Salt Hydrate Eutectic Thermal Energy Storage for Building Thermal Regulation



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Project Summary

Timeline:

Start date: 04/01/2020 (NEW START) Planned end date: 07/31/2023

Key Upcoming Milestones:

- 1. Identify 3-4 salt hydrate eutectics in each of target ranges; 08/01/2021
- 2. Demonstrate decreased supercooling to <5 C in 2-5 eutectics; 08/01/2021
- Demonstrate the general ability to shape stabilize nitrate and chloride salt hydrates (up to 100 cycles, >90 vol.% PCM; 02/01/2022
- Demonstrate robust micro-encapsulation with >95 vol.% and <5% degradation over 10³ melt cycles; 08/01/2022

Budget:

Total Project \$ to Date:

- DOE: \$466,749.
- Cost Share: \$ 163,058.

Total Project \$:

- DOE: \$1,546,556.
- Cost Share: \$386,639.

Key Partners/Co-PI's: 4 Dept's, 6 PI's/Co-PI's

Dr. Choongho YU (TAMU, Mech Eng; R3).

Dr. Emily **PENTZER** (TAMU, Mat Sci Eng/Chem; R4).

Dr. Svetlana SUKHISHVILI (TAMU, Mat Sci Eng; R5).

Dr. Jonathan FELTS (TAMU, Mech Eng; R6).

Dr. Charles CULP (TAMU, Architecture; R6).

Project Outcome:

Low-cost, high energy-density, stable inorganic thermal energy storage material systems for integrating with HVAC: 5-25 °C



Team

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Challenge

Problem Definition: Intermittent power generation and variable demand have resulted in variable electricity pricing schemes where peak prices are 5 to 40 ¢/kWh higher than off-peak prices. The largest single segment of residential electricity consumption (~30 to 35%) corresponds to building HVAC and commercial consumption is similar.

To capture this differential pricing, consumers require a low cost, dependable approach to displace use of HVAC to off-peak hours. Current technologies are limited by the capability of available thermal energy storage materials.







Approach + Takeaway



Progress: R1 Identify stable salt hydrate eutectics



Accomplishments:

- Complete evaluation of pseudo-binary nitrate eutectics
- ID 2 to 4 compounds in both high T and low T ranges for HVAC: subset of 3-4 eutectics <\$10/kWh_{th}

Next Steps:

- Evaluate mixed nitrate-chloride systems
- Move to deep eutectics to fine-tune eutectic temperature







Experimentally Measured





Progress: R2 Decrease undercooling to <5 C

High throughput evaluation of Nucleation Agents (NA's)





Accomplishments:

- Applied lattice-matching technique to different inorganic salt hydrates and salt hydrate eutectics.
- For eutectics of interest: ID'd appropriate NA's ٠ with $\Delta T < 5$ C (at ~mL volumes).

Next Steps:

- Evaluate cycling, stability over time (degradation of NA activity).
- Continue to develop NA's for new eutectics. •

Scaled Volume / Cycling



Progress: R3 Stable thermal conductivity enhancement



(a) Highly porous bare EG-750 matrix. (b) Paraffin infiltrated EG-750 matrix. (c) Compacted EG/paraffin wax composite. (d) Composite sectioned down the middle to analyze anisotropy of k-value.

Accomplishments:

- Evaluate composite synthesis protocols and effect on latent heat/ thermal conductivity.
- Quantify degradation with cycling in different composites to >5k cycle.
- Demonstrated infiltration of salt hydrates in expanded graphite matrix

Next Steps:

• Evaluate high thermal conductivity salt hydrate composites.



Long-term cycling behavior:



Progress: R4 Microencapsulation of Salt Hydrate Eutectics



Accomplishments:

- Development of PCM-agnostic approach to microencapsulation based
 on Pickering emulsions
- Applicable to different polymer coatings as well
- Stable for small numbers of cycles
- Cooperative nucleation effect (decrease undercooling) Next Steps:
- Evaluate aging/stability with:
 - Cycling
 - Different environmental relative humidities

300 µm





Progress: R5 Decrease undercooling to <5 C





Accomplishments:

- Physical gels using hydrogen-bonded-network forming polymers poly(vinyl alcohol) (PVA).
- Addition of small amounts of borax raises transition temperature of salogels above Tm of a PCM
- Increase in PH further strengthens salolgels

Next Steps:

Apply this technique to salt hydrate eutectics

Shape Stabilization with minimal impact on latent heat



3 wt% PVA-0.32 wt% borax-CaN.4H salogel



PVA-borax-CaNH systems after 7th cycle of melting/ crystallization and during melting in the 8th cycle



Progress: R6 Transient Thermal Modeling



Impact

1) Towards DOE *Energy Storage Grand Challenge roadmap goal* of \$0.05/kWh levelized cost of storage for long-duration stationary applications, a 90% reduction from 2020 baseline costs by 2030.

Table 1. Levelized Cost of Thermal Energy Storage				$\Sigma(Capital_{t} + 0 \& M_{t} + Fuel_{t}) \cdot (1 + r)^{2}$
t	Cost of Storage Media			$LCOS = \frac{\Delta(cup + u_t) + (1 + r)^{-t}}{\sum MWh_t \cdot (1 + r)^{-t}}$
(yrs)	\$5/kWh _{th}	\$10/kWh _{th}	\$15/kWh _{th}	• $r = 8\%$
10	\$ 0.019	\$ 0.037	\$ 0.056	• $t = 10$ to 30 yr Example diacharge (m = cooler (m × ortilization)
20	\$ 0.013	\$ 0.025	\$ 0.038	 Energy alsonarge/yr = cycles/yr × utilization Cvcles/vr = 80
30	\$ 0.011	\$ 0.022	\$ 0.033	• Utilization = 50%

2) Towards 1-2 yr simply payback cost for installed thermal energy storage.

Table 2. Simple payback (in yrs) assuming utilization of 0.5			
Δ	Cost of Proposed Technology		
\$/kWh	\$5/kWh _{th}	\$10/kWh _{th}	\$15/kWh _{th}
\$0.10	1.3	2.5	3.8
\$0.20	0.6	1.3	1.9
\$0.30	0.4	0.8	1.3

$\begin{bmatrix} Simple \\ Payback \\ (Yr) \end{bmatrix} = \begin{bmatrix} I \\ c \\ c \end{bmatrix}$	$ \left[\begin{array}{c} \text{Cost} \\ \text{Sproposed Tech} \\ \text{S/kWh}_{\text{th}} \end{array} \right] / \left[\begin{array}{c} \text{Cost} \\ \text{Savings} \\ (\$/kWh/Yr) \\ \end{array} \right] $	
Cost Savings (\$/kWh/Yr)	$= \begin{bmatrix} \text{Savings from} \\ \text{a single cycle} \\ (\$/kWh_{\text{displaced}}) \end{bmatrix} \times \begin{bmatrix} \text{Number of} \\ \text{cycles} \\ (\text{cycle s/y r}) \end{bmatrix} \times \begin{bmatrix} \text{Utilization} \\ \frac{kWh_{\text{displaced}}}{kWh_{\text{instation}}} \end{bmatrix}$	n aced alled

Stakeholder Engagement

This is an early stage project (just completed yr 1).

Industrial Advisory Board:

- NETenergy
- Viking Cold Solutions
- RGEES, LLC. // Akuratemp
- Thornton Tomasetti
- Dow Packaging
- Advanced Cooling Technologies

- Thermavant
- Latent Heat Sol'ns
- CALMAC
- CAVU group
- NREL
- ORNL

Remaining Project Work



Ultimate Success: A suite of robust low-cost materials solutions spanning 5 to 25 C

EVALUATE ADDITIONAL EUTECTICS:

- Displace costly components
- Fine-tune eutectic temperature

STABILITY/RATE-DEPENDENT SOLIDIFICATION

• Focus on eutectics of interest

R3 HIGH

R1

R2

R4

R5

HIGH THERMAL CONDUCTIVITY COMPOSITES

• Translate results to salt hydrate systems

MICROENCAPSULATION

- Domonotroto gonorality of to
- Demonstrate generality of techniqueTest robustness of microencapsulations

SHAPE STABILIZATION

 Translate initial results to eutectics of interest

R6 COMPONENT-LEVEL MODELING

 Optimize conductivity/latent heat trade offs for particular applications

Thank You

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REFERENCE SLIDES

Project Budget

Project Budget:

Variances: 4 mo. No-Cost Extension (04/2021 - 07/2021) to account for COVID-related delays to project start

Cost to Date: \$466,749 (DOE)/ \$163,058 (cost-share)

Additional Funding: N/A

Budget History					
04/01/2020 - FY 2020 (past)		FY 2021 (current)		FY 2022 – 07/31/2023 (planned)	
DOE	Cost-share	DOE	Cost-share	DOE	Cost-share
\$466,749.	\$163,058.	\$533,090.	\$110,212.	\$546,717.	\$113,369.

Project Plan and Schedule

3.7 Gantt Chart

R1.1 Ternary NO₃ eutectics R1.2 Ternary Cl eutectics R1.3 Mixed ternary eutectics R2.1 Identify nucleation catalysts R2.2 Stability of nucl. Catalysts **R3.1** Optimize graphite layer R3.2 Fabricate multi-layer composites R3.3 Fabricate encapsulated composites R4.1 Encapsulation of salt hydrates R4.2 Cyclic performance of capsules R4.3 Incorporate nucl. catalysts in capsules R5.1 Form PVA/PEO salogels **R5.2** Add crosslinker to tune T_{gel} R5.3 Validate long-term stability of gels R6.1 Lumped thermal model R6.2 Component-level thermal model R6.3 Validation of TES components



Start date: 04/01/2020 (**NEW START**) Planned end date: 07/31/2023

(4 mo. no-cost extension at end of yr 1 related to slow-start due to limitations in lab associated with COVID protocols.)

G/NG 1	Yr1, Q4	Are the posed PCM discovery approaches converging on acceptable PCMs? Identify eutectic nitrate PCMs which maintain high volumetric density (>80 kWh/m ³), while reducing cost to <\$10/kWh.
G/NG 2	Yr1, Q4	Is undercooling generally surmountable in salt hydrate PCMs through ID of appropriate nucleation catalysts? Demonstrate DT < 5 C in (a) ~2 to 5 nitrate hydrate eutectic phases, and (b) ~2 to 5 nitrate chloride eutectic phases of interest.
G/NG 3	Yr2, Q2	Can lamellar graphitic PCM composites increase directional thermal transport, while sustaining repeated melt cycles? Demonstrate effective thermal conductivity of >1 W-m ⁻¹ -K ⁻¹ for >90 vol.% PCM, and survivability for >1000 melt/freeze cycles with minimal degradation of thermal properties.
G/NG 4	Yr2, Q4	Are capsules of salt hydrates with less than 10 vol% shell stable to extended aging? Microencapsulation of target salt hydrate phase with >95% yield, and <5% degradation of DH _{fus} over extended aging (up to ~3 to 6 mo.).
G/NG 5	Yr2, Q2	Are polymer-based salogels capable of shape stabilization for both nitrate and chloride hydrate eutectics? Demonstrate thermo-reversible gelation in salogel consisting of >90 vol.% PCM for proto-typical nitrate hydrate eutectic and chloride hydrate eutectics.