

## Radiation Tolerance of Nanostructured Ceramic/Metal Composites

**Michael Nastasi**

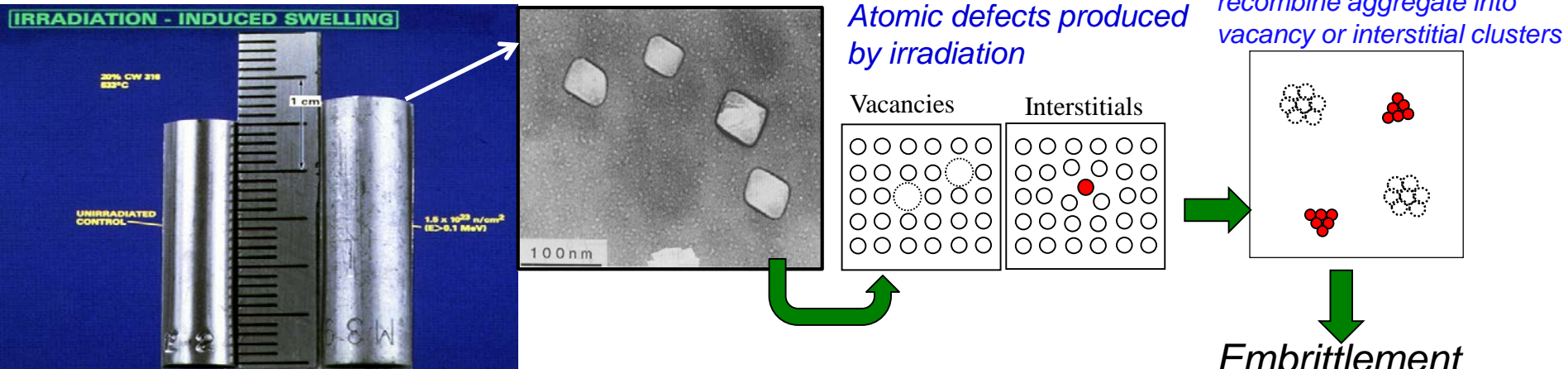
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University of Nebraska-Lincoln

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Work supported by DoE Office of Nuclear Energy, Nuclear Energy Enabling Technologies,  
award DE-NE0000533

# What is the problem

## Structural materials are prone to radiation damage: *void swelling and embrittlement*

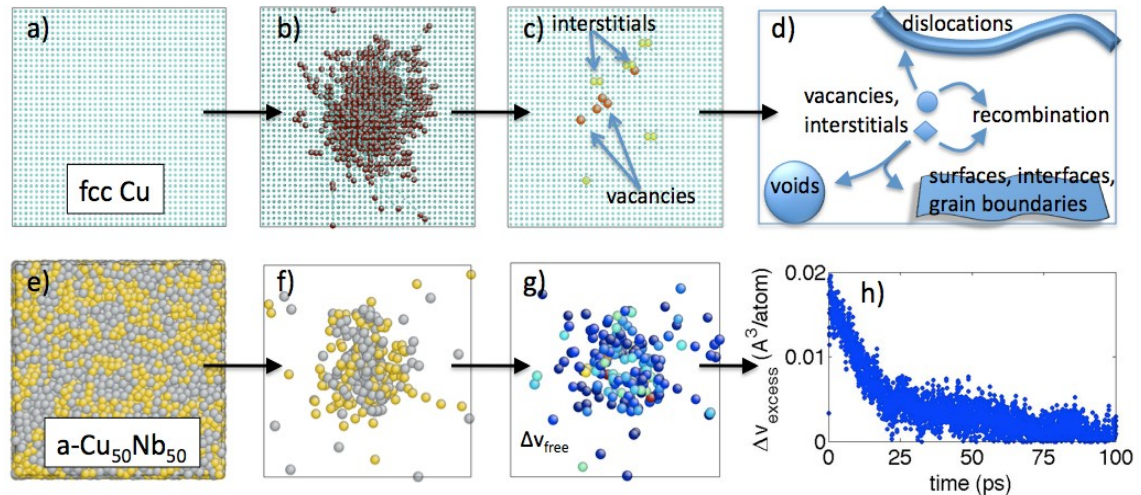
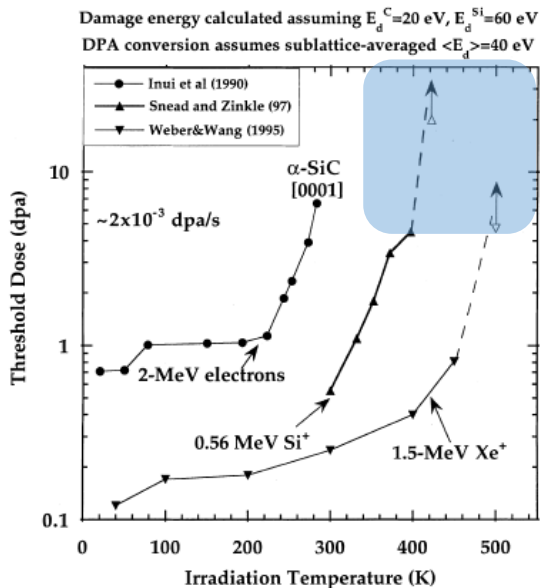


D.L. Porter and F. A. Garner, J. Nuclear Materials, **159**, p. 114 (1988)  
D.J. Bacon and Y.N. Osetsky, Int. Mater. Rev., **47**, p. 233 (2002).  
H. Trinkaus and B.N. Singh, J. Nuclear Materials, **323**, p. 229 (2003).

## How to design radiation damage tolerant materials?

# From crystalline to amorphous materials

**Upper limit**



## Questions:

- How do amorphous materials respond to radiation damage
- Are the interfaces between crystalline and amorphous materials also effective defect sinks?

L.L. Snead *et al.*, Nucl. Instr. and Meth. in Phys. Res. B 141, p. 123 (1998).

R. E. Baumer *et al.*, Materials Research Letters, 2, p 221 (2014).

# Team capabilities

**nanostructured  
amorphous-  
ceramic/metal  
composites**

## Sample Synthesis

1. Sputtering (thin film)
2. Pyrolysis, plasma-enhanced sintering (bulk)

## Ion Irradiation and He Ion Implantation

1. Si ions: 0.5-3 MeV,
2. Fe ions: 0.5-4 MeV.
3. He ions: 50-140 keV

## Microstructure and Mechanical Properties Evaluation

1. TEM, SEM, XRD
2. Nanoindentation testing

## Atomistic Modeling

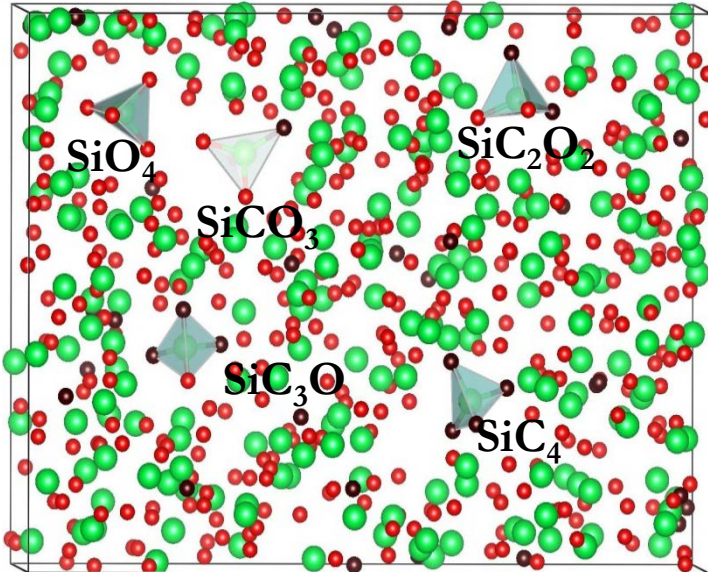
1. First principles density functional theory (DFT)
2. First principles molecular dynamics (MD) simulations

# Outline

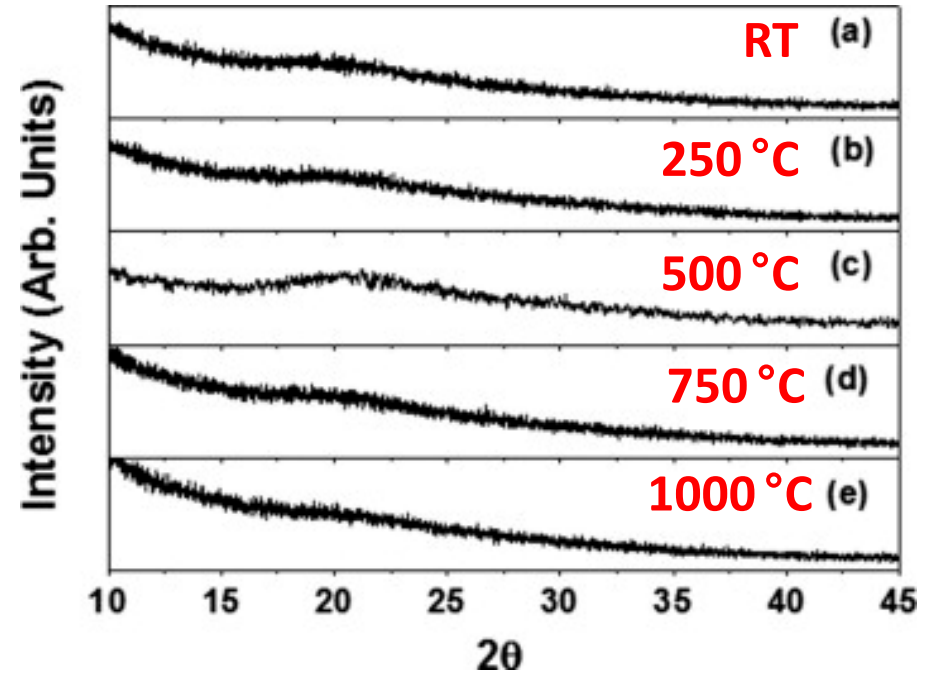
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- **Radiation tolerance of amorphous SiOC**
- Radiation tolerance of Fe/SiOC nanocomposites
- Modelling
- Summary

# Thermal stability of amorphous SiOC



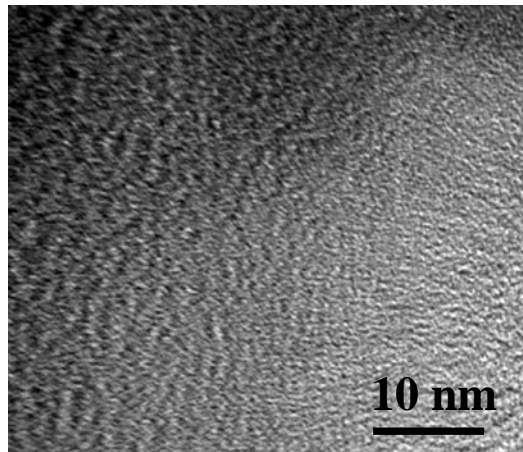
Green: Si, Red: O, and Brown: C  
Five types of tetrahedra units



**Amorphous SiOC is stable >1000 °C and with good oxidation and creep resistance**

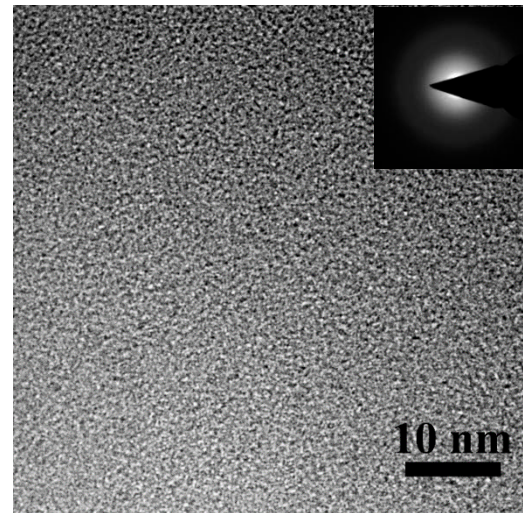
# Irradiation stability of amorphous SiOC

*As-prepared*



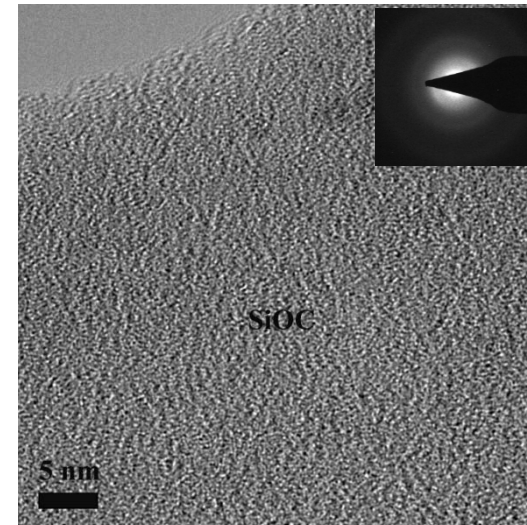
600 °C irradiation

100 keV *He*, 20 dpa



300 °C irradiation

1 MeV *Kr*, 5 dpa



***SiOC remains amorphous!!!***

# Summary of Kr and He irradiation

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Ion species	Kr	He
Acceleration voltage	1 MeV	120 keV
Cascade type	Large damage cascade zones	Scattered point defects
Irradiation temperature	RT to 300 °C	RT to 600 °C
Dose	Up to 5 dpa	Up to 20 dpa
Crystallization	No	No
Void formation	No	No



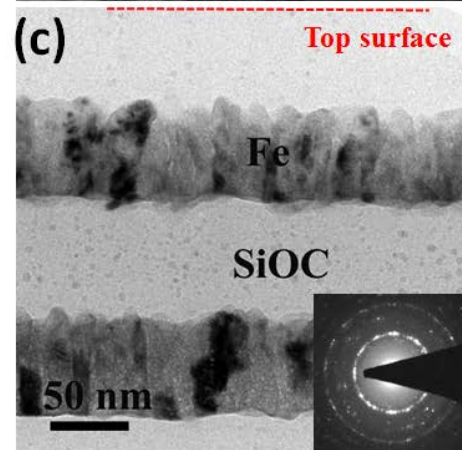
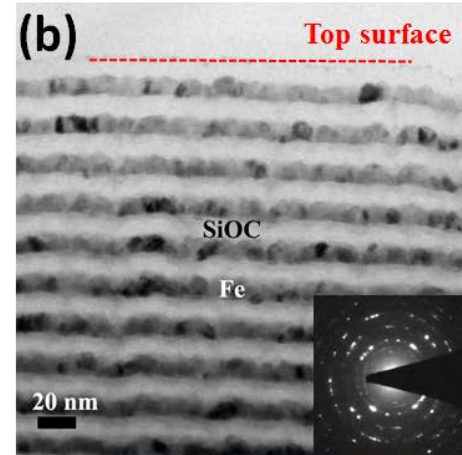
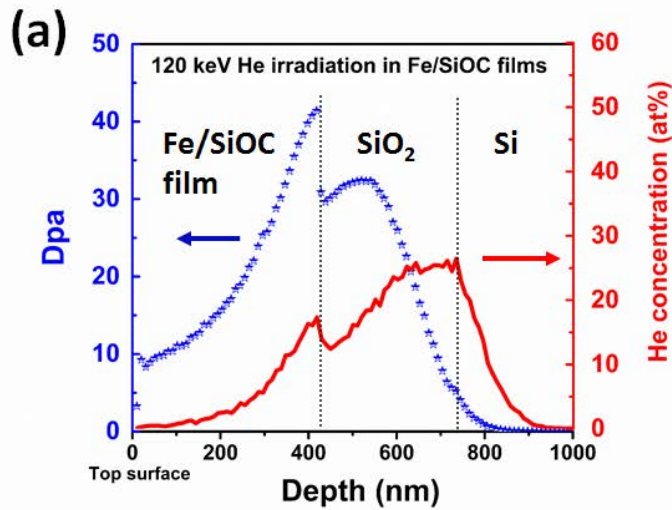
# Outline

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- **Radiation tolerance of Fe/SiOC nanocomposites**
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# Development of Fe/SiOC composite

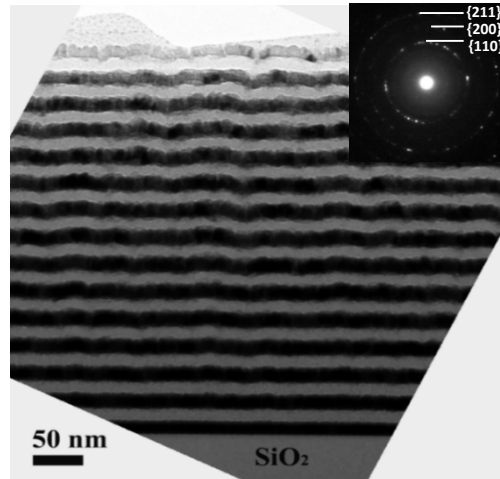
Goal: To possess good mechanical properties, be capable of operation at temperatures greater than 500°C, and have extreme radiation tolerance.



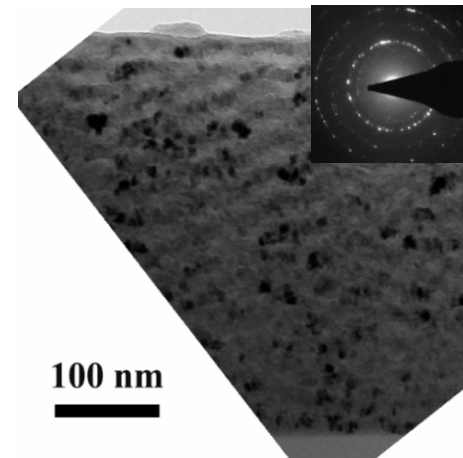
Fe/SiOC interface acts as an efficient point defects sink and results in good radiation stability of Fe/SiOC composite system.

# Temperature-dependent Kr irradiation

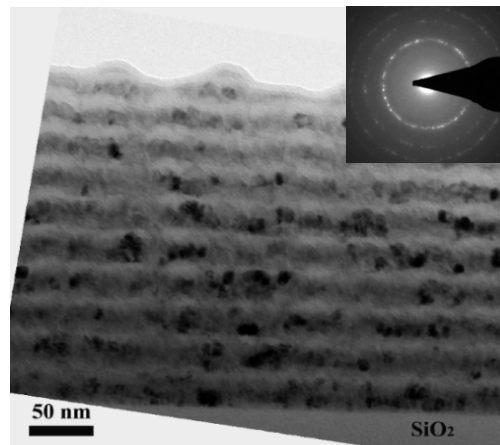
(a) Virgin



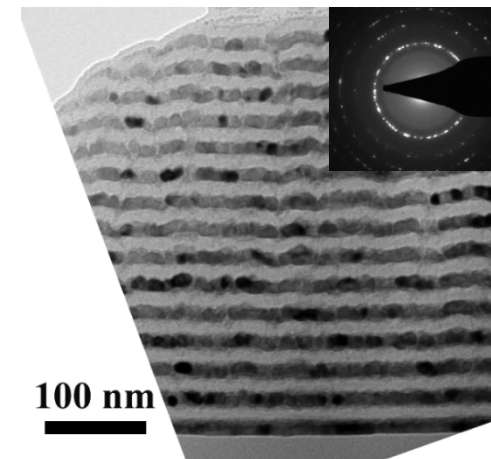
(b) 50K,  $4 \times 10^{14}$  ions/cm<sup>2</sup>



(c) RT,  $4 \times 10^{14}$  ions/cm<sup>2</sup>



(d) 300 °C,  $8 \times 10^{14}$  ions/cm<sup>2</sup>



**More radiation stability at elevated temperatures (thermodynamically stable) which benefits engineering applications.**

# Outline

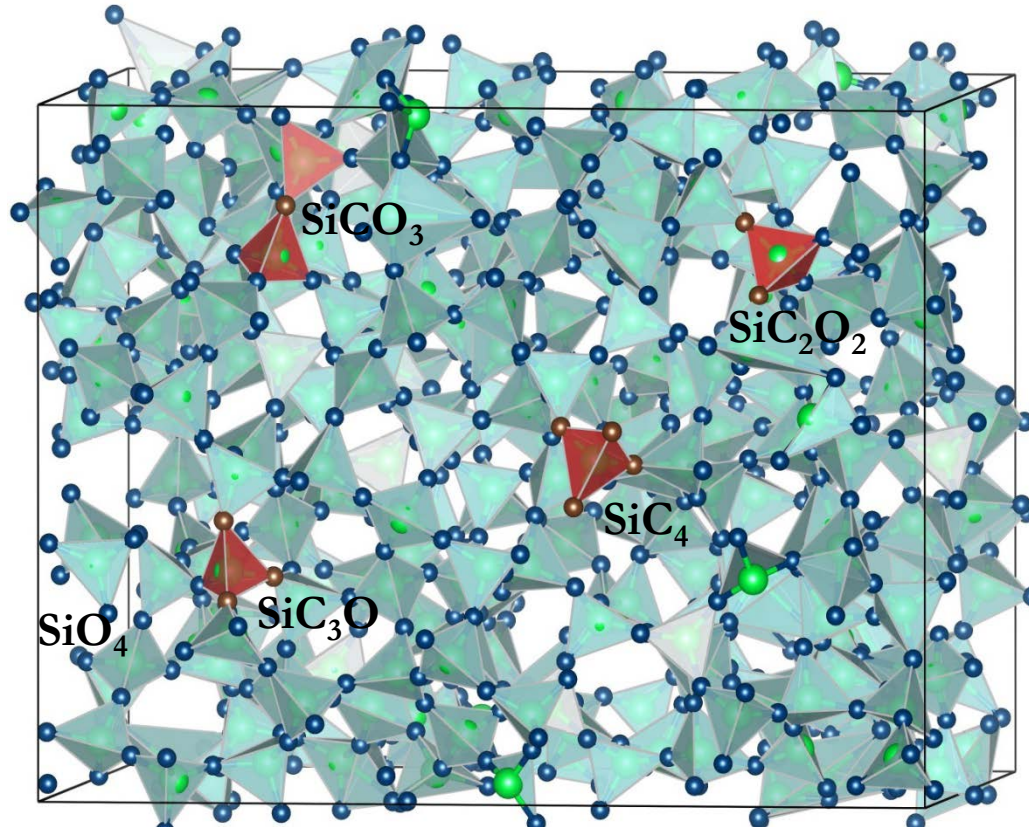
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# Atomic structure of SiOC

Amorphous SiOC = **continuous random network of SiOC** + free C phase + segregated SiC  
Non-crystalline: disordered, but not random

- Short range order ---  $\text{SiC}_n\text{O}_{4-n}$  tetrahedra ( $n$  ranges from 0 to 4)
- Medium range order --- tetrahedra are corner-sharing



Si atoms in  
**Green**

O atoms in  
**Blue**

C atoms in  
**Brown**

How is the C distributed?

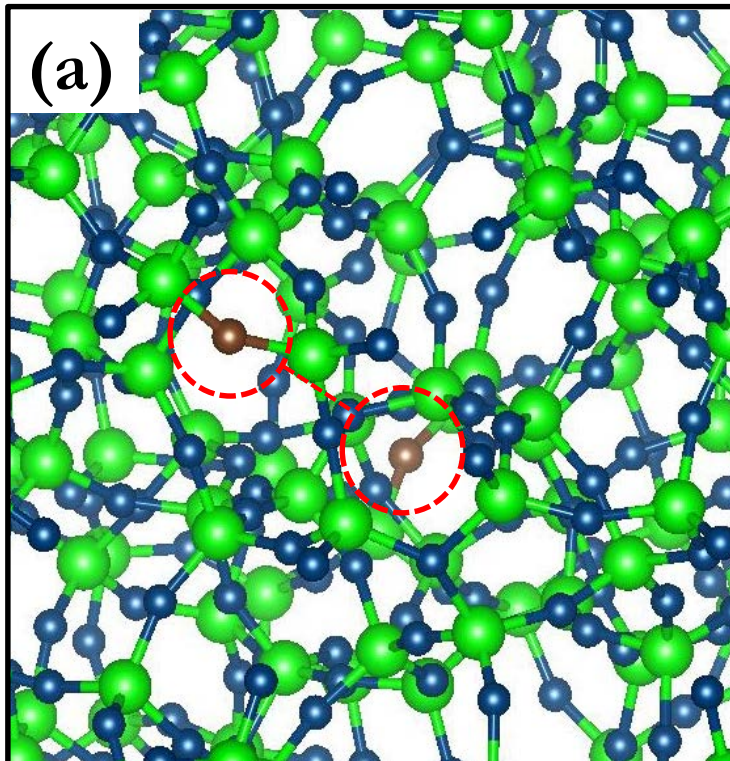
# Generation of amorphous SiOC

Classical melting and quenching **fails** for SiOC because of CO/CO<sub>2</sub> formation

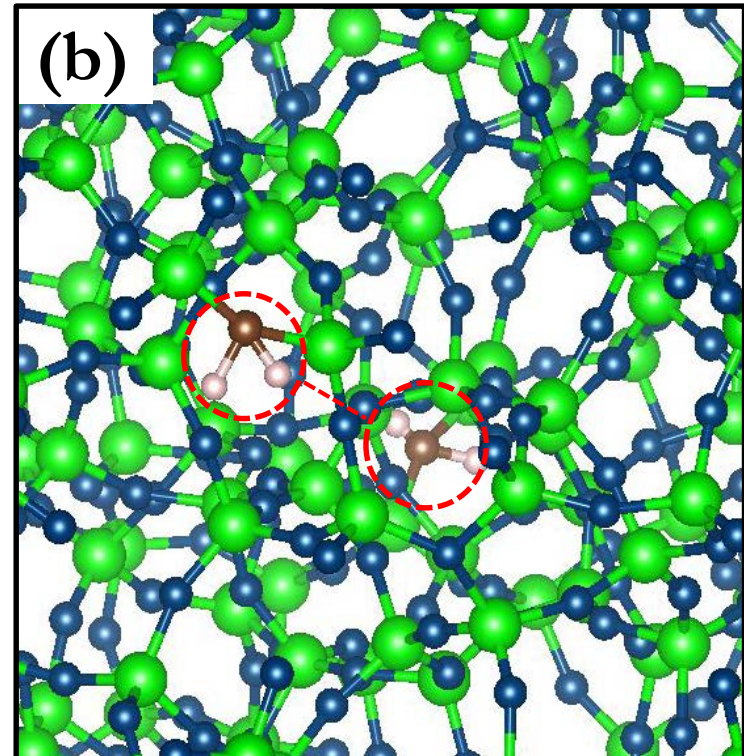
Only thermally stable below ~ 1500 K; Phase separation before melting

We start from amorphous SiO<sub>2</sub> (SiO<sub>2</sub> and SiOC share the same basic tetrahedral units)

Introduce C (a) or CH<sub>2</sub> (b) as dopants to replace O



(a) Replace O with C



(b) Replace O with CH<sub>2</sub>

# C-C interactions in SiOC (DFT)

(a) C replacing O

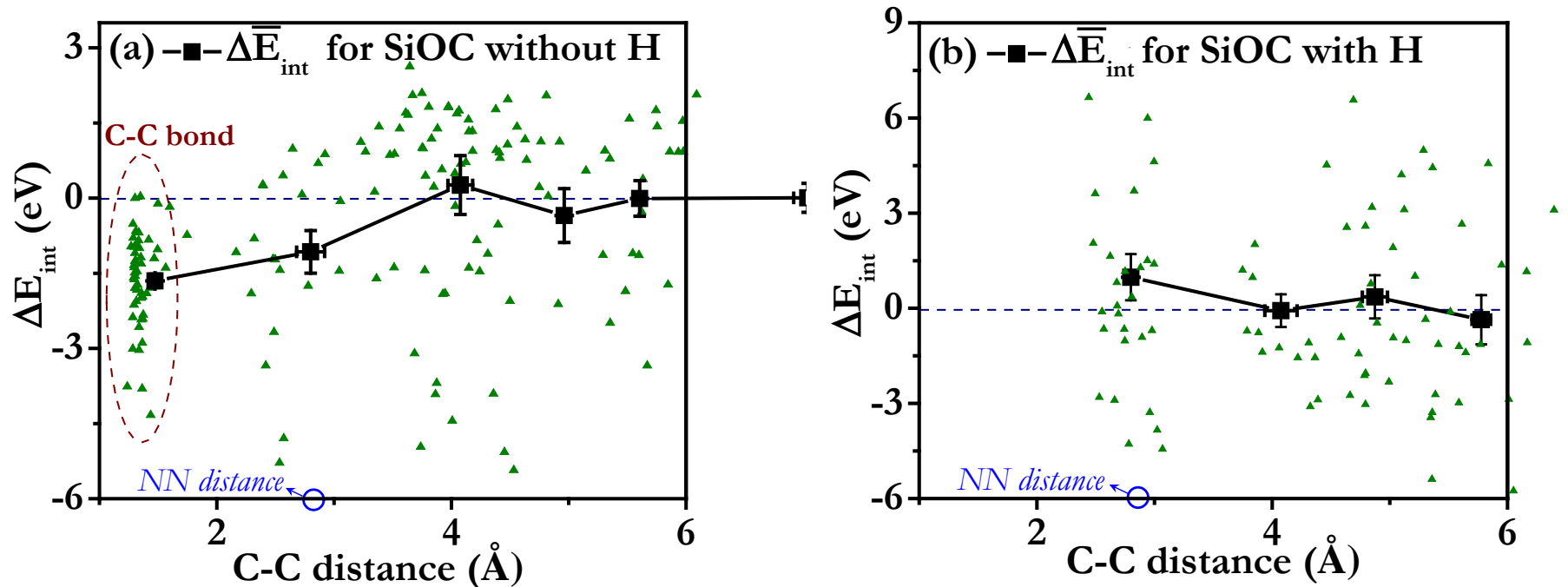
C-C interaction **attractive** at nearest neighbor (NN) distance

→ Clustering of C atoms

(b) CH<sub>2</sub> replacing O

C-C interaction **repulsive** at NN distance

→ Uniform dispersal of C atoms

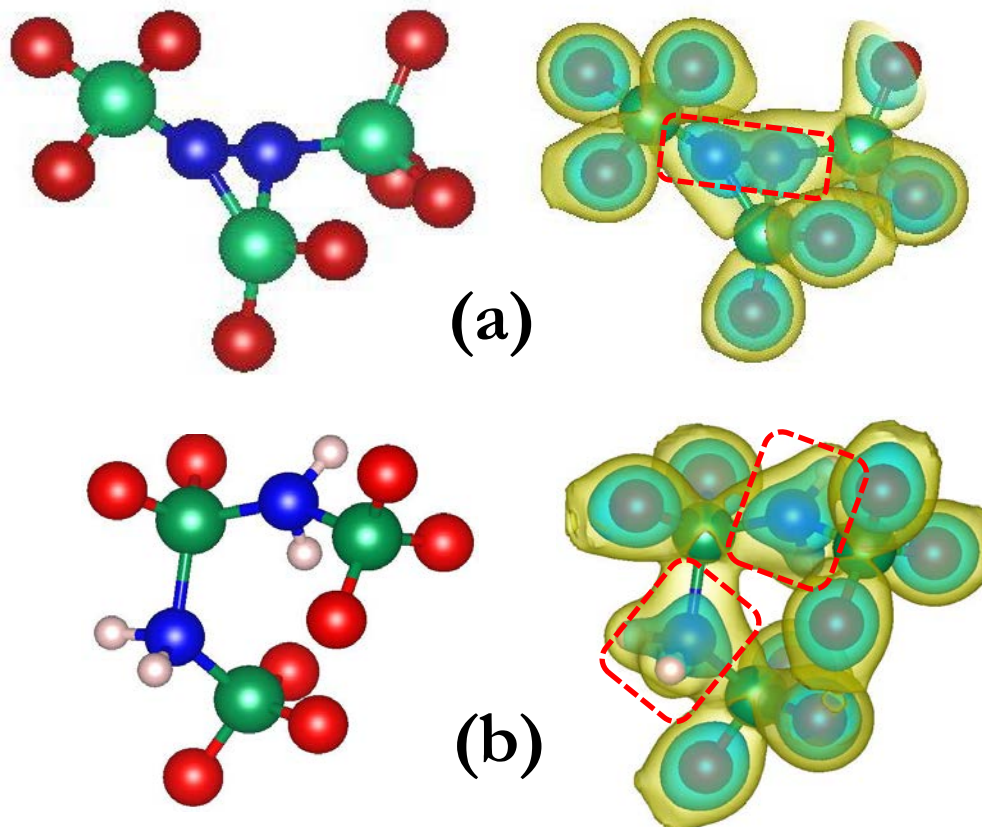


# Bond configurations of C atoms

(a) Without H: Direct C-C bond formation

Initiation of phase separated C network

(b) With H: No direct C-C bonding

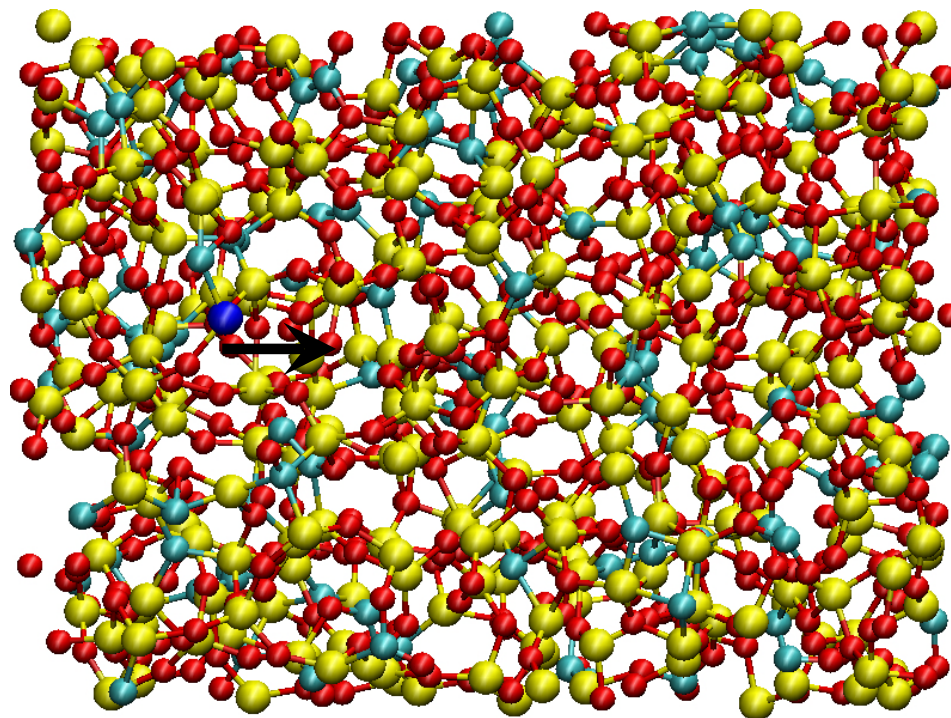
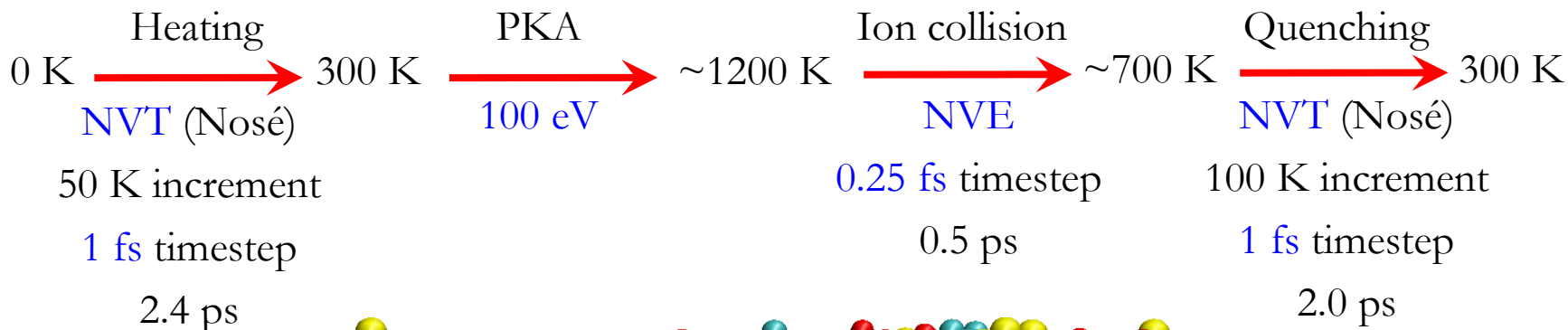




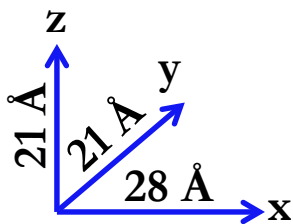
# Knock-on damage in SiOC: first principles molecular dynamics (MD)

DFT MD using VASP: PAW-PBE, 500 eV, Gamma point only

Primary knock-on atom (PKA) kinetic energy: 100 eV



Si atoms in **Yellow**  
O atoms in **red**  
C atoms in **Cyan**  
PKA atom in **Blue**



# Potential energy change due to knock-on

## ➤ $\text{SiO}_2$

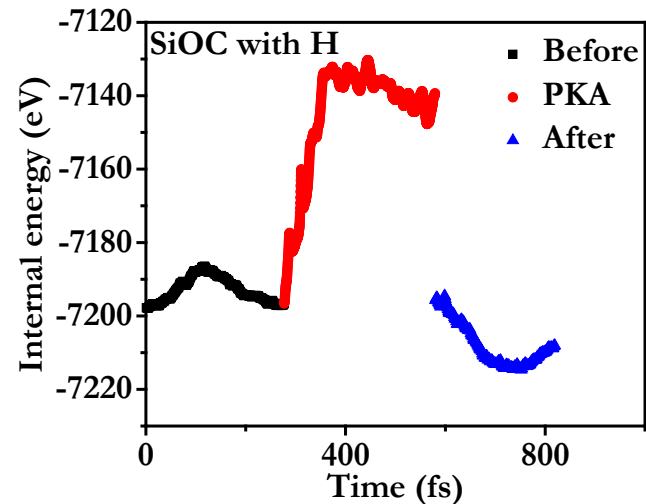
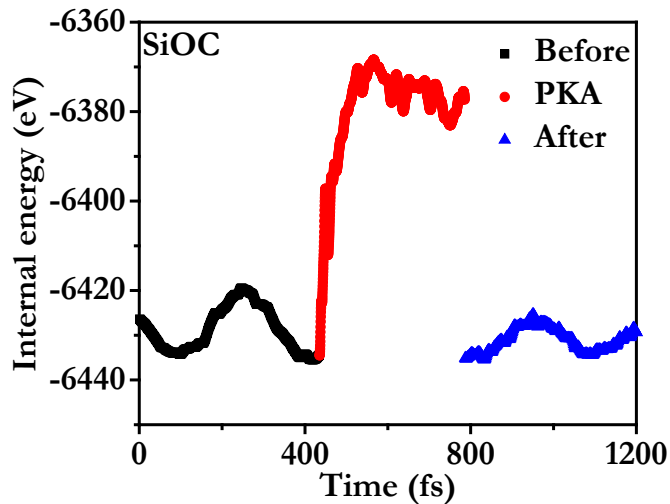
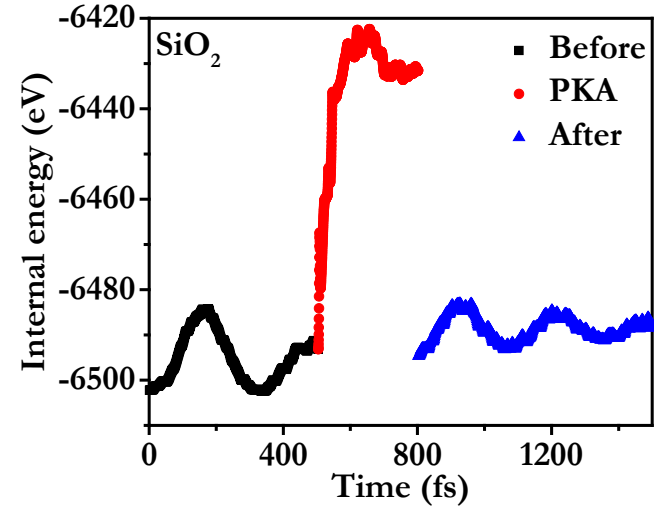
Potential energy slightly *increases*

--- Less stable after knock-on

## ➤ $\text{SiOC}$ and $\text{SiOC}$ with H

Potential energy the same or lower

--- Stable upon knock-on



# Bond breaking and reformation due to PKA

Number of bonds broken and reformed

➤ SiO<sub>2</sub>

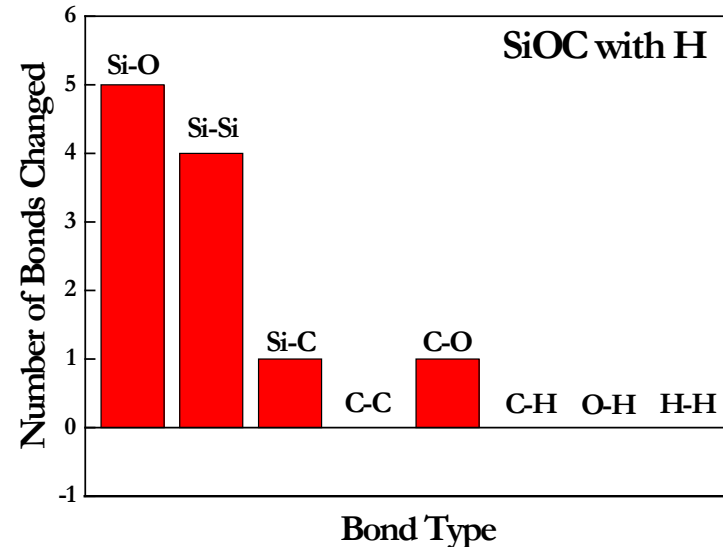
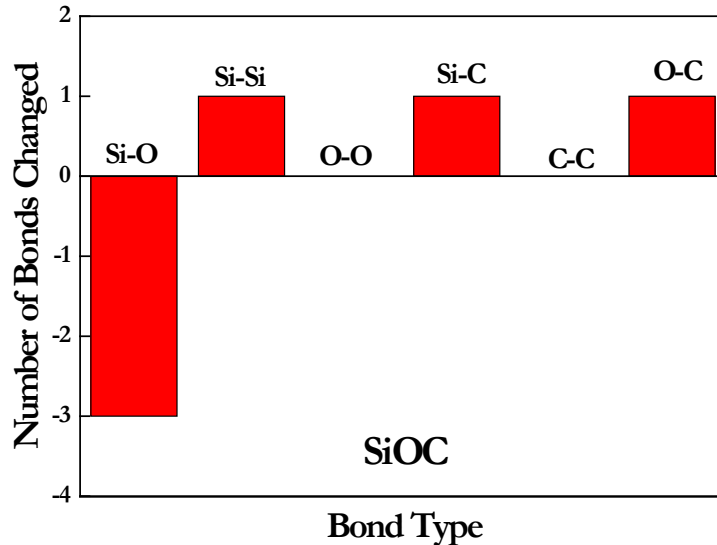
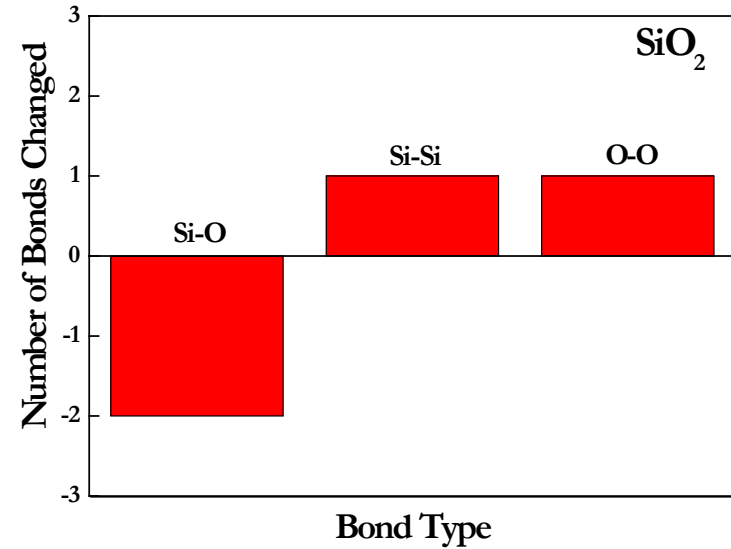
--- 26 broken, 26 reformed

➤ SiOC

--- 36 broken, 36 reformed

➤ SiOC with H

--- 36 broken, 48 reformed



# Summary: Experimental

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- **Amorphous SiOC is radiation stable (He and Kr)**
  - I. **20 dpa at 600 °C (He)**
  - II. **5 dpa at 300 °C (Kr)**
  
- **Irradiation stability of Fe/SiOC nanocomposite**
  - I. **Room temperature stability up to ~40 dpa.**
  - II. **Amorphous SiOC/crystalline Fe interface is demonstrated as defect sinks.**
  - III. **Enhanced stability at elevated temperature.**

# Summary: Modelling

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## Amorphous SiOC

- Without H, C-C interaction is attractive, C tends to cluster
- With H, C-C interaction is repulsive, C tends to disperse

*Sci. Rep. 5, 13051 (2015)*

## Knock-on damage in amorphous SiO<sub>2</sub>, SiOC, and SiOC with H:

No major structural change up to 1200 K --- thermally stable

### ➤ SiO<sub>2</sub>

Broking of Si-O bonds, formation of Si-Si and O-O bonds

### ➤ SiOC and SiOC with H

No individual CO, CO<sub>2</sub> or H<sub>2</sub>O molecules formed due to ion irradiation

# Going Forward

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- **Year 1:** Evaluation of the role of Fe/SiOC interfaces on defect mitigation out to harsher environments with >300 dpa and >500 °C. Evaluate the role of SiOC and Fe/SiOC interfaces on He incorporation. Determine mechanical properties (hardness, elastic modulus, fracture toughness and creep resistance) of SiOC and SiOC/Fe composites of various compositions, layer thicknesses, and volume fractions as a function of irradiation damage levels and irradiation temperatures. Continued development of empirical potentials and atomic structure descriptions for the amorphous alloys.
- **Year 2:** Further optimize compositions of SiOC ceramics and layered structures of Fe/SiOC to achieve the maximum radiation tolerance, and determine the roles of Fe and SiOC volume fractions on overall radiation tolerance, swelling resistance, and He solubility. Continue mechanical property evaluations. Multiscale modeling through integration of first principles calculations and molecular dynamics simulations to shed light onto the interactions of defects and gas atoms with interfaces.
- **Year 3:** Detailed experimental studies of irradiated composites using Fe(Cr) accompanied by modeling of cascade damage, defect behavior, swelling behavior, and diffusion. Mechanical property evaluations. Integration of modeling and experiments to shed light onto fundamentals and identify governing factors, which determine the maximum radiation tolerance of the composite materials.