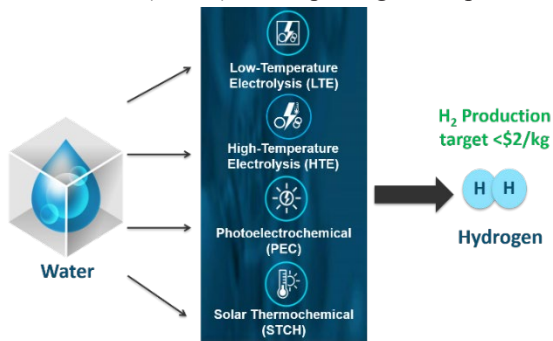


## CONSORTIUM HydroGEN

HydroGEN is a consortium of five DOE National Laboratories working together to advance water-splitting (AWS) technologies for clean, sustainable hydrogen production. EERE's Hydrogen and Fuel Cell Technologies Office launched HydroGEN in 2016 to overcome R&D challenges in AWS and meet DOE's H<sub>2</sub> cost goal of less than \$2 per kilogram. Producing H<sub>2</sub> from water using renewable electricity, nuclear power, or direct sunlight enables opportunities for large-scale energy storage, grid resiliency, and a clean, equitable energy future.

### Collaborative Research Capabilities

As a collaborative effort among National Laboratories, academia, and industry, HydroGEN is addressing materials barriers associated with low-temperature electrolysis (LTE), high-temperature electrolysis (HTE), photoelectrochemistry (PEC), and solar thermochemical (STCH) water splitting for H<sub>2</sub> production.



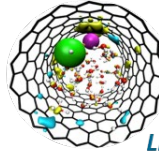
HydroGEN brings together a network of unique, world-class capabilities in materials theory and computation, advanced materials synthesis, characterization, and analysis to support:

- **30+** projects awarded through DOE funding opportunity announcements (FOAs)
- **5** collaborative multi-lab R&D projects
- **4** multi-agency projects with the National Science Foundation's Designing Materials to Revolutionize and Engineer our Future program

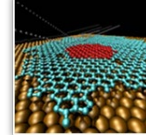
### SUPPORTING UNIVERSITY & INDUSTRY LED R&D

HydroGEN provides academia, industry, and other national labs with streamlined access to world-class experimental and computational capabilities for accelerated materials discovery. Labs and project partners collaborate through personnel, capabilities, materials, and data sharing.

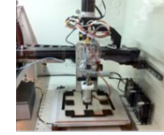
#### Theory & Computation



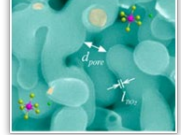
**LLNL**  
Bulk & interfacial models of aqueous electrolytes



**SNL**  
LAMMPS classic molecular dynamics modeling



**NREL**  
High-throughput spray for electrode fabrication

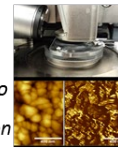


**LLNL**  
Conformal ultrathin TiO<sub>2</sub> ALD coating on bulk nanoporous Au

#### Characterization & Analytics

##### BNL

Atomic force microscopy: in-situ & operando material characterization



##### INL

TAP reactor: extracting quantitative kinetic data



### NATIONAL LAB-DIRECTED R&D

Collaborative multi-lab projects leverage HydroGEN capabilities and expertise to address specific research gaps critical to enabling low-cost H<sub>2</sub> through improved performance and durability. Efforts include theory-guided R&D where models and experimental testing are combined to analyze performance and durability. The consortium also aims to exploit similarities in technical challenges while leveraging simulation capabilities across all four AWS pathways.



### ACCOMPLISHMENTS

- Achieved 70% PEM electrolyzer cell efficiency, while improving durability and reducing cost
- Demonstrated metal-supported oxygen-conducting solid oxide electrolyzer cell (o-SOEC) with dramatically improved stability
- Scaled up baseline cell by 8X w/ 9% STH efficiency & 100-hour stability for an integrated PV-PEC system
- Discovered new STCH compounds with H<sub>2</sub> production capacities exceeding state-of-the-art at lower reduction temperatures

### DATA HUB & BENCHMARKING

The HydroGEN Data Hub is a robust and secure resource for hosting experimental and modeling/simulation data that is searchable, shareable, usable, and freely leveraged by the community. HydroGEN is also engaging the research community in developing universal standards and best practices for materials benchmarking and reporting for all four advanced water splitting technologies.



# Projects

## LOW TEMPERATURE ELECTROLYSIS - LTE

|  |   |
|--|---|
| <b>Proton Energy Systems</b>           | High Efficiency PEM Water Electrolysis Enabled by Advanced Catalysts, Membranes and Processes                     |
| <b>Northeastern University</b>         | Developing Novel Platinum Group Metal-Free Catalysts for Alkaline Hydrogen and Oxygen Evolution Reaction          |
| <b>Argonne National Lab</b>            | PGM-Free OER Catalysts for PEM Electrolyzer   |
| <b>Los Alamos National Lab</b>         | High-Performance Ultralow-Cost Non-Precious Metal Catalyst System for AEM Electrolyzer                            |
| <b>Los Alamos National Lab</b>         | Scalable Elastomeric Membranes for Alkaline Water Electrolysis  |
| <b>Georgia Institute of Technology</b> | High-Performance AEM LTE with Advanced Membranes, Ionomers and PGM-Free Electrodes                                |
| <b>University of Oregon</b>            | Pure Hydrogen Production Through Precious-Metal-Free Membrane Electrolysis of Dirty Water                         |
| <b>Chemours</b>                        | Performance and Durability Investigation of Thin, Low Crossover Proton Exchange Membranes for Water Electrolyzers |

## HIGH TEMPERATURE ELECTROLYSIS - HTE

|                                     |  |
|-------------------------------------|--|
| <b>University of Connecticut</b>    | Proton-Conducting Solid Oxide Electrolysis Cells for Large-Scale Hydrogen Production at Intermediate Temperatures                            |
| <b>Northwestern University</b>      | Degradation Characterization and Modeling of a New Solid Oxide Electrolysis Cell Utilizing Accelerated Life Testing                          |
| <b>UTRC</b>                         | Thin-Film, Metal-Supported High-Performance and Durable Proton-Solid Oxide Electrolyzer Cell   |
| <b>Saint-Gobain</b>                 | Development of Durable Materials for Cost Effective Advanced Water Splitting Utilizing All Ceramic Solid Oxide Electrolyzer Stack Technology |
| <b>West Virginia University</b>     | Intermediate Temperature Proton-Conducting Solid Oxide Electrolysis Cells with Improved Performance and Durability                           |
| <b>University of South Carolina</b> | A Multifunctional Isostructural Bilayer Oxygen Evolution Electrode for Durable Intermediate-Temperature Electrochemical Water Splitting      |
| <b>Redox</b>                        | Scalable High-H <sub>2</sub> Flux, Robust Thin Film Solid Oxide Electrolyzer   |
| <b>Nexceris</b>                     | Advanced Coating to Enhance the Durability of SOEC Stacks  |

## PHOTOELECTROCHEMICAL - PEC

|   |   |
|---|---|
| <b>Rutgers University</b>               | Best-in-Class PGM-Free Catalyst Integrated Tandem Junction PEC Water Splitting Devices  |
| <b>Stanford University</b>              | Protective Catalyst Systems on III-V and Si-based Semiconductors for Efficient, Durable Photoelectrochemical Water Splitting Devices                      |
| <b>University of Hawaii</b>             | Novel Chalcopyrites For Advanced Photoelectrochemical Water Splitting   |
| <b>University of Michigan</b>           | Monolithically Integrated Thin-Film/Silicon Tandem Photoelectrodes for High Efficiency and Stable Photoelectrochemical Water Splitting                    |
| <b>University of California, Irvine</b> | Development of Composite Photocatalyst Materials that are Highly Selective for Solar Hydrogen Production and their Evaluation in Z-Scheme Reactor Designs |
| <b>Rice University</b>                  | Highly Efficient Solar Water-Splitting Using 3D/2D Hydrophobic Perovskites with Corrosion Resistant Barriers  |
| <b>University of Toledo</b>             | Perovskite/Perovskite Tandem Photoelectrodes For Low-Cost Unassisted Photoelectrochemical Water Splitting   |

## SOLAR THERMOCHEMICAL - STCH

|  |  |
|--|--|
| <b>Colorado School of Mines</b>            | Accelerated Discovery of Solar Thermochemical H <sub>2</sub> Production Materials via High-Throughput Computational and Experimental Methods |
| <b>University of Colorado</b>              | Computationally Accelerated Discovery and Experimental Demonstration of High-Performance Materials for Advanced STCH Hydrogen Production     |
| <b>Northwestern University</b>             | Transformative Materials for High-Efficiency Thermochemical Production of Solar Fuels  |
| <b>Arizona State University</b>            | Mixed Ionic Electronic Conducting Quaternary Perovskites: Materials by Design for Solar Thermochemical H <sub>2</sub>                        |
| <b>University of Florida</b>               | A New Paradigm for Materials Discovery and Development for Lower Temperature and Isothermal Thermochemical H <sub>2</sub> Production         |
| <b>University of California, San Diego</b> | New High-Entropy Perovskite Oxides with Increased Reducibility and Stability for Thermochemical Hydrogen Generation                          |

## BENCHMARKING

|                              |  |
|------------------------------|--|
| <b>Proton Energy Systems</b> | Benchmarking Advanced Water Splitting Technologies: Best Practices in Materials Characterization |
|------------------------------|--|