West Virginia University

NATIONAL

TECHNOLOGY LABORATORY





John Hu DOE Ammonia for H₂@Scale Workshop

MICROWAVE CATALYTIC

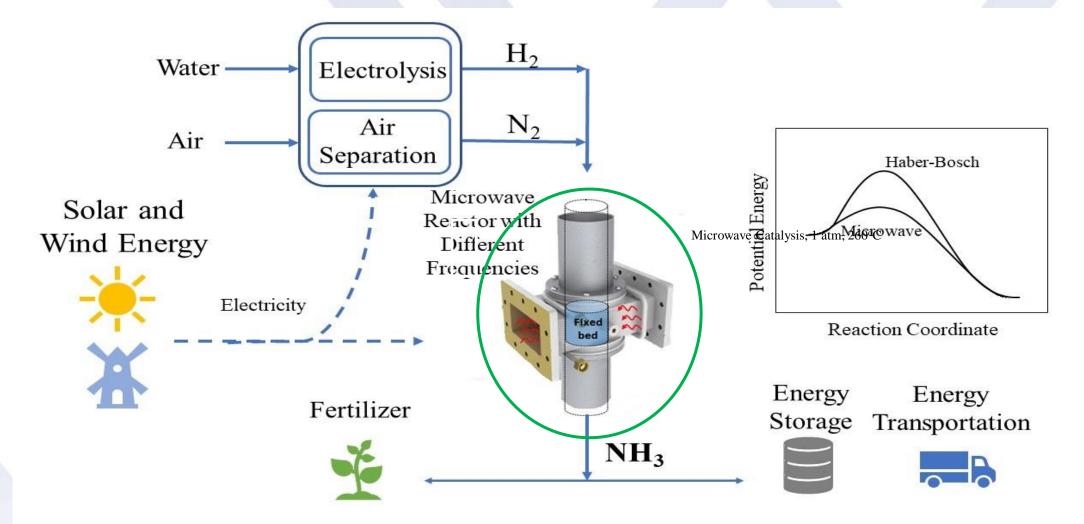
SYNTHESIS OF AMMONIA

May 6, 2021

West Virginia University National Energy Technology Laboratory (NETL) Malachite Technologies Shell



Renewable Power to Carbon-Neutral Liquid Fuel





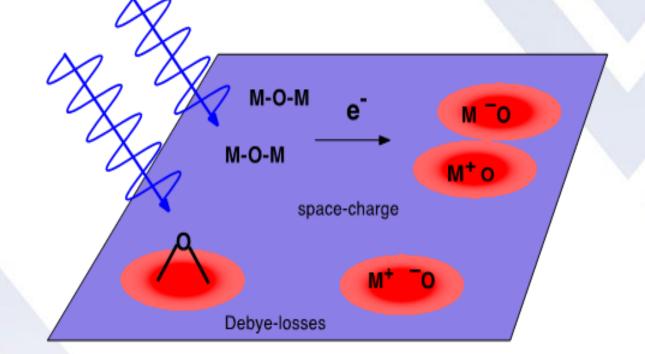
Ambient Pressure Ammonia Synthesis

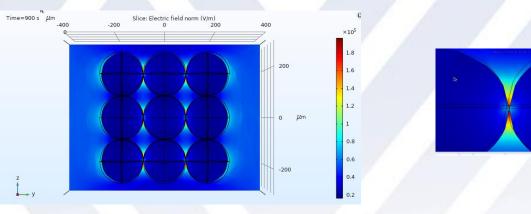
- Haber-Bosch process (3000 psig, 500°C, 1000 ton/day) difficult to scale down economically.
- Renewable power: 5 MW-100 MW, this is equivalent to 5-100 ton NH₃ per day.
- Renewable power –dealing with challenge of intermittent nature



Microwave Catalysis-Activation of N₂

Space-charge and Debye dielectric loss mechanisms for microwaves interacting with a catalyst surface for selective bond activation of reactant molecules

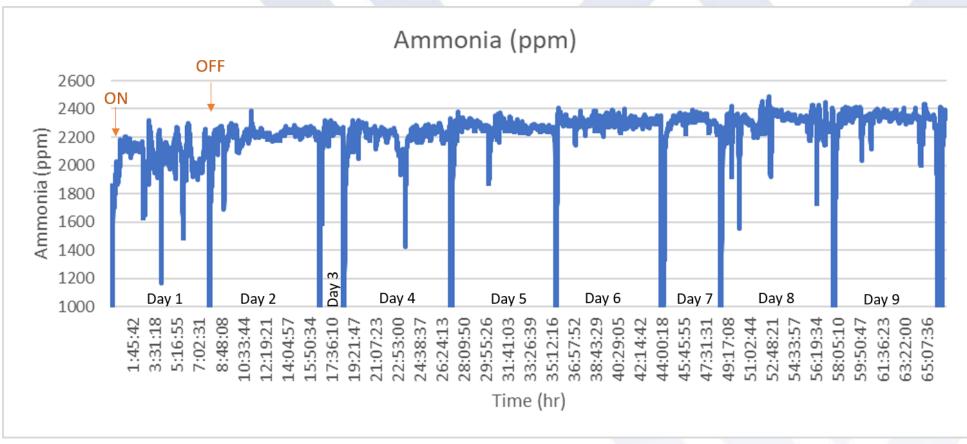




Strong e-field between particles could ionize N2 locally



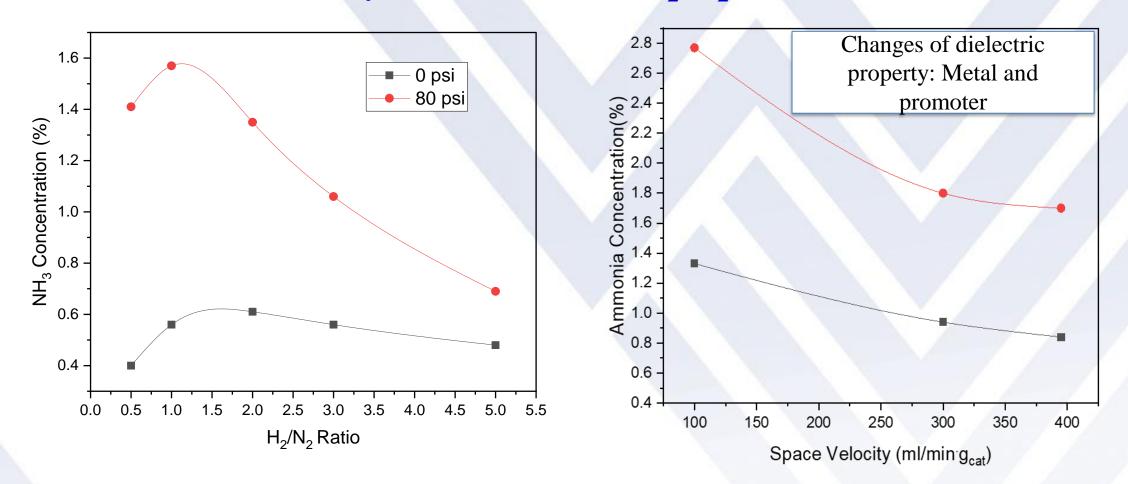
Tolerance of Intermittent Power Supply (260°C, ambient Pressure)



Microwave catalytic synthesis of ammonia under repeatedly startup and shutdown. During startup, reactor was back on-line within minutes. After 9 days, no performance deviation from the initial activity.



Haber Bosch: $3H_2 + N_2 = 2NH_3$ Microwave Catalysis: Pressure and H_2/N_2 Ratio (5.65 GHz, 260°C)

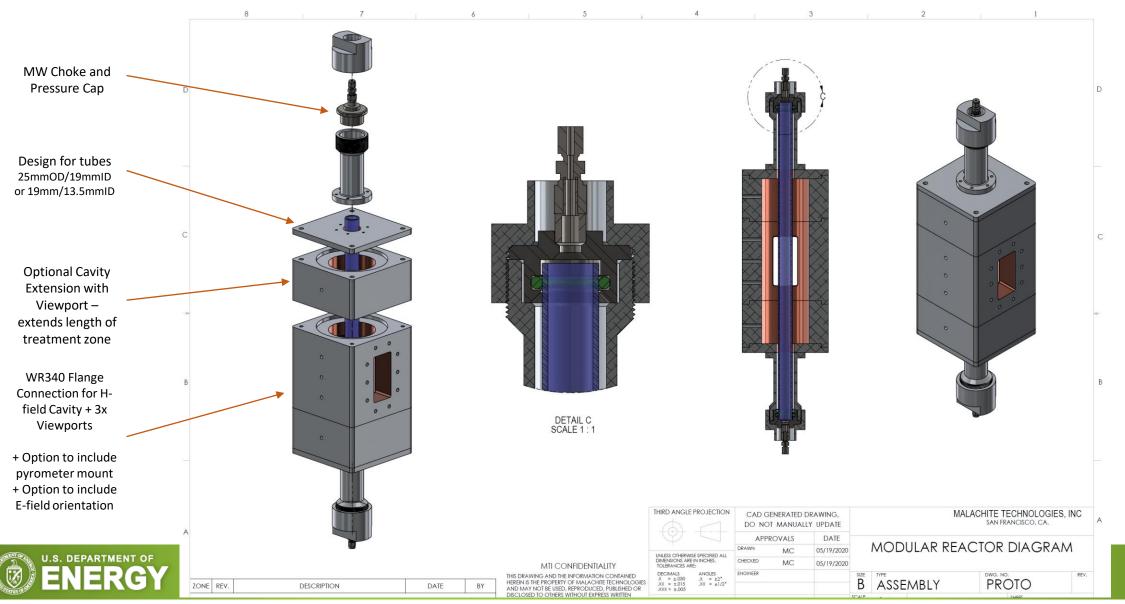


Accomplished: ammonia production rate: 0.3 gram/gram catalyst.hr



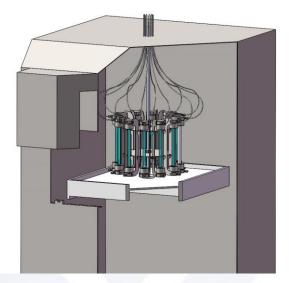
H-Field Reactor Design- Kilograms NH₃ /day





High Pressure Microwave Reactor





Rated: 500 psig, 1000°C. The first-of-its-kind in the field

Multimode Reactor Design Concept



Simulation and Technoeconomic Analysis

Detailed MISP of the Cases Minimum Setting						
Case	RM (MM\$/year)	UC (MM\$/year)	AD (MM\$/year)	Pressure (psig)	Single Pass H ₂ Conversion (%)	MSP (\$/kWh)
Case 1	24.6	6.73	6.63	80	8.08	0.123
Case 2	24.5	6.13	6.62	80	18.1	0.12
Case 3	26.0	6.72	4.51	300	10.3	0.12
Case 4	25.5	5.68	3.95	300	22.6	0.113
Case 5	26.8	6.24	3.06	500	11.4	0.117
Case 6	25.4	5.71	3.33	500	25.3	0.111
Case 7	24.0	5.91	2.86	500	30.8	0.106
Base Case 1	24.6	5.99	3.77	3970	34.1	0.111
Base Case 2	25.2	13.0	3.88	3970	8.08	0.136

Detailed MSP of the Cases

Minimum Selling Price, MSP $NH_3\left(\frac{\$}{kWh}\right) = \frac{(AD + UC + RM)}{(PC) * LHV_{NH3} * 8000}$

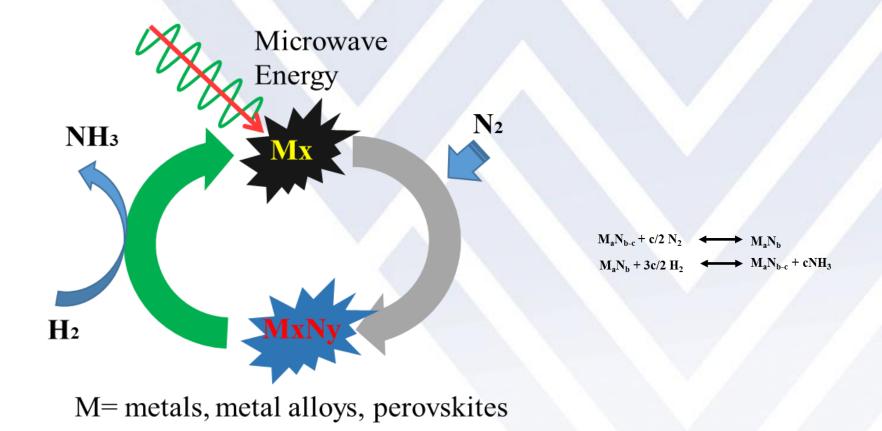
where AD = Annualized Direct CostUC = Utilities Cost RM = Raw Material Cost PC = Plant Capacity LHV_{NH3} = Ammonia lower heating value

Plan of the next step: 1 ton per day pilot demonstration. Forming an investable business entity.



Case 1-7 has favorable MSP(MSP < \$0.13/kWh)

Future Development-Metal Nitride Reaction under Microwave





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