

A Transient Kinetic Approach to Catalytic Materials For Energy-Efficient Routes to Ammonia, Ethylene and Related Chemicals

Contract #32132

**Chemical Catalysts Research and Testing/Idaho National Laboratory,
Argonne National Laboratory
Project Period 10/1/2016 to 9/30/2018**

Rebecca Fushimi, Idaho National Laboratory

U.S. DOE Advanced Manufacturing Office Program Review Meeting
Washington, D.C.
July 17-19, 2018

This presentation does not contain any proprietary, confidential, or otherwise restricted information.

Project Overview

Timeline

- Project commenced October 2016
- Projected end date September 2018
- Project 90% complete

Budget

	FY 16 Costs	FY 17 Costs	FY 18 Costs	Total Planned Funding (FY 19- Project End Date)
DOE Funded	\$850 k	\$3.0 M	\$2.7 M	\$6.55 M
Project Cost Share	—	—	—	—

Barriers

- Molecular-level knowledge of catalyst control of ammonia synthesis and methane activation chemistries.

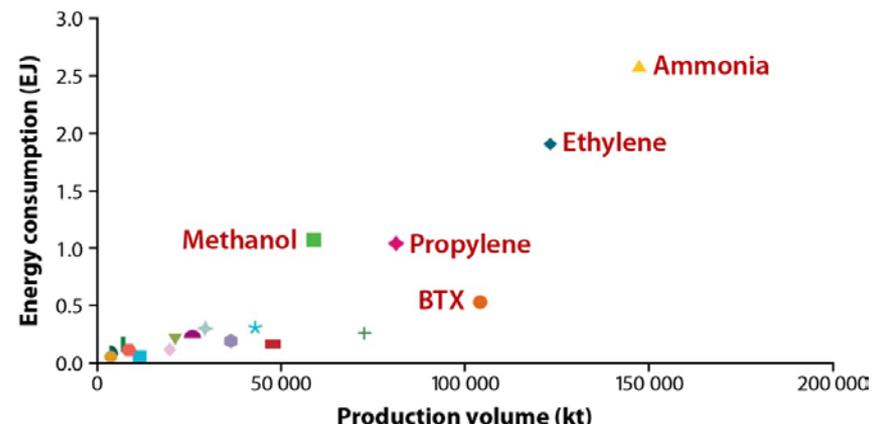
Partners

- Idaho National Laboratory
 - Transient kinetics
- Argonne National Laboratory
 - Atomic layer deposition of catalyst materials
- California Institute of Technology
 - Atomistic modeling
- University of Virginia
 - Homogeneous catalysis
- Georgia Institute of Technology
 - Data science and analytics

Project Objective

- Challenge: Reduce energy intensity chemical manufacturing

- Ammonia: 2% of the world's energy use
- Ethylene: 30% energy saving with catalysis over current steam cracking practice

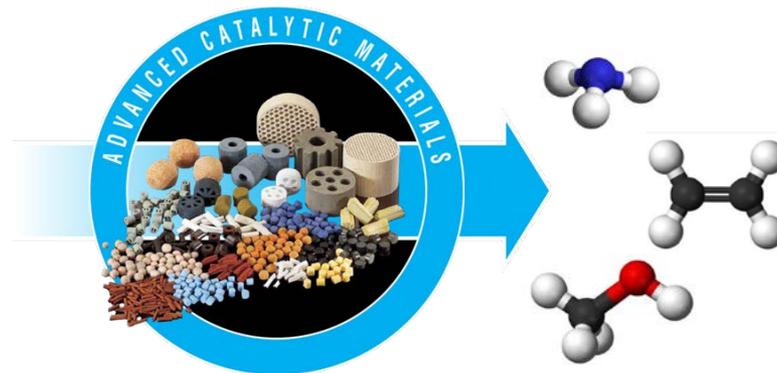


International Energy Agency, Technology Roadmap, Energy and GHG Reduction in the Chemical Industry via Catalytic Processes, 2013.

- Barrier: Catalyst development is primarily trial-and-error

- Complex multistep reaction mechanism
- Complex multicomponent, ill-defined materials

- Goal: Accelerate catalyst development through molecular-level knowledge provided by *transient kinetics*

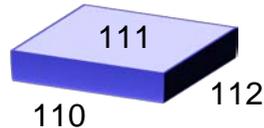


Technical Innovation

- Current practice of catalyst development

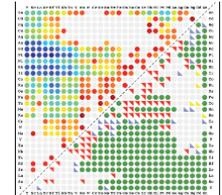
Surface Science

- Detailed kinetics
- Model materials



High-Throughput

- Basic kinetics
- Industrial materials



Curtarolo, Stefano, et al. *Nature materials* 12.3 (2013): 191-201.

- New Approach: *Transient kinetics*

Temporal Analysis of Products (TAP) Reactor System

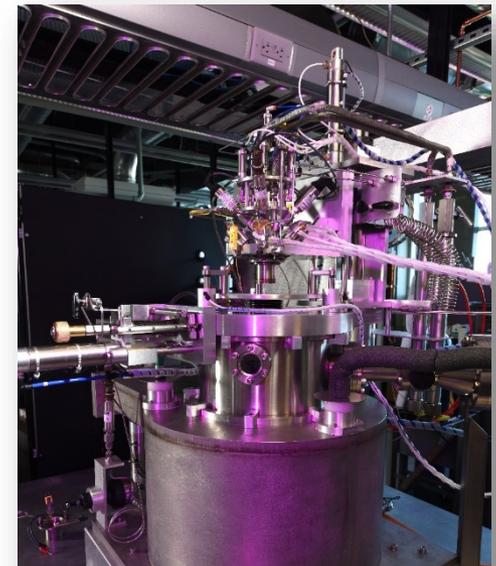
- Detailed kinetics : Complex mechanism
- Industrial materials : Complex materials

Exploiting data science tools around microkinetics

Understanding *how* and *why* materials function

- Technology Impact

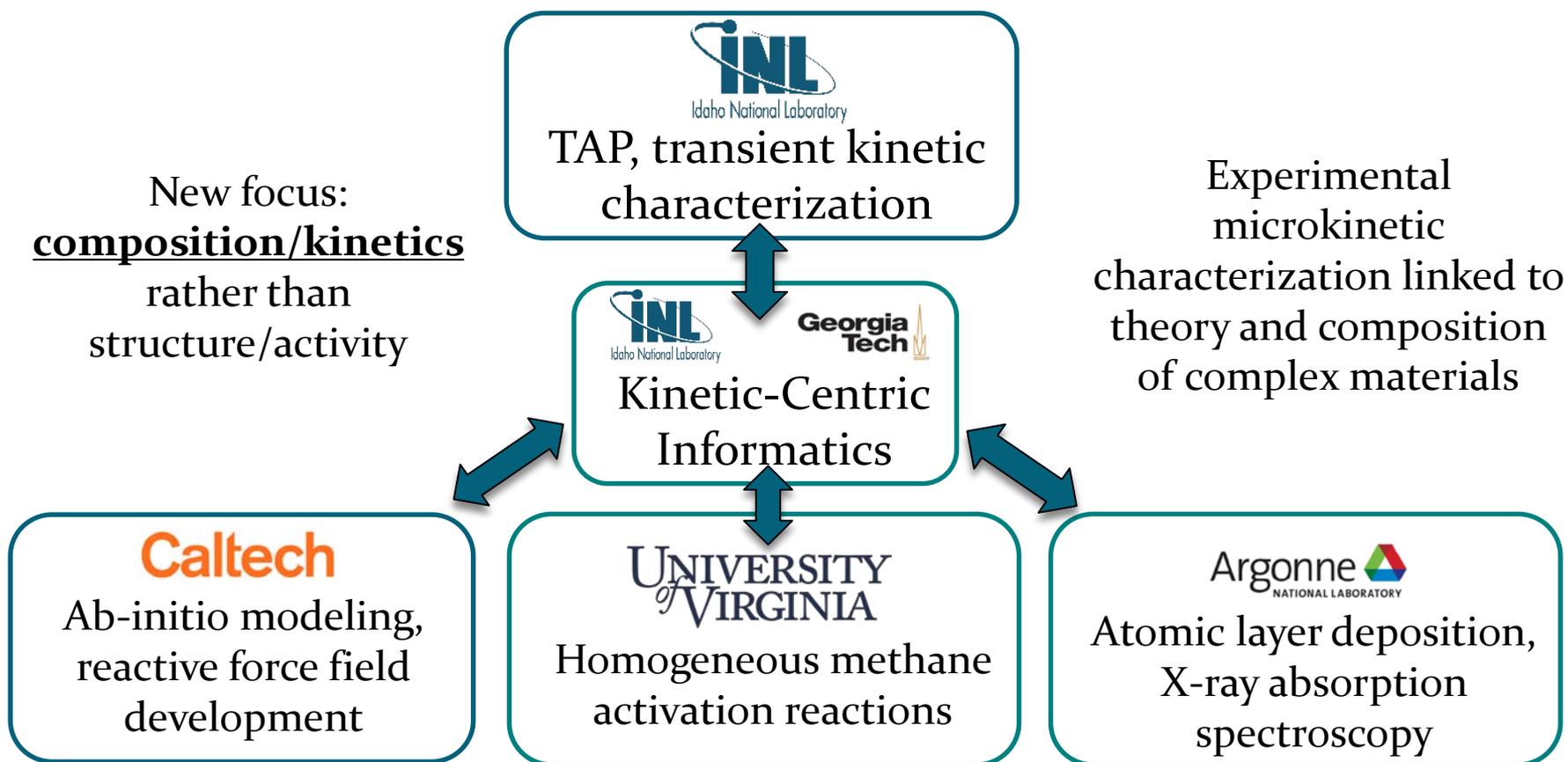
- Advanced catalysts (lower temperature & pressure)
- Accelerating the catalyst development cycle



INL: 2 of 3 TAP
Reactors in the US

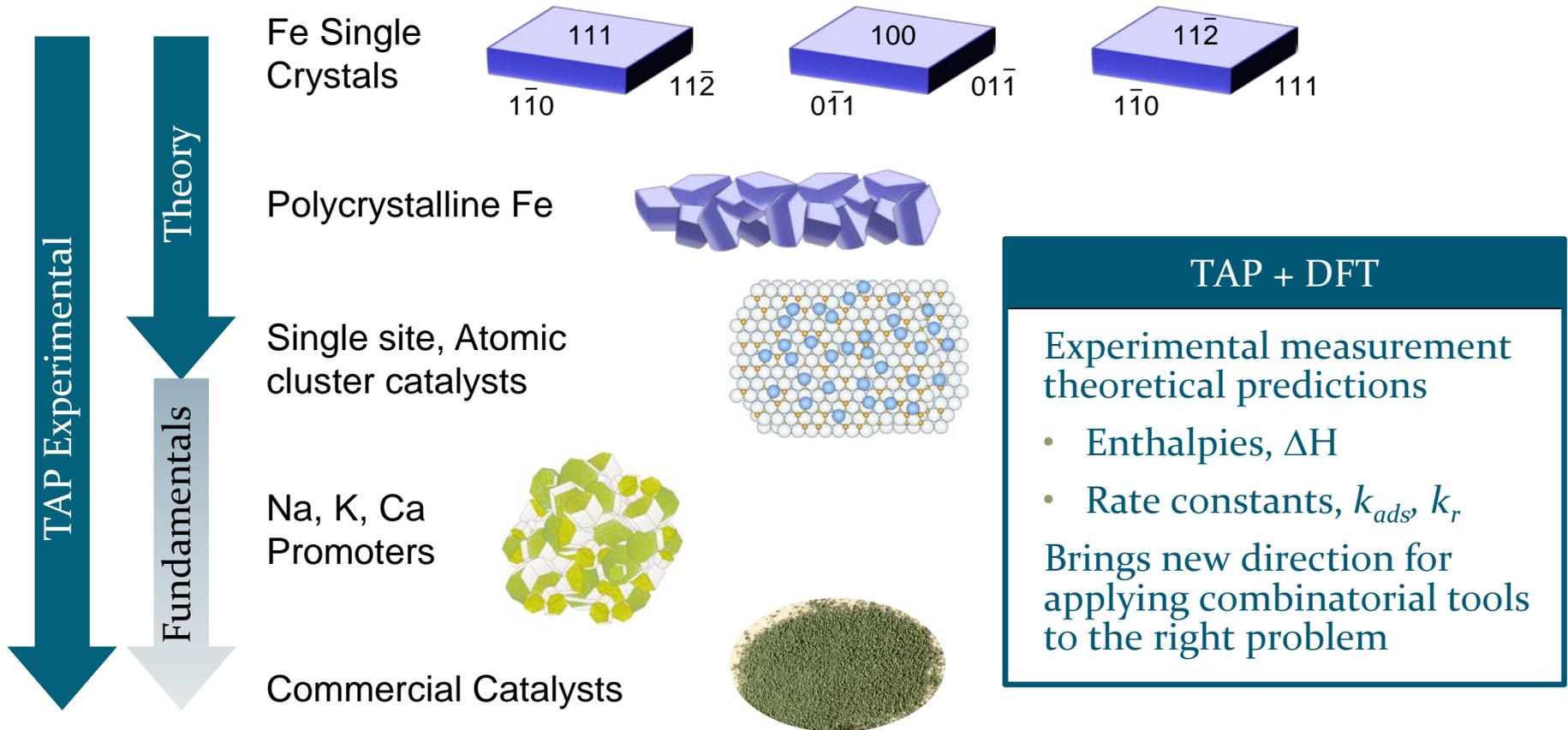
Technical Approach

- A new paradigm for catalyst development:
Transient kinetic experiments
- Kinetic-Centric Informatics tool:
More meaningful connections from complex data sources



Technical Approach

- Ammonia synthesis from 200 to 20 bar
 - Elementary reaction steps, N-N rupture, N-H formation
 - Incremental surface coverage change, N, NH, H, etc.
 - Material complexity

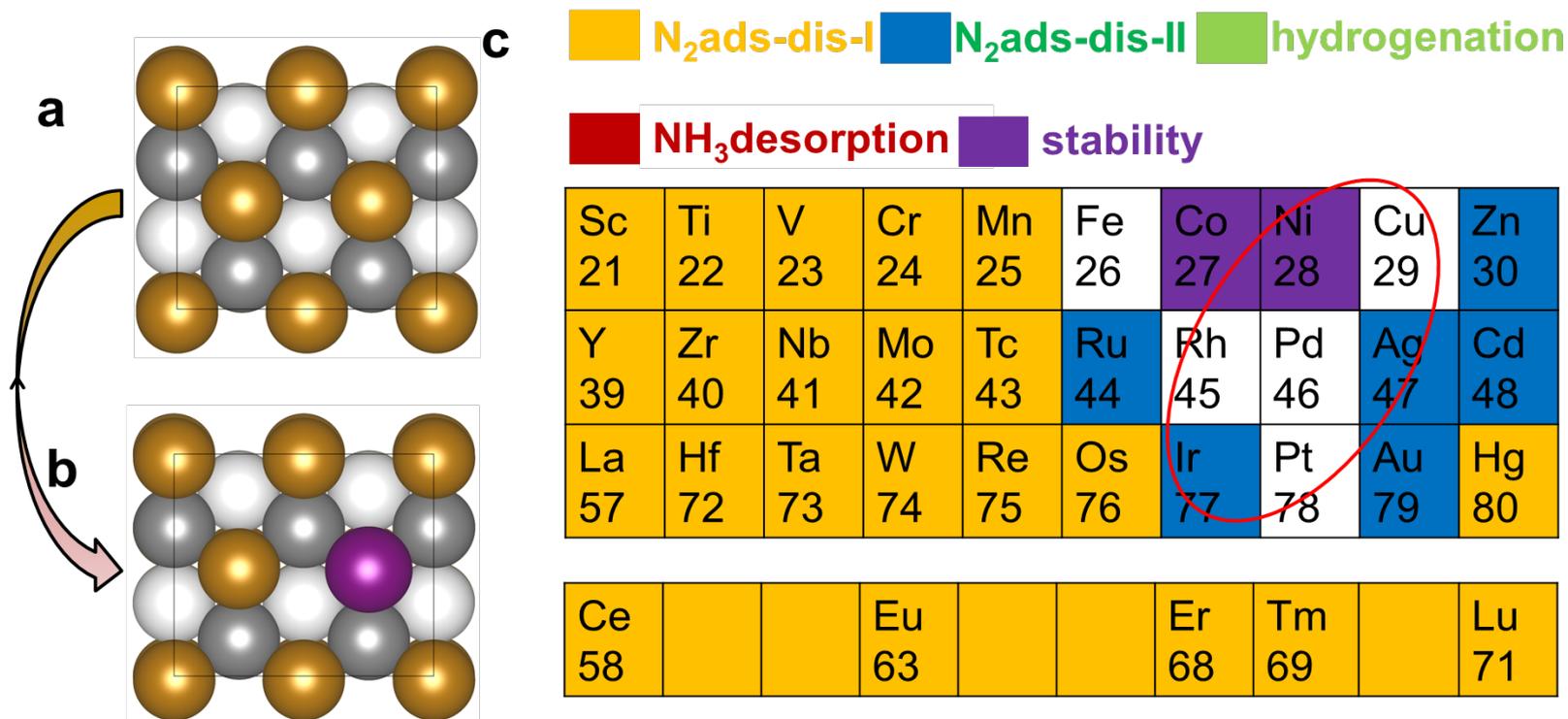


Results and Accomplishments

- **Atomistic Modeling:**

- Complete *ab initio* modeling of 17 key steps of ammonia synthesis on Fe(111)
 - Qian, J., An, Q., Fortunelli, A., Nielsen, R.J. and Goddard III, W.A., (2018) Journal of the American Chemical Society, 140(20), pp.6288-6297.
 - Modeling of Fe-alloys predict increase in NH_3 rate by 18,000x
 - Initial results of *in silico* screening of 34 alloy compositions

Colors indicate at which step dopants were eliminated in high throughput screening

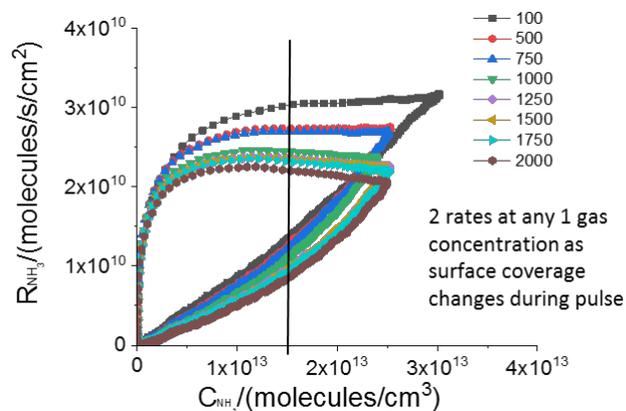


Results and Accomplishments

- **Accomplishments Transient Kinetics:**

- TAP transient kinetic testing of NH_3 decomposition supporting atomistic predictions for NH_3 synthesis
- Alloy testing: Materials prepared via incipient wetness, strong electrostatic adsorption, electroless deposition
- Developed new data analysis techniques of transient data
 - Software for rate/concentration analysis of exit flux
 - Captures intrinsic rate constants as the pulse changes the concentrations
 - TAP becomes a high-throughput device for sampling kinetic states.

Medford, A.J., Kunz, M.R., Ewing, S.M., Borders, T. and Fushimi, R.R., 2018. *ACS Catalysis*. Accepted.



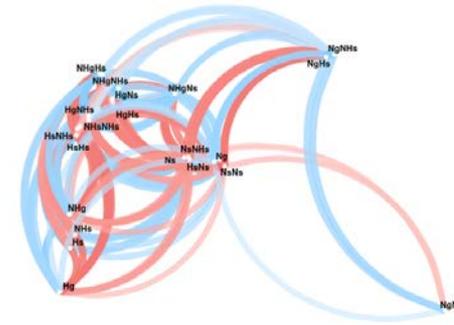
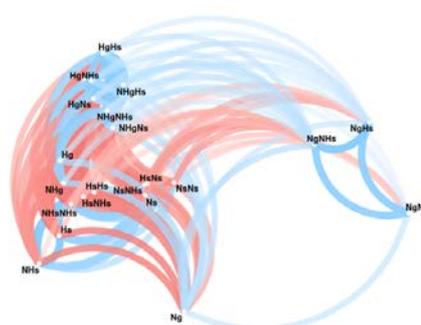
New Transient Analysis Theory and Software

R. Kunz, E.A. Redekop, T. Borders, L. Wang, G.S. Yablonsky, R. Fushimi, (2018) *Chem. Eng. Sci.* Submitted

Materials perform differently at a global level

Iron Catalyst

Iron Alloy Catalyst



New correlation structure analysis uses transient data to understand which reaction steps are important

Transition and Deployment

- Industry Engagement

- 1 IDR, 7 publications submitted, 9 conference presentations
- Industry Workshop, June 7, 2017
- CRADA in development

Representatives from 12 companies attending



Developing a unique capability for industrial catalyst development

- Sustainment Model:

- Industrial Utilization: project results that speak to proprietary needs
- A specialized team surrounding transient kinetics
- Accelerate and derisk proprietary projects
- Advance the state-of-the-art: theoretical methods, advanced instrumentation, transient kinetics, high throughput *in silico* and experiment, data analytics

Questions?
