

# 2.3.1.318 Syngas Derived Mixed Olefin Oligomerization for Sustainable Aviation Fuel

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Technology Area Session: Catalytic Upgrading

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## Project Overview

**Develop and demonstrate mixed olefin (C<sub>2</sub>-C<sub>5</sub>) co-oligomerization to enable multiple renewable feedstocks to produce SAF**

### Project Goal

Develop a C<sub>2</sub>-C<sub>5</sub> co-oligomerization catalyst and demonstrate an efficient path to generating sustainable aviation fuel (SAF) from syngas via a mixed olefin intermediate

### Outcome

- Co-oligomerization catalyst development that is stable for mixed olefins derived from methanol to olefin process
- Produce >1 L finished jet fuel from this process that meet the ASTM standards for Tier α and β analysis

### Relevance

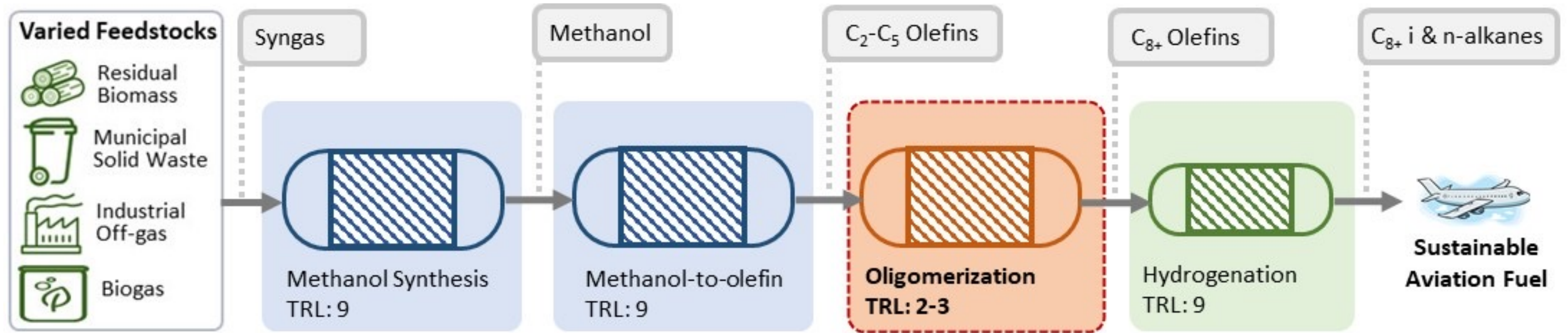
Develop technologies that can achieve greenhouse gas (GHG) reduction of >70% for SAF production from renewable feedstocks



Source: <https://www.energy.gov/sites/default/files/2022-09/beto-saf-gc-roadmap-report-sept-2022.pdf>

## 1-Approach

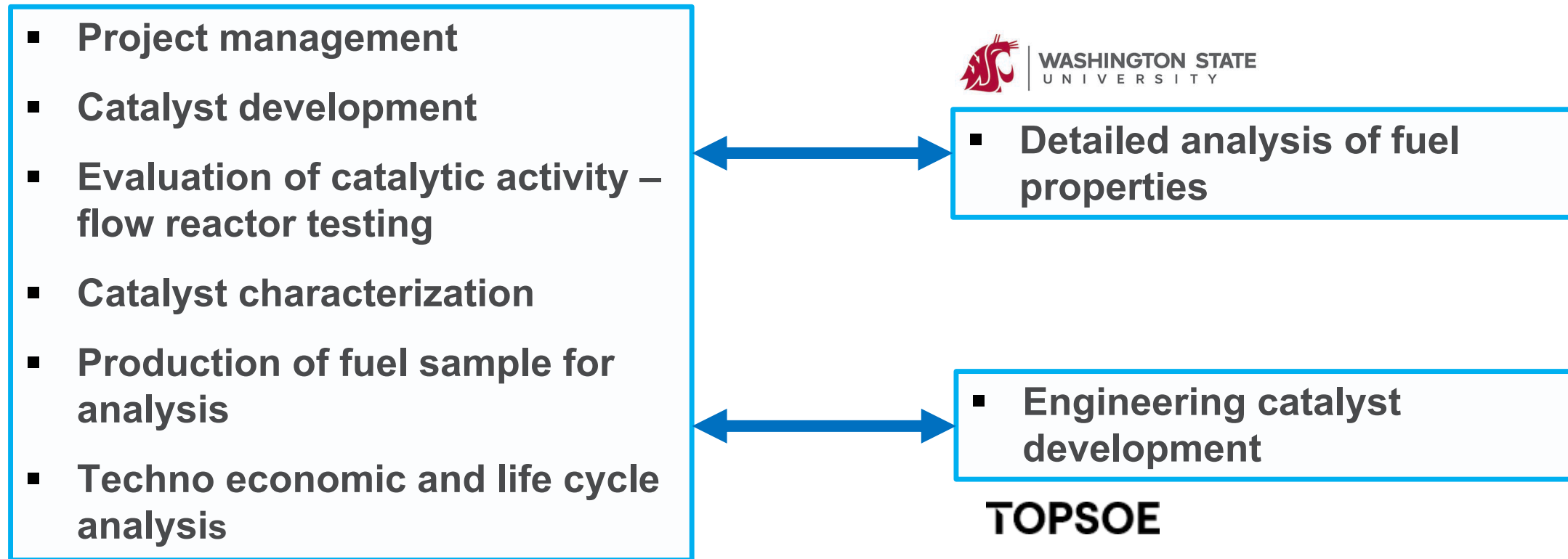
# Co-oligomerization of mixed olefins to demonstrate efficient path to produce SAF via syngas



- Leveraging existing commercial processes and feedstocks will be the most efficient path toward producing SAF in the near-term
- Developing mixed olefin co-oligomerization is key to achieve an end-to-end commercial pathway for producing SAF from syngas derived from various renewable feedstock

## 1-Approach (Project Management)

# Integrated work between PNNL, WSU and Haldor Topsoe



- Integrated workflow and handoff points between the partners based on the core capability and the technical expertise
- Regular meetings between the partners for the technical updates

## 1 – Approach (Project Management)

# Diversity, Equity, and Inclusion (DEI) Plan: Hired a summer intern through PNNL's diversity internship program

### Project DEI Task: Hire at least one student from groups under-represented in STEM

- PI of this project participated in [The Energy and Environment Diversity Internship Program \(EEDIP\)](#) program designed for students passionate about environmental and clean energy science
- This is a competitive opportunity supports traditionally under-represented students in target technical areas through a 10-to-12-week paid internship

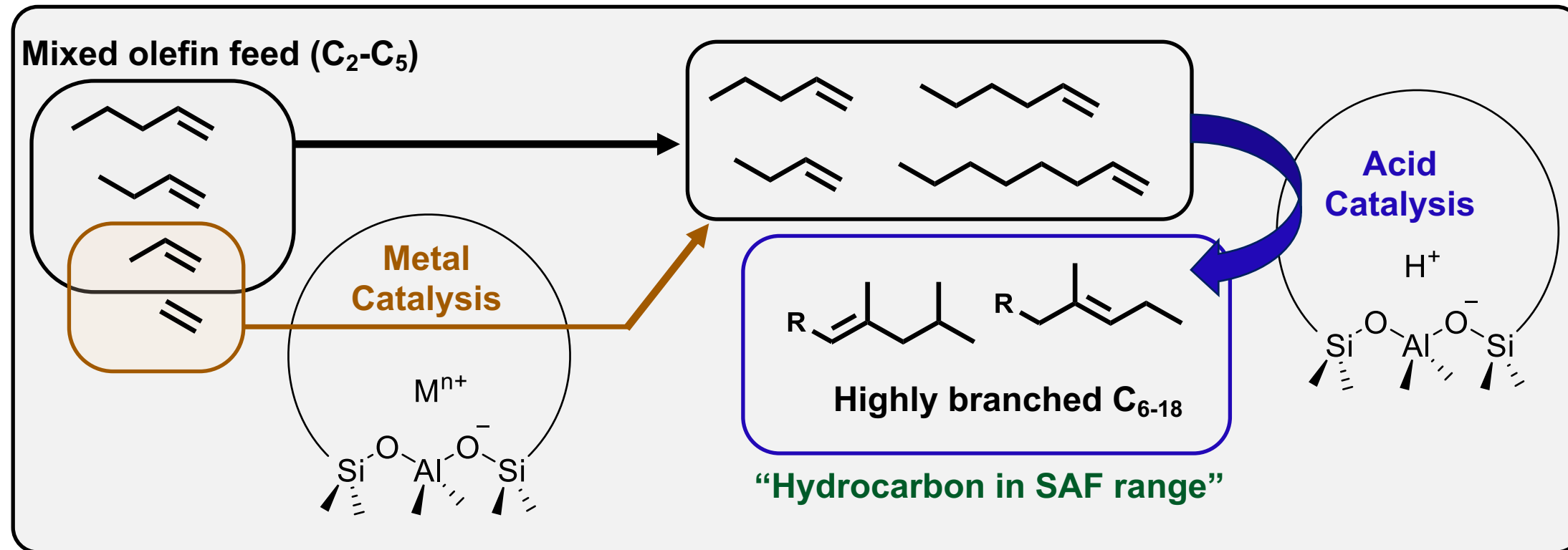


### Outcome:

- Hired a Bachelor of Science Chemical Engineering student through PNNL's EEDIP program
- Student is planned to work at PNNL between June – August 2023 to participate in the co-oligomerization catalyst/process development activities

## 1. Approach

**Requires hybrid catalyst containing both metal and acid sites**



- Activation of C<sub>2</sub> and C<sub>3+</sub> goes through different reaction mechanism and active sites
- Integrating both metal and acid catalysis pathways are key to facilitate co-oligomerization of C<sub>2</sub> and C<sub>3+</sub> olefins to produce SAF

## 1. Approach

# Addressing risk and measuring progress

▪ **Milestone (Sep 2022):** Produce 100 mL of finished jet fuel sample from the representative Methanol to Olefin (MTO) feedstock **Completed**

**Milestone (Sep 2023):** Demonstrate the integration between Methanol-to-Olefin (MTO) reactor and oligomerization reactor to evaluate the feedstock impurity effect of the catalyst performance **Ontrack**

**Milestone (Sep 2024):** Demonstrate extended operation (>500 hours) of an integrated process using engineered catalysts with on-stream regeneration **Ontrack**

### Risk Mitigation:

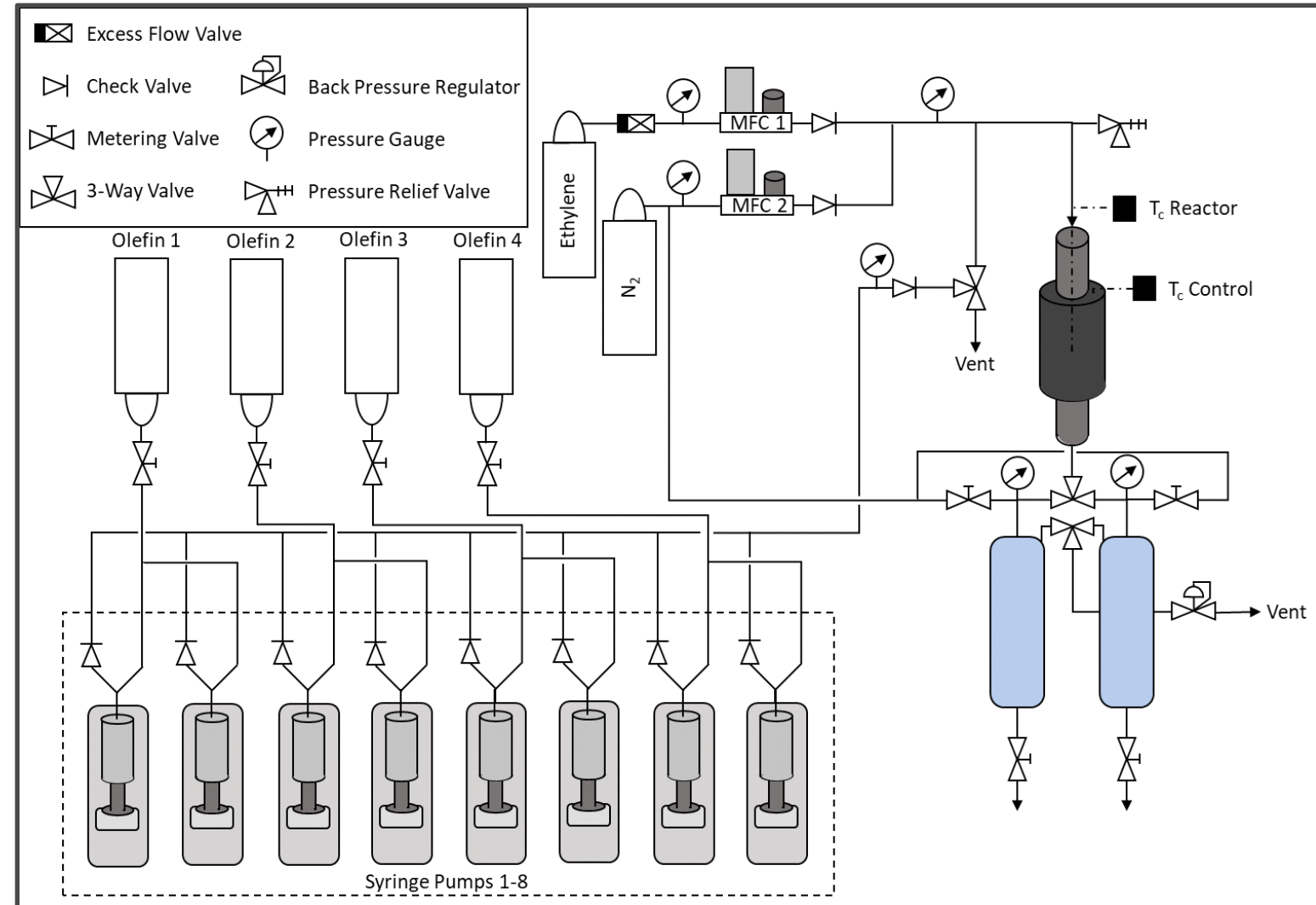
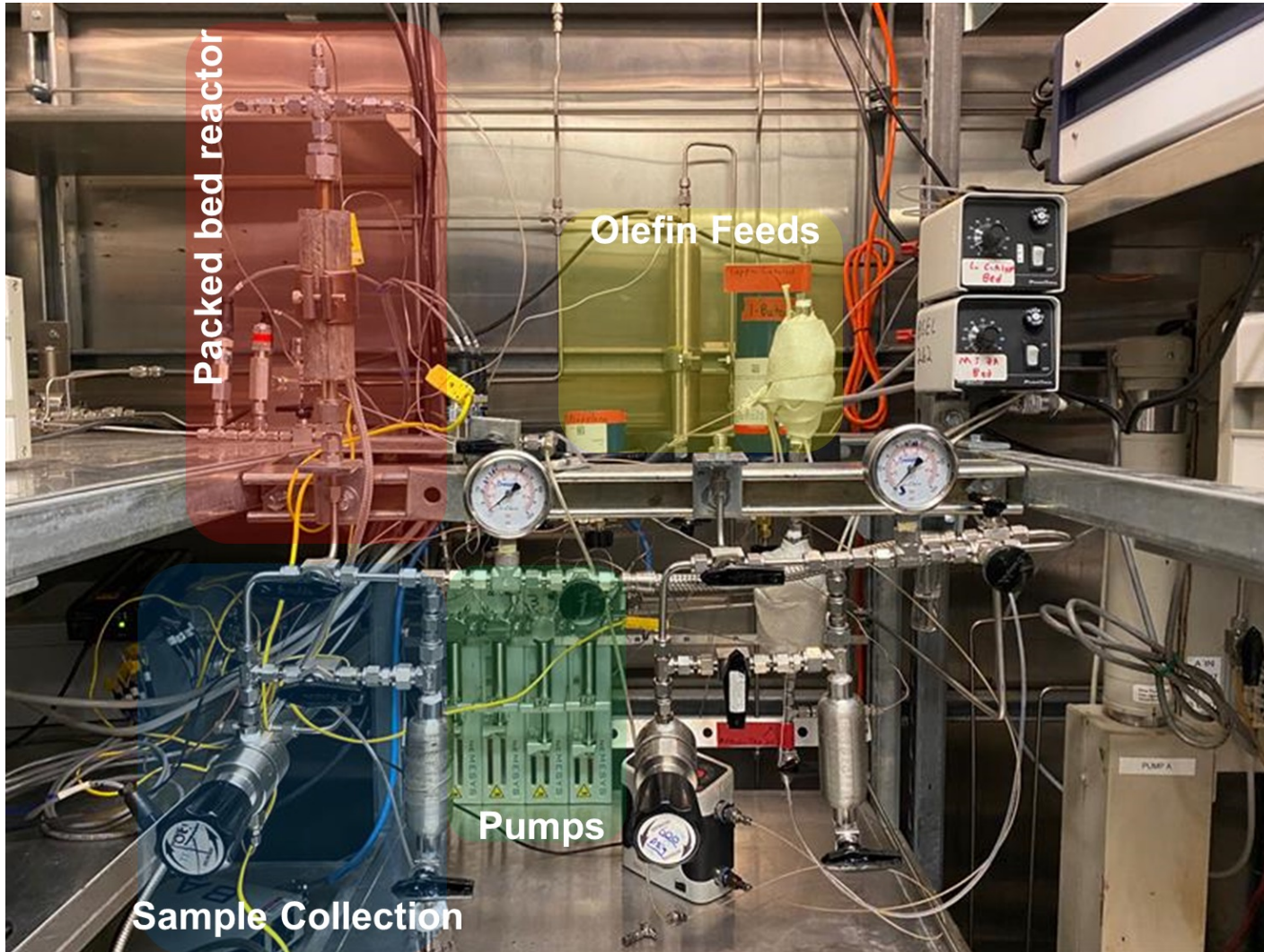
*Low product yield and catalyst deactivation on single-step mixed olefin oligomerization:*

Perform oligomerization in staged two-zone reactor with independently optimized catalyst, still achieving goal of significantly reducing process intensity.

**Go/ No-Go Completion:** Production of 100 mL of finished jet fuel starting from MTO mixed olefin feed and complete analysis of Tier  $\alpha$  and Tier  $\beta$  properties meeting ASTM standards at  $\geq 50\%$  blend level. **December 2022**

## 2. Progress and Outcome

# Construction of new reactor system that handles multiple olefin feed

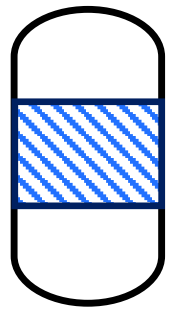


- Constructed new reactor systems to carry out co-oligomerization of C<sub>2</sub>-C<sub>5</sub> mixed olefins

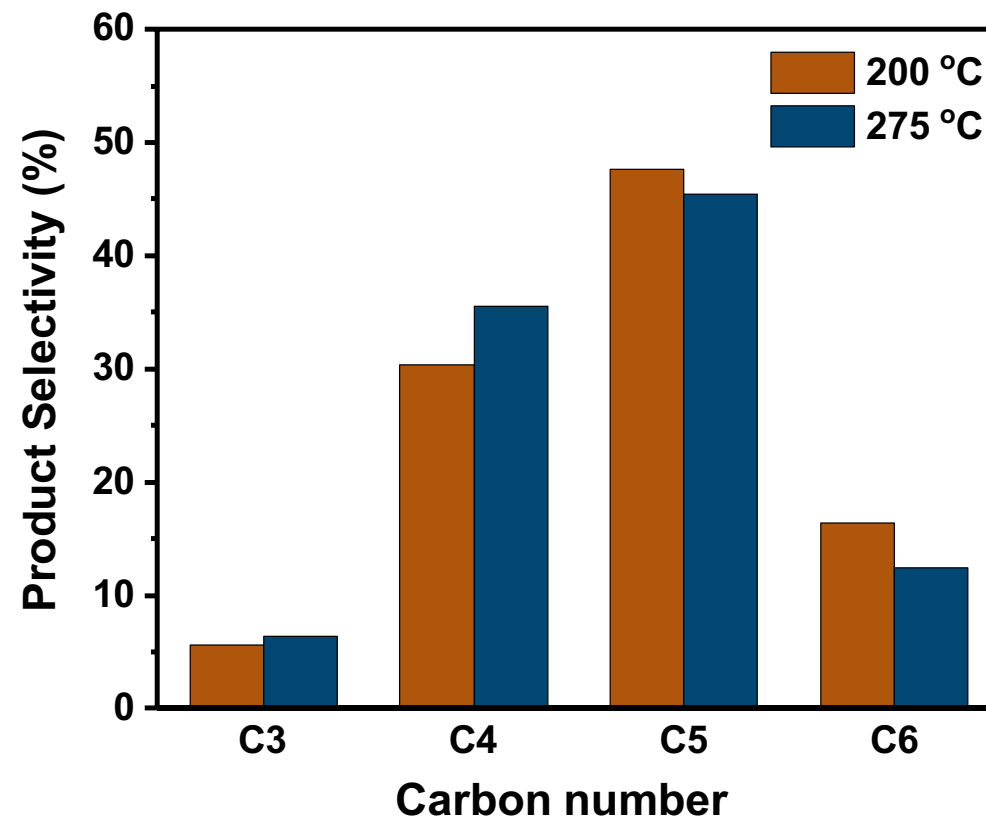
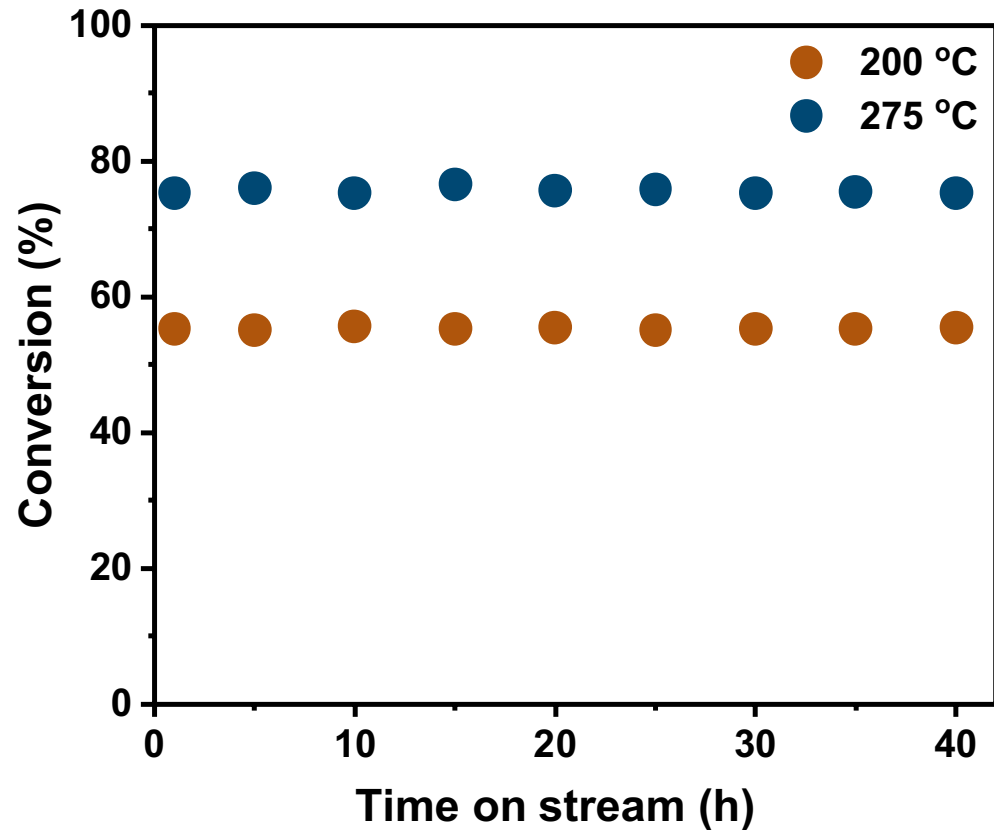


## 2. Progress and Outcome

# Metal catalyst alone doesn't promote the chain growth beyond dimerization



Metal supported catalyst



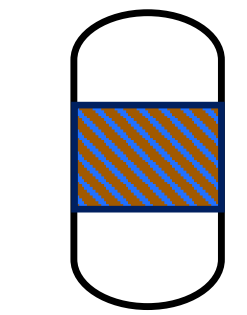
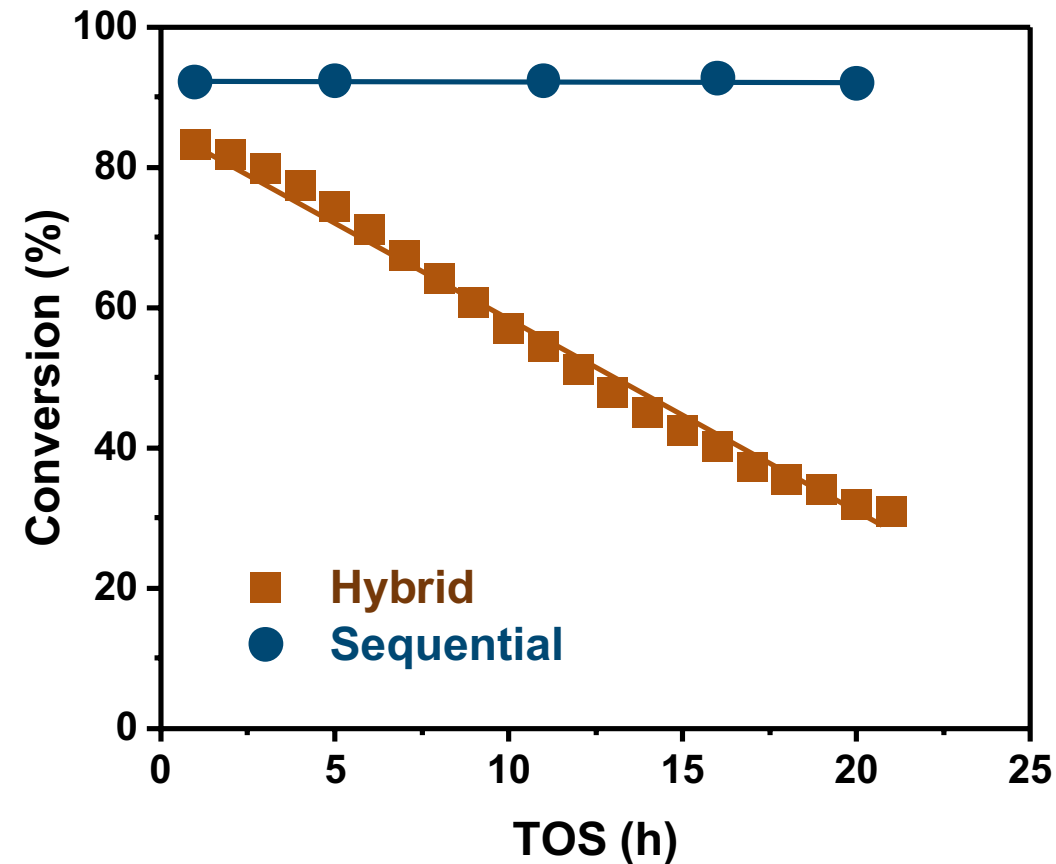
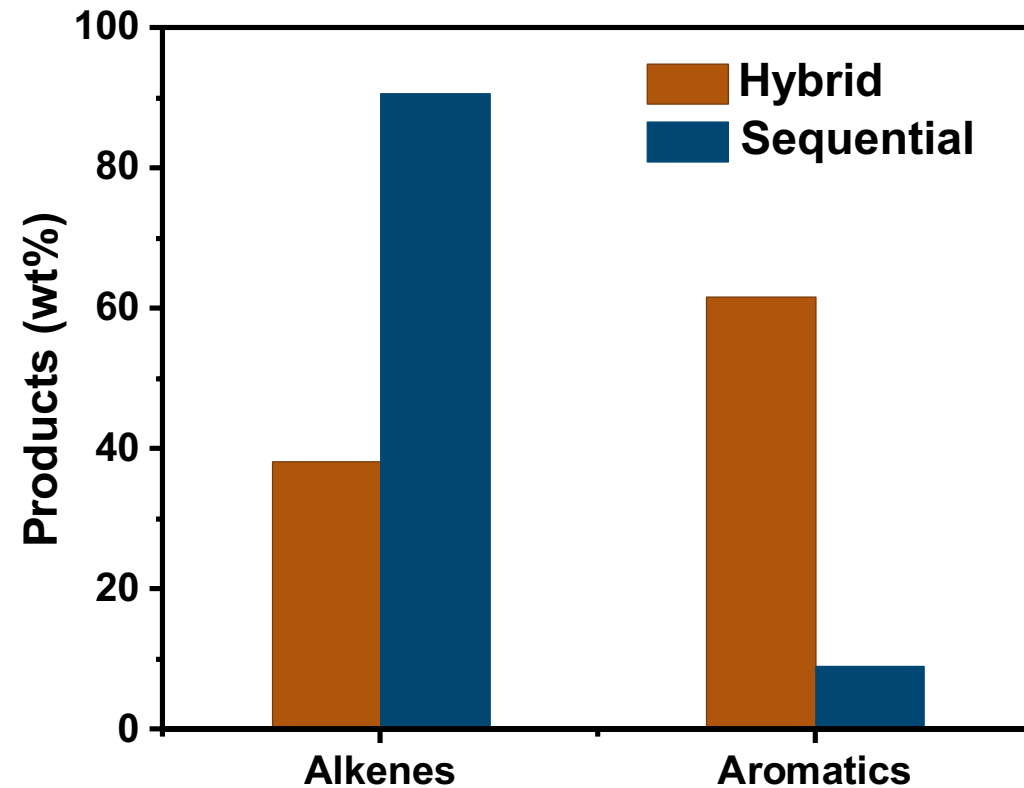
*Operating Conditions: WHSV: 0.8 h<sup>-1</sup>, Feedstock: equimolar C<sub>2</sub>-C<sub>3</sub>, Pressure: 100 psi, Temperature: 200 - 275 °C*

- Co-oligomerization of ethylene (C<sub>2</sub><sup>=</sup>) and propylene (C<sub>3</sub><sup>=</sup>) using the baseline metal catalyst at varying temperature in a plug flow reactor system

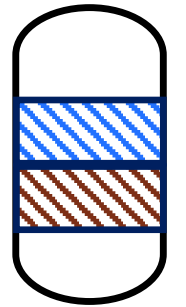
- Products obtained from C<sub>2</sub><sup>=</sup> and C<sub>3</sub><sup>=</sup> co-oligomerization are primarily between C<sub>4</sub>-C<sub>6</sub> range with C<sub>5</sub> being the major product
- Significant cross oligomerization between C<sub>2</sub><sup>=</sup> and C<sub>3</sub><sup>=</sup> with minimal chain growth
- Outcome:** Optimized metal composition and loading for co-oligomerization of C<sub>2</sub><sup>=</sup> and C<sub>3</sub><sup>=</sup>

## 2. Progress and Outcome

# Sequential catalyst bed is active and selective to the oligomerized products



Hybrid: Metal supported on zeolite



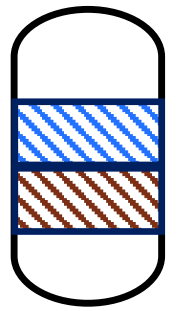
Sequential: Metal followed by acid catalyst

*Operating Conditions: WHSV: 0.8 h<sup>-1</sup>, Feedstock: equimolar C<sub>2</sub>-C<sub>3</sub>, Pressure: 100 psi, Temperature: 275 °C*

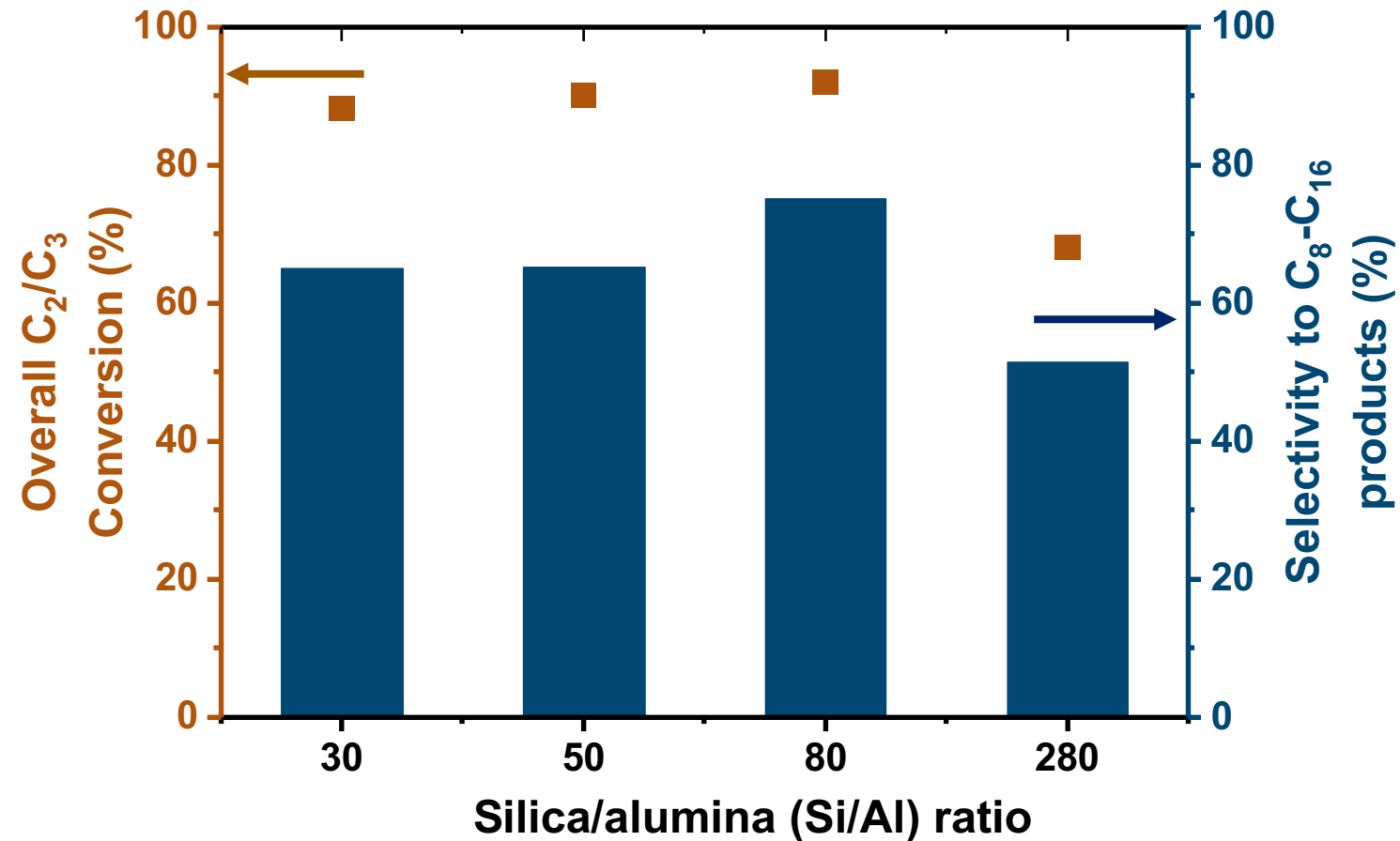
- Both hybrid and sequential catalyst exhibited ~70% selectivity to C<sub>8</sub>-C<sub>16</sub> compounds
- Hybrid catalyst produced higher fraction of light olefins and **6x higher aromatics** compared to sequential catalyst
- Hybrid catalyst suffers severe deactivation** compared to sequential catalyst

## 2. Progress and Outcome

**Moderate acidity is required to maintain the balance between product selectivity and conversion**



**Sequential:**  
Metal followed by  
acid catalyst

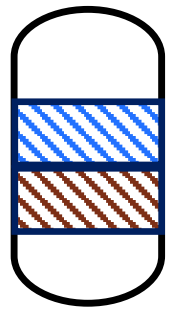


*Operating Conditions: WHSV: 0.8 h<sup>-1</sup>, Feedstock: equimolar C<sub>2</sub>-C<sub>3</sub>, Pressure: 100 psi, Temperature: 275 °C*

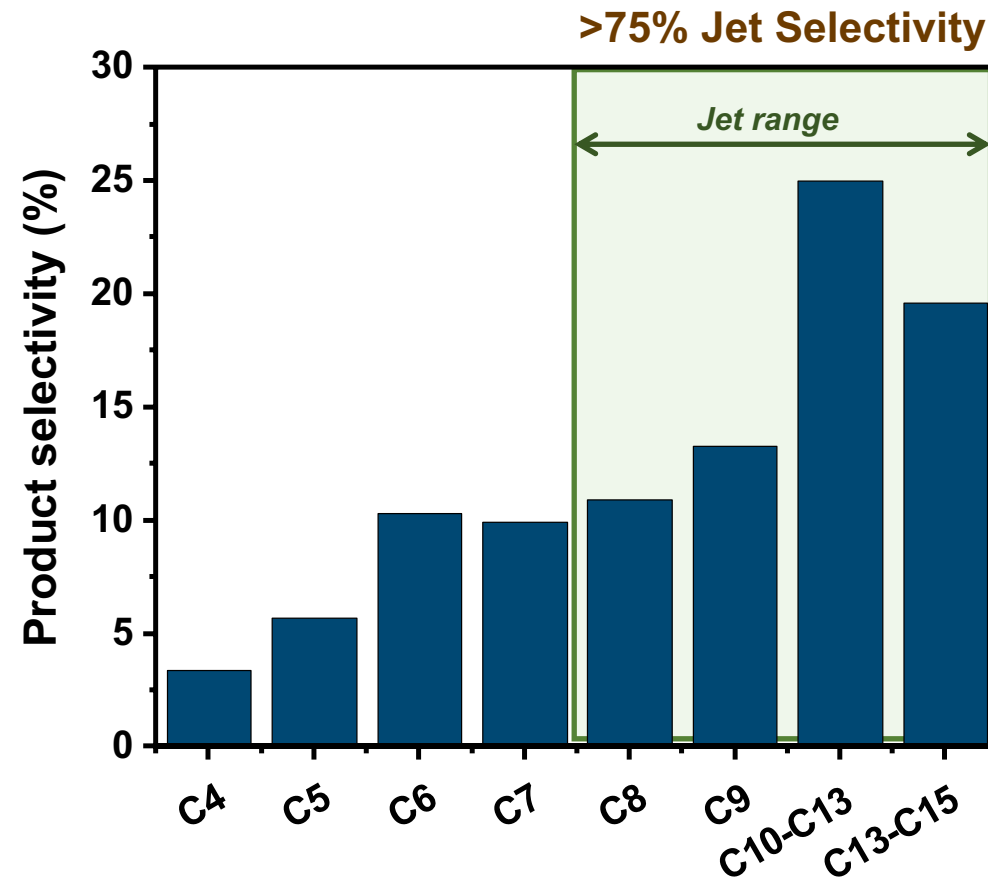
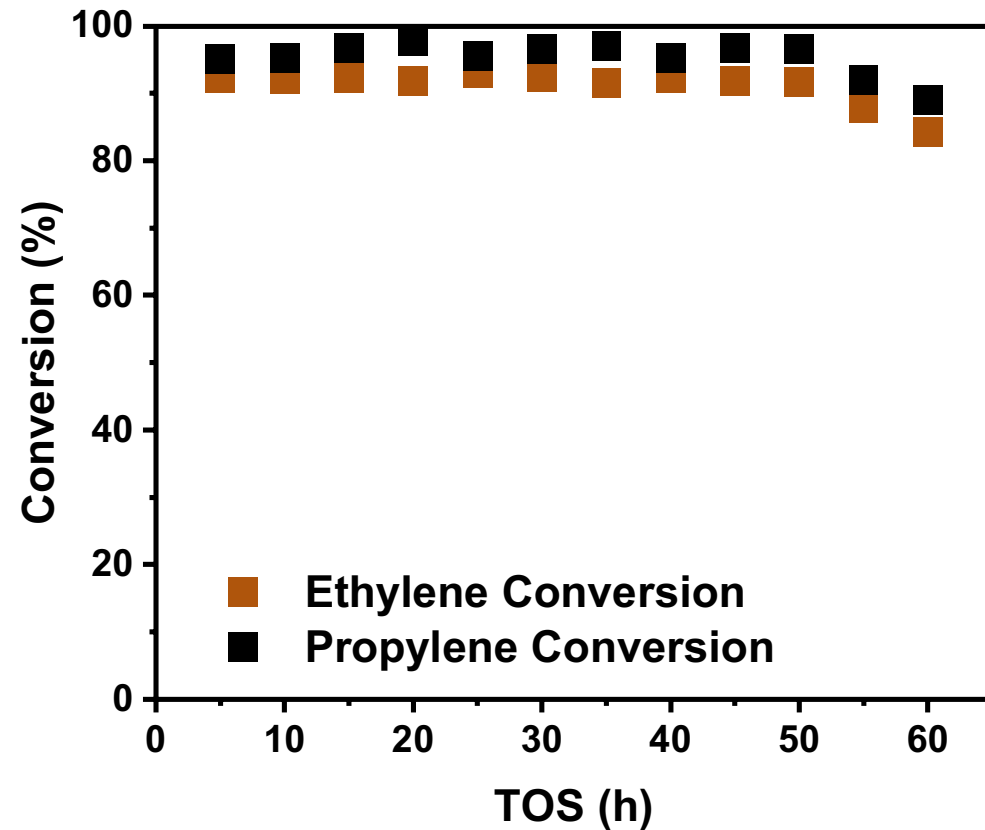
- Catalyst with moderate acidity exhibited both higher conversion and selectivity to desired jet range products
- Higher acidity (Si/Al:30) tend to have cracking as side reaction

## 2. Progress and Outcome

# Demonstrated the sequential catalyst bed system with high selectivity and activity



**Sequential:**  
Metal followed by  
acid catalyst



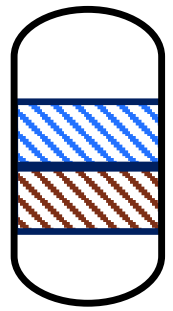
- Single step co-oligomerization of  $C_2^=$  and  $C_3^=$  using sequential bed containing metal and acid catalyst

*Operating Conditions: WHSV: 0.8 h<sup>-1</sup>, Feedstock: equimolar C<sub>2</sub>-C<sub>3</sub>, Pressure: 100 psi, Temperature: 275 °C*

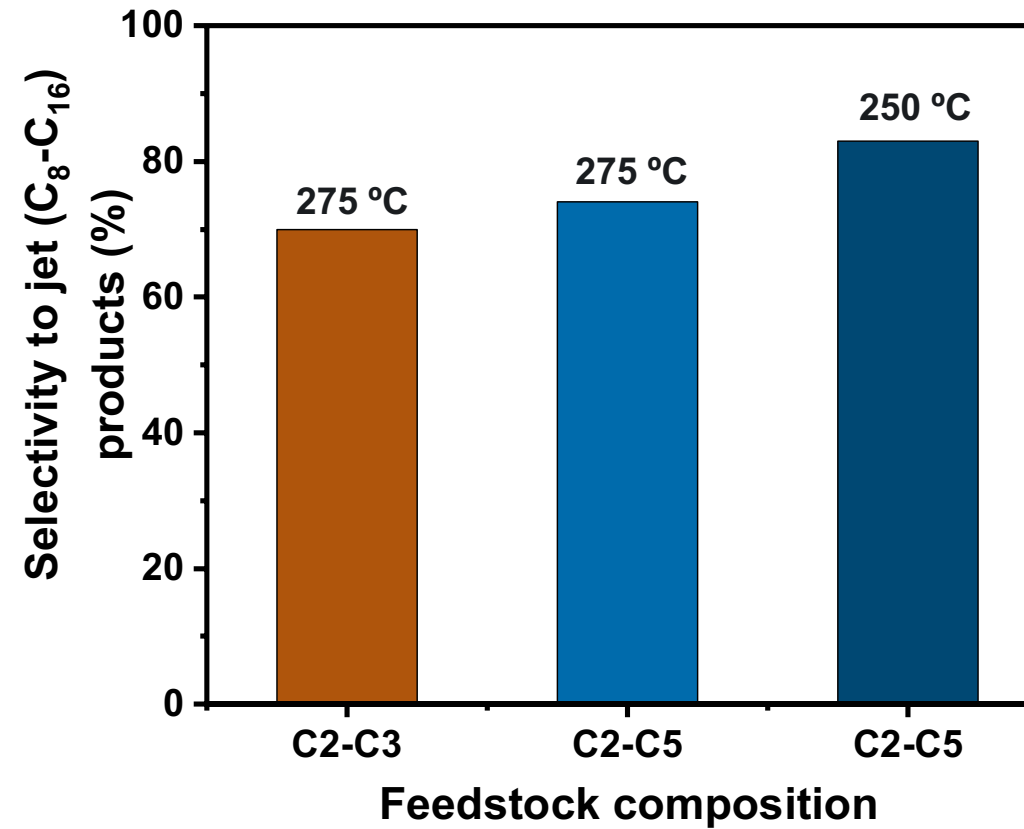
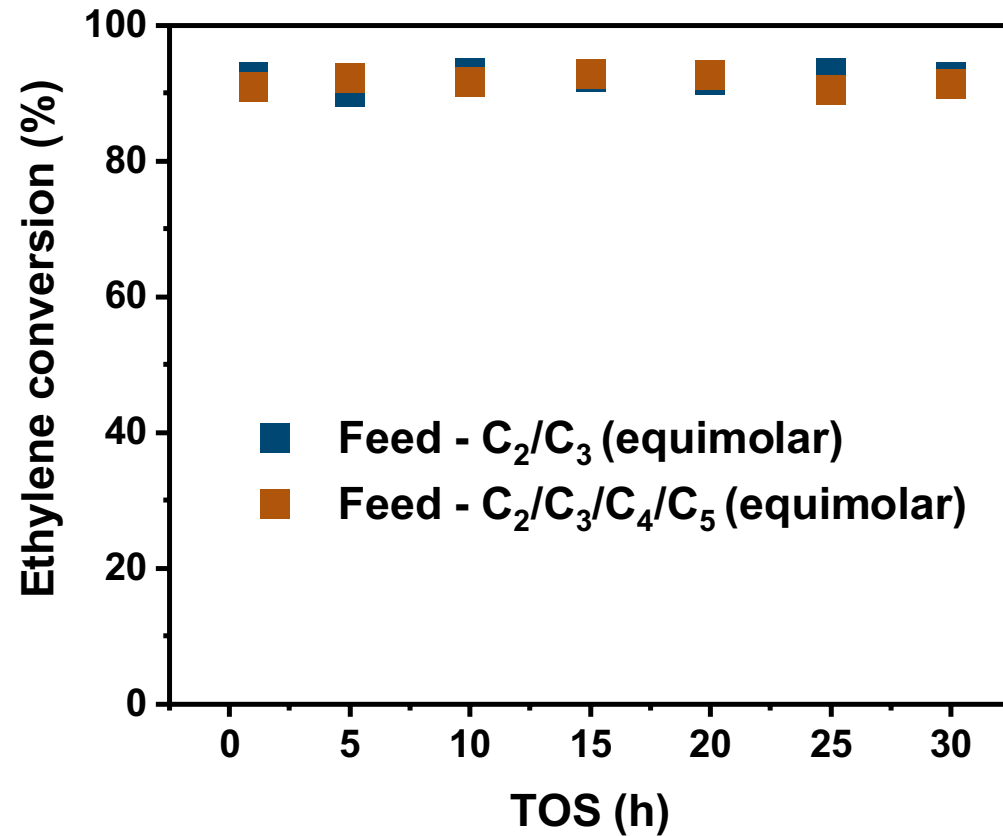
- >90% conversion of both ethylene and propylene; >75% selectivity to jet range (C<sub>8</sub>-C<sub>16</sub>) olefins
- Stability of the sequential catalyst system— ~ 50 h continuous time on stream

## 2. Progress and Outcome

# Co-oligomerization of mixed olefins ( $C_2-C_5$ ) performs similar to the $C_2-C_3$



**Sequential:**  
Metal followed by  
acid catalyst

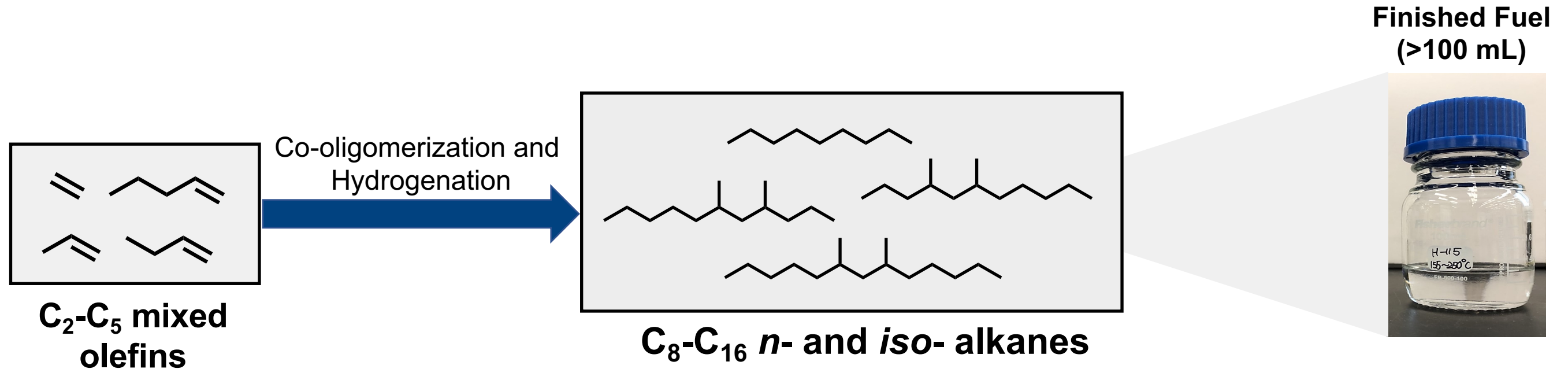


**Operating Conditions:** WHSV:  $0.8 \text{ h}^{-1}$ , Feedstock: equimolar  $C_2-C_5$ , Pressure: 100 psi, Temperature: 250 - 275 °C

- Composition of olefin feedstock merely affect ethylene conversion and the selectivity to jet range ( $C_8-C_{16}$ ) remains the same
- Higher selectivity obtained at lower temperature could be attributed to lower cracking activity
- Demonstrated co-oligomerization of mixed olefin ( $C_2-C_5$ ) as feedstock**

## 2. Progress and Outcome

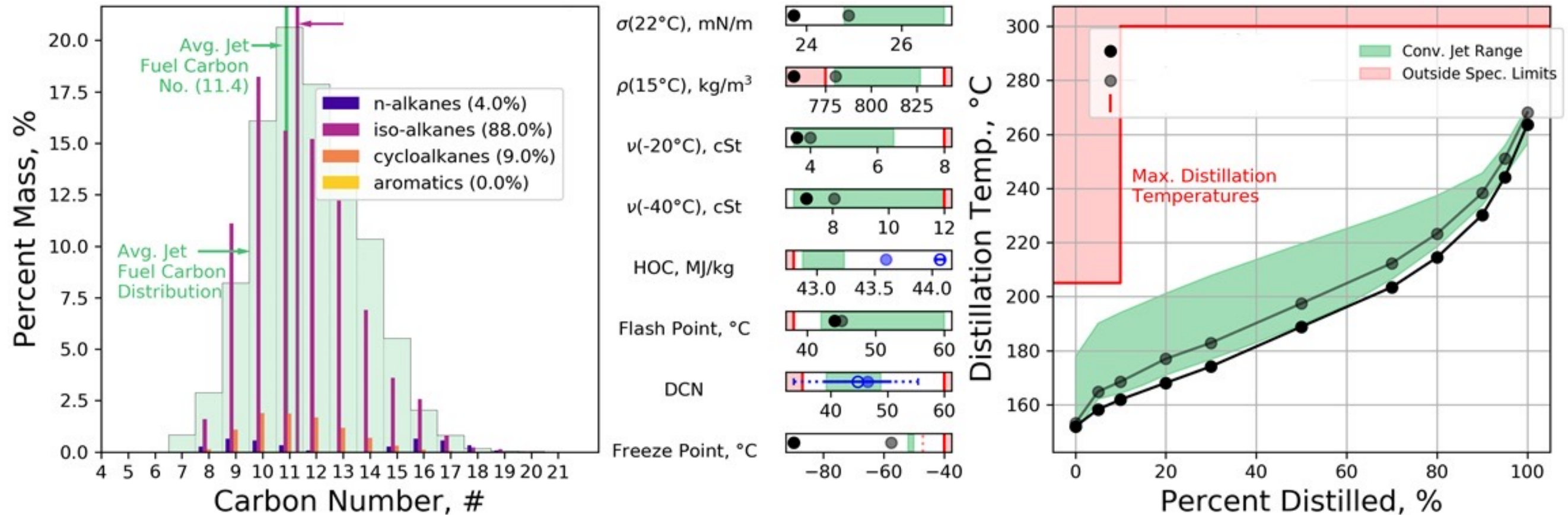
**Produced SAF sample >100mL for the fuel property analysis**



- Completed the co-oligomerization followed by hydrogenation to generate the finished fuel
- Generated > 100mL of finished fuels to conduct the initial Tier  $\alpha$  and Tier  $\beta$  sustainable aviation fuel property analysis at Washington State University

## 2. Progress and Outcome

# Co-oligomerized samples met the Tier $\alpha$ and Tier $\beta$ fuel property requirements at 50% blend level



- Density of neat fuel low – further adjustment in composition (n-, iso- and cycloalkane) is required
- Properties of jet fuel produced by co-oligomerization process meeting Tier  $\alpha$  and Tier  $\beta$  ASTM standard at  $\geq 50\%$  blend level

## 2. Progress and Outcomes

# Research progress and timeline



**FY22:** Optimization hybrid and /or sequential catalyst to demonstrate co-oligomerization of mixed olefins (C<sub>2</sub>-C<sub>5</sub>) to produce ~75% jet range compounds.

- Developed sequential catalyst containing both metal and acid sites and demonstrated co-oligomerization
- **Successfully completed Go/No-Go**

**FY23:** Demonstrate 100 h time-on-stream using an integrated reactor system (metal and acid catalyst bed) and fuel properties meeting Tier  $\alpha$  and Tier  $\beta$  ASTM standards, and complete process analysis

- Catalyst deactivation – spent catalyst characterization
- Fine tuning catalyst structure and process parameters – improve catalyst lifetime
- Detailed process analysis

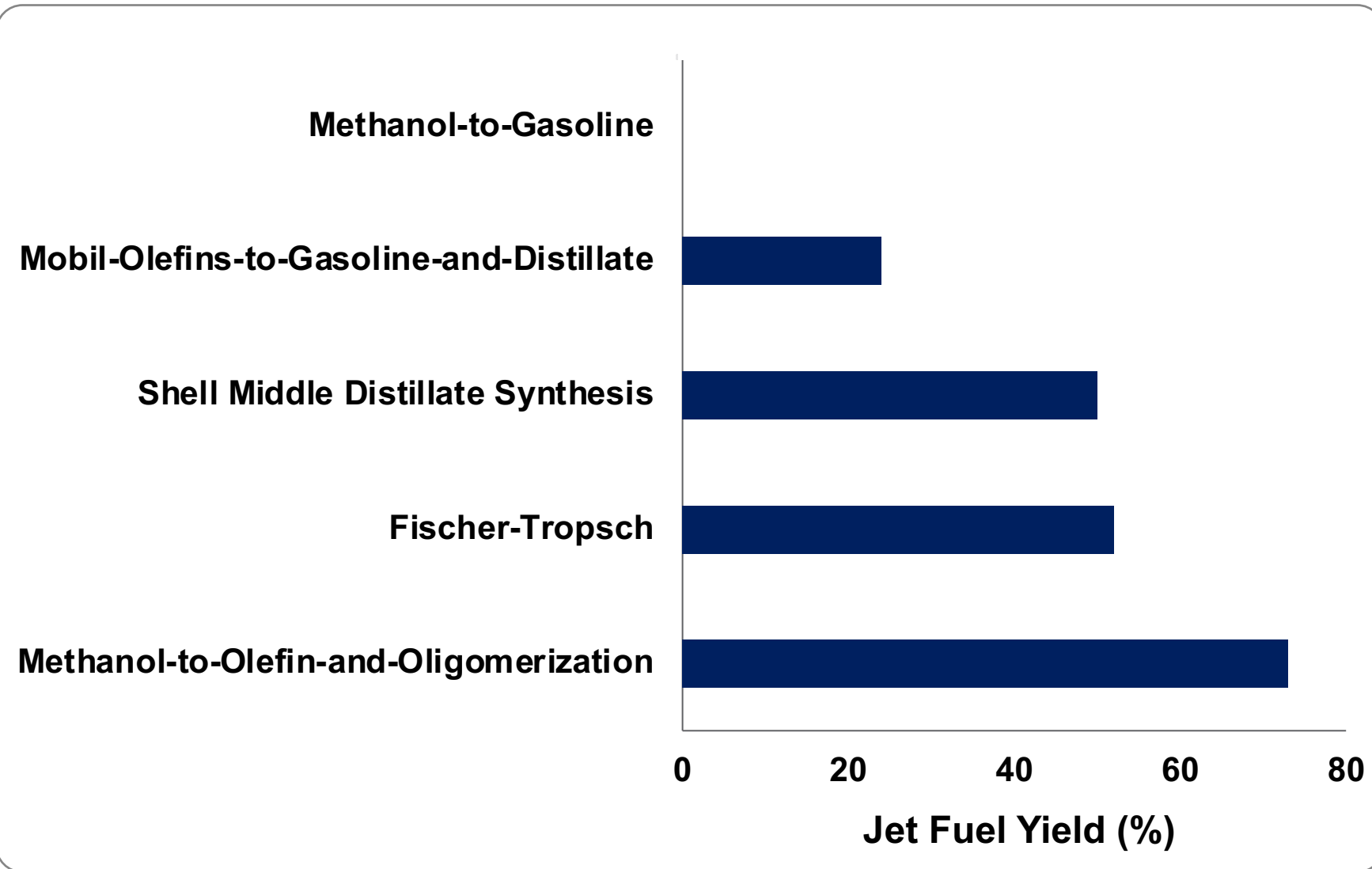
- Synthesize powdered catalyst and provide to industrial partner
- Engineered catalyst testing and demonstration

**FY24:** Engineered catalyst development, process integration, and demonstration



### 3. Impact

## Supports near term decarbonization of aviation industry by maximizing SAF yield

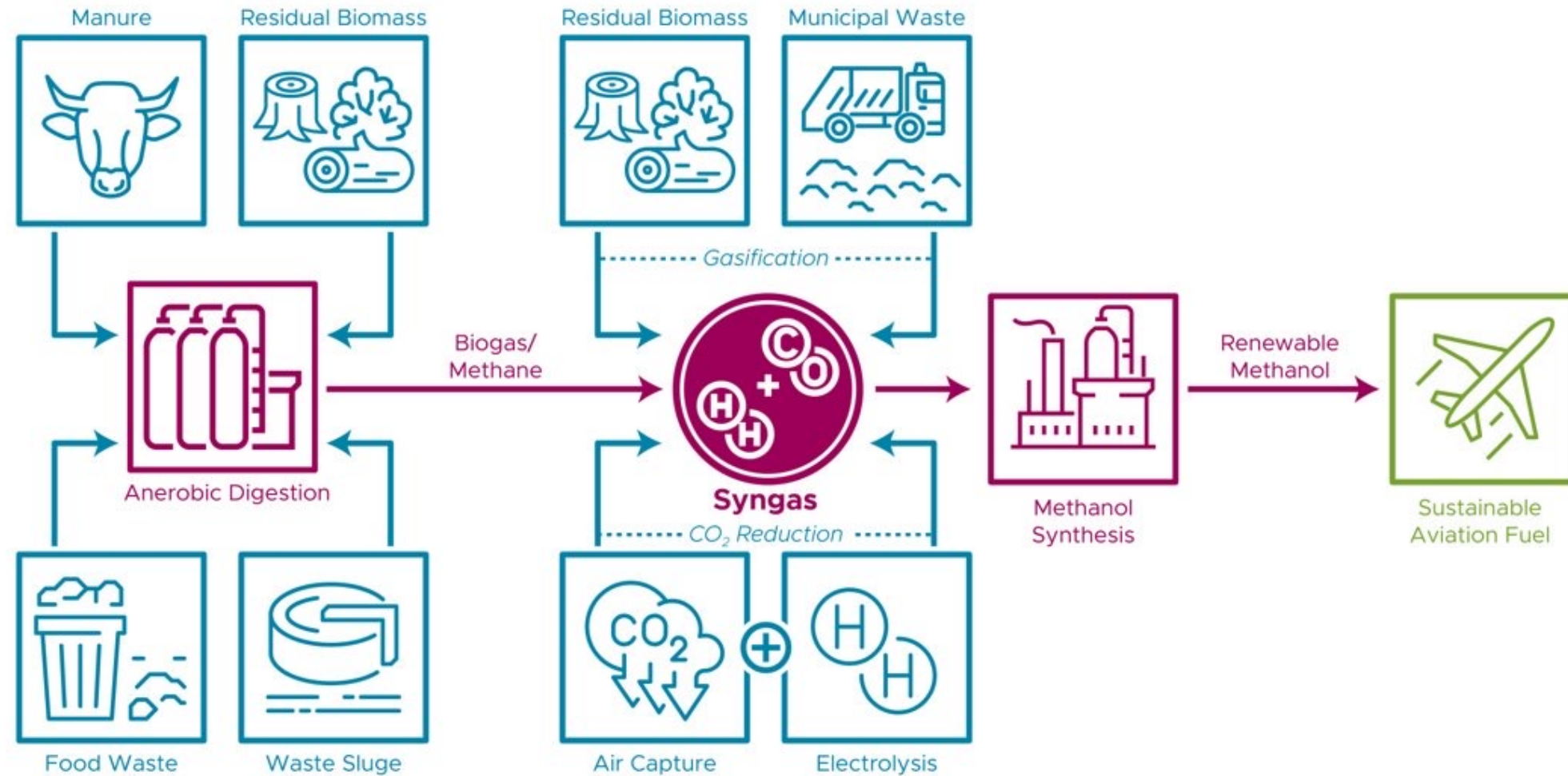


*Jet fraction (SAF) yield from the major syngas conversion technologies compared to the Methanol-to-Olefin-and-Oligomerization (this work).*

- Maximize the carbon efficiency from syngas to SAF and reduce/ eliminate the other fuel fractions
- Leverage existing commercial process for the transformation of syngas intermediate to produce SAF
- Potential to de-risk and integrate the unit operations at a rapid rate to support the aviation industries net zero emission goals
- Industrial institution interest (to convert methanol to SAF)

### 3. Impact

## Supports the long-term aviation industry decarbonization goals



- Syngas obtained from waste source (e.g., MSW) can potentially provide ~65% of the global aviation fuel requirement by 2050
- Renewable electricity can enable the CO<sub>2</sub> conversion to syngas and then to SAF – further increase the global SAF contribution by 2050

## Summary

# Enable the technology towards commercialization

### Overview

Develop a  $C_2$ – $C_5$  co-oligomerization catalyst and demonstrate an efficient pathway to produce sustainable aviation fuel (SAF) from syngas via a mixed olefin intermediate

### Approach

Developing catalyst that enables integrating both metal and acid catalysis pathways which are key to facilitate co-oligomerization of  $C_2$  and  $C_{3+}$  to produce SAF

### Impact

Demonstrated co-oligomerization process with mixed olefins ( $C_2$ – $C_5$ ) as feedstock to produce jet range hydrocarbons – Meeting Tier  $\alpha$  and Tier  $\beta$  standard at  $\geq 50\%$  blend level

### Progress & Outcome

Developed the hybrid catalyst system and optimized reaction condition to achieve  $C_2$ – $C_5$  co-oligomerization with  $\geq 75\%$  selectivity to jet range ( $C_8$ – $C_{16}$ ) products

### Future Work

Demonstrate co-oligomerization to produce jet range product with catalyst lifetime  $>500$  h using engineered catalyst

# Quad Chart Overview

## Timeline

- Project start date: October 2022
- Project end date: September 2025

	FY 22	Total Award
DOE Funding	\$530,000	\$1,630,000 (FY 2022-2024)
Project Cost Share*	NA	NA

**TRL at Project Start: 2**  
**TRL at Project End: 4**

## Project Goal:

Develop sustainable aviation fuel (SAF) production process from syngas derived mixed olefins and demonstrate the commercial path to achieve 70% reduction in greenhouse gas emissions.

## End of Project Milestone:

Production of 1 L of finished jet fuel from syngas derived mixed olefins with fuel properties meeting Tier  $\alpha$  and Tier  $\beta$  standard and completion of 500 hours continuous operation (with intermittent regeneration) using engineered catalysts with >75% yield to fuel-range products.

**Funding Mechanism: AOP Lab Call**

## Project Partners

- Washington State University
- Topsoe

## Acknowledgement

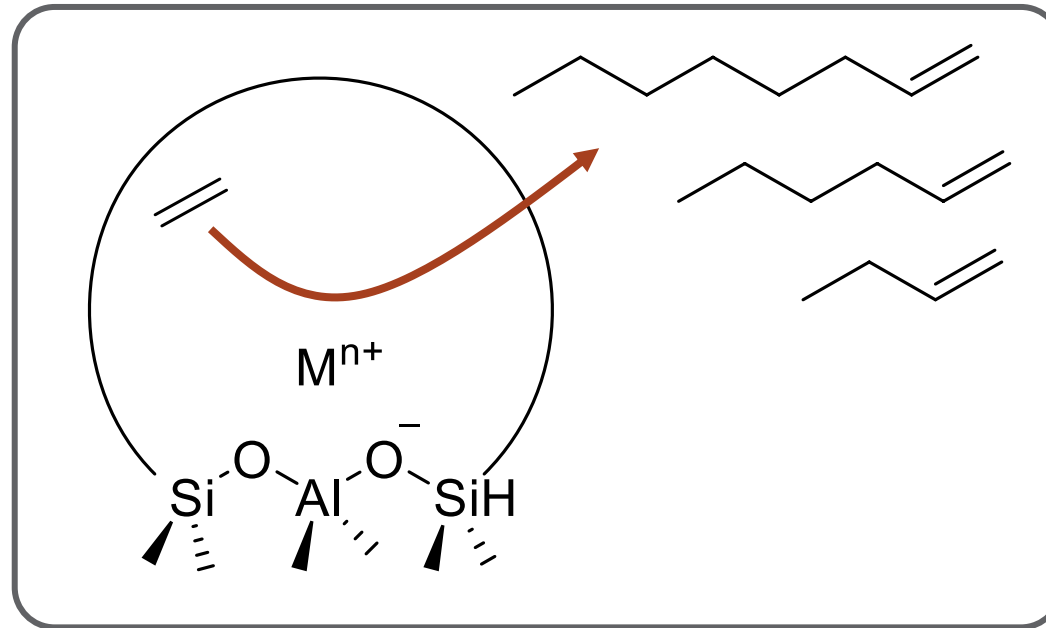
- ❑ **Pacific Northwest National Laboratory:** Udishnu Sanyal, Mond Guo, Anthony Giduthuri, Laura Meyer
- ❑ **Washington State University:** Joshua Heyne, Harrison Yang,
- ❑ **Topsoe:** Pablo Beato, Esben Taarning
- ❑ **Bioenergy Technology Office:** Sonia Hammache

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**Thank you**



### C<sub>2</sub> oligomerization



### C<sub>3+</sub> oligomerization

