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LI-CYCLE HUB PROJECT August 6, 2021 CHEMICALS & REAGENTS – REVISED FOR SEQR



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Organic Reagents

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- Solvent Extraction Organics



USE OF REAGENTS

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The purpose of this presentation is to provide an overview of the reagents that will be used at Li-Cycle's North America Hub #1

- Li-Cycle's Hub will use various reagents to extract and recover nickel sulfate, cobalt sulfate, lithium carbonate, manganese carbonate, and other metal products from black mass concentrate.
- Black mass concentrate will be shipped to the Hub from other facilities for processing and is derived from the recycling of lithium-ion batteries.
- As part of this presentation, Li-Cycle have identified the risks associated with reagents that will be used at the Hub and the measures that will be implemented to eliminate and/or abate those risks.





INORGANIC REAGENTS

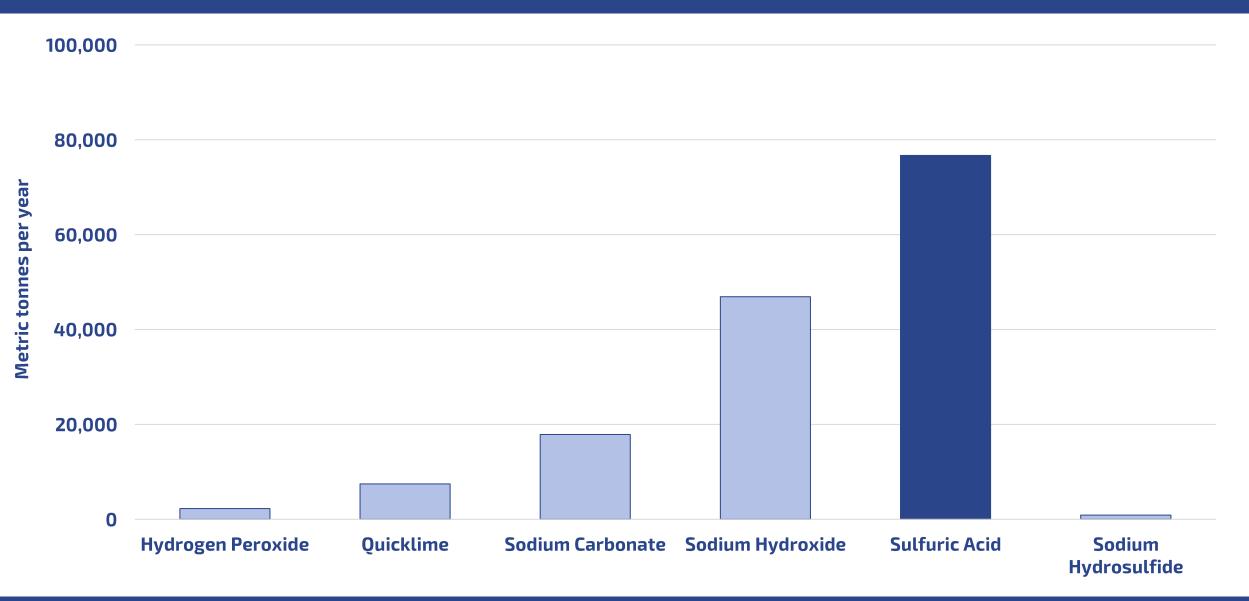
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93% SULFURIC ACID

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REAGENT USAGE (Average)



93% SULFURIC ACID



Physical Properties

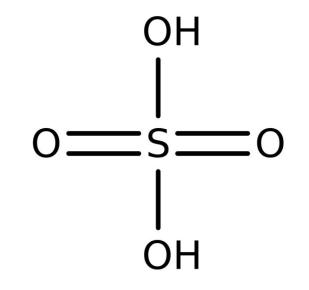
- Molecular form: H₂SO₄
- Colorless, odorless, oily liquid
- Denser than water (SG = 1.83)
- Highly corrosive
- Oxidant and strongly acidic

Incompatible Chemicals

- Strong bases
- Strong acids
- Hydration reaction releases heat

Industrial Uses

- Fertilizer production (ammonium sulfate/phosphate)
- Chemical industry (detergents, pharma)
- Water treatment
- Paints, coatings





General Hazards

Flammable Properties:	Does not burn
Extinguishing Media:	Not combustible. Use extinguishing agent suitable for surrounding fire. Only use water to keep non-leaking, fire-exposed containers cool. Do not use water or water-based extinguishing agents.
Specific Hazards:	Contact with water causes frothing and spattering. Reacts with some metals to produce flammable gas. In a fire, the following hazardous materials may be generated: sulfur oxides.
Chemical Stability:	Normally stable
Incompatible Materials:	Reacts with many chemicals, including, water. Corrosive to aluminum alloys, carbon steel, and other metals.
Hazardous Decomposition Products:	None known
Possibility of Hazardous Reactions:	None known

93% SULFURIC ACID

Specific Hazards

- Contact with water causes frothing and spattering
- Reacts with some metals to produce flammable gas
- In a fire, the following hazardous materials may be generated: sulfur oxides

Li-Cycle Mitigations for Specific Hazards

- Controlled injection of sulfuric acid into process tanks
- Instrumentation (i.e., pH, temperature), monitoring, and control systems (i.e., cooling water coils for select process tanks)
- Low temperature aqueous operation (<175°F)
- ~12 months of pilot plant operation using sulfuric acid



Health Hazards



Skin:	Corrosive
Eyes:	Corrosive
Inhalation:	not expected to be a hazard unless heated or misted. Toxic / fatal if inhaled.
Ingestion:	Corrosive
Carcinogenicity	Mists are carcinogenic to humans.
/ Teratogenicity Embryotoxicity	Not known to harm the unborn child
Reproductive Toxicity:	Not known to be a reproductive hazard
Mutagenicity:	Not known to be a mutagen



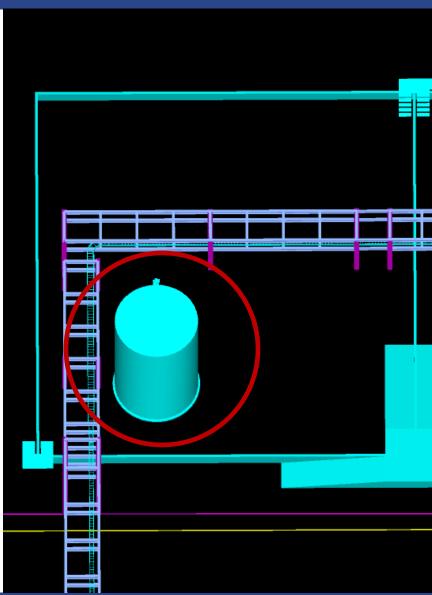
93% SULFURIC ACID

Bulk Delivery

- Rail delivery: 85 tons each car
- Three (3) tank cars per day
- Roughly 22 tank cars per week
- Unloaded by dedicated pumping system from rail cars

Bulk Storage

- One (1) bulk storage tank installed
- ~2.5 days of storage
- Space left for installation of 2nd tank, if needed
- Separate spill containment area with 3 ft walls



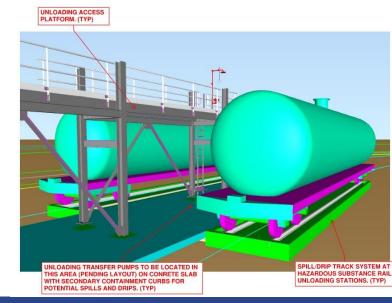


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93% SULFURIC ACID

Rail Unloading

- Separate spill containment under sulfuric acid rail cars & pumping systems
- Overhead unloading arms
- Spill containment system will be coated to resist sulfuric acid corrosion. Coating to be discussed with supplier.







PVS CHEMICALS – BUFFALO, NY

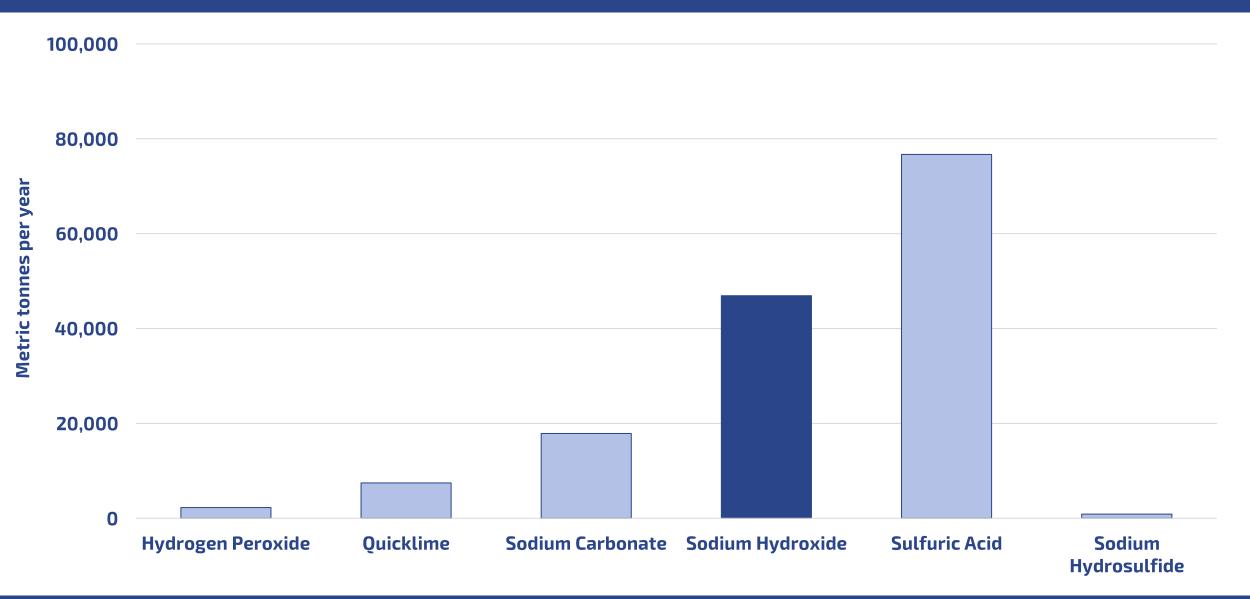






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REAGENT USAGE (Average)



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Physical Properties

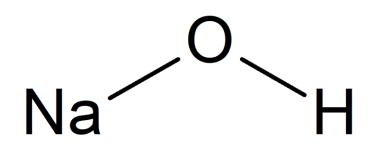
- Molecular form: NaOH
- Known as 'caustic soda'
- Colorless, odorless,
- Denser than water (SG = 1.50)
- Highly corrosive
- Strongly basic

Incompatible Chemicals

- Strong bases
- Strong acids
- Hydration reaction releases heat

Industrial Uses

- Chemical industry (detergents, sodium salts)
- Water treatment
- Pulp & paper
- Cleaning agents





General Hazards

Flammable Properties:	Does not burn
Extinguishing Media:	Not combustible. Use extinguishing agent suitable for surrounding fire. Do not use carbon dioxide
Specific Hazards:	Contact with water causes frothing and spattering. Reacts with metals to produce flammable gas. Toxic sodium oxide fumes can be generated at high temperatures.
Chemical Stability:	Normally stable
Incompatible Materials:	Highly reactive. Reacts with many chemicals, including, water, organic acids, inorganic acids, oxidizing agents, metals. Corrosive to aluminum alloys, carbon steel, and other metals.
Hazardous Decomposition Products:	None known
Possibility of Hazardous Reactions:	None known

Specific Hazards

- Contact with water causes frothing and spattering
- Reacts with metals to produce flammable gas
- Toxic sodium oxide fumes can be generated at high temperatures

Li-Cycle Mitigations for Specific Hazards

- Controlled injection of sodium hydroxide into our process tanks
- Instrumentation and monitoring (i.e., pH, temperature)
- Low temperature aqueous operation (<175°F)
- ~12 months of pilot plant operation using sodium hydroxide





Health Hazards



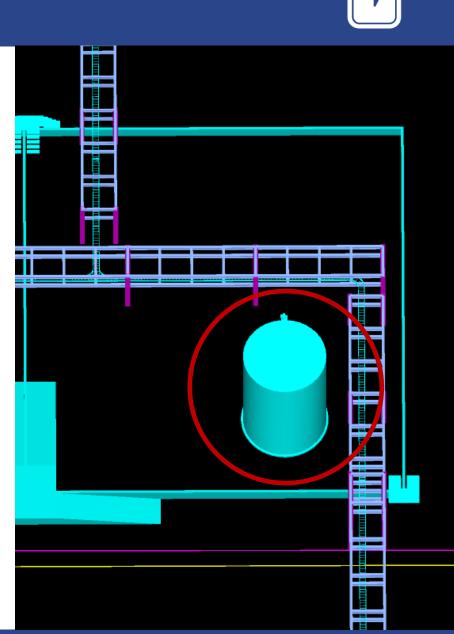
Skin:	Corrosive
Eyes:	Corrosive
Inhalation:	Not expected to be an inhalation hazard unless it becomes an airborne dust or mist. Can cause severe irritation of the nose and throat.
Ingestion:	Corrosive
Carcinogenicity	Not known to cause cancer
Embryotoxicity	Not known to harm the unborn child
Reproductive Toxicity:	Not known to be a reproductive hazard
Mutagenicity:	Not known to be a mutagen

Bulk Delivery

- Rail delivery: 85 tons each car
- 2 tank cars per day
- Roughly 14 tank cars per week
- Unloaded by dedicated pumping system from rail cars

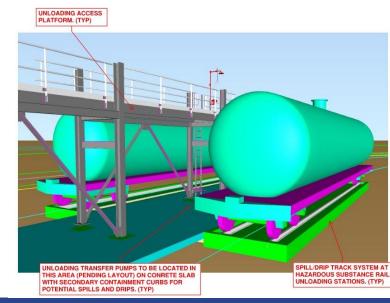
Bulk Storage

- One (1) bulk storage tank installed
- ~3.5 days of storage
- Space left for installation of 2nd tank, if needed
- Separate spill containment area with 3 ft walls



Rail Unloading

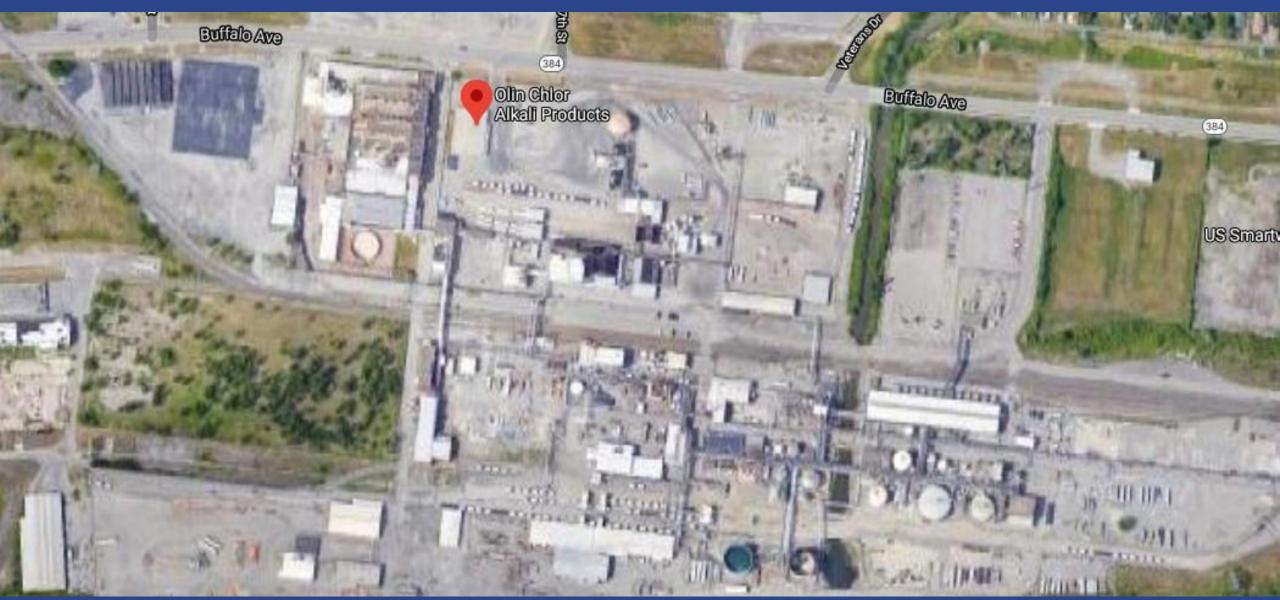
- Separate spill containment under sodium hydroxide rail cars & pumping systems
- Overhead unloading arms
- Spill containment system will be coated to resist sodium hydroxide corrosion. Coating to be discussed with supplier.





OLIN – NIAGARA FALLS, NY

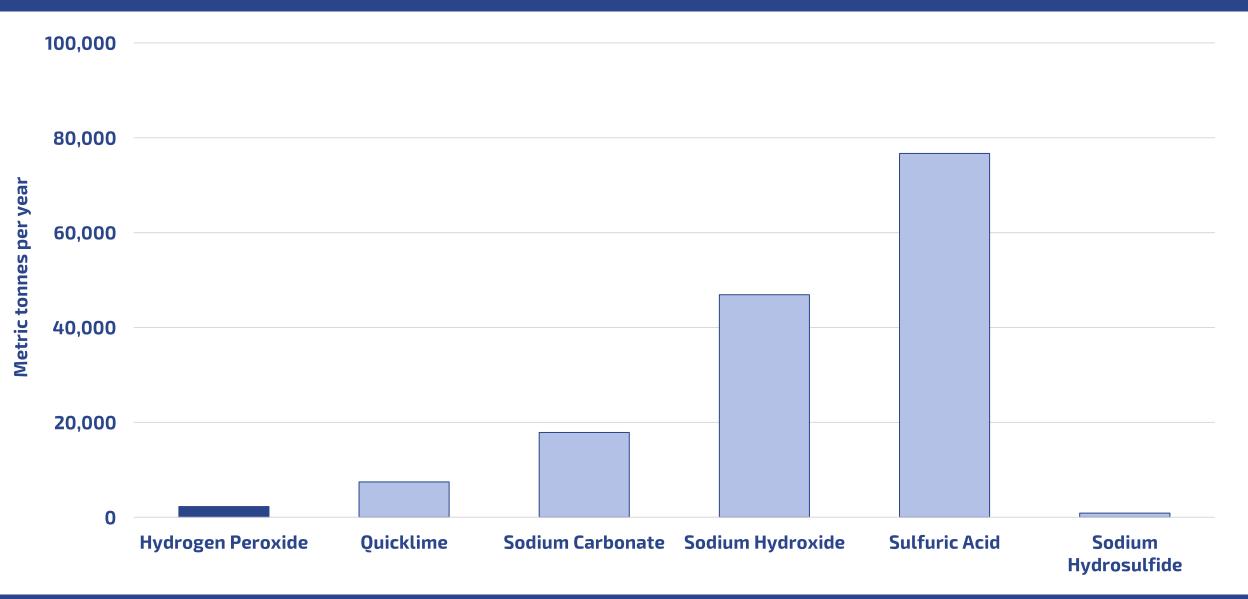






Private & Confidential

REAGENT USAGE (Average)





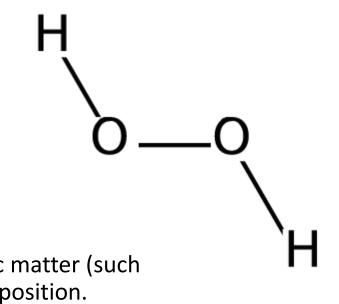
- Molecular form: H_2O_2
- Colorless, odorless, liquid
- Denser than water (SG = 1.20)
- Reactive and oxidizing agent
- Acidic

Incompatible Chemicals

- Strong bases (unless in a controlled process)
- Strong acids
- Combustible materials
- Contact with metals, metallic ions, alkalis, reducing agents and organic matter (such as alcohols or terpenes) may produce self-accelerated thermal decomposition.

Industrial Uses

- Bleaching
- Disinfection
- Propellant (at concentrations > 70%)





General Hazards

Flammable Properties:	Non flammable
Extinguishing Media:	Water. Do not use any other substance.
Specific Hazards:	In closed unventilated containers, risk of rupture due to the increased pressure from decomposition. Contact with combustible material may cause fire.
Chemical Stability:	Stable under normal conditions. Decomposes on heating. Stable under recommended storage conditions.
Incompatible Materials:	Combustible materials. Contact with metals, metallic ions, alkalis, reducing agents and organic matter (such as alcohols or terpenes) may produce self-accelerated thermal decomposition.
Hazardous Decomposition Products:	On decomposition product releases oxygen which may intensify fire.
Possibility of Hazardous Reactions:	Contact with organic substances may cause fire or explosion. Avoid excessive heat, contamination, exposure to UV-rays, pH variations.

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Incompatible Materials

- Contact with metals, metallic ions, alkalis, reducing agents and organic matter (such as alcohols or terpenes) may produce self-accelerated thermal decomposition.
- Strong acids and/or strong bases
- Combustible materials

Li-Cycle Mitigations for Specific Hazards

- No combustible materials stored nearby
- Distance between hydrogen peroxide storage and solvent extraction area
- Pipelines will be self-draining to the extent feasible.
- Pipelines and tanks will be equipped with a pressure relief mechanism per industry best practices (hydrogen peroxide suppliers already engaged for design input)
- ~12 months of pilot plant operation using hydrogen peroxide

Health Hazards



Skin:	Corrosive
Eyes:	Corrosive
Inhalation:	Not expected to be an inhalation hazard unless it becomes an airborne mist. Can cause severe irritation of the nose and throat.
Ingestion:	Corrosive
Carcinogenicity	Not known to cause cancer in humans
Embryotoxicity	Not known to harm the unborn child
Reproductive Toxicity:	Not known to be a reproductive hazard
Mutagenicity:	Not known to be a mutagen



Bulk Delivery

- Rail delivery: 85 tons each car
- Roughly 1 tank car per week
- Unloaded by dedicated pumping system from rail cars

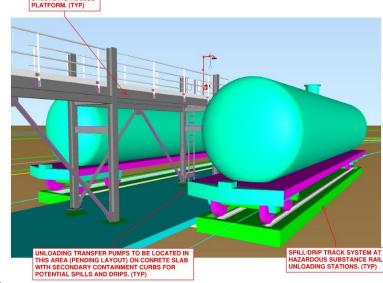
Bulk Storage

- One (1) bulk storage tanks
- ~12 days of storage
- Separate spill containment area with 3 ft walls



Rail Unloading

- Separate spill containment under hydrogen peroxide rail cars & pumping systems
- Overhead unloading arms
- Spill containment system will be coated to be compatible with hydrogen peroxide. Material/coating to be discussed with supplier.



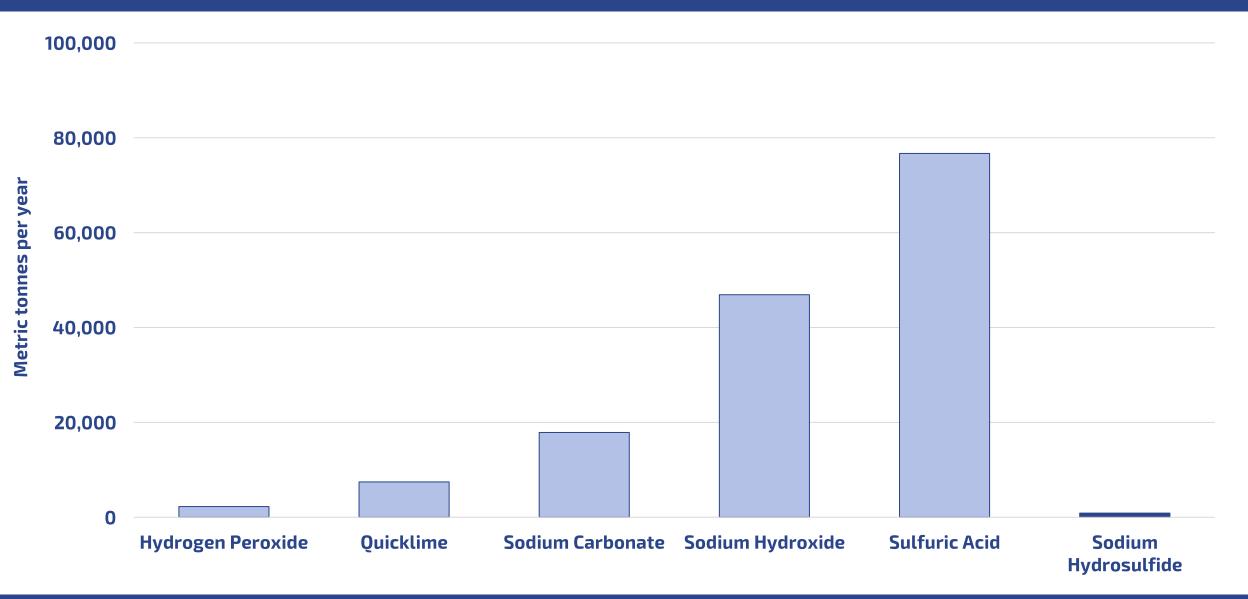




SODIUM HYDROSULFIDE HYDRATE

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REAGENT USAGE (Average)





Physical Properties

- Molecular form: NaHS•2H₂O
- Solid (flake) material, yellow

Incompatible Chemicals

- Strong acids (unless in a controlled process)
- Strong bases (unless in a controlled process)
- Strong oxidizers (unless in a controlled process)
- Heat & sources of ignition
- Do not store in contact with metals such as copper, zinc, or aluminum. Stainless steel is preferred. Carbon steel is acceptable.

Industrial Uses

- Flotation agent
- Metals precipitant
- Cloth & paper manufacturing

Na — S

• XH₂O



Fire Fighting Measures Hazards

	Non-combustible – sodium hydrosulfide does not burn but may decompose upon heating to produce corrosive and/or toxic fumes.
Autoignition:	Not applicable.
Extinguishing Media:	Water. A heavy fog of water may be effective in knocking down vapors.

Li-Cycle Mitigations for Fire Hazards

- Dedicated storage area for sodium hydrosulfide hydrate bulk bags
- Low temperature aqueous operation (<175°F)
- Instrumentation and monitoring throughout the process area (i.e., gas monitors).
- Gas monitors will be for detecting very low levels of hydrogen sulfide gas (H₂S) and will have an audible siren and visual light alert.



Specific Hazards & Chemical Stability

Specific Hazards:	Self-contained breathing apparatus and full protective clothing must be worn in case of fire. Poisonous sulfur dioxide gas will be generated if this product burns.		
	Stable under normal conditions. Do not store near acids. Keep away from heat and sources of ignition.		

Li-Cycle Mitigations for Specific Hazards & Chemical Compatibility

- No acid pipelines will be run through the area using sodium hydrosulfide.
- Instrumentation and monitoring throughout the process area (i.e., gas monitors). Gas monitors will have an audible siren and visual light alert in the event of low-level H₂S gas detection.
- No heat or sources of ignition in the areas storing or using sodium hydrosulfide.



Hazardous Reactions & Decomposition Products

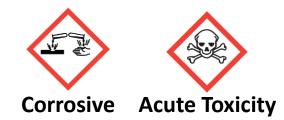
Incompatible Materials:	Strong oxidizers, acids, bases, certain metals	
Hazardous Decomposition Products:	Hydrogen sulfide (H ₂ S)	
Possibility of Hazardous Reactions:	Toxic H ₂ S gas when mixed with an acid or exposed to high heat sources such as a fire.	

Li-Cycle Mitigations for Specific Hazards & Chemical Compatibility

- No acid pipelines will be run through the areas storing or using sodium hydrosulfide.
- Instrumentation and monitoring throughout the process area (i.e., gas monitors). Gas monitors will
 have an audible siren and visual light alert in the event of low-level H₂S gas detection.
- All incompatible chemicals (sulfuric acid, hydrogen peroxide) are stored away from the area.
- All equipment will be closed and vented to a caustic scrubber for H₂S control. The scrubber will be connected to emergency/back-up power.



Health Hazards



Skin:	Corrosive
Eyes:	Corrosive
Inhalation:	Severe respiratory distress because of corrosivity.
Ingestion:	Headache, nausea, dizziness, confusion, and painful alkali burns to the esophagus.
Carcinogenicity	Not known to cause cancer in humans
Embryotoxicity	Not known to harm the unborn child
Reproductive Toxicity:	Not known to be a reproductive hazard
Mutagenicity:	Not known to be a mutagen

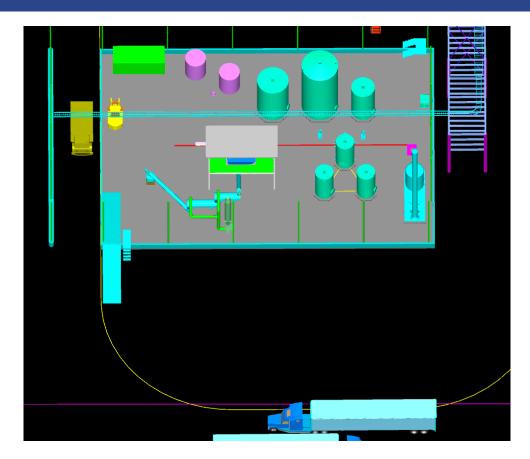
SODIUM HYDROSULFIDE HYDRATE

Bulk Delivery

- Truck delivery: 1 ton bags (~24 bags per ISO container)
- Roughly ½ truck per week
- Fully contained bag breaking system
- All equipment connected to emissions control device (i.e., caustic scrubber)

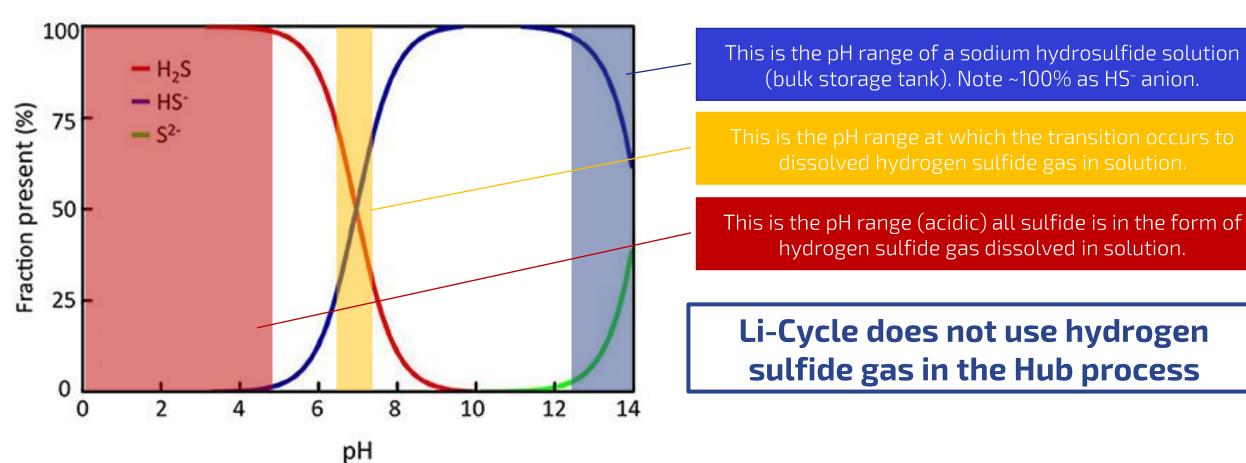
Bulk Storage

- Batch tank
- 30 hrs (maximum 3,500 gallons)
- No incompatible chemicals stored in, or passing through, sodium hydrosulfide area.
- Equipment will be constructed in secondary containment.





Sodium hydrosulfide is formed via the part neutralization of hydrogen sulfide (H₂S) with sodium hydroxide (NaOH)



The Hub process requires the injection of sodium hydrosulfide in to an acidic solution. Minimal H₂S generation is expected. Why does this work?

- 1. The acidic solution being processed has a high concentration of dissolved metals
- 2. Hydrosulfide prefers to precipitate metals rather than form H_2S in a metals-rich solution
- 3. Proper mixing and chemical injection helps to reduce H₂S formation
- 4. A scrubber will be used to capture H_2S gas that is produced

Me + HS⁻ → MeS +



Li-Cycle piloted this process without issue by applying appropriate control devices and engineering controls

HYDROGEN SULFIDE GAS



Physical Properties

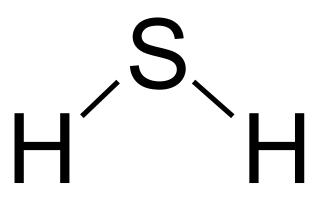
- Molecular form: H₂S
- Colourless gas
- Rotten eggs odor
- Denser than air (1.36 kg/m³)
- Highly corrosive
- Strongly basic

Incompatible Chemicals

- Highly reactive
- Increased risk of fire and explosion on contact with: metal oxides, oxidizing agents, strong bases
- In the presence of water, corrosive to: carbon steel
- Not corrosive to: aluminum alloys

Industrial Uses

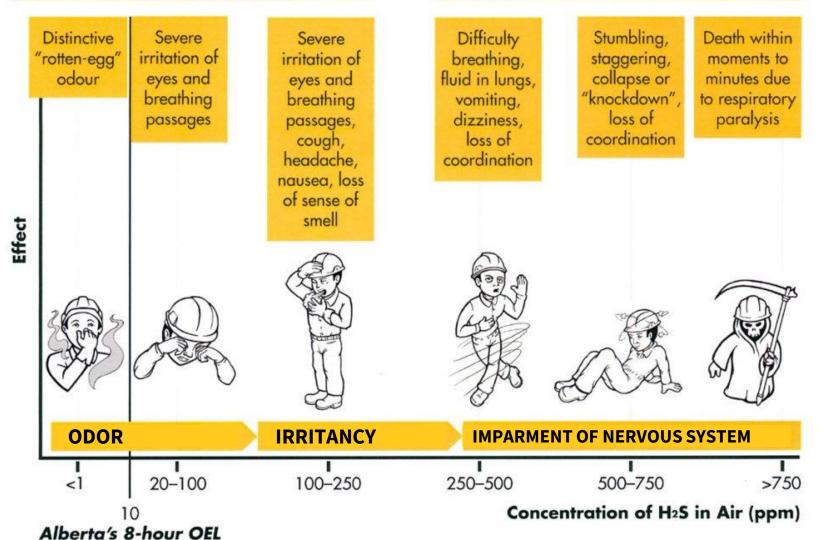
- Flotation agent
- Metals precipitant
- Cloth & paper manufacturing



EFFECTS OF EXPOSURE TO H₂S



EFFECTS OF H₂S EXPOSURE



Li-Cycle Hub Process

- Caustic scrubber inlet
 - 0.05 vol% = 50 ppm
- Caustic scrubber efficiency
 - >95% (conservative)
- Caustic scrubber outlet
 - > 0.00025 vol% = 2.5 ppm
- Air dispersion modelling inprogress

HYDROGEN SULFIDE EMISSIONS CONTROL DEVICE

Emissions Control Device

- Packed bed / venturi scrubber
- The scrubber will remove H₂S and sulfur odors from tank vents prior to venting air to atmosphere
- Water recirculated with pH control (sodium hydroxide injection)
- pH: 8 to 12
- Duty/standby fans
- Emergency/back-up power
- Scrubber sizing & arrangement in-progress

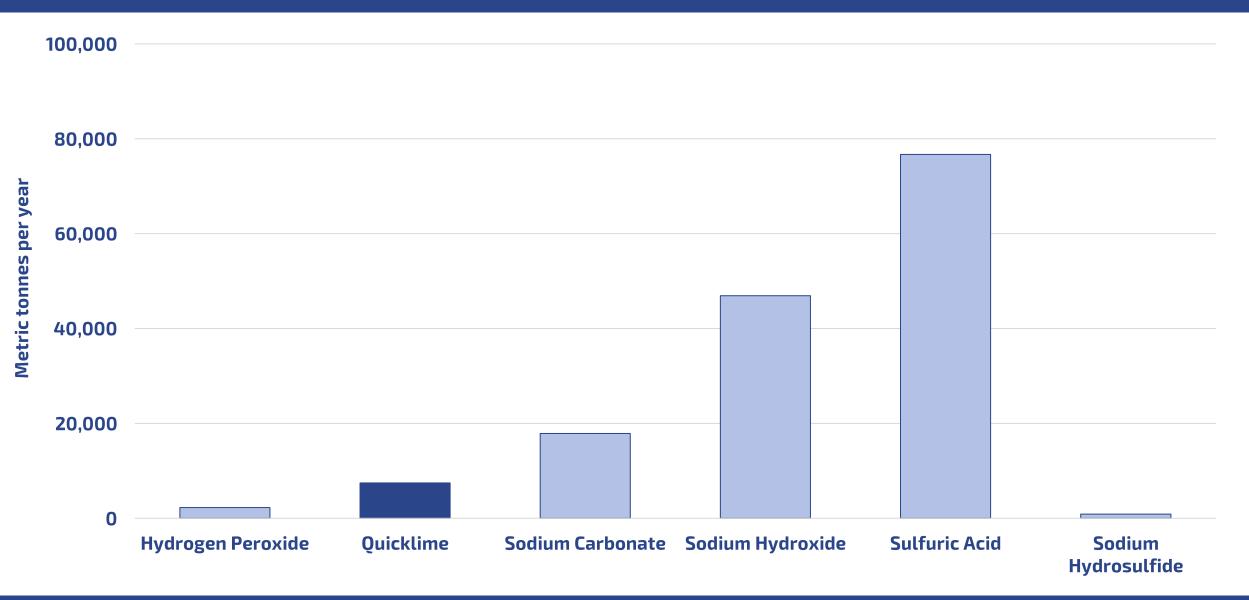




CALCIUM OXIDE (QUICKLIME)

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REAGENT USAGE (Average)





Physical Properties

- Molecular form: CaO
- White solid powder or pebble
- Caustic, alkaline
- Odorless
- Specific gravity: 1.6 2.8

Incompatible Chemicals

- Water (unless in a controlled process)
- Acids (unless in a controlled process)
- Oxidizing materials (Hydrogen peroxide storage away from lime)
- Reactive fluorinated compounds (none @ Hub)
- Reactive brominated compounds (none @ Hub)
- Reactive powdered metals (none @ Hub)
- Reactive phosphorous compounds (none @ Hub)

Industrial Uses

- pH regulator
- Cement additive
- Pulp & paper
- Mining
- Plaster
- Chemical production



Camo



General Hazards

Flammable Properties:	Non flammable, non combustible	
Extinguishing Media:	Use dry chemical or CO ₂ fire extinguisher to extinguish the surrounding fire. Do not use water, unless it is added in excess to flood the fire.	
Specific Hazards:	Store in a cool, dry, and well-ventilated location. Do not store near acids or other incompatible materials. Keep away from moisture. Do not store or ship in aluminum containers.	
Chemical Stability:	Chemically stable, but reacts with water to form calcium hydroxide while generating heat. Quicklime also reacts with carbon dioxide to form calcium carbonate.	
Incompatible Materials:	See previous slide.	
Hazardous Decomposition Products:	Does not occur.	
Possibility of Hazardous Reactions:	See previous slide – avoid incompatible materials.	



Health Hazards



Skin:	Corrosive, skin irritation or burning		
Eyes:	Corrosive, serious eye damage		
Inhalation:	Respiratory irritation		
Ingestion:	Severe irritation or burning of gastrointestinal tract if swallowed		
Carcinogenicity	May cause cancer if inhaled. Risk of cancer depends on duration and level of exposure.		
Embryotoxicity	No known significant effects or critical hazards.		
Reproductive Toxicity:	No known significant effects or critical hazards.		
Mutagenicity:	No known significant effects or critical hazards.		

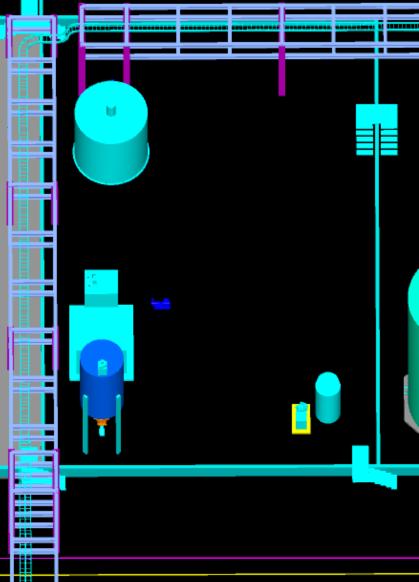
CALCIUM OXIDE (QUICKLIME)

Bulk Delivery

- Rail delivery: 100 tons each car
- Hopper car
- Roughly 1.8 railcars per week
- Unloaded by a contained/enclosed pneumatic conveying system

Bulk Storage

- One (1) bulk storage silo
- Larger of 150 tonnes or 7 days
- Storage silo equipped with dust control





CALCIUM OXIDE (QUICKLIME)



Rail Unloading

- Unloaded by pneumatic conveying system so that calcium oxide will not be exposed to precipitation
- Ground-mounted system similar to photo
- Control device on bulk silo to control dust during unloading



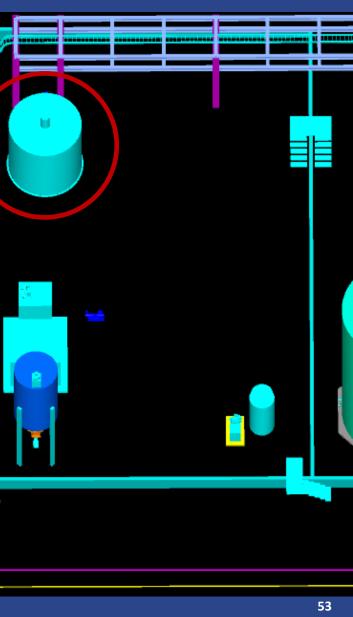
CALCIUM OXIDE (QUICKLIME)

Hydrated Lime - Slurry Preparation

- Rail unloaded to bulk storage silo
- Quicklime conveyed (fully enclosed screw conveyor) to lime slaker
- Lime slaker allows controlled addition of water:

Calcium oxide + water \rightarrow calcium hydroxide + heat

- Temperature controlled process to optimizing lime slaking
- Slaked lime slurry (calcium hydroxide/hydrated lime) pumped to storage tank
- Ring main used for delivering hydrated lime to process users
- Hydrated lime will be distributed as a 10 to 20 wt% solution

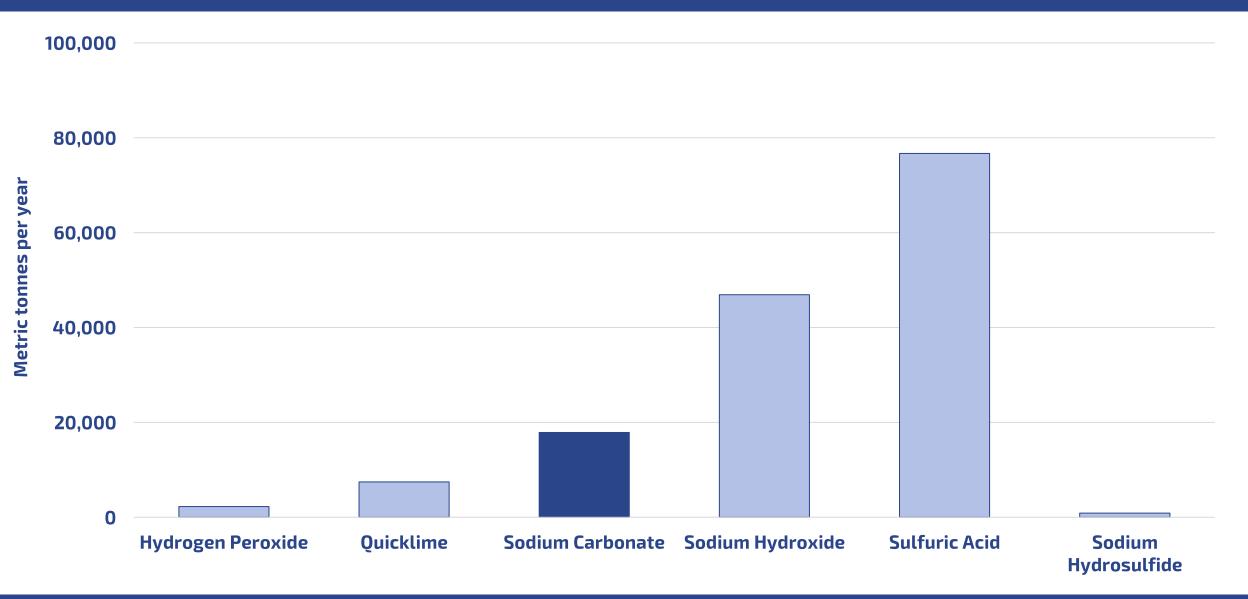




SODIUM CARBONATE (SODA ASH)

Private & Confidential

REAGENT USAGE (Average)





Physical Properties

- Molecular form: Na₂CO₃
- White solid powder
- Alkaline when dissolved in water
- Odorless
- Specific gravity: 1.9 2.5

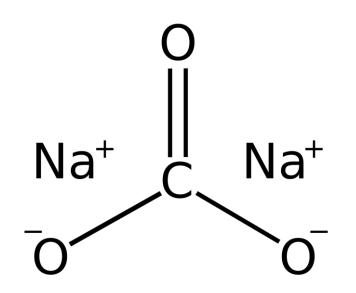
Incompatible Chemicals

- Acids (unless in a controlled process)
- Powdered aluminum
- CO₂ generation occurs when mixed with acidic materials

Industrial Uses

- Glass manufacturing
- Water softening
- Detergents
- Paper production
- Acidity regulator
- Carbonate source







General Hazards

Flammable Properties:	Non flammable, non combustible			
Extinguishing Media:	Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide.			
Specific Hazards:	Reacts on exposure to water with some metals. CO ₂ generation occu when mixed with acidic materials.			
Chemical Stability:	Chemically stable. Decomposes by reaction with strong acid.			
Incompatible Materials:	See previous slide.			
Hazardous Decomposition Products:	Does not occur.			
Possibility of Hazardous Reactions:	See previous slide – avoid incompatible materials.			



Health Hazards



Skin:	Not irritating
Eyes:	Inflammation/damage of the eye tissue. Corrosion of the eye tissue.
Inhalation:	Irritation of the respiratory tract. Irritation of the nasal mucous membranes. Respiratory difficulties.
Ingestion:	Nausea. Vomiting. Abdominal pain. Irritation of the gastric/intestinal mucosa.
Carcinogenicity	No known significant effects or critical hazards.
Embryotoxicity	No known significant effects or critical hazards.
Reproductive Toxicity:	No known significant effects or critical hazards.
Mutagenicity:	No known significant effects or critical hazards.

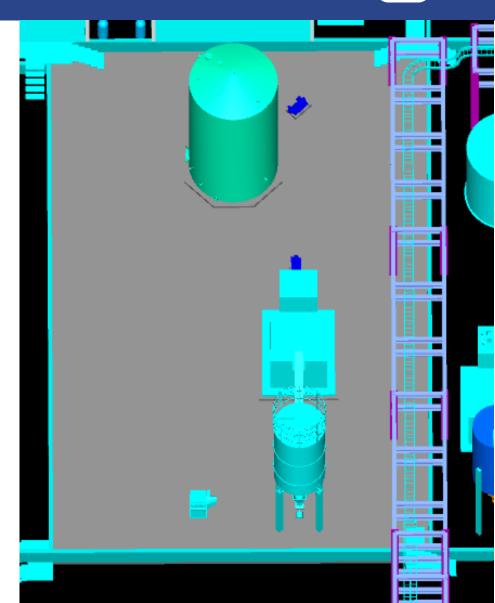
SODIUM CARBONATE (SODA ASH)

Bulk Delivery

- Rail delivery: 100 tons each car
- Hopper car
- Roughly 4.5 railcars per week
- Unloaded by enclosed pneumatic conveying system

Bulk Storage

- One (1) bulk storage silo
- Larger of 300 tonnes or 5 days





SODIUM CARBONATE (SODA ASH)



Rail Unloading

- Unloaded by pneumatic conveying system
- Ground-mounted system similar to photo
- Control device on bulk silo to control dust during unloading

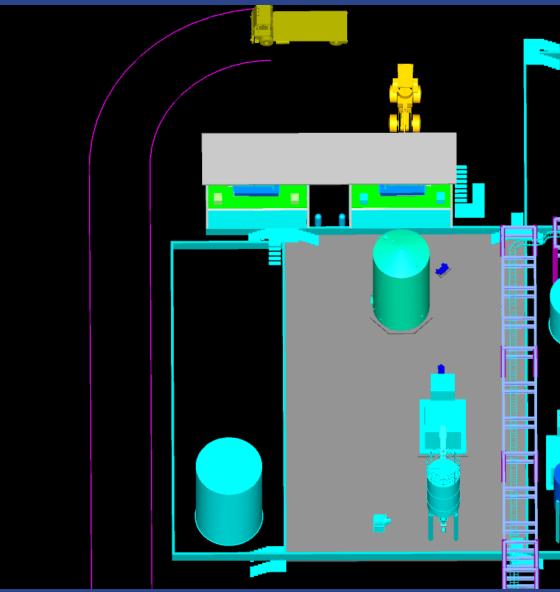


SODIUM CARBONATE (SODA ASH)



Soda Ash – Solution Preparation

- Rail unloaded to bulk storage silo
- Soda ash conveyed (screw conveyor) to makedown tank and mixed with water
- Dissolved soda ash is filtered to remove grit / solid impurities
- Soda ash solution is pumped to storage tank and then on to process users
- Containment area for tanks, silo, etc





CHEMICAL COMPATIBILITY

Private & Confidential

BASIC COMPATIBILITY TABLE



	93% Sulfuric Acid	50% Sodium Hydroxide	50% Hydrogen Peroxide	Sodium Sulfide Hydrate	Quicklime	Soda Ash
93% Sulfuric acid	_	Heat Violent Rxn	Heat Explosion risk	Heat Toxic gas Flammable gas	Heat	Heat
50% Sodium hydroxide	Heat Violent Rxn	_	_	Heat	_	_
50% Hydrogen peroxide	Heat Explosion risk	_	_	Heat Fire Flammable gas	Gas (oxygen)	Gas (oxygen)
Sodium sulfide hydrate	Heat Toxic gas Flammable gas	_	Heat Fire Flammable gas	_	_	_
Calcium oxide (Quicklime)	Heat	_	_	_	_	_
Sodium carbonate (Soda Ash)	Heat	_	_	_	_	_

CHEMICAL MIXING RISKS

Chemical Bulk Storage

- Catastrophic rupture of a properly designed and fabricated bulk storage tank is very low.
- Multiple bulk storage tanks rupturing simultaneously is *extremely* unlikely.

Rail Unloading

- Unloading of rail cars is expected to be one of the highest risk chemical-related activities at the Hub.
- Written unloading Standard Operating Procedures (SOPs) and employee training
- HAZOP Study to identify additional safeguards to include in design.





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DESIGN CONCEPTS IN-PROGRESS

Plans for ensuring safe rail unloading

- Instrumentation on all unloading lines with setpoints and interlock to unloading pumps
 - pH / density / conductivity
 - Visual Meter Displays at unloading stations
 - High level and High-High level alarms
- Security systems for full cars (e.g., perimeter fencing, CCTV, 24/7 site security)
- Each chemical has dedicated unloading system with different pipe sizes
- Separate spill containment systems with compatible liners (where applicable) for each reagent for the rail cars and unloading stations
- Spill prevention & Emergency Respond planning and procedures
- Checklists, procedures, and training (administrative controls)







ORGANIC REAGENTS

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SOLVENT EXTRACTION OVERVIEW

Private & Confidential

H U B P R O C E S S

- The Li-Cycle process produces an aqueous solution that contains a mixture of concentrated dissolved metals. This aqueous solution is called 'PLS' or Pregnant Leach Solution.
- The first few process steps recover some of these metals and remove impurities from the PLS.
- Solvent eXtraction (SX) is then used to recover key products from the PLS, mainly:
 - > Cobalt sulfate, manganese sulfate, nickel sulfate



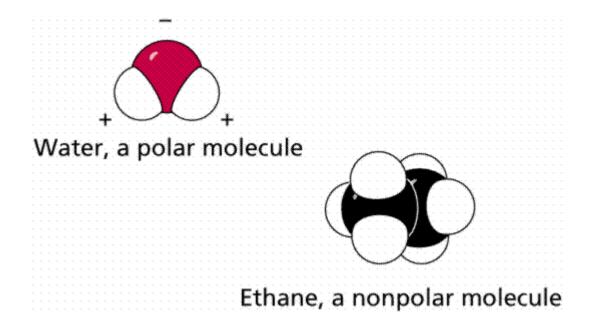


WHAT IS SOLVENT EXTRACTION?



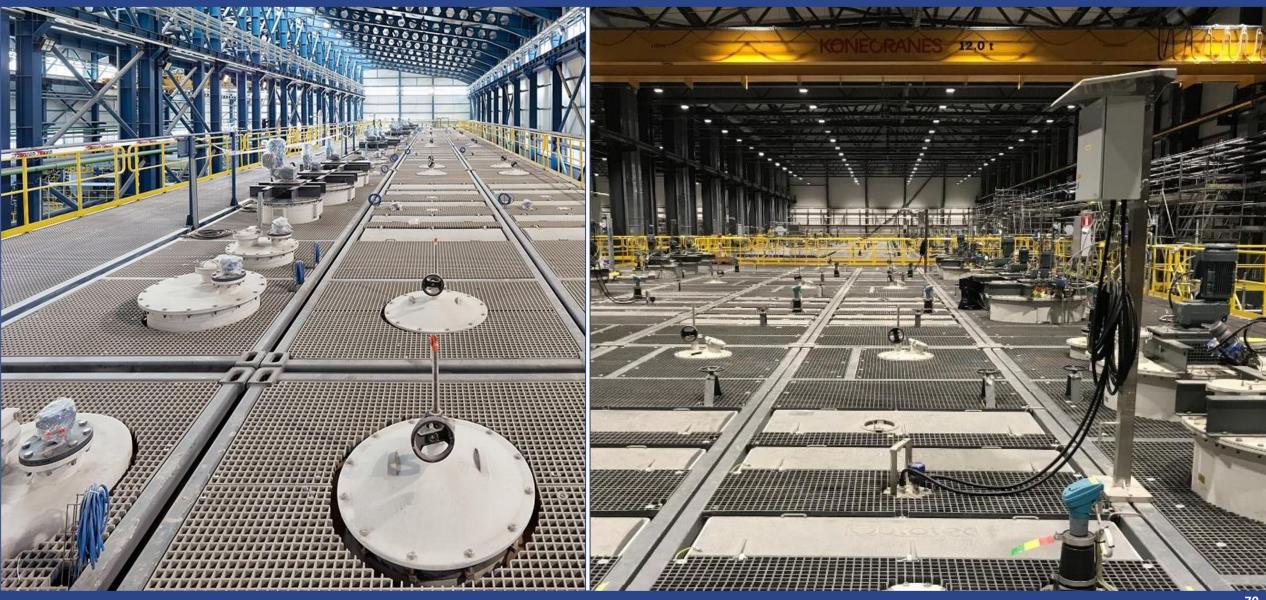
A liquid-liquid extraction method to separate metals based on their relative solubilities in two different immiscible liquids, usually water (polar) and an organic solvent (non-polar)

- Polar bonds form when two bonded atoms share electrons unequally.
- Non-polar bonds form between two atoms that share their electrons equally.



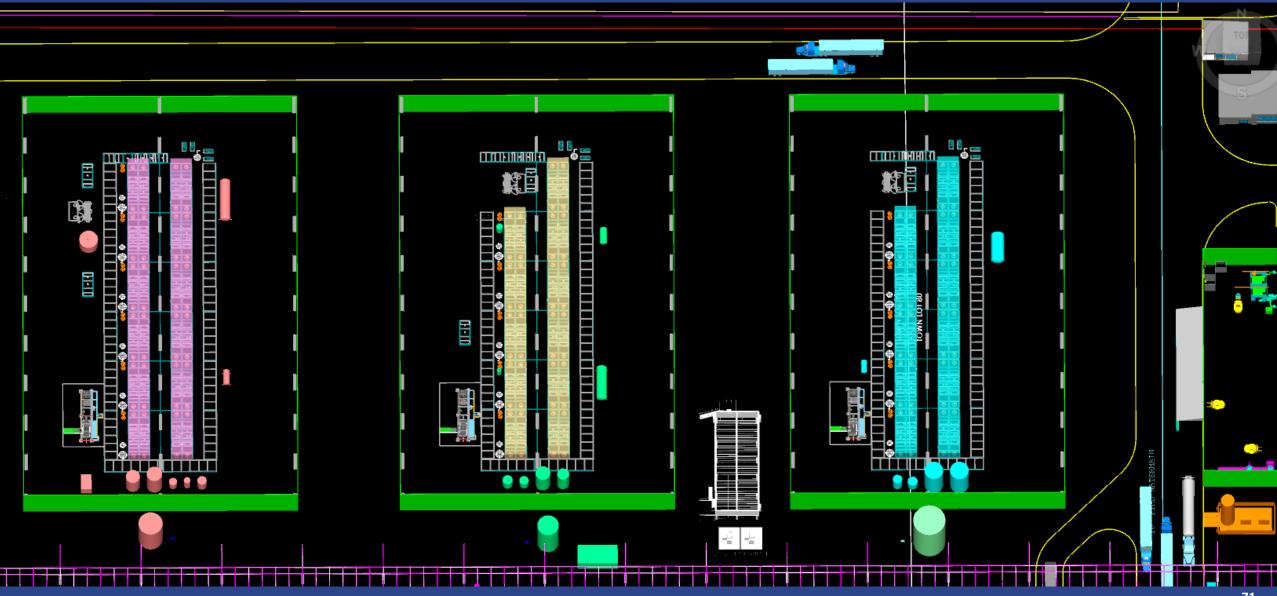
WHAT DOES THIS LOOK LIKE?





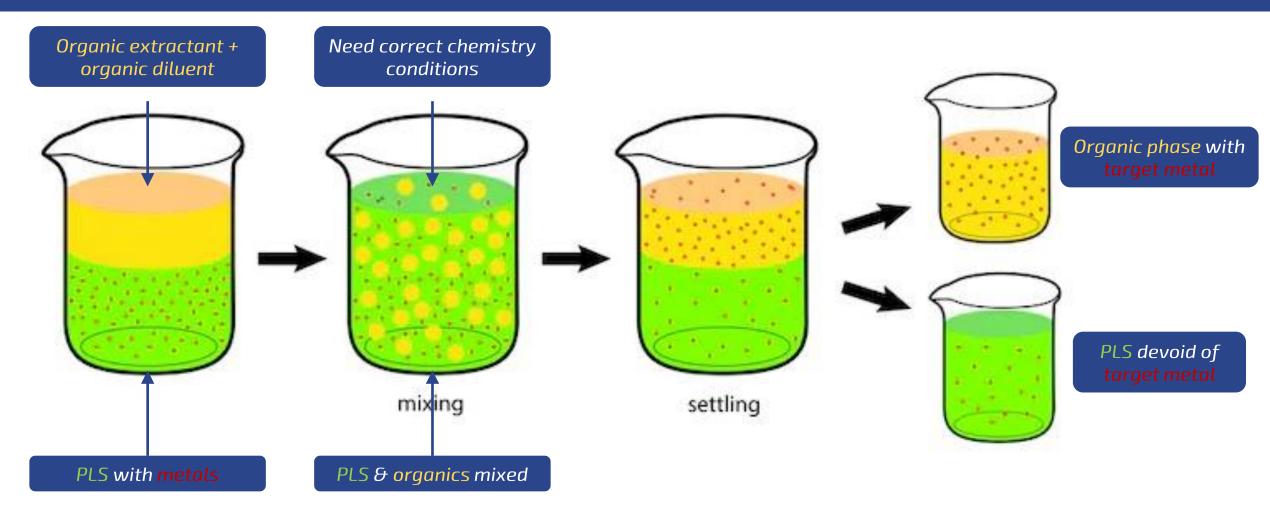
3 D M O D E L





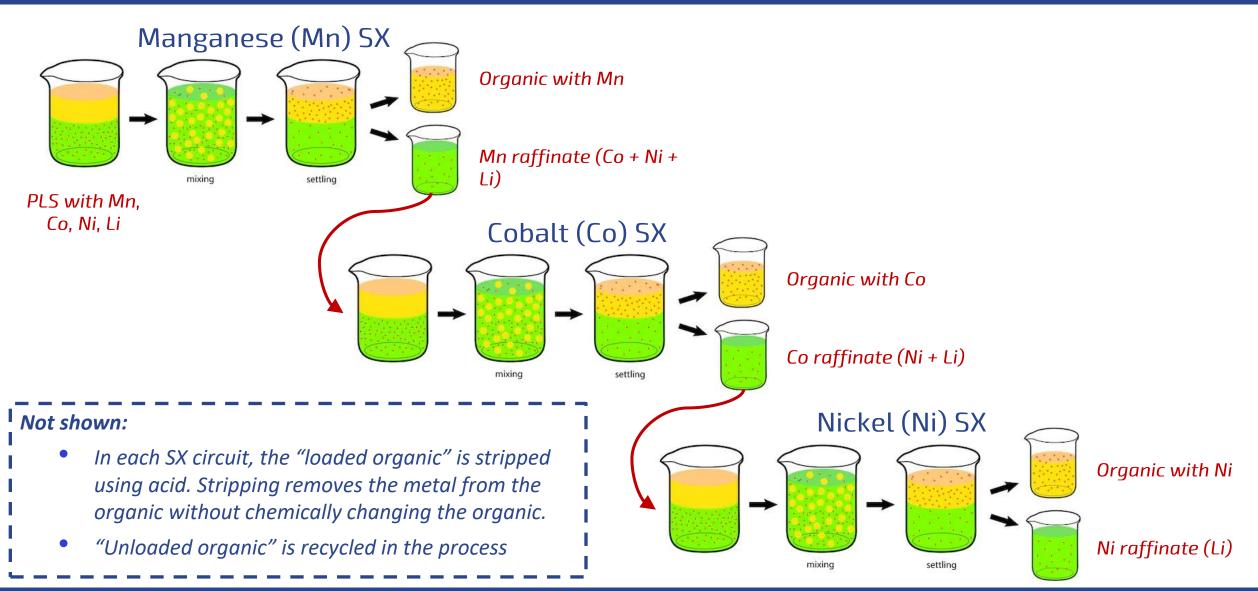
HOW DOES SX WORK?





HOW DOES SX WORK?





SOLVENT EXTRACTION EMISSIONS CONTROL DEVICE

Emissions Control Device

- Activated carbon air scrubber
- Remove volatile organic carbon vapors from tank vents prior to venting air to atmosphere
- Continuous differential pressure measurement
- Continuous organic compound monitoring in air outlet
- Emergency/back-up power
- Duty/standby fans
- Scrubber sizing & arrangement in-progress





SOLVENT EXTRACTION FIRE PROTECTION

Solvent Extraction Fire Protection Design

- Design currently in-progress.
- Petroleum hydrocarbon make-up (minor amounts) will be done via drums and/or totes on a 'just in time' basis.
- Fire fighting foam is typical for solvent extraction systems. Currently assumed to be the design basis.
- Grounded tanks & well ventilated building (through control device).
- No incompatible chemicals piped through the SX building (i.e., hydrogen peroxide).
- Area classification for SX building will be part of the design.
- Ignition sources will be mitigated or eliminated in SX buildings.
- Results of the Fire Protection Study will be a part of the design.







SOLVENT EXTRACTION ORGANICS

Private & Confidential



Combustible Liquid (per NFPA 30)

Class	Flashpoint	Examples	
Class II	≥ 100°F and ≤ 140°F	Diesel fuel, kerosene	
Class IIIA	≥ 140°F and ≤ 200°F	Linseed oil, mineral oil	
Class IIIB	≥ 200°F	Ethylene glycol (anti-freeze), glycerine	

Flammable Liquid (per NFPA 30)

Class	Flashpoint	Boiling Point	Examples
Class IA	< 73°F	< 100°F	Ethyl ether, heptane, pentane
Class IB	< 73°F	≥ 100°F	Acetone, ethanol, IPA
Class IC	≥ 73°F and < 100°F	All	Mineral spirits, isobutyl alcohol



Generally three types of organics in each solvent extraction circuit

	Manganese Circuit	Cobalt Circuit	Nickel Circuit
Diluent	Exxsol D80 / Shellsol D80	Exxsol D80 / Shellsol D80	Exxsol D80 / Shellsol D80
Extractant	Proprietary	Proprietary	Proprietary
Modifier	Proprietary	Proprietary	Proprietary

Diluent is a Class IIIA combustible, like linseed oil and mineral oil, and is not a flammable.

Total organics inventory in each solvent extraction circuit (preliminary design)

	Manganese Circuit	Cobalt Circuit	Nickel Circuit
Diluent &			
Extractant &	75,000 gallons	100,000 gallons	150,000 gallons
Modifier			
No. of tanks	12	12	12

DILUENT – EXXSOL D80

Physical Properties

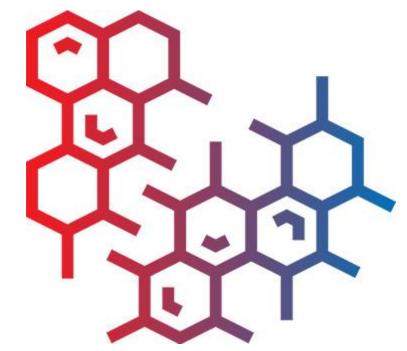
- Petroleum hydrocarbon solvent with aromatics removed
- Normal paraffins, isoparaffins and cycloparaffins
- Very low levels of aromatic hydrocarbons
- Colorless, mild solvent odor
- Less dense than water (SG = 0.798)

Incompatible Chemicals

- Fluid is combustible (Class IIIA)
- Relatively high vapor pressure / evaporation rate
- Handle with adequate ventilation and in areas where ignition sources have been removed
- The flash point is approximately 176°F / 80°C

Industrial Uses

- Paints, coatings
- Printing inks
- Mining solvent extraction



DILUENT - EXXSOL D80



General Hazards

Flammable Properties:	Material can accumulate static charges which may cause an ignition. Material can release vapors that readily form flammable mixtures. Vapor accumulation could flash and/or explode if ignited. Combustible.		
Extinguishing Media:	Use water fog, foam, dry chemical or carbon dioxide (CO ₂) to extinguish flames. Do not use straight streams of water.		
Specific Hazards:	Combustible (Class IIIA)		
Chemical Stability:	Material is stable under normal conditions.		
Incompatible Materials:	Avoid heat, sparks, open flames and other ignition sources. Strong oxidizers.		
Hazardous Decomposition Products:	Material does not decompose at ambient temperatures.		
Possibility of Hazardous Reactions:	Hazardous polymerization will not occur.		

Physical Properties

- Petroleum hydrocarbon
- Clear to light yellow
- Odorless to faint odor
- Slightly less dense than water

Incompatible Chemicals

- Extractants are combustible (Class IIIB)
- Low vapor pressure / evaporation rate
- Avoid strong oxidizing agents and strong bases
- The flash points:

(1) 279ºF /	(2) 226°F /	(3) 252ºF /
137ºC	108°C	122°C

Industrial Uses

Mining – solvent extraction



General Hazards

Flammable Properties:	No data in SDS		
Extinguishing Media:	Use water fog, foam, dry chemical or carbon dioxide (CO ₂) to extinguish flames. Do not use straight streams of water.		
Specific Hazards:	Combustible (Class IIIB)		
Chemical Stability:	Material is stable under normal conditions.		
Incompatible Materials:	Avoid heat, sparks, open flames and other ignition sources. Strong oxidisers and strong bases.		
Hazardous Decomposition Products:	Material does not decompose at ambient temperatures.		
Possibility of Hazardous Reactions:	Oxides of carbon. Oxides of phosphorous.		

Extractants are a Class IIIB combustible, like anti-freeze or glycerine, and are not flammable.

MODIFIER - PROPRIETARY

4

Physical Properties

- Petroleum hydrocarbon solvent
- Colorless
- Odorless
- Slightly less dense than water

Incompatible Chemicals

- Fluid is combustible (Class IIIB)
- Low vapor pressure / evaporation rate
- Avoid strong oxidizing agents and strong bases
- Flash point: 295°F / 146°C

Industrial Uses

- Solvent
- Plasticizer
- Adhesive additive
- Anti-foam for paints
- Mining solvent extraction



General Hazards

Flammable Properties:	No data in SDS		
Extinguishing Media:	Use water fog, foam, dry chemical or carbon dioxide (CO ₂) to extinguish flames. Do not use straight streams of water.		
Specific Hazards:	Combustible (Class IIIB)		
Chemical Stability:	Material is stable under normal conditions.		
Incompatible Materials:	Avoid heat, sparks, open flames and other ignition sources. Strong oxidisers.		
Hazardous Decomposition Products:	Material does not decompose at ambient temperatures.		
Possibility of Hazardous Reactions:	Combustion products: Carbon monoxide (CO). Carbon dioxide (CO2). Oxides of phosphorus. Phosphorus trihydride (phosphine).		

Modifier is a Class IIIB combustible, like anti-freeze or glycerine, and is not flammable.



Source: ICE Internationa

YS PETRO TERMINALS & CAPACITY

Petroleum storage terminals in Finger Lakes Region

- 13 total terminals
- Total regional capacity: 2,379,000 bbl (99,918,000 gallons)
- Average: 7,686,000 gallons per terminal total

Li-Cycle Hub Petroleum Hydrocarbon **Extraction Circuits**

- 3 extraction circuits
- 36 tanks total
- Total organic inventory: ~350,000 gallons
- Average: ~6,950 gallons per tank



Li-Cycle Tank Information - Hazardous Material Storage Tanks

Tank Number	Tank Description	Vol. gal (Upper Limit)	Comments	Classification	
120-TK-003	Stage 1 Leach Filter Feed Tank	52,800	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
			Contains sulfuric acid - regulated by NYCRR part 597 as a hazardous substance.		
120-TK-004	Pregnant Leach Solution Tank	18,700	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
			Contains sulfuric acid - regulated by NYCRR part 597 as a hazardous substance.		
120-TK-008	Graphite Residue Filter Teed Tank	25,300	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
			Contains sulfuric acid - regulated by NYCRR part 597 as a hazardous substance.		
210-TK-001	Cu Precipitation Feed Tank	151,800	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
			Contains sulfuric acid - regulated by NYCRR part 597 as a hazardous substance.		
210-TK-005	Cu Precipitation Filter Feed Tank	22,000	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
			Contains sulfuric acid and sodium hydrosulfide - both of which are regulated by NYCRR part 597 as a hazardous substance.		
210-TK-006	Sulfide Filtrate Tank	39,600	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
			Contains sulfuric acid - which is regulated by NYCRR part 597 as a hazardous substance.		
310-TK-005	AL-Fe Precipitation Filter Feed Liquor Tank	93,500	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
310-TK-006	Gypsum Filtrate Tank	57,200	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
310-TK-008	Gypsum Wash Filtrate Tank	12,100	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
410-TK-001	Mn SX Feed Tank	174,900	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
410-TK-050	Raffinate Filter Backwash Solution Tank	8,800	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
510-TK-001	Co SX feed Tank	66,000	Contains Li, Ni, Co	Storage Tank - Process Solution, Hazardous Materials	
510-TK-004	Co SX Spent Scrub Solution Tank	3,300	Contains Co	Storage Tank - Process Solution, Hazardous Materials	
510-TK-064 520-TK-001	Raffinate Filter Backwash Solution Tank	8,800	Contains Li, Ni	Storage Tank - Process Solution, Hazardous Materials	
610-TK-001	Cobalt Sulfate Crystallizer Feed Tank Ni SX Feed Tank	74,800	Contains Co Contain Li, Ni	Storage Tank - Process Solution, Hazardous Materials Storage Tank - Process Solution, Hazardous Materials	
610-TK-001	Ni SX Spent Scrub Solution Tank	6,600	Contain Li, Ni	Storage Tank - Process Solution, Hazardous Materials	
610-TK-055	Ni SX Raffinate Filtrate Tank	29,700	Contain Li	Storage Tank - Process Solution, Hazardous Materials	
610-TK-056	Raffinate Filter Backwash Solution Tank	8,800	Contain Li	Storage Tank - Process Solution, Hazardous Materials	
620-TK-001	Nickel Sulfate Crystallizer Feed Tank	74,800	Contains sulfuric acid and nickel sulfate- regulated by NYCRR part 597 as a hazardous substance.	Storage Tank - Process Solution, Hazardous Materials	
805-TK-006	Impurity Rejection Filtrate Tank	103,400	Contains Li Contains sodium hydroxide - regulated by NYCRR part	Storage Tank - Process Solution, Hazardous Materials	
805-TK-007	Impurity Rejection Filter Feed Tank	47,300	597 as a hazardous substance. Contains Li	Storage Tank - Process Solution, Hazardous Materials	
810-TK-007	Glauber's Salt Crystallizer Centrate Tank		Contains Li	Storage Tank - Process Solution, Hazardous Materials	
810-TK-004 820-TK-001	SSA Crystallizer Centrate Tank	ТВО	Contains Li	Storage Tank - Process Solution, Hazardous Materials Storage Tank - Process Solution, Hazardous Materials	
910-TK-005	Crude Lithium Carbonate Centrifuge Feed Tank	39,600	Contains Li	Storage Tank - Process Solution, Hazardous Materials	
910-TK-005 910-TK-006	Crude Lithium Carbonate Centrifuge Feed Tank	18,700	Contain Li	Storage Tank - Process Solution, Hazardous Materials	
915-TK-005	Fluoride Precipitation Filter Feed Tank	18,700	Contain El Contains sodium hydroxide - regulated by NYCRR part 597 as a hazardous substance.	Storage Tank - Process Solution, Hazardous Materials Storage Tank - Process Solution, Hazardous Materials	

915-TK-006	Fluoride Precipitation Filtrate Tank	17,600	Contains sodium hydroxide - regulated by NYCRR part	Storage Tank - Process Solution, Hazardous Materials
			597 as a hazardous substance.	
920-TK-007	Lithium Bicarbonate Ion Exchange Feed Tank	26,400	Contains Li	Storage Tank - Process Solution, Hazardous Materials
920-TK-008	Purified Lithium Bicarbonate Storage Tank	26,400	Contains Li	Storage Tank - Process Solution, Hazardous Materials
930-TK-006A	Pure Lithium Carbonate Centrate Tank #1	27,500	Contains Li	Storage Tank - Process Solution, Hazardous Materials
930-TK-006B	Pure Lithium Carbonate Centrate Tank #2	27,500	Contains Li	Storage Tank - Process Solution, Hazardous Materials



New York State Department of Environmental Conservation Division of Environmental Remediation

Hazardous Substance Bulk Storage Application

CBS Number:

Pursuant to the Hazardous Substance Bulk Storage Law, Article 40 of ECL and 6 NYCRR 596-599

(See instructions and please be sure to complete Sections A, B & C)

Return Completed Form & Fees To: NYSDEC Spill Prevention & Bulk Storage Section 625 Broadway, 11th Floor Albany, NY 12233-7020



See

ction A	- Fa	cility/F	Property	Owner/	/Contact	Information

Transaction		Facility Name:			Tax	<u>x Map</u>	Т	TYPE OF CHEMICAL STORAGE FACILITY (Check only one)								
Type: 1		Li-Cycle North Ame	erica Hub #	1	Bor	rough/Section	١ř	01=Storage Terminal/Petrol. Distributor		etail Gasoline Sales						
	F	Facility Address (Physical Ad						03=Other Retail Sales		[anufacturing(non-chemical)						
1) Initial/New	А	50 & 205 McLaughlin R	oad Extension		Blo	ock:	ᅴᄂ	05=Utility		rucking/Transportation/Fleet						
Facility		Facility Address (cont.):			Lot	t		07=Apartment/Office Building	$\square 08=Sc$	0 1						
2) Change of	С						ᅴ┝	09=Farm		rivate Residence						
Ownership	T	City:				ZIP										
3) Tank	•	Rochester				14606		11=Airline/Air Taxi/Airport		Chemical Distributor						
Installation,	L	County:	Township or	1	Facility P	Phone Number:		13=Municipality		ailroad						
Closing, or	Ι	Monroe		8	877-54	42-9253		20=Chemical Manufacturing		wimming Pools (Other than						
Repair		Facility Operator:						25=Auto Service/Repair (No Gasoline Sales)	Munic	cipality)						
4) Information	Т	Li-Cycle North Ame	rica Hub In	C				26=Religious (Church, Synagogue, Mosq	ue, Templ	e, etc.)						
Correction	Y	LI-OYCIE NOITH AILE		0.				27=Hospital/Nursing Home/Health Care	28=C	emetery/Memorial						
								52=Marina	□ 99=C	Other (Specify):						
5) Renewal							En	nergency Contact Name:		gency Telephone Number:						
Provide property		Facility (Property) Owner (fro	om Deed):					ackie Jordan		, <u>,</u> ,						
owner information		Ridgeway Properties 1 LLC c		nstruction Manage	ement		Lb	ereby certify, under penalty of law, that all of the in	formation n	rovidad on this form is true						
here and tank		Facility Owner Address (Stre	et and/or P.O. Bo	oxes):				d correct. False statements made herein may be								
owner information n Section C.	0	1010 Lee Road						ivil violation in accordance with applicable state ar								
ii Section C.	W	City:		State:	ZIP Coc	de:	_									
Fransaction type		Rochester		NY	14606		Na	me of Property Owner or Authorized Representative:	Amo	ount Enclosed: \$						
1, 2 and 5: attach	Ν	Owner Telephone Number:					_			Ψ						
a copy of cover	Е	585 440 4900					Titl	e:								
bage, table of contents, and		303 440 4900					_									
signature page	R	Type of Owner (check only one): 3 Local Govern				t	Sig	nature:	Date:							
from spill		1 Private Resident	4	Federal C	Governme	ent	518		Date	-						
prevention report.		2 State Government	5	Corporate	e/Comme	ercial/Other										
Official Use	С	(Please keep this information	up to date.)													
Only	O R	Facility Contact Person Nam	e: Chris Bieder	man, Chief Te	chnolog	gy Officer										
Date Received:	R E	Contact Person Company Na	me: Li-Cycle	North Amer	rica Hu	ub, Inc.										
Date Processed:	S P	Address: 2351 Royal	Windsor D	rive Unit 10	0											
Amount	P O N	Address (cont.):														
Received: \$	D															
Reviewed By:	E N C	City/State/ZIP Code: Miss	issauga, C	ntario Can	iada, L	L5J 4S7										
Rev. 6/14/2021	C E															

Section B - Tank Information

(Please use the key located on the last page to complete

Registration Expiration Date:

<u>each item/column)</u>

(1)	(2)	(3)	(4)	(5)	(6)	(8)	(9)	(10)	(11)	(12)	(13)	(14)	(16)	(17)	(18)	(19)	(20)				
Action	Tank Number	Tank Location	Status	Installation, Out-of-service Or Permanent Closure Date (mm/dd/yyyy) Application will be returned if blank	Capacity (Gallons)	Tank Type	Tank Internal Protection	Tank External Protection	Tank Secondary Containment	Tank Leak Detection	Tank Overfill Prevention	Tank Spill Prevention	Piping Location	Piping Type	Piping External Protection	Piping Secondary Containment	Piping Leak Detection	Hazardous Substance Name (List <u>all</u> Part 597 Substances, if more than 3 please list on separate sheet)	CAS Number	% of Haz Sub	Tank Fee \$
1	120-TK-003	02	01	Installation TBD	47,539	03	00	00	10	06	02 99	00	01	06	00	00	00	 Please see attached Table 1 3) 	Table 1	Table 1	\$125
1	120-TK-004	02	01	Installation TBD	16,316	06	00	00	10	06	02 99	00	01	06	00	00	00	 Please see attached Table 1 3) 	Table 1	Table 1	\$125
1	120-TK-008	02	01	Installation TBD	22,030	06	00	00	10	06	02 99	00	01	06	00	00	00	 Please see attached Table 1 3) 	Table 1	Table 1	\$125
1	210-TK-001	02	01	Installation TBD	137,047	03	00	00	10	06	99	00	01	07	00	00	00	 Please see attached Table 1 3) 	Table 1	Table 1	\$125
1	210-TK-005	02	01	InstallationTBD	19,277	06	00	00	10	06	02 99	00	01	07	00	00	00	1) Please see attached 2) ^{Table 1} 3)	Table 1	Table 1	\$125
1	210-TK-006	02	01	InstallationTBD	35,695	06	00	00	10	06	02 99	00	01	07	00	00	00	1) Please see attached 2) Table 1 3)	Table 1	Table 1	\$125
1	210-TK-009	02	01	Installation TBD	11,750	03	00	00	10	06	02	00	01	07	00	00	00	1) sodium hydrosulfide 2) 3)	16721-80-5	10	\$125

Note: If you need to add tanks to your registration, write them in using blank lines above. Attach additional sheets as needed. Blank Section B is available at <u>http://www.dec.ny.gov/docs/remediation_hudson_pdf/cbsregab.pdf</u>

Section B - Tank Information

(Please use the key located on the last page to complete

Registration Expiration Date:

<u>each item/column)</u>

(1)	(2)	(3)	(4)	(5)	(6)	(8)	(9)	(10)	(11)	(12)	(13)	(14)	(16)	(17)	(18)	(19)	(20)				
Action	Tank Number	Tank Location	Status	Installation, Out-of-service Or Permanent Closure Date (mm/dd/yyyy) Application will be returned if blank	Capacity (Gallons)	Tank Type	Tank Internal Protection	Tank External Protection	Tank Secondary Containment	Tank Leak Detection	Tank Overfill Prevention	Tank Spill Prevention	Piping Location	Piping Type	Piping External Protection	Piping Secondary Containment	Piping Leak Detection	Hazardous Substance Name (List <u>all</u> Part 597 Substances, if more than 3 please list on separate sheet)	CAS Number	% of Haz Sub	Tank Fee \$
1	310-TK-005	02	01	Installation TBD	93,500	03	00	00	10	06	02 99	00	01	07	00	00	00	1)nickel sulfate 2) 3)	7758-98-7	8.16	\$125
1	310-TK-006	02	01	Installation TBD	57,200	03	00	00	10	06	02 99	00	01	07	00	00	00	1) nickel sulfate 2) 3)	7758-98-7	8.73	\$125
1	310-TK-008	02	01	Installation TBD	10,868	06	00	00	10	06	02 99	00	01	06	00	00	00	 1) sulfuric acid 2) nickel sulfate 3) 	7664-93-9 7758-98-7		\$125
1	410-TK-050	02	01	Installation TBD	8,800	06	00	00	10	06	99	00	01	07	00	00	00	1) nickel sulfate 2) 3)	7758-98-7	8.09	\$125
1	510-TK-004	02	01	Installation TBD	3,300	06	00	00	10	06		00	01	07	00	00	00	 nickel sulfate 3) 	7758-98-7	0.38	\$125
1	510-TK-064	02	01	Installation TBD	8,800	06	00	00	10	06	99	00	01	07	00	00	00	 nickel sulfate 3) 	7758-98-7		
1	610-TK-004	03	01	Installation TBD	5,453	06	00	00	10	06	99	00	01	07	00	00	00	 1) sulfuric acid 2) nickel sulfate 3) 	7664-93-9 7758-98-7		\$125

Note: If you need to add tanks to your registration, write them in using blank lines above. Attach additional sheets as needed. Blank Section B is available at <u>http://www.dec.ny.gov/docs/remediation_hudson_pdf/cbsregab.pdf</u>

Section B - Tank Information

(Please use the key located on the last page to complete

Registration Expiration Date:

<u>each item/column)</u>

(1)	(2)	(3)	(4)	(5)	(6)	(8)	(9)	(10)	(11)	(12)	(13)	(14)	(16)	(17)	(18)	(19)	(20)				
Action	Tank Number	Tank Location	Status	Installation, Out-of-service Or Permanent Closure Date (mm/dd/yyyy) Application will be returned if blank	Capacity (Gallons)	Tank Type	Tank Internal Protection	Tank External Protection	Tank Secondary Containment	Tank Leak Detection	Tank Overfill Prevention	Tank Spill Prevention	Piping Location	Piping Type	Piping External Protection	Piping Secondary Containment	Piping Leak Detection	Hazardous Substance Name (List <u>all</u> Part 597 Substances, if more than 3 please list on separate sheet)	CAS Number	% of Haz Sub	Tank Fee \$
1	620-TK-001	02	01	Installation TBD	67,866	06	00	00	10	06	99	00	01	07	00	00	00	1)sulfuric acid 2) nickel sulfate 3)	7664-93-9 7758-98-7	0.09 21.28	\$125
1	805-TK-006	02	01	Installation TBD	93,268	03	00	00	10	06	99	00	01	07	00	00	00	1) sodium hydroxide 2) 3)	1310-73-2	0.02	\$125
1	915-TK-005	02	01	Installation TBD	18,700	03	00	00	10	06	02 99	00	01	06	00	00	00	1) sodium hydroxide 2) 3)	1310-73-2		\$125
1	915-TK-006	02	01	Installation TBD	17,600	03	00	00	10	06	02 99	00	01	06	00	00	00	1) sodium hydroxide 2) 3)	1310-73-2	0.06	\$125
1	2110-TK-001A	02	01	Installation TBD	113,736	01	00	01	10	06	02 99	02	01	01	00	00	00	1) sulfuric acid 2) 3)	7664-93-9	>93	\$125
1	2120-TK-001	02	01	Installation TBD	31,377	03	00	00	10	06	02 99	02	01	07	00	00	00	1) hydrogen peroxide 2) 3)	7722-84-1	50.00	
1	2150-TK-001A	02	01	Installation TBD	93,268	03	00	00	10	06	02 99	02	01	07	00	00	00	1) sodium hydroxide 2) 3)	1310-73-2	50.00	\$125

Note: If you need to add tanks to your registration, write them in using blank lines above. Attach additional sheets as needed. Blank Section B is available at <u>http://www.dec.ny.gov/docs/remediation_hudson_pdf/cbsregab.pdf</u>

Hazardous Substance Bulk Storage Application

Section C - Tank Ownership Information (for CBS tanks listed in Section B)

Tank Own Check box if same as If tank owner is different from pro			ation below:	Ch If tank owner is	Owner. t inform	ation below:						
Tank Owner Name (Company/Individual)):			Tank Owner Name (Company/Individual):								
Li-Cycle North America Hub #1												
Contact Person: Chris Biederman, Chief Techno	ology Officer			Contact Person:								
Tank Owner Address: 2351 Royal Windsor Drive Unit 10)			Tank Owner Address:								
City: Mississauga	State: Ontario, Canada	ZIP: L5	J 4S7	City:								
-	Contact Person ema	il:		Contact Person Telephone Number: Contact Person email:								
Check box if this owner If not, list tanks own	Tanks Owned owns all tanks at the ned by this owner be		у.	Specific Tanks Owned Check box if this owner owns all tanks at this facility. If not, list tanks owned by this owner below:								
Tank Number:				Tank Number:								
Name of Class B (Daily On-Site) Operator:			Authorization No:	Name of Class B (Daily C	On-Site) Operator:				Authorization No:			
Name of Class A (Primary) Operator:			Authorization No:	Name of Class A (Primary) Operator: Authoriza								

HAZARDOUS SUBSTANCE BULK STORAGE APPLICATION - SECTION B - TANK INFORMATION - CODE KEYS

Action (1)

- 1. Initial Listing
- 2. Add Tank
- 3. Close/Remove Tank
- 4. Information Correction
- 5. Repair/Reline Tank

<u>Tank Location (3)</u>

- Aboveground-contact w/soil
 Aboveground-contact w/
- impervious barrier 3. Aboveground on saddles, legs,
- stilts, rack or cradle 4. Partially buried tank /10% or more
- below ground.
- Underground including vaulted with no access for inspection
 Aboveground in Subterranean
- Vault w/access for inspections.

Status (4)

- 1. In-service
- 2. Out-of-service
- 3. Closed-Removed
- 4. Closed- In Place
- 5. Tank converted to Non-Regulated use

Tank Type (8)

01. Steel/Carbon Steel/Iron
02. Galvanized Steel Alloy
03. Stainless Steel Alloy
04. Fiberglass Coated Steel
05. Steel Tank in Concrete
06. Fiberglass Reinforced Plastic (FRP)
07. Plastic
08. Equivalent Technology
09. Concrete
10. Urethane Clad Steel

Internal Protection (9)

00. None 01 Epoxy Liner 02. Rubber Liner 03. Fiberglass Liner (FRP) 04. Glass Liner 99. Other-Please list:* **External Protection (10/18)**

01. Painted/Asphalt Coating

02. Original Sacrificial Anode

03. Original Impressed Current

07 Retrofitted Sacrificial Anode

08. Retrofitted Impressed Current

Tank Secondary Containment

00. None

04. Fiberglass

06. Wrapped (Piping)

99. Other-Please list:*

01. Diking (AST Only)

02. Vault (w/access)

05. Synthetic Liner

07. Excavation Liner

(AST Only)

(AST Only)**

99. Other - Please list*

01. Interstitial Electronic

Monitoring

04. Groundwater Well

03. Vapor Well

Gauge)

00. None

03. Vault (w/o access)

04. Double-Walled (UST Only)

06. Remote Impounding Area

09. Modified Double-Walled

10. Impervious Underlayment

11. Double Bottom (AST Only)**

Tank Leak Detection (12)

02. Interstitial Manual Monitoring

05. In-Tank System (Auto Tank

06. Impervious Barrier/Concrete Pad (AST Only)99. Other-Please list: *

12. Double-Walled (AST Only)

(11)

00. None

05. Jacketed

09. Urethane

<u>Overfill Protection (13)</u>

00. None
01. Float Vent Valve
02. High Level Alarm
03. Automatic Shut-Off
04. Product Level Gauge (AST Only)
05. Vent Whistle
99. Other-Please list;*

Spill Prevention (14)

00. None 01. Catch Basin 02. Transfer Station Containment 99. Other-Please list:*

Piping Location (16)

00. No Piping01. Aboveground02. Underground/On-ground03. Aboveground/Underground Combination

Piping Type (17)

- 00. None
- 01. Steel/Carbon Steel/Iron
- 02. Galvanized Steel
- 03. Stainless Steel Alloy
- 04. Fiberglass Coated Steel
- 05. Steel Encased in Concrete 06. Fiberglass Reinforced Plastic
- (FRP)
- 07. Plastic
- 08. Equivalent Technology
- 09. Concrete
- 10. Copper
- Flexible Piping
 Other-Please list:*

Piping Secondary Containment (19)

00. None 01. Diking (Aboveground Only) 02. Vault (w/access) 04. Double-Walled (Underground Only) 06. Remote Impounding Area 07. Trench Liner 12. Double-Walled (Aboveground Only) 99. Other-Please list: * Pipe Leak Detection (20) 00. None 01. Interstitial Electronic Monitoring 02. Insterstitial Manual Monitoring 03. Vapor Well 04. Groundwater Well 07. Pressurized Piping Leak Detector 09. Exempt Suction Piping 99 Other-Please list*

* If other, please list on a separate sheet including tank number,

** Each of these codes must be combined with code 01 or 06 to meet compliance requirements.