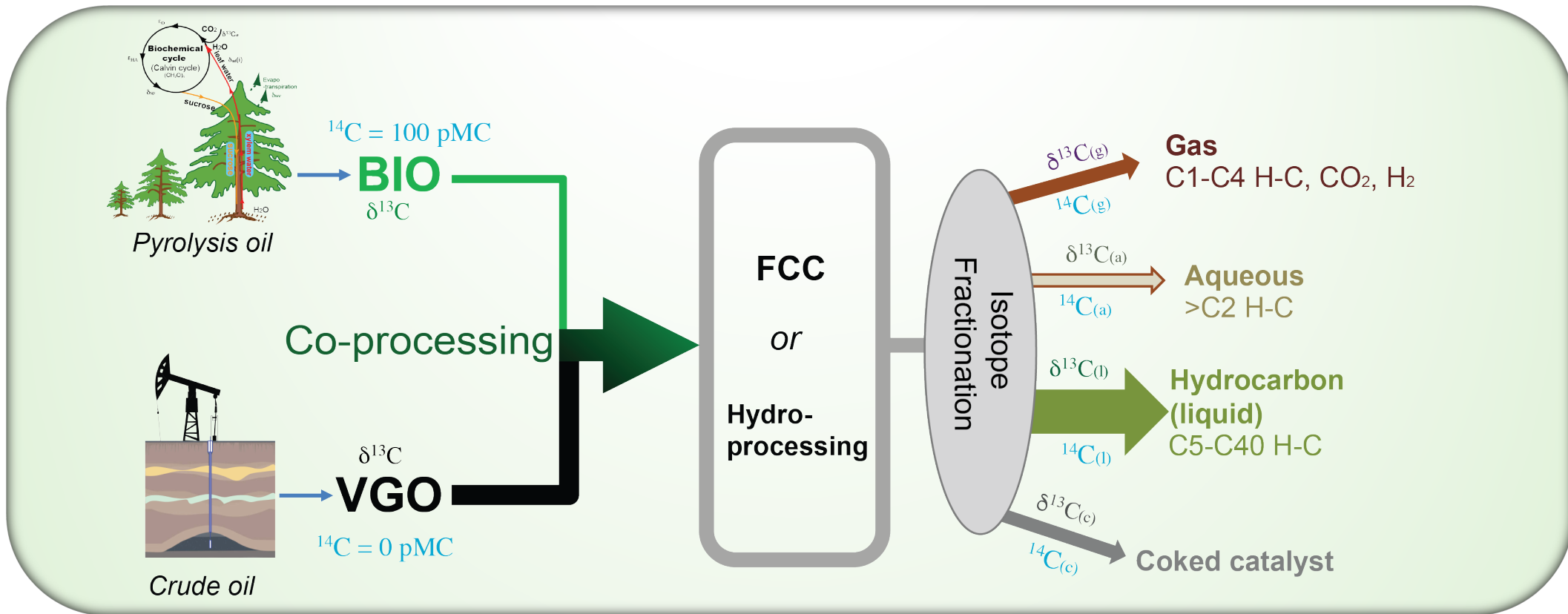
Last Week's Webinar (9/20/2023)

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Biogenic Carbon Tracking and Measurement by LSC ^{14}C and IRMS $\delta^{13}\text{C}$ in Co-processing

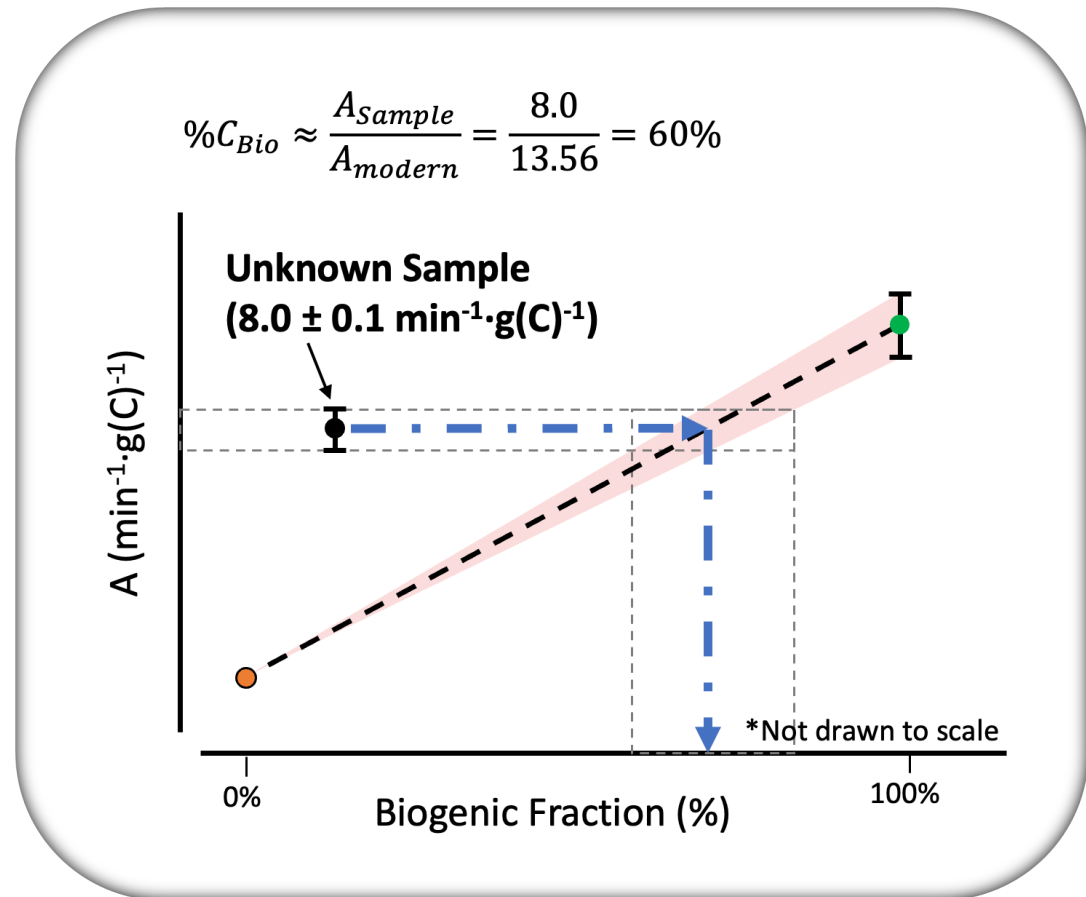
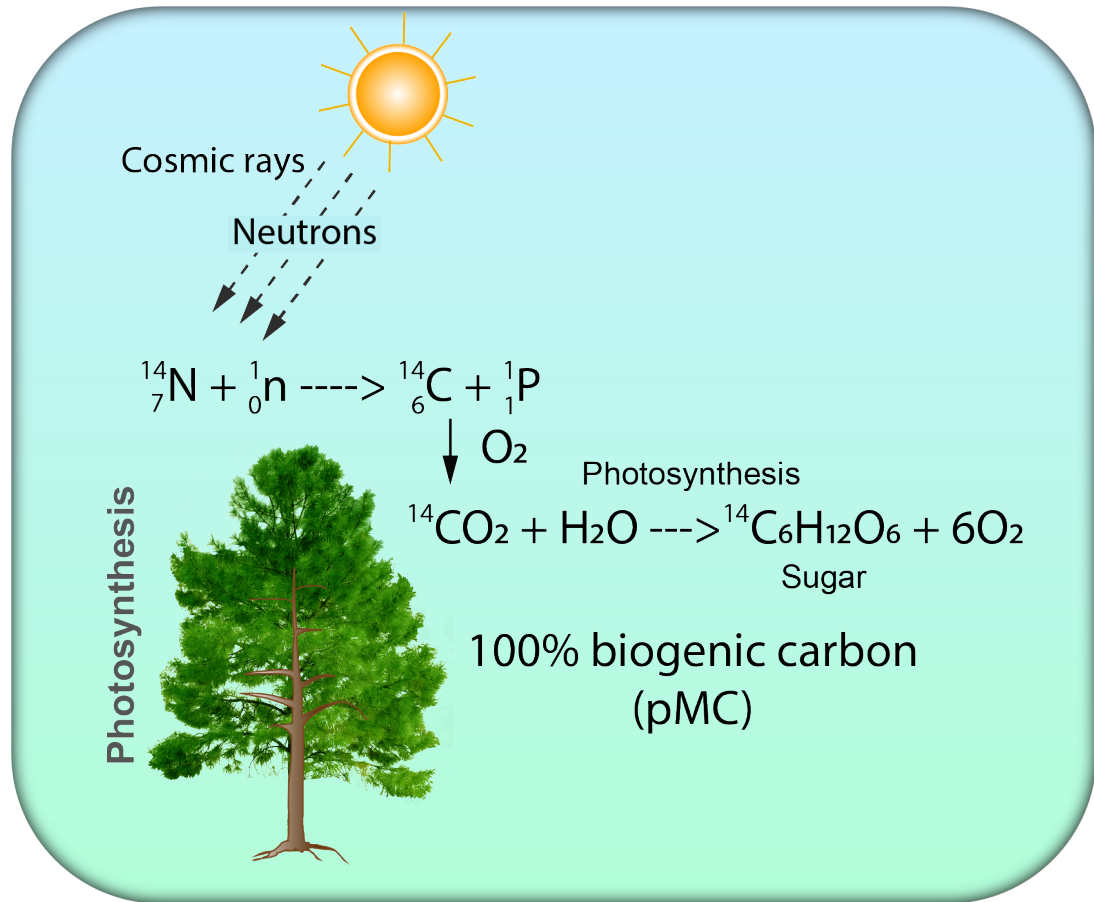
LSC: Liquid Scintillation Counting; IRMS: Isotope Ratio Mass Spectrometry



pMC: Percent Modern Carbon ^{14}C : radiocarbon isotope $\delta^{13}\text{C}$: stable carbon isotope ratio

- Petroleum contains ^{12}C and ^{13}C but not ^{14}C . Modern vegetation contains ^{12}C , ^{13}C , and ^{14}C .
- Modern vegetation contains 100 pMC ^{14}C (100% biogenic carbon).
- Fossil fuel contains 0 pMC ^{14}C due to ^{14}C decay with half-life=5,730-year (0% biogenic carbon)

Radiocarbon ^{14}C Formation and Determination of Biogenic Carbon by Measuring Radioactive Decay Activity (A)



- ^{14}C is produced by cosmic rays and incorporated in plants through photosynthesis
- Biogenic C content is determined by measuring ^{14}C activity (A) in plants.
- Petroleum A is zero, while modern plant A is 13.56 (100% biogenic carbon)

LANL Liquid Scintillation Counting (LSC)

Quantulus™ GCT 6220



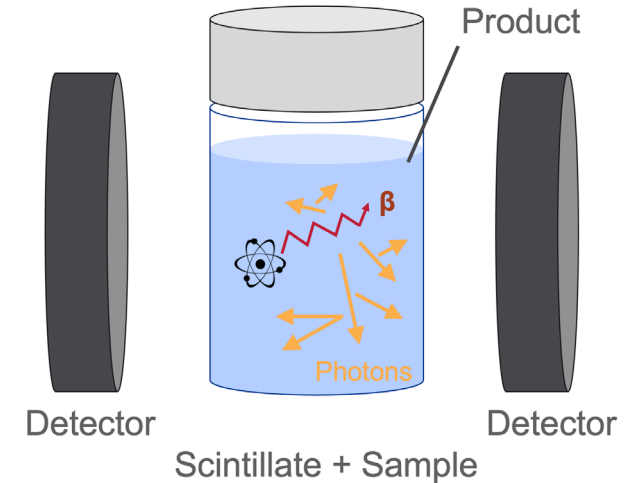
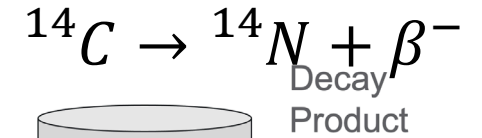
Quantulus™ 1220



HiDex 300SL



Small footprint



^{14}N : nitrogen isotope.

β^- : beta particle emission

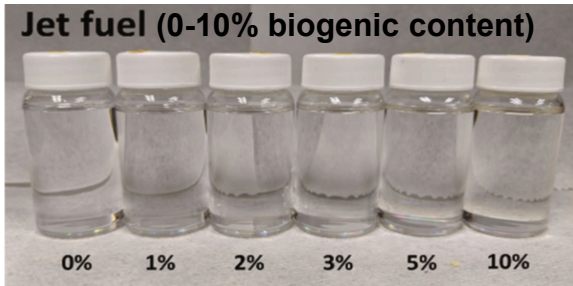
LSC Analysis

- Mix sample with scintillate
- Scintillate produces photons when decay occurs
- Number of photons emitted are related to energy of decay products
- PMTs detect amount of photons

LSC Complications

- Scintillate can produce photons semi-randomly (Background)
- Photons “blocked” by color and/or chemical makeup of sample (Efficiency)

1. Near-clear fuels (petroleum jet fuel blended with SAF)

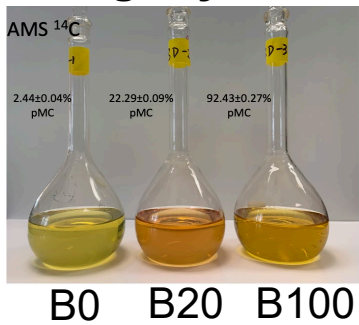


Direct LSC Measurement

Optimized data range, counting time, and data interpretation. Achieved +/-0.4 % points relative to ASTM 6866 (^{14}C AMS) in the range of 1-10% biocarbon.
Lee & Li et al 2022 (Fuel)

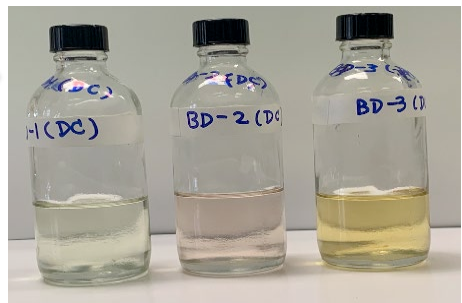


2. Lightly colored fuels (petroleum diesel and petroleum diesel/biodiesel blend, and B100 biodiesel)



Decolorization

Adsorbents



LSC Measurement

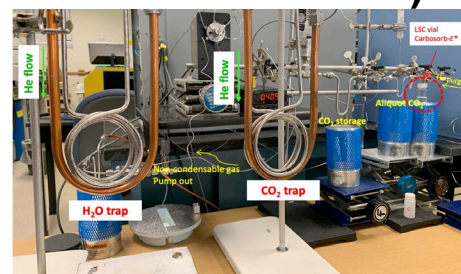
Decolorized fuels can be analyzed in LSC, but decolorization reduced biogenic carbon.
Lee & Li et al 2022 (Energy & Fuel)

3. Dark fuels, and solids (bio intermediates and feeds)



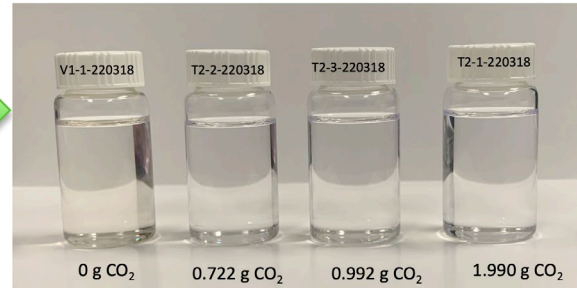
CO₂ Conversion

Oxy-combustion



CO₂ capture

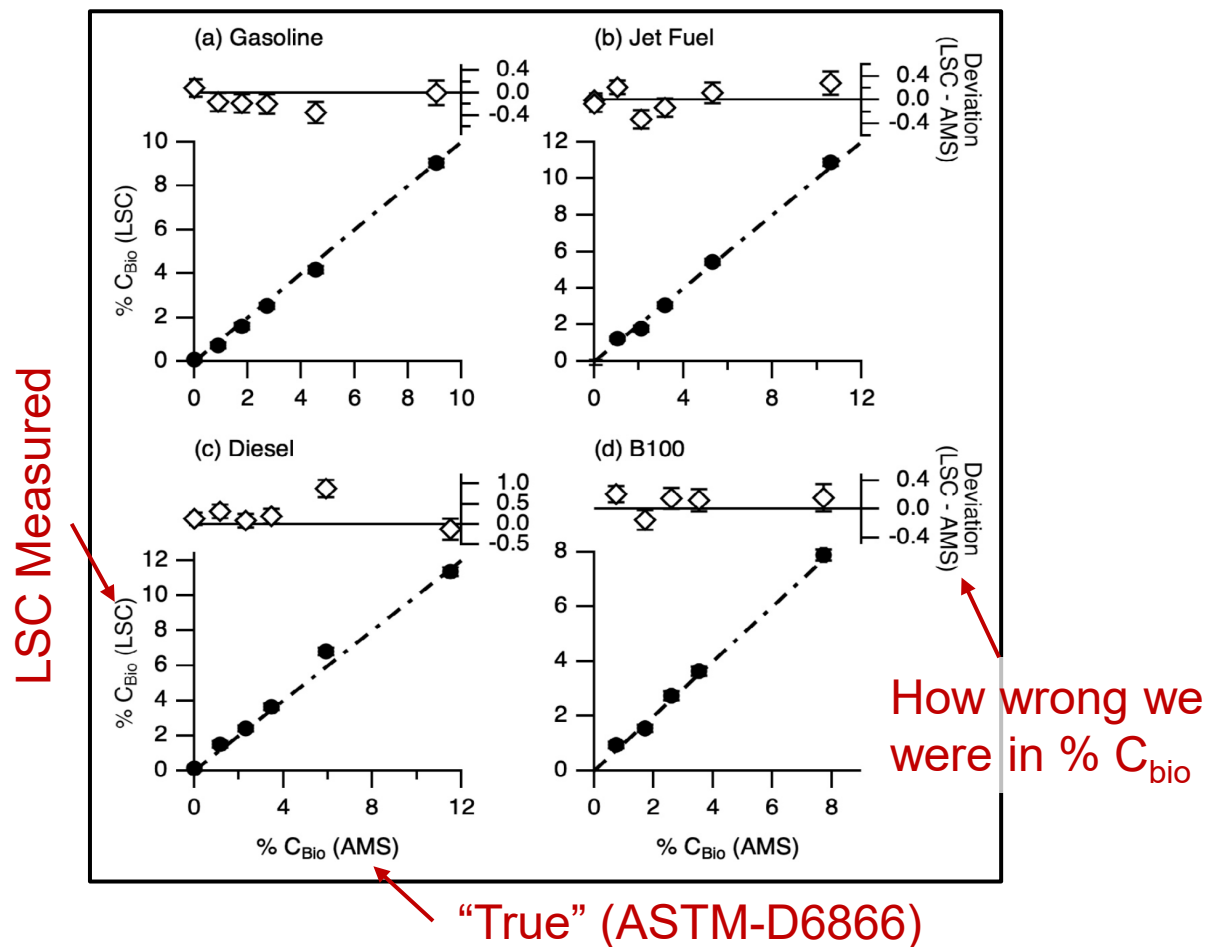
In absorbent/scintillate



LSC Measurement

Direct LSC measurement of near-clear samples of petroleum fuels blended with biofuels works

Correlation between LSC and AMS

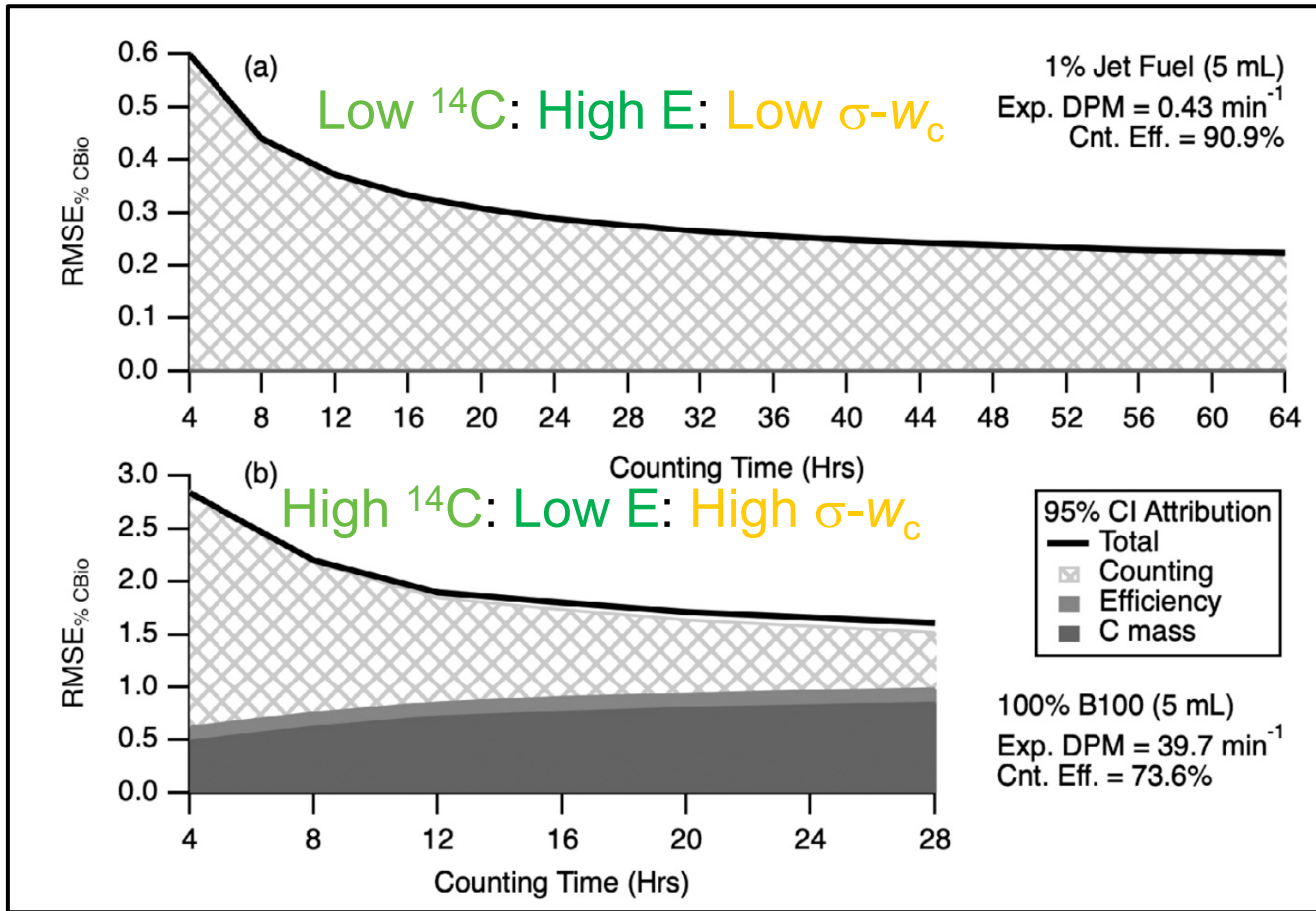


Direct Measurement Method

- “No Prep” approach: no sample preparation is needed, mixing fuel with scintillator directly.

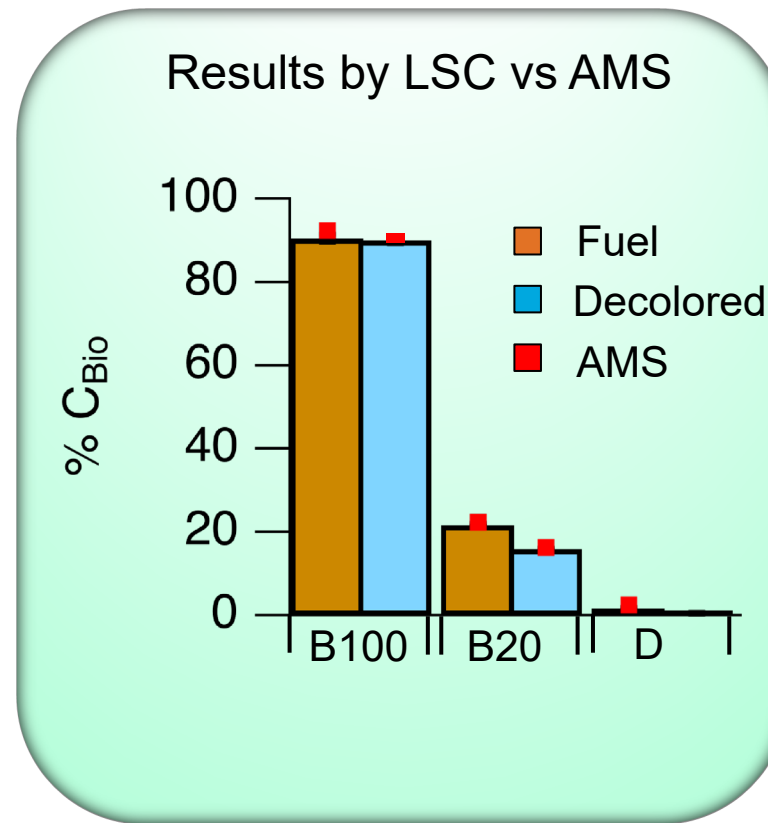
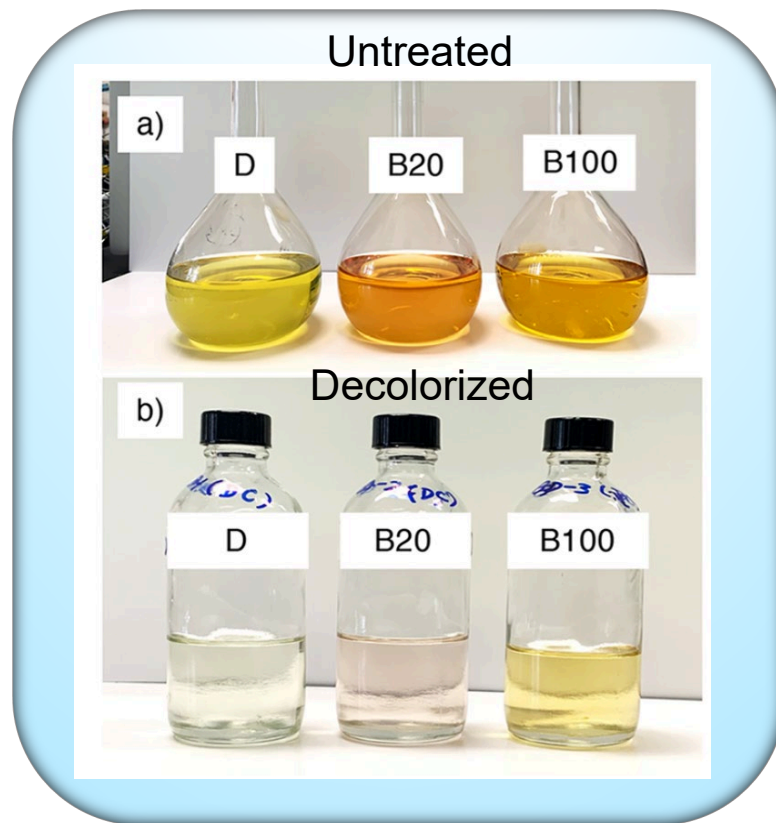
Results

- Very small errors
- Accuracy better than 0.4%
- Counting Efficiency > ~80%



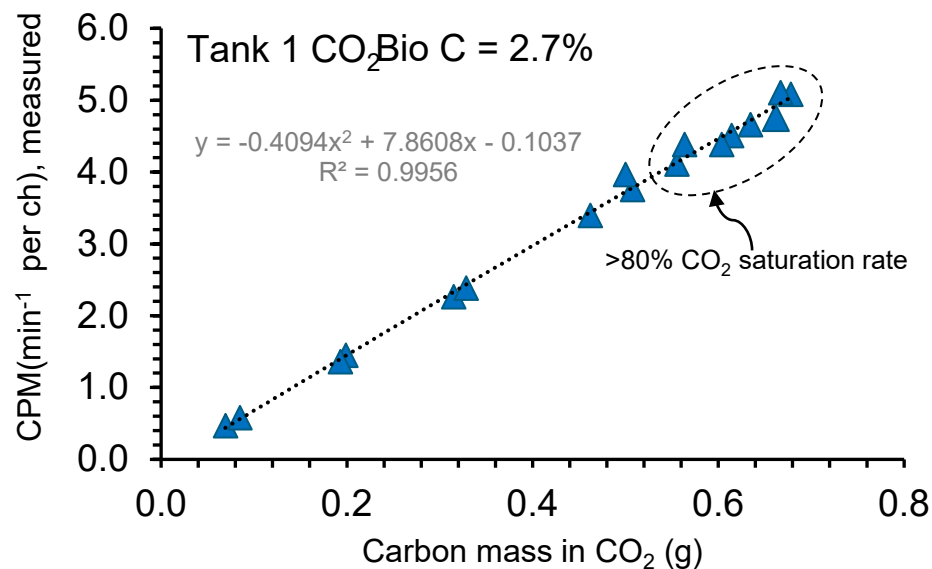
- Longer counting time, smaller uncertainty (RMSE: Root-Mean-Square Error).
- Uncertainty depends on the following fuel properties
 - Low biogenic carbon and high counting efficiency results in low uncertainty
 - High biogenic carbon and low counting efficiency results in high uncertainty
 - Higher carbon fraction means higher ^{14}C and thus higher counting efficiency.
 - Darker samples exhibit less counting efficiency, thus larger uncertainty.
- Uncertainty for the high-efficiency sample is negligible.

Lee & Li et al. 2022 (Fuel)

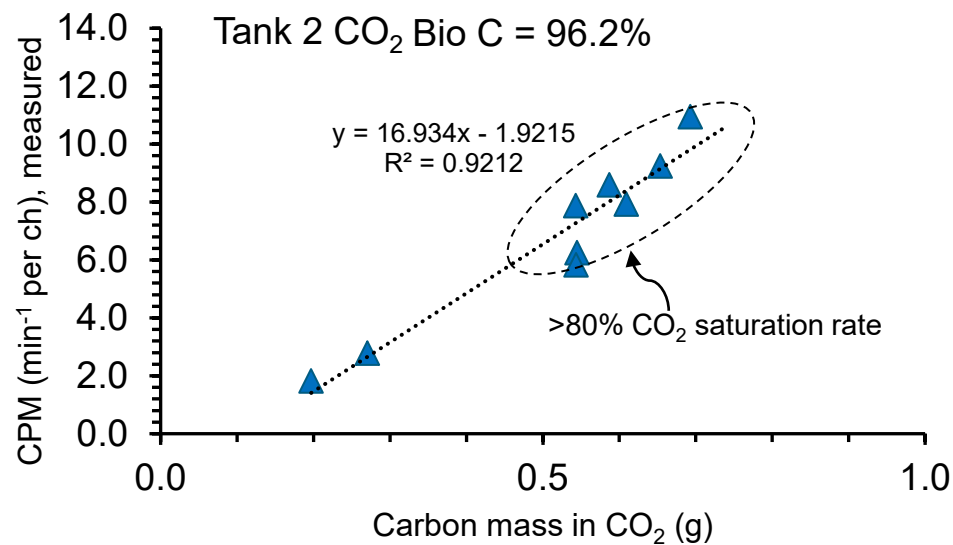
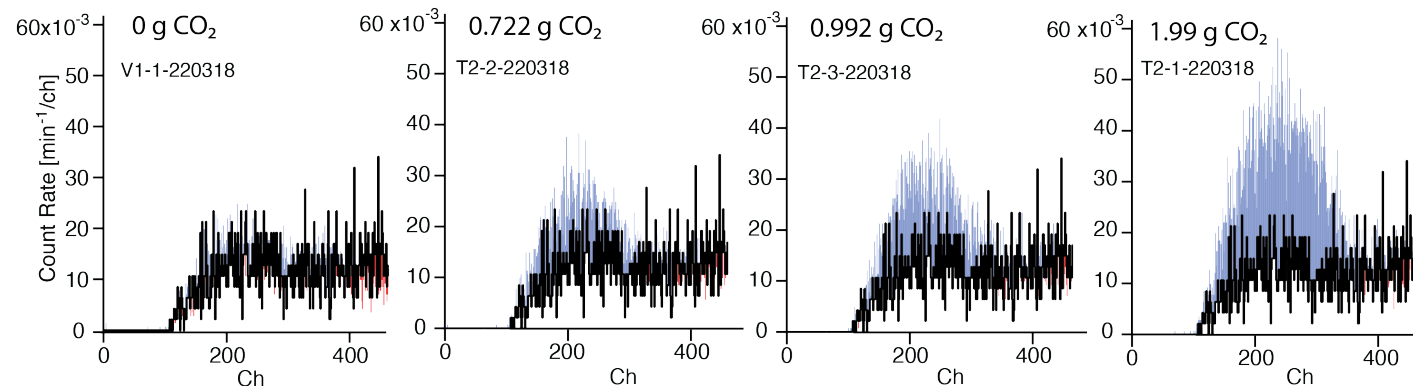


D: petroleum Diesel B20: bio-Diesel with 20% biogenic carbon B100: bio-Diesel with 100% biogenic carbon

- Investigated different techniques: Adsorbents, Catalytic, Oxidation. Adsorbents w/ Clay and Silica Gel was most effective.
- Decolorization reduced uncertainty w/out affecting %C_{Bio} interpretation for B100.
- Decoloring preferentially removed biogenic carbon in B20. We do not expect blended, co-processed samples to be affected by this artifact.

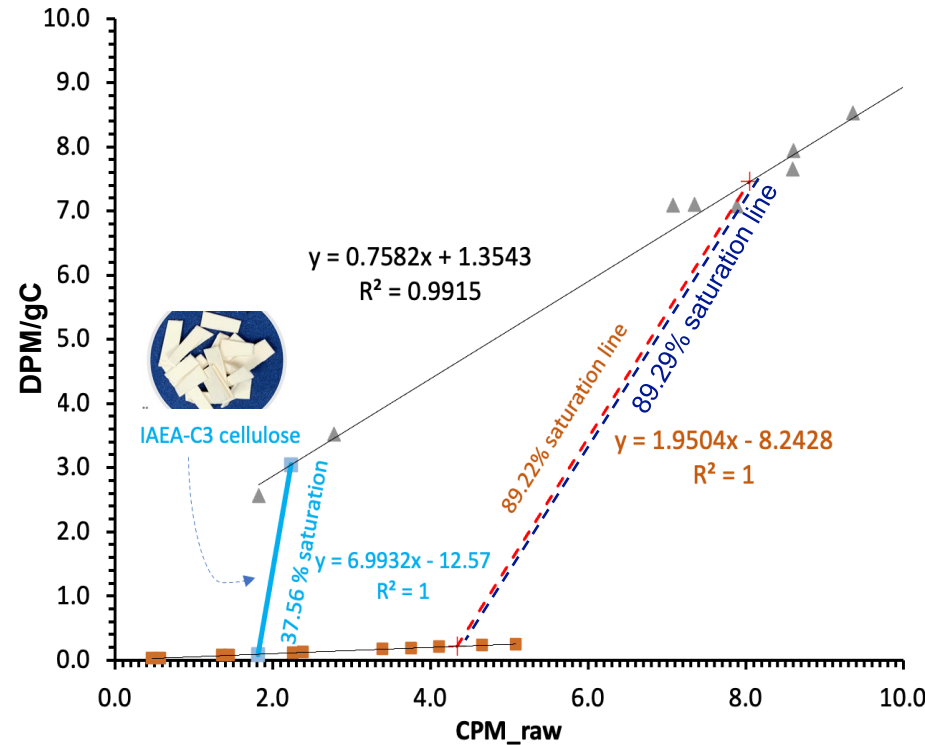
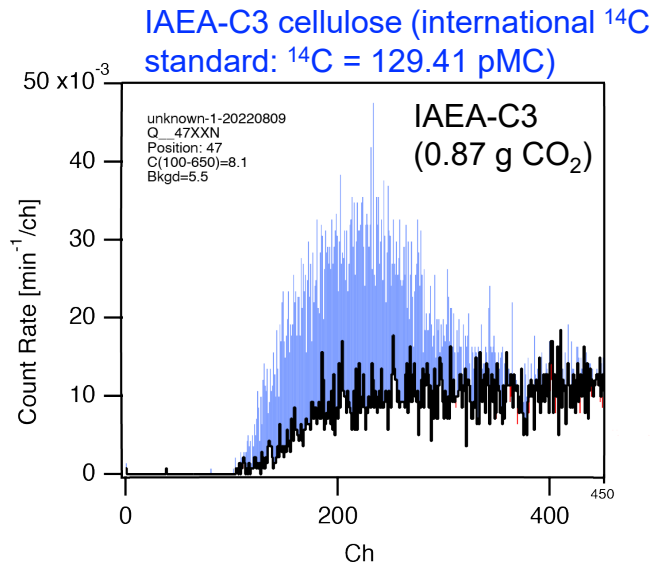
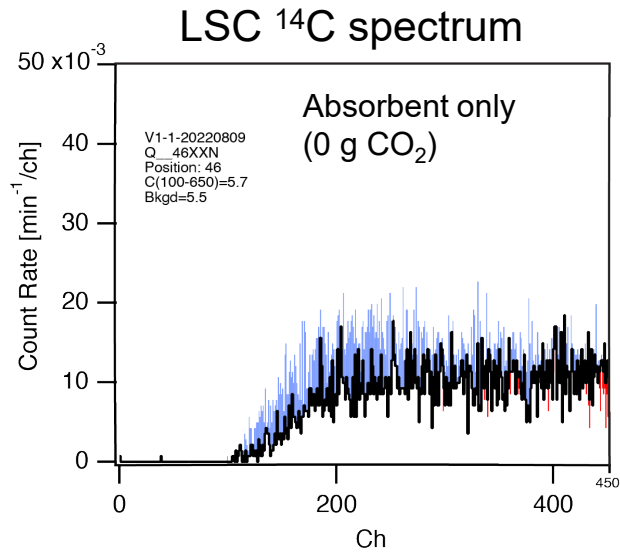


LSC ¹⁴C spectra (samples containing various amounts of CO₂ with 96.2% biogenic C)



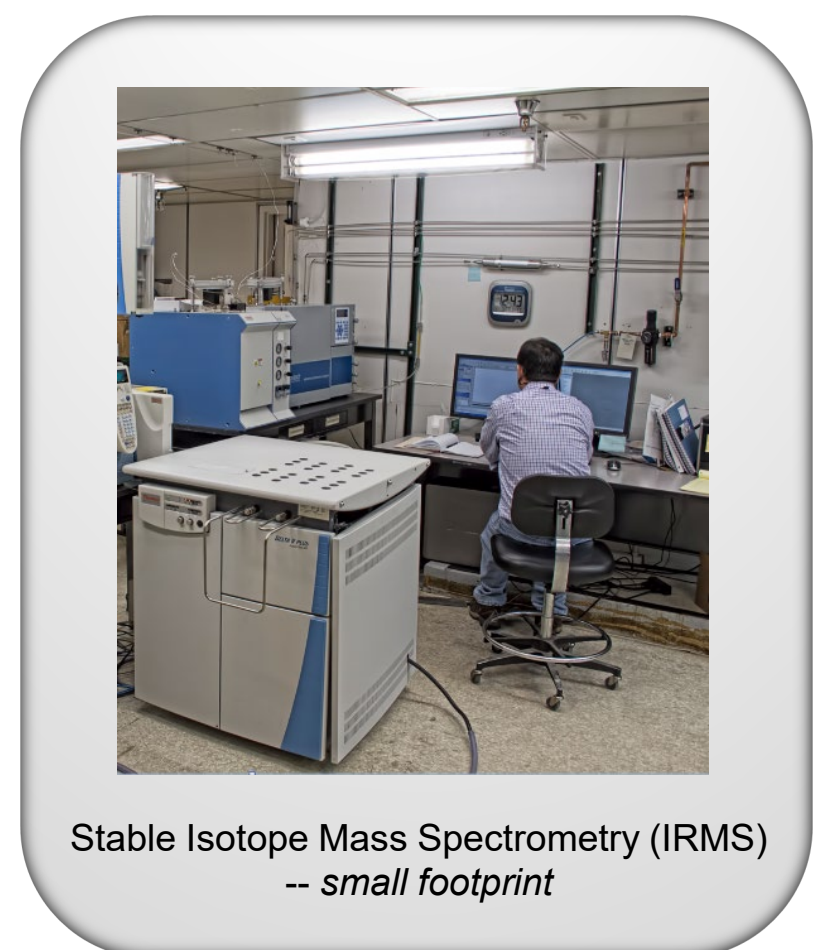
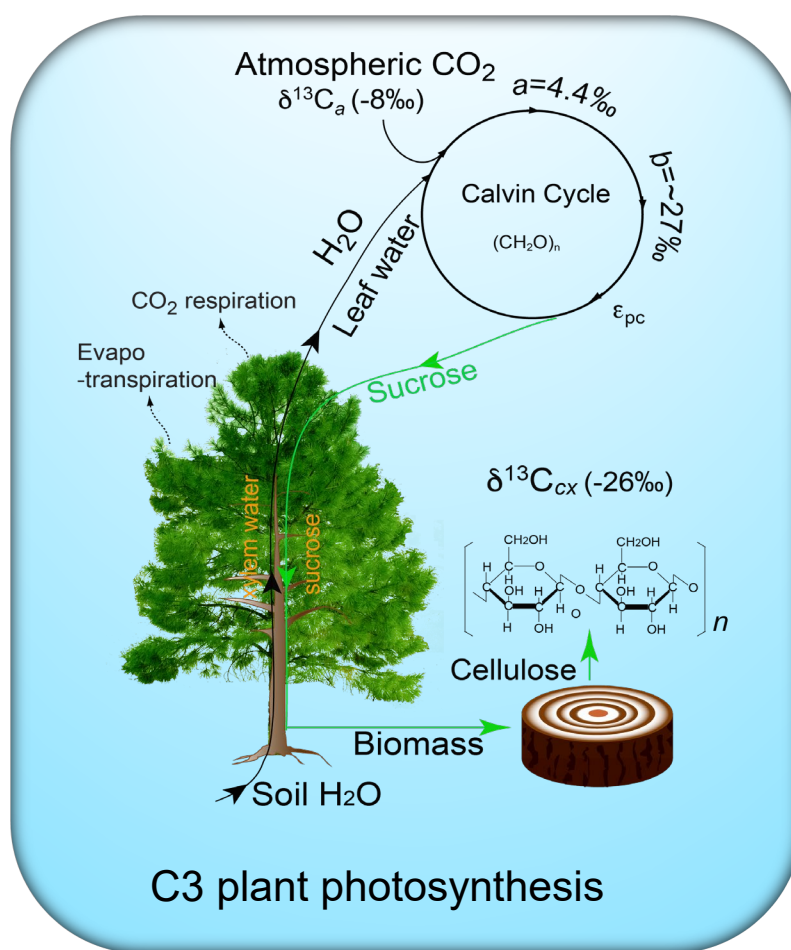
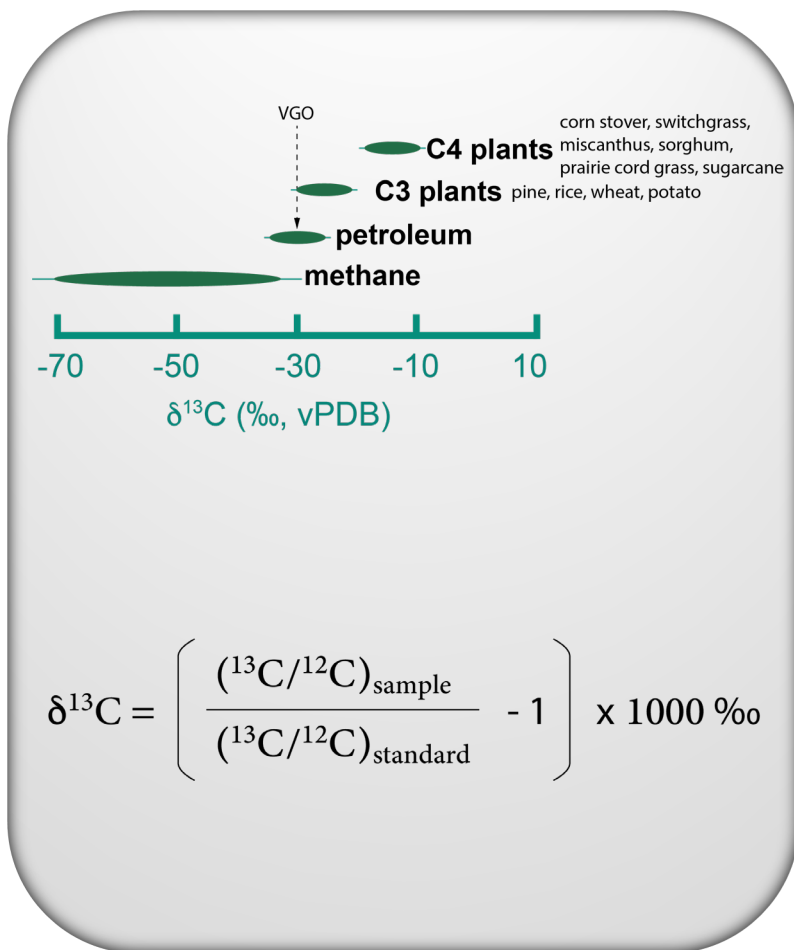
- Two tank CO₂ gases with very different biogenic carbon content are used to develop the calibration curves.
- Quantitative absorption of CO₂ in amine has been achieved.
- Very low ¹⁴C signal (0.1 g C with 2.7% Bio C) can be meaningfully detected through CO₂-Amine capture.

Verification of Calibration Line for Biogenic Carbon Measurement in Solid Biomass and CO₂ Gas Samples



Sample ID	Saturation rate	CO ₂ (g)	Bio C%, LSC	Expected %
IAEA-C3	37.56%	0.87	130.87	129.41
CO ₂ gas 1	89.22%	2.222	76.9	75.9
CO ₂ gas 2	89.29%	2.171	48.1	47.3

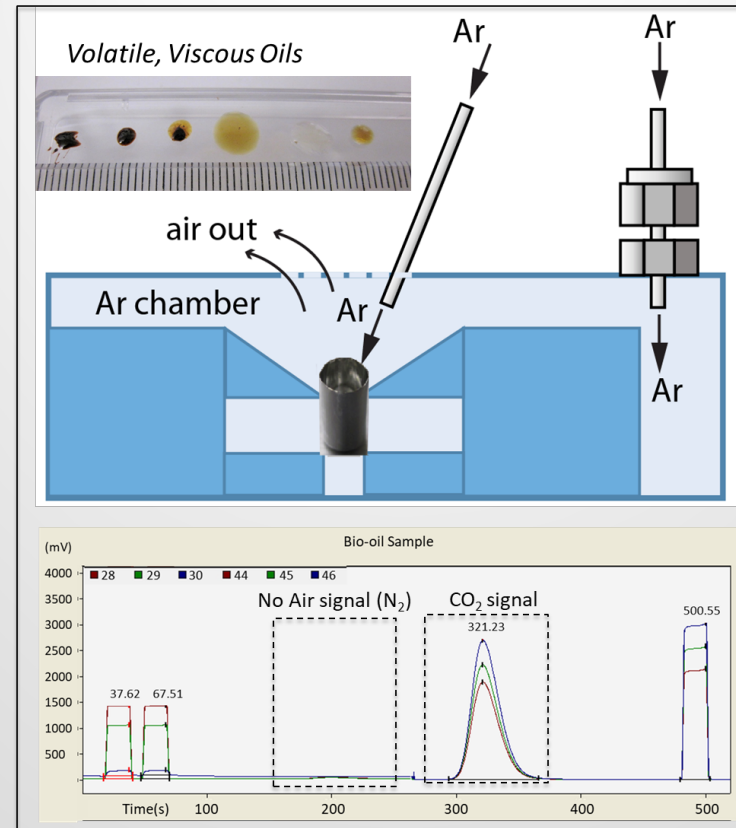
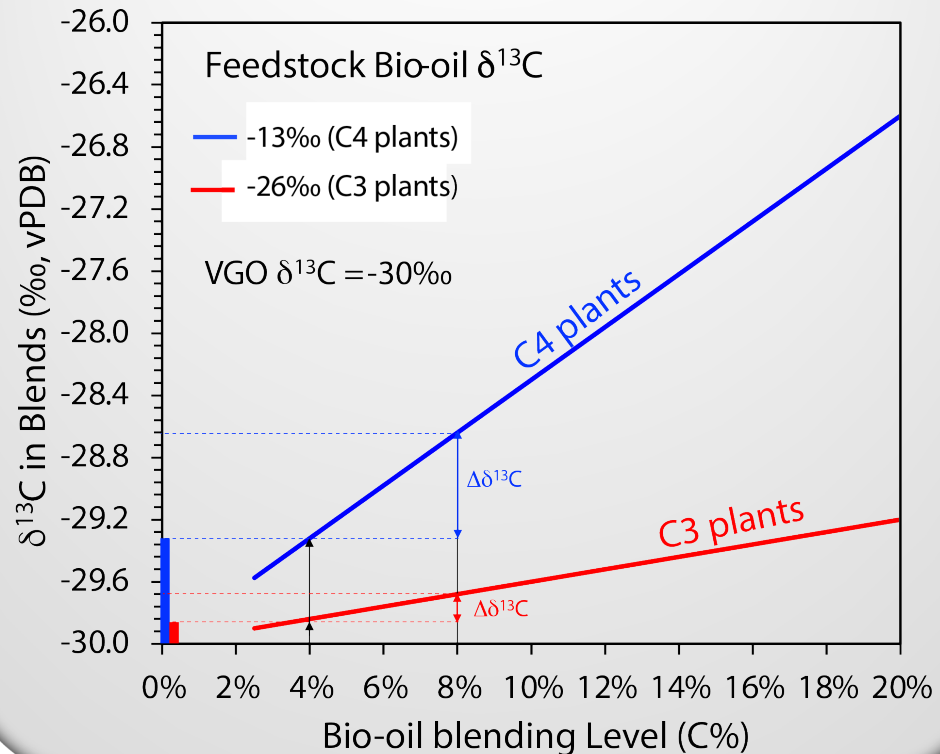
- CO₂ conversion method works for all types of fuels and feedstock (clear, moderately colored, dark, and solid).
- Future work:
 - Test fuels with low biogenic carbon content and repeatability.
 - Reduce instrument background noise (currently ~3 cpm at Los Alamos).
 - Develop high capacity CO₂ absorption media.



- Stable carbon method is based on $\delta^{13}\text{C}$ difference between two feedstocks.
- C4 plant-derived bio-oil has better traceability for biogenic carbon distribution.
- $\delta^{13}\text{C}$ can be easily measured using a IRMS.

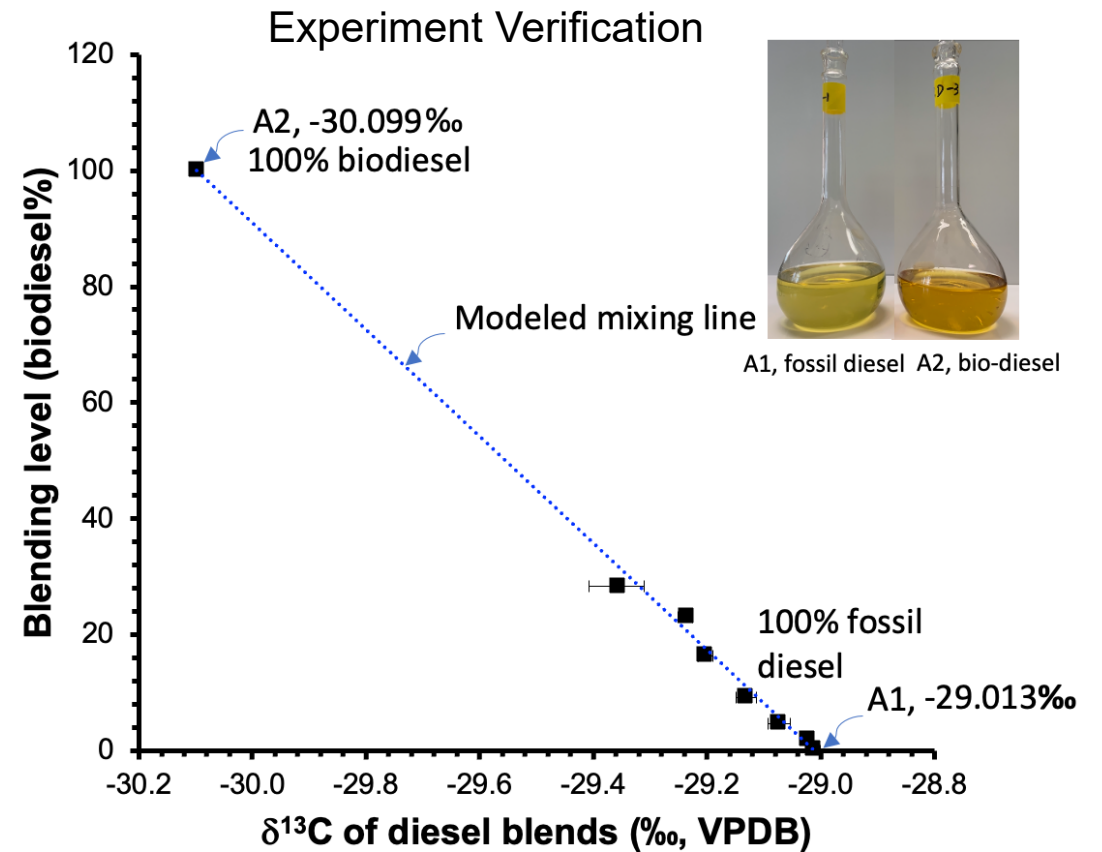
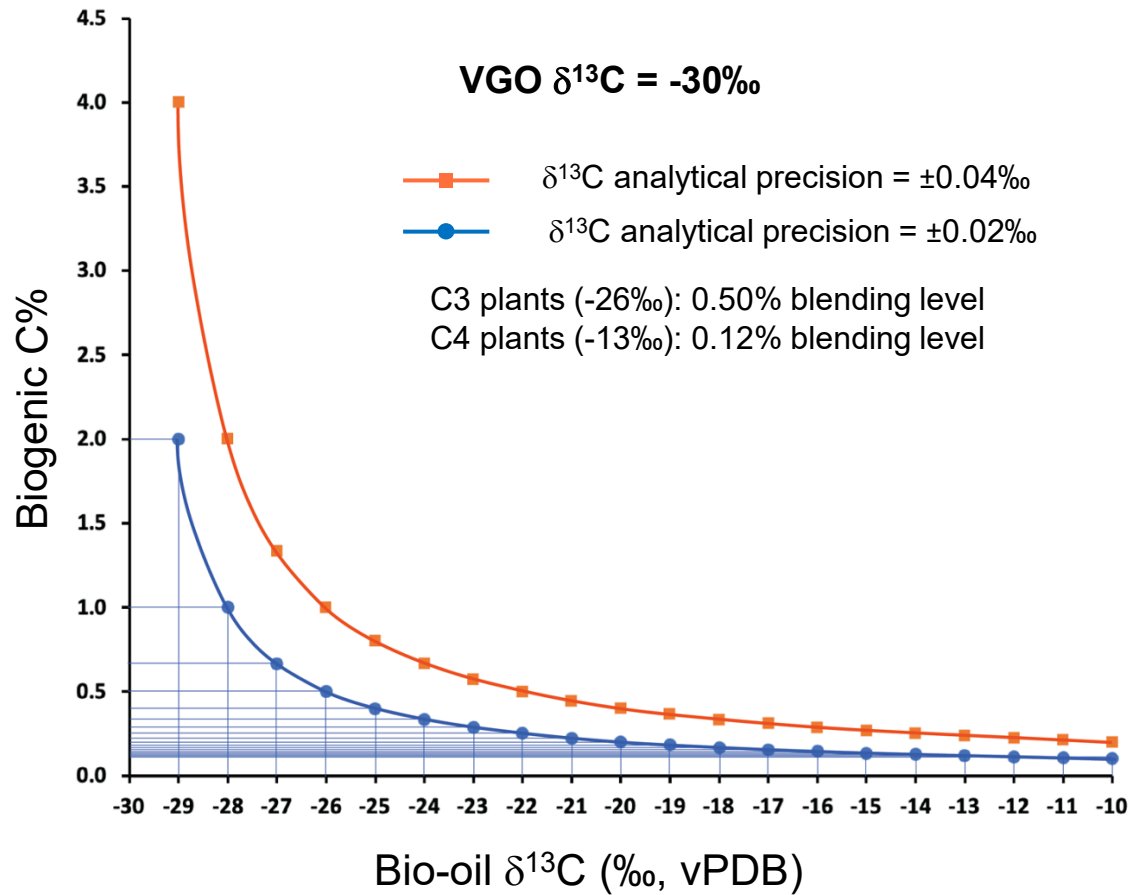
Sensitivity of $\delta^{13}\text{C}$ for Tracking Biogenic Carbon and Sample Preparation Protocol

Sensitivity of $\delta^{13}\text{C}$ method



- Volatile, viscous samples are difficult to measure due to fractionation.
- Sealing in tin capsules under argon eliminates volatilization and prevents atmospheric contamination
- Solid standards can be run concurrently

Geeza & Li et al. 2020 (Fuel)

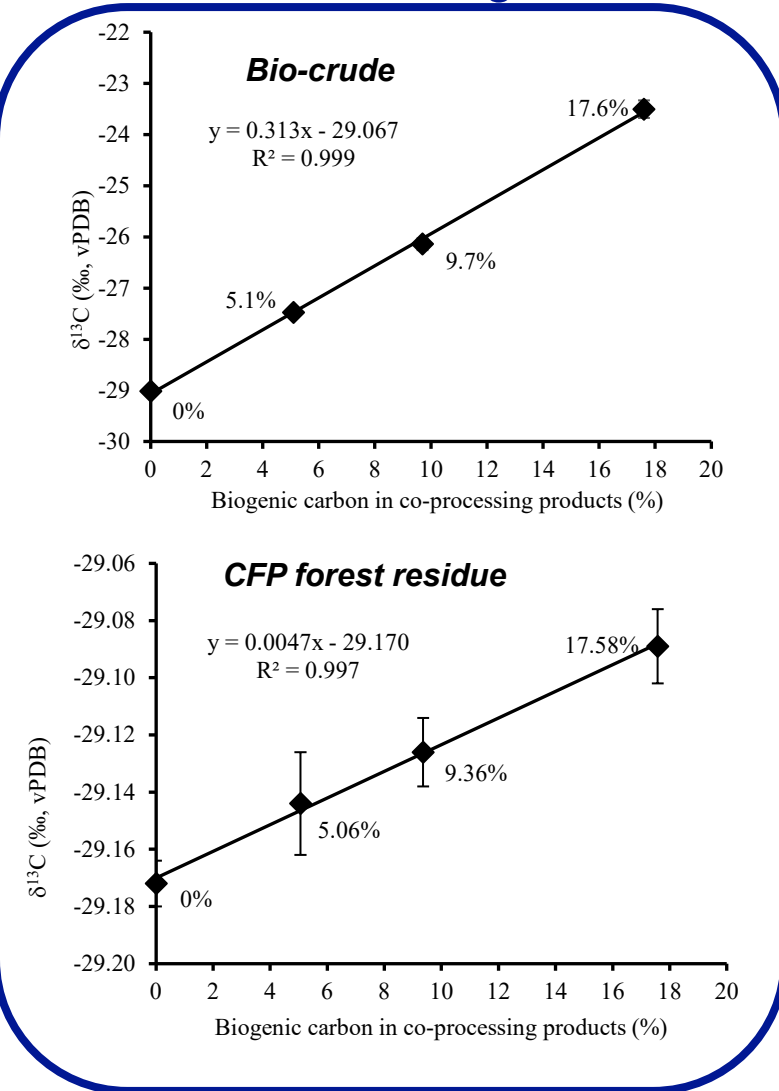


$$C_{\text{bio}}\% = \left[\frac{\delta^{13}\text{C}_{\text{mix}} - \delta^{13}\text{C}_{\text{fos}}}{\delta^{13}\text{C}_{\text{bio}} - \delta^{13}\text{C}_{\text{fos}}} \right] \times 100\%$$

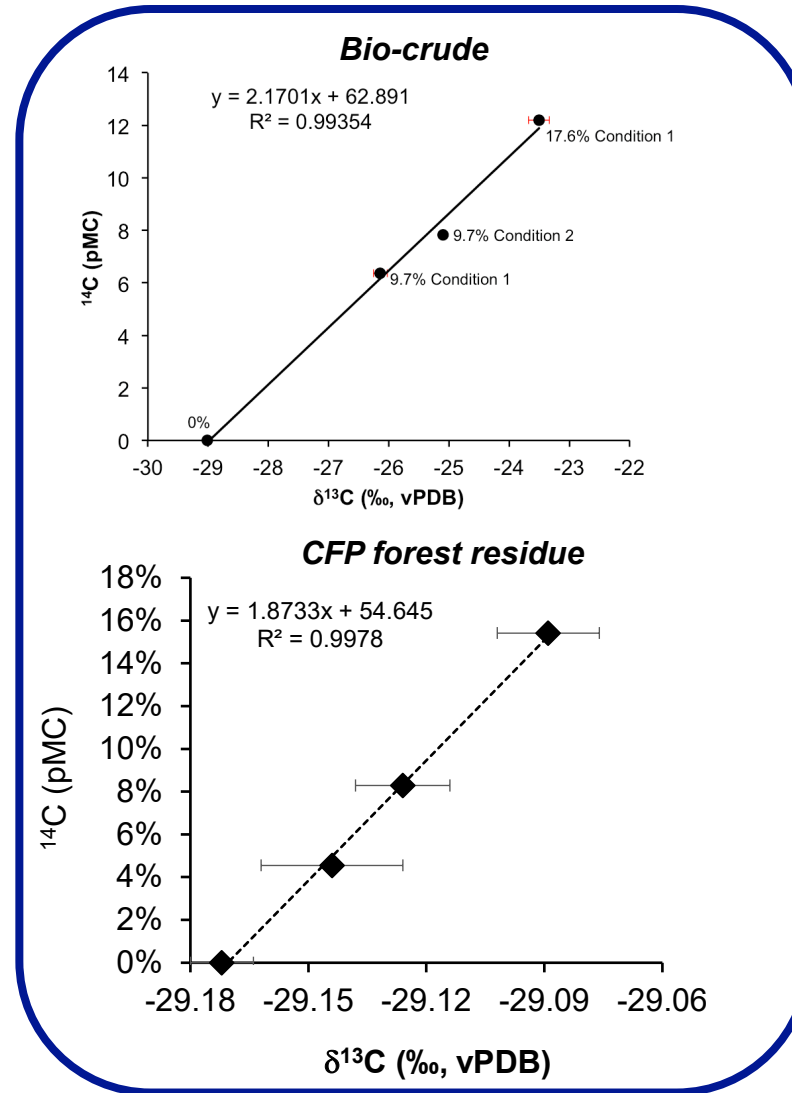
- Limit of Detection depends on analytical uncertainty.
- 0.5% blending level can be detected for C3 plant-derived bio-oil.
- 0.12% blending level can be detected for C4 plant-derived bio-oil.

Significant Correlation Between Biogenic C% and $\delta^{13}\text{C}$ in Co-processed Products Revealed

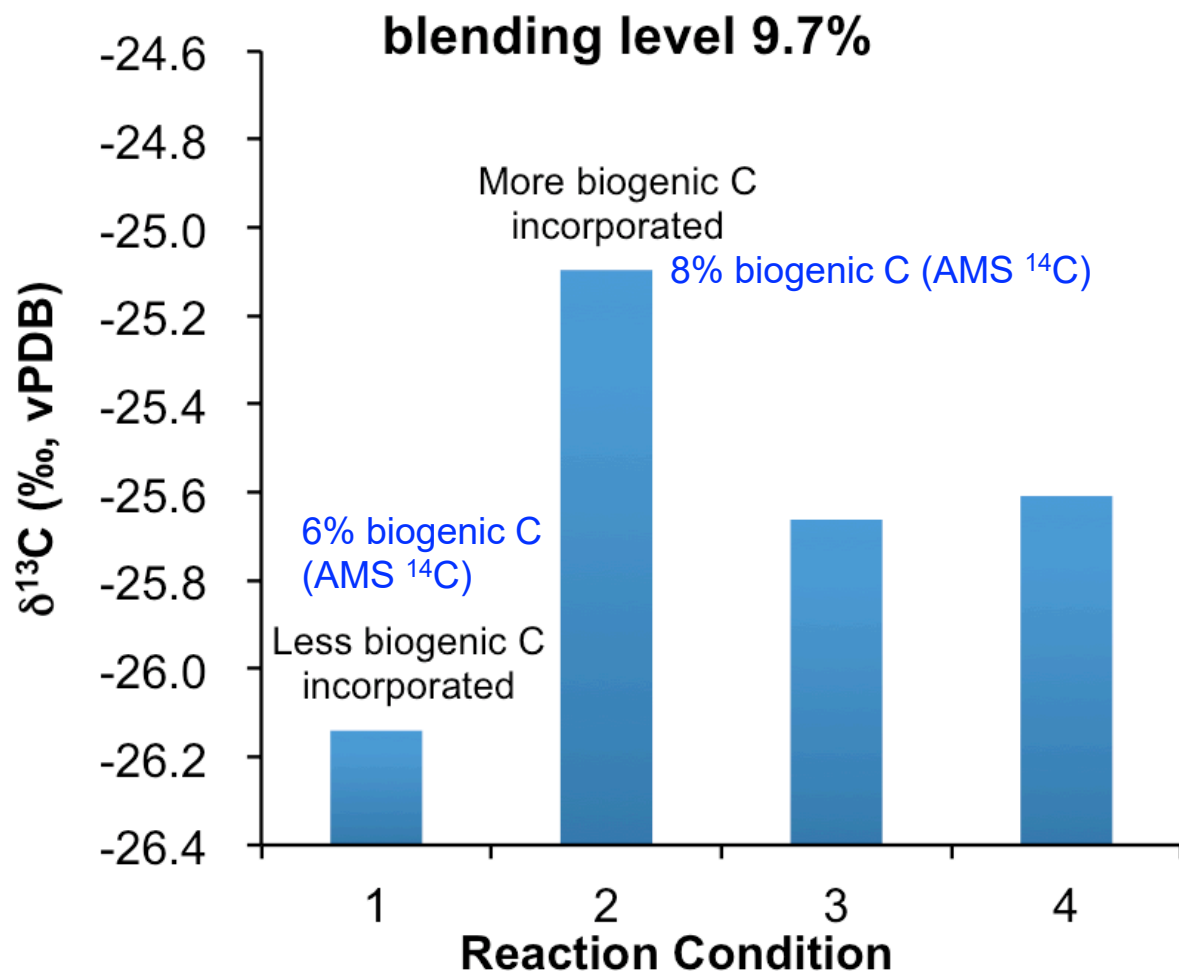
$\delta^{13}\text{C}$ and biogenic C%



AMS ^{14}C verification

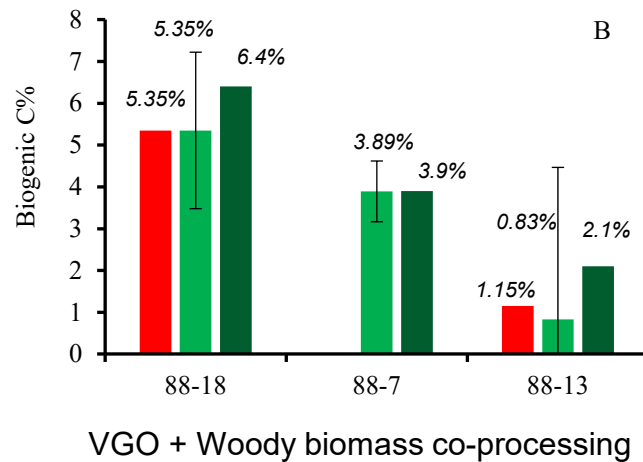
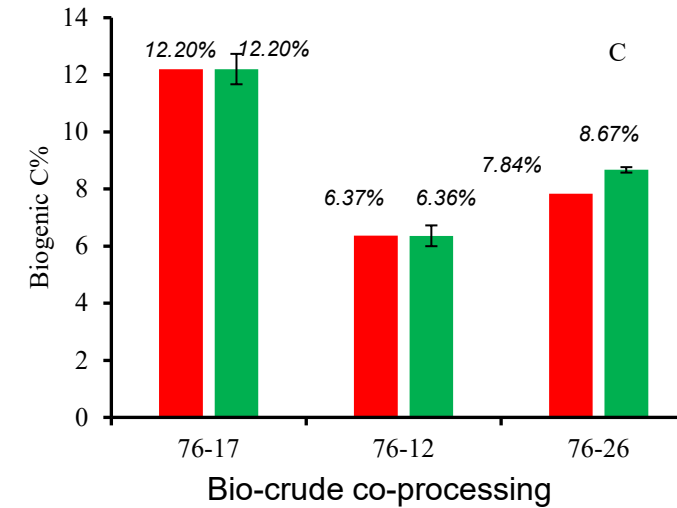
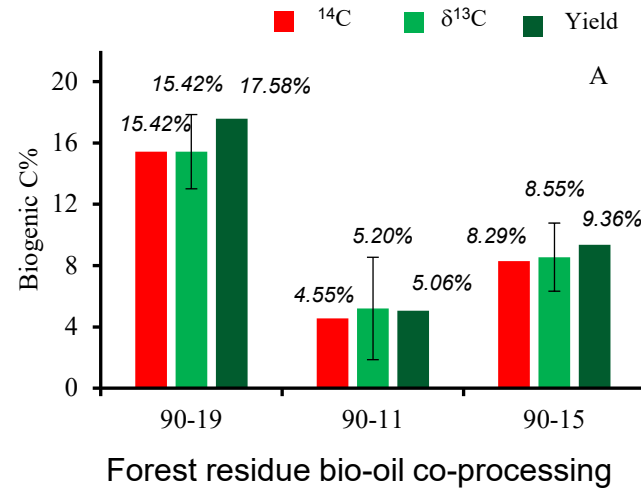


- Conducted co-processing experiments on bio-crude, FP, and CFP bio-oil.
- High precision $\delta^{13}\text{C}$ analysis revealed significant correlation with biogenic C% in co-processed products.
- $\delta^{13}\text{C}$ applicability for biogenic C tracking is confirmed by ^{14}C AMS method.





- Bio-crude co-processing with VGO at the same blending level (9.7%) under different co-processing reaction conditions (T, P, WHSV, H_2/VGO).
- Reaction Condition 2 shows the highest biogenic carbon incorporation.
- $\delta^{13}\text{C}$ analysis can be used to guide the optimization of co-processing parameters.

Quantification of Biogenic Carbon in Co-processed Products by $\delta^{13}\text{C}$ Analysis



- Quantification of biogenic carbon in co-processed products can be achieved through $\delta^{13}\text{C}$ analysis. Large uncertainty for quantifying C3-plant co-processing.
- C isotope fractionation factor is not affected by the bio-oil blending levels in co-processing.
- $\delta^{13}\text{C}$ analysis can be a viable way to track biogenic C in co-processing, **but feedstocks must be available for $\delta^{13}\text{C}$ analysis.**

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Senior Scientist

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Optical approach for tracking biocarbon in a refinery setting using $\delta^{13}\text{C}$ values

Interest of application

- *Interest from Industry Advisory Board*
- *Fast, reliable, and inexpensive biocarbon trackers*
- *Onsite and near real time analytical methods*
 - *Low blend ratios of biocarbon*

Application of Optical Spectroscopy to Co-Processing

Overarching objective

- Can optical spectroscopy monitor $\delta^{13}\text{C}$ as well as isotope ratio mass spectrometry?
 - Transportation fuels
 - Low biocarbon ratios (0% to 10%)
- Can optical spectroscopy be adapted for refinery applications?
 - Challenging environment
 - Near real time measurements

$\delta^{13}\text{C}$ can be used for Tracking Biocarbon

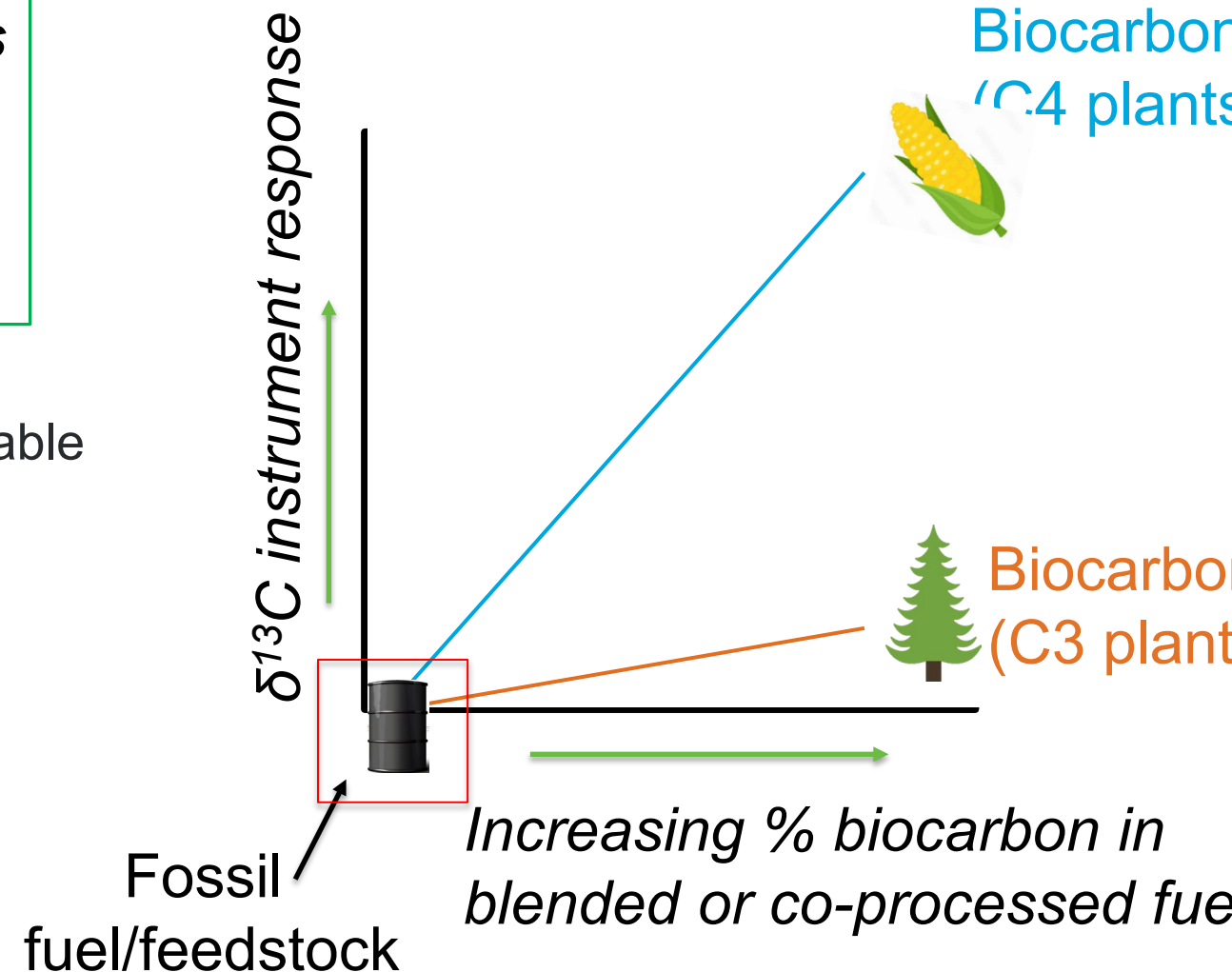
Natural abundance

^{12}C : 98.89 %
 ^{13}C : 1.108 %
 ^{14}C : 1×10^{-10} %

$\delta^{13}\text{C}$ is a *ratio of ratios*

$$\delta^{13}\text{C} = \left[\frac{(^{13}\text{C}/^{12}\text{C})_{\text{sample}}}{(^{13}\text{C}/^{12}\text{C})_{\text{standard}}} - 1 \right] \times 1000 \text{ ‰}$$

- Measured $\delta^{13}\text{C}$ correlates with percent renewable carbon in blended and co-processed fuels (i.e., linear relationship)
- $\delta^{13}\text{C}$ data can help monitor and optimize incorporation of biocarbon during coprocessing (e.g., prevention of loss to gas phase or coke)



Instrumentation for $\delta^{13}\text{C}$ Measurements

Isotope Ratio Mass Spectrometry (IRMS)



- Gold standard for $\delta^{13}\text{C}$ measurements
- High precision
- Collection to data time: ~1 to 2 days
 - Analysis time: 30 minutes
- More expensive laboratory setup
- Stable laboratory environment needed
- Larger instrument (101 cm x 135 cm x 136 cm)

Optical Spectroscopy (TILDAS)



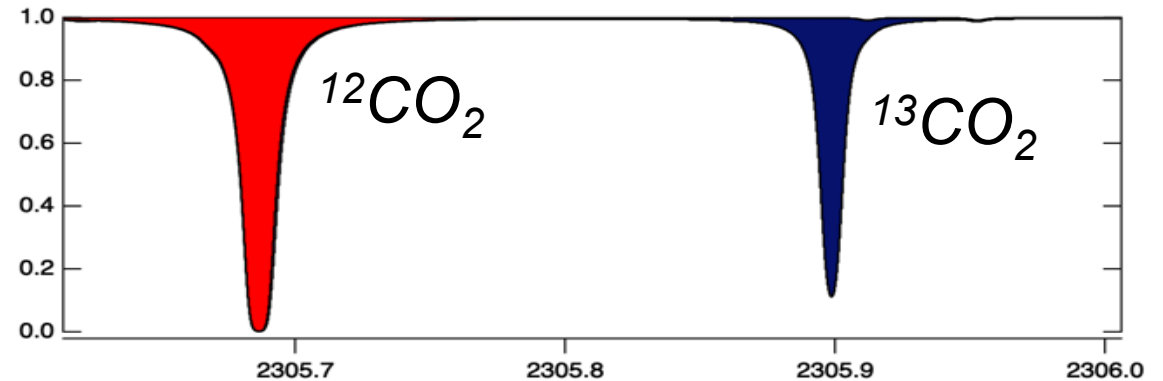
- Less expensive
- High precision
- Collection to data time: 20 to 60 minutes
- Smaller footprint (44 cm x 66 cm x 27 cm)
- Rugged design, field deployable
- Potential for online analysis & automation

Optical Spectroscopy: Principles of TILDAS Operation

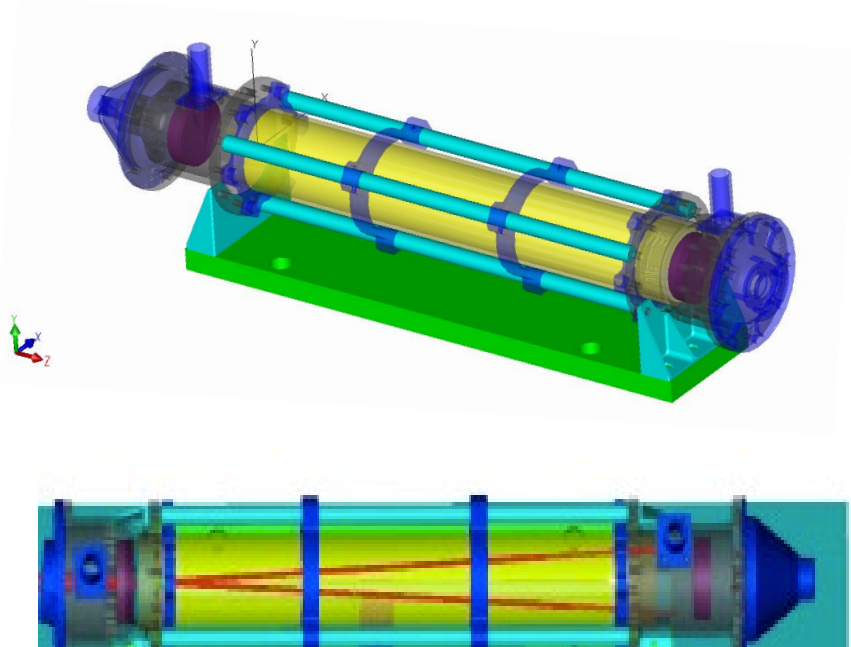
Tunable Infrared Laser Direct Absorption Spectroscopy (TILDAS)

- Sample converted to CO₂ by combustion
 - *Solid, liquid, gas samples are possible*
- Light absorption measured as wavelength
- Light transmission used to calculate $\delta^{13}\text{C}$

Light Transmission



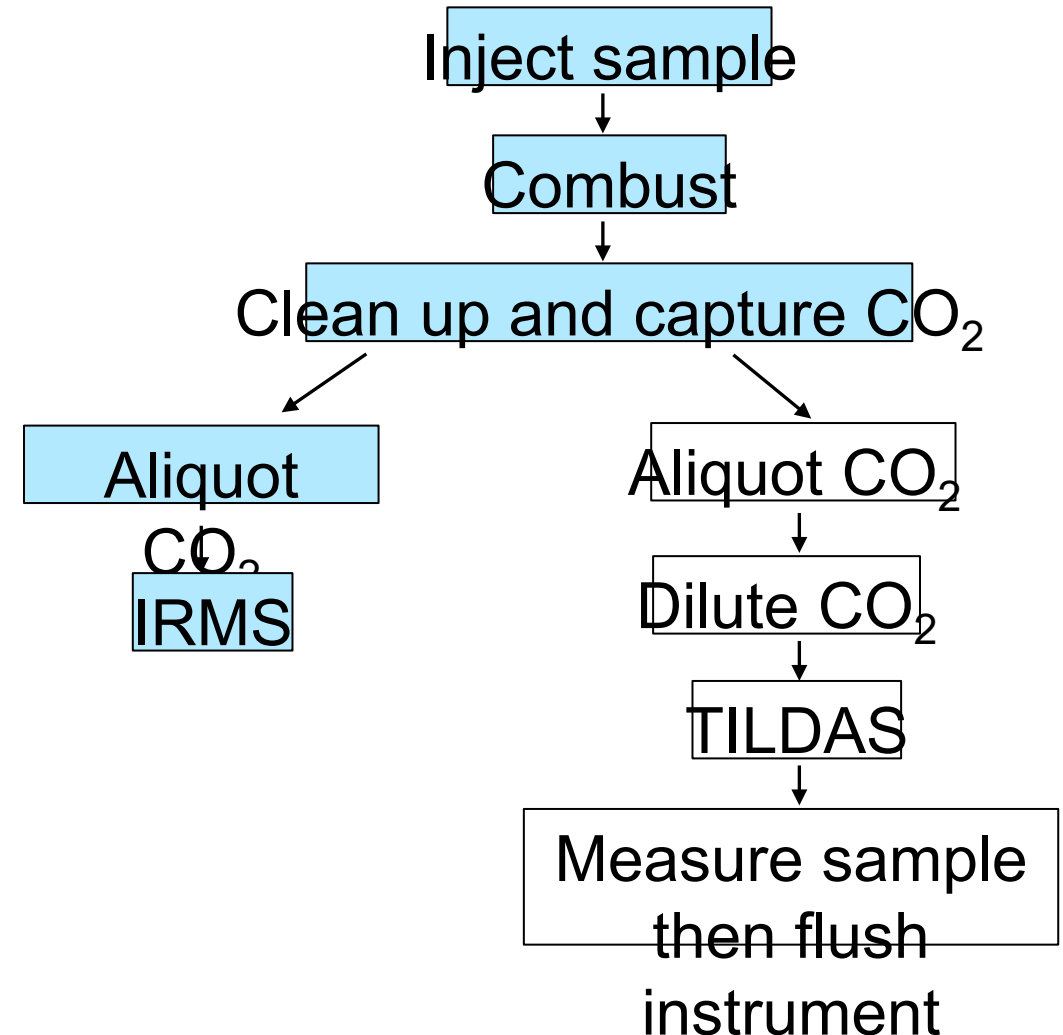
Wavenumber (cm⁻¹)



Method Development: Sample Preparation and Analysis

Can optical spectroscopy monitor $\delta^{13}\text{C}$ as well as isotope ratio mass spectrometry?

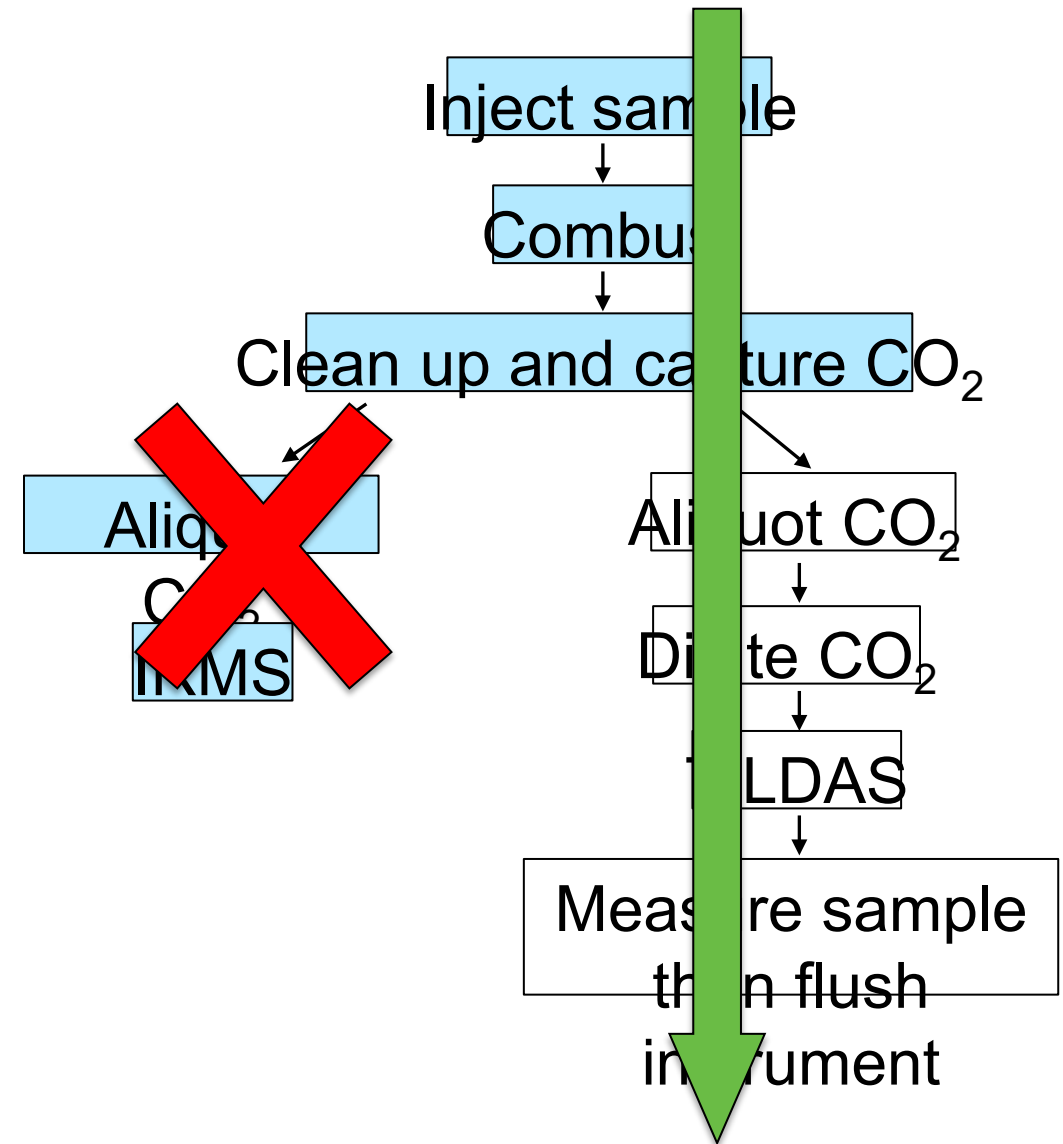
- Transportation fuels
- Low biocarbon ratios (0% to 10%)



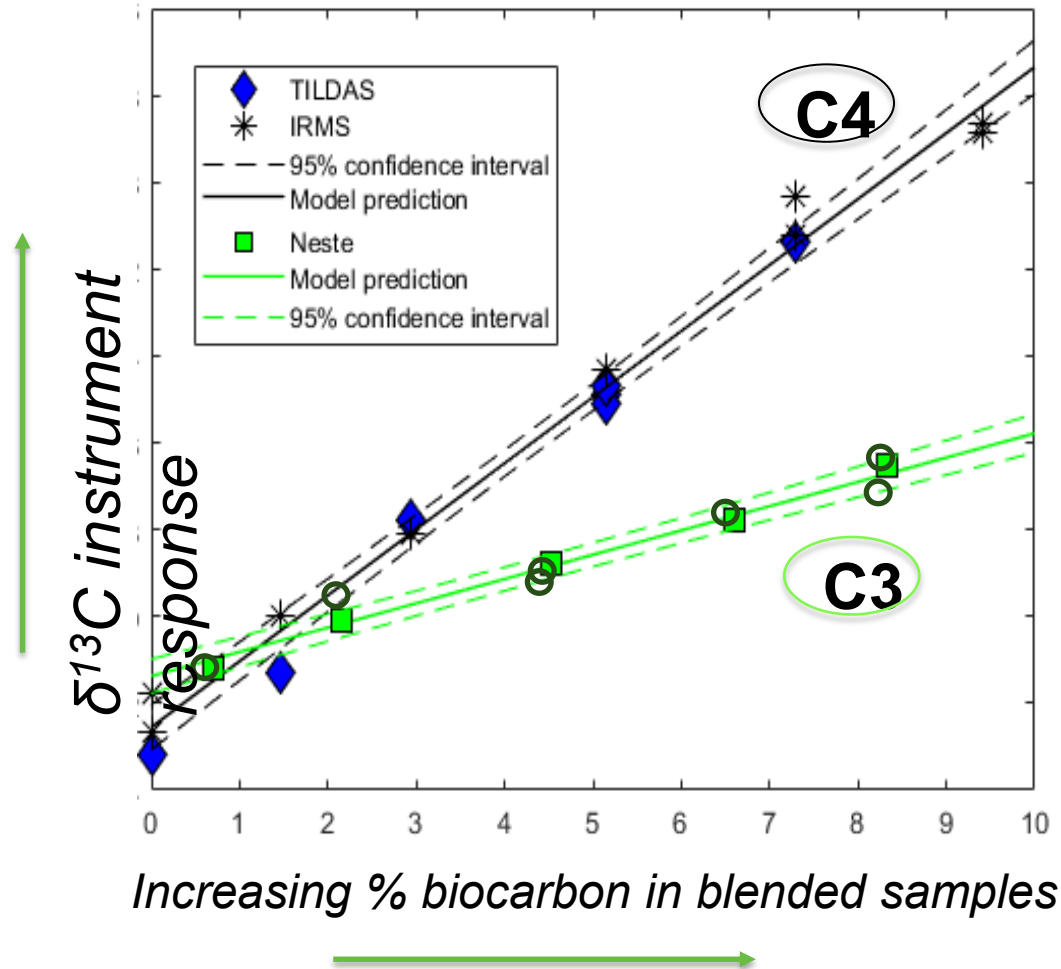
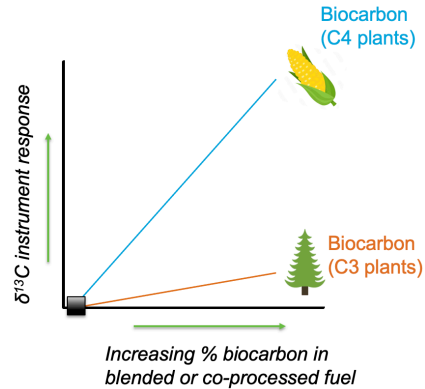
Method Development: Optimizing Sample Preparation and Analysis

Can optical spectroscopy be adapted for refinery applications?

- Challenging environment
- Near real time measurements



TILDAS $\delta^{13}\text{C}$ Data are Comparable to IRMS Data at Low Blend Ratios



- Fossil fuel and biofuel blended at low biocarbon ratios (0% - 10%).
- Strong correlation between methods
 - R^2 close to one
 - Tight 95% confidence intervals
- ***TILDAS and IRMS yield similar results for tracking low levels of biocarbon in transportation fuels***

Optical Spectroscopy in the Refinery Setting

- **Some instrument requirements for measuring $\delta^{13}\text{C}$ at a refinery**

- ✓ Stable, onsite/near real time method
- ✓ Complete sample preparation and sample analysis
- ✓ Consistent, high precision measurements
- ✓ Ideally inexpensive, fast, and automated analytical system

- **Project status**

- TILDAS has the potential for near real time analysis
- Integrating sample preparation and TILDAS to simulate near real time measurements
- More development needed before deployment
- Need to partner for insights on how this method can best be employed at refineries

Acknowledgements

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James Lee
Douglas Ware
Thomas Geeza
Oleg Maltseve
Jacob Helper



Scott Herndon
David Nelson

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- Laser Spectroscopy and Stable Carbon Isotopes for Tracking Biocarbon

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- • Q&A

